

# Compressible flows: important definitions

Cristopher Morales Ubal

c.m.ubal@gmail.com

January 20, 2025

## 1 Definitions

1. Total Pressure:

$$p_{\text{total}} = p \cdot \left(1 + \frac{\gamma}{2} \cdot \text{Ma}^2\right) \quad (1)$$

2. Total Temperature:

$$T_{\text{total}} = T \cdot \left(1 + \frac{(\gamma - 1)}{2} \cdot \text{Ma}^2\right) \quad (2)$$

3. Mach Number:

$$\text{Ma} = \frac{\|\vec{u}\|}{c} \quad (3)$$

4. Speed of sound:

$$c = \sqrt{\gamma RT} \quad (4)$$

5. Ratio of specific heats:

$$\gamma = \frac{c_p}{c_v} \quad (5)$$

6. Entropy of a mixture:

$$s = \sum_{i=1}^{N_s} \left[ s_i^0 + \int_{T_0}^T \frac{c_{p,i}}{T} dT \right] Y_i - \frac{R}{M} \ln \left( \frac{p}{p_0} \right) \quad (6)$$

7. Total chemical energy of a mixture:

$$e = \int_{T_0}^T c_v dT - RT_0 + \sum_{i=1}^{N_s} h_{f,i}^0 Y_i + \frac{1}{2} \vec{u} \cdot \vec{u} \quad (7)$$

8. Static energy of a mixture:

$$e_s = \int_{T_0}^T c_v dT - RT_0 = \int_{T_0}^T c_p dT - RT \quad (8)$$

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## 2 Energy equation for multi-component flows

For compressible multicomponent flows, we have the following energy equation: For reacting flows, we have the energy equation for the total non-chemical energy  $E$  [1] given by:

$$\frac{\partial \rho E}{\partial t} + \nabla \cdot (\rho \vec{u} H) = \dot{\omega}_T + \nabla \cdot (\kappa \nabla T) - \nabla \cdot \left( \rho \sum_{i=1}^N h_{s,i} Y_i \vec{V}_i \right) + \nabla \cdot (\boldsymbol{\tau} \cdot \vec{u}) \quad (9)$$

where  $H$  is the total non-chemical enthalpy:

$$H = E + \frac{p}{\rho} \quad (10)$$

and  $E$  is the total non-chemical energy:

$$E = e_s + \frac{1}{2} \vec{u} \cdot \vec{u} = \int_{T_0}^T c_p dT - RT + \frac{1}{2} \vec{u} \cdot \vec{u} \quad (11)$$

The conservative variable for compressible solvers are usually:

$$\vec{U} = \begin{pmatrix} U_1 \\ U_2 \\ U_3 \\ U_4 \\ U_5 \end{pmatrix} = \begin{pmatrix} \rho \\ \rho u \\ \rho v \\ \rho w \\ \rho E \end{pmatrix} \quad (12)$$

then, the working variables  $\vec{V}$  can be written in terms of the conservative variables as follows:

$$\vec{V} = \begin{pmatrix} \rho \\ u \\ v \\ w \\ E \end{pmatrix} = \begin{pmatrix} U_1 \\ U_2/U_1 \\ U_3/U_1 \\ U_4/U_1 \\ U_5/U_1 \end{pmatrix} \quad (13)$$

## 3 Heat flux jacobian

For implicit solver, the heat flux Jacobians computations are needed. Following the approach explained in [2], we can compute the heat flux jacobian as follows:

$$\frac{\partial \bar{F}^v(\vec{U}, \nabla \vec{U})}{\partial \vec{U}} = \frac{\partial \bar{F}^v(\vec{V}, \nabla \vec{V})}{\partial \vec{V}} \cdot \frac{\partial \vec{V}}{\partial \vec{U}} \quad (14)$$

For the heat flux in the energy equation (9), we have  $\vec{U} = (U_1, U_2, U_3, U_4, U_5)^T$ , and  $V = (T, \rho)$ . From (11), we can rewrite it in terms of the conservative variables as follows:

$$\int_{T_0}^T c_p dT - RT = \frac{U_5}{U_1} - \frac{1}{2U_1^2} (U_2^2 + U_3^2 + U_4^2) \quad (15)$$

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Then, the temperature is implicitly defined as a function of the conservative variables  $\vec{U}$ . Hence, we can compute the partial derivative of  $T$  with respect to the conservative variables as follows:

$$\frac{\partial T}{\partial U_1} = \frac{1}{c_v} \left( -\frac{U_5}{U_1^2} + \frac{(U_2^2 + U_3^2 + U_4^2)}{U_1^3} \right) \quad (16)$$

$$\frac{\partial T}{\partial U_2} = \frac{1}{c_v} \cdot -\frac{U_2}{U_1^2} \quad (17)$$

$$\frac{\partial T}{\partial U_3} = \frac{1}{c_v} \cdot -\frac{U_3}{U_1^2} \quad (18)$$

$$\frac{\partial T}{\partial U_4} = \frac{1}{c_v} \cdot -\frac{U_4}{U_1^2} \quad (19)$$

$$\frac{\partial T}{\partial U_5} = \frac{1}{c_v} \cdot \frac{1}{U_1} \quad (20)$$

Likewise, we have:

$$\frac{\partial \rho}{\partial U_1} = 1, \quad \frac{\partial \rho}{\partial U_i} = 0, \quad \forall i = 2, 3, 4, 5 \quad (21)$$

## References

- [1] T. Poinso **and** D. Veynante. *Theoretical and Numerical Combustion*. Edwards, 2005. ISBN: 9781930217102. URL: <https://books.google.nl/books?id=cqFDkeVABYoC>.
- [2] Enrico Rinaldi, Rene Pecnik **and** Piero Colonna. “Exact Jacobians for implicit Navier–Stokes simulations of equilibrium real gas flows”. **in** *Journal of Computational Physics*: 270 (2014), **pages** 459–477. ISSN: 0021-9991. DOI: <https://doi.org/10.1016/j.jcp.2014.03.058>. URL: <https://www.sciencedirect.com/science/article/pii/S0021999114002459>.