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Catalyst-Free α -Alkylation- α -Hydroxylation of Oxindole with Alcohols

Siwei Wu, Wei Song, Runyu Zhu, Jingwen Hu, Lin Zhao, Zheyao Li, Xinhong Yu,* Chengcai Xia,* and Jianhong Zhao*



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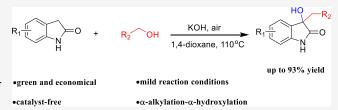
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ABSTRACT: 3-Alkyl-3-hydroxyoxindoles, a subclass of oxindole products, have antioxidant, neuroprotective, anticancer, and anti-HIV activities. In this study, a green and economical protocol for the synthesis of 3-alkyl-3-hydroxyoxindoles is developed for the first time via α -alkylation- α -hydroxylation of oxindole with benzyl alcohols without using any transition-metal catalysts in yields of 29–93%.



■ INTRODUCTION

Natural oxindole products present diverse chemical and biological properties and can be widely used in pharmaceutical agents.¹ 3-Alkyl-3-hydroxyoxindoles and 3-aryl-3-hydroxyoxindoles belong to a subclass of oxindole products (Figure 1).

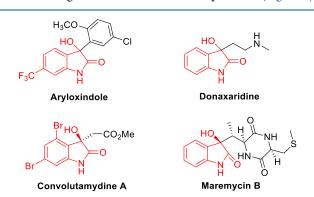


Figure 1. Natural bioactive oxindole products.

These natural compounds have various biological activities, such as antioxidant, neuroprotective, anticancer, and anti-HIV activities. 3-Alkyl/aryl-3-hydroxyoxindole products are important intermediates for the synthesis of complex compounds as well. Consequently, a number of synthetic strategies have been developed for the synthesis of this class of heterocycle. The traditional synthetic method of 3-substituted-3-hydroxyoxindoles is a nucleophilic addition to isatins including aldol reaction and metal-mediated alkylation of isatins and so on. Hayashi's group reported the aldol reaction of isatins with aldehydes using proline as the additive, while the reaction needs further reduction to afford 3-alkyl/aryl-3-hydroxyoxindoles. As to the metal-mediated alkylation of isatins, Grignard-type and Barbier-type reactions are among the most fundamental reactions. Feringa and and co-workers

developed Rh-catalyzed 1,2-addition of boronic acids to isatins to obtain various 3-aryl-3-hydroxyoxindole. However, the disadvantage of this method is the use of metal catalysts and the formation of metal byproducts. An alternative method to synthesize the target compounds is electrophilic oxidation of 3-alkyl- or 3-aryl-2-oxindoles,⁷ which undergoes the process of cyclization and rearrangement. Du⁸ and co-workers reported phenyliodine bis(trifluoroacetate) (PIFA) oxidation of anilide derivatives for the synthesis of 3-hydroxy-2-oxindole without using any metal catalysts; besides, this method can be used for the synthesis of spirooxindoles.

Recently, alcohols have been effectively used as versatile alkylating reagents^{9–12} in several borrowing hydrogen concepts because of their environmentally benign properties, the only byproduct of this transformation being water. Gnanaprakasam¹¹and co-workers reported the direct synthesis of 3-alkyl/ aryl-3-hydroxyoxindoles from 2-oxindoles using alcohols as alkylating reagents and metal as catalysts. Xu's group 12 reported the alkylation of ketones with alcohols. Dehydrative α -alkylation reactions of ketones with alcohols are now realized under simple, practical, and green conditions without using external catalysts. 12 However, novel green methods for the synthesis of 3-alkyl/aryl-3-hydroxyoxindoles from 2-oxindoles without using any transition metal catalyst are greatly demanded and highly challenged. Herein, we report a potassium hydroxide-mediated C3-alkylation and C3-hydroxylation of 2-oxindoles with alcohols in simple conditions (Scheme 1).

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Scheme 1. Synthetic Routes of 3-Alkyl/Aryl-3-Hydroxyoxindoles

Previous work

$$\begin{array}{c|c}
 & R_2 & OH \\
\hline
 & Ru cat. Bu^l OK
\end{array}$$

This work

RESULTS AND DISCUSSION

To realize our idea, 2-oxindole 1 and benzyl alcohol 2 were selected as the model substrate (Table 1) in the presence of bases in air. We first screened the effect of solvents on the C-H double functionalization of 2-oxindole. Fortunately, desired product 3a was obtained in 64% yield in 1,4-dioxane (entry 2). The desired product was obtained in 43 and 31% in toluene and PhCl, respectively, while other solvents were ineffective. Then, we studied different strengths of bases. Strong bases can afford the product in good yields (Table 1, entries 2, 11, 14, and 15) whether it is organic or inorganic bases, and KOH was the best among them. In addition, temperature also played an

important role in this reaction, and higher yields were obtained at higher temperatures (Table 1, entries 18–20). Furthermore, this reaction failed to occur when using 1.0 equiv KOH (Table 1, entry 21).

With the optimal conditions in hand, we examined the substrate scope of this reaction (Scheme 2). First, we investigated a broad range of primary benzyl alcohols. Electron-donating substituted benzyl alcohols such as 4methoxybenzyl alcohol and 4-methylbenzyl alcohol gave moderate yields (3b, 3h). When ortho-substituted benzyl alcohols were used, due to steric hindrance, lower yields were obtained (3g, 3e). While electron-withdrawing substituted benzyl alcohols and halogen-substituted benzyl alcohols afforded desired products in high yields (3c, 3d, 3i-3k). In addition, we also investigated other aromatic heterocyclic alcohols such as 2-thiophene methanol, 2-pyridylmethanol, and 2-furan methanol, and these substrates can all react successfully with 2-oxindole (3n-3p). Furthermore, we examined several aliphatic alcohols. When propanol and pentanol were used as substrates, target products were formed with longer reaction times (3q), and methanol and ethanol were not effective in this reaction. Unfortunately, when choosing 1-phenylethanol and isopropyl alcohol as substrates, no reaction was detected. We suspected that secondary alcohols were less reactive compared with primary benzyl alcohols. Then, we examined the versatility of 2-oxindole with different substituents, and most of the substrates could provide good yields (3r-3u). N-

Table 1. Optimization of Reaction Conditions^a

| | | | | Ja | |
|-----------------|-------------------|-------------------|-------|------|------------------------|
| entry | solvent | base | atm. | T/°C | yield (%) ^b |
| 1 | CH₃CN | КОН | air | 110 | NR |
| 2 | 1,4-dioxane | КОН | air | 110 | 64 |
| 3 | EtOH | КОН | air | 110 | NR |
| 4 | DMF | КОН | air | 110 | NR |
| 5 | DMSO | КОН | air | 110 | NR |
| 6 | THF | КОН | air | 110 | NR |
| 7 | toluene | КОН | air | 110 | 43 |
| 8 | CHCl ₃ | КОН | air | 110 | NR |
| 9 | THF | КОН | air | 110 | NR |
| 10 | PhCl | КОН | air | 110 | 31 |
| 11 | 1,4-dioxane | NaOH | air | 110 | 50 |
| 12 | 1,4-dioxane | K_2CO_3 | air | 110 | NR |
| 13 | 1,4-dioxane | Cs_2CO_3 | air | 110 | NR |
| 14 | 1,4-dioxane | tBuOK | air | 110 | 57 |
| 15 | 1,4-dioxane | <i>t</i> BuONa | air | 110 | 52 |
| 16 | 1,4-dioxane | DIPEA | air | 110 | NR |
| 17 | 1,4-dioxane | Et ₃ N | air | 110 | NR |
| 18 | 1,4-dioxane | КОН | air | 100 | 55 |
| 19 | 1,4-dioxane | КОН | air | 90 | 48 |
| 20 | 1,4-dioxane | КОН | air | 80 | 35 |
| 21 ^c | 1,4-dioxane | КОН | air | 110 | NR |
| 22 | H_2O | КОН | air | 110 | NR |
| 23 | 1,4-dioxane | КОН | N_2 | 110 | NR |

^aReaction conditions: a mixture of **1a** (0.50 mmol), **2a** (0.75 mmol), and base (1.00 mmol) in the solvents (5 mL) was stirred at screening of temperature for 20 h. ^bIsolated yields. ^c**1a** (0.50 mmol), **2a** (0.75 mmol), and KOH (0.50 mmol).

Scheme 2. Scope of Substrates ab

Scheme 3. Scope of Other Substrates

protected 2-oxindole only gave trace products. Additionally, we examined other substrate scope of this reaction. (Scheme 3). Unfortunately, ketones yielded lower yields and produced more other compounds, possibly due to their multiple reactive sites. Benzofuran-2(3*H*)-one, amides, and nitrile compounds did not produce target products.

We propose the reaction mechanism shown in Scheme 4. As shown in Scheme 4, we hypothesize that alcohol 2 will be oxidized to aldehyde 3 through Meerwein—Pondorf—Verley—Oppenauer type (MPV-O type) and then condensed with 1 to obtain intermediate 4, and then, it will be reduced with alcohol 2 through MPV-O type to obtain alkylated 2-oxindole 5, thus regenerating aldehydes for the catalytic cycle. Next, compound 5 undergoes base-mediated deprotonation to give enolate 6 and then reacts with air to generate compound 7 that may abstract a proton from 5 to form 8, which is subsequently cleaved by reductant enolate 6 to give the expected product 9.

CONCLUSIONS

In conclusion, we have developed a transition-metal-free, efficient method for α -C-H hydroxylation and α -C-H alkylation of various oxindoles and alcohols using inexpensive KOH as a base and environmentally benign atmospheric air as an oxidant. This methodology delivers a broad array of substrates and provides an alternate route for the synthesis of 3-alkyl/aryl-3-hydroxyoxindoles. The mechanistic study is in progress in our laboratory.

■ EXPERIMENTAL SECTION

General Methods. Commercial reagents were purchased from TCI, Acros, Accela, and Adamas, and used without further purification unless otherwise stated. Solvents, unless otherwise specified, were of reagent grade and distilled once prior to use. ¹H and ¹³CNMR spectra were recorded on Bruke Avance-400 (400 MHz), and tetramethylsilane (TMS) was used as a reference. Chromatography was carried out with silica gel (300–400 mesh) using mixtures of petroleum ether (b.p. 60–90 °C) and ethyl acetate

^aReaction conditions: a mixture of 1 (0.50 mmol), 2 (0.75 mmol), and KOH (1.00 mmol) in the solvents (5 mL) was stirred 110 $^{\circ}$ C for 20 h. ^bIsolated yields.

MPV-O type Model

Scheme 4. Proposed Reaction Mechanism 10,12,13

as eluents. High-resolution mas spectrometry (HRMS) was carried out on Micromass GCTTM gas chromatograph mass spectrometer.

General Procedure for Synthesis of 3a-4b. A mixture of 1 (0.50 mmol, 1.00 equiv), 2 (0.75 mmol, 1.50 equiv), KOH (1.00 mmol, 2.00 equiv) in 1,4-dioxane (5 mL) was stirred at 110 °C under oil bath for 20 h. After the reaction completed, the mixture was filtered and the filtrate was washed by CH₂Cl₂ and water. The organic phase was dried over anhydrous magnesium sulfate and concentrated under vacuum. The crude product was purified by column chromatography on silica gel (Petroleum ether/EtOAc = 10:1) to obtain 3a-4b.

Synthesis of 3-Benzyl-3-Hydroxyindolin-2-One (3a). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3a; 3a was isolated as a white solid (77 mg, 64% yield); Mp.122-124 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.06 (s, 1H), 7.12 (d, J = 7.5 Hz, 2H), 7.10–7.06 (m, 3H), 6.95– 6.86 (m, 3H), 6.61 (d, J = 7.5 Hz, 1H), 6.14 (s, 1H), 3.17 (d, J = 12.6 m)Hz, 1H), 3.01 (d, J = 12.6 Hz, 1H). $^{13}C\{^{1}H\}$ NMR (100 MHz, DMSO- d_6) δ [ppm]: 179.2, 142.1, 135.5, 131.4, 130.6, 129.3, 127.9, 126.8, 125.0, 121.7, 109.7, 77.1, 43.9. This compound was known: Chaudhari, M. B.; Sutar, Y.; Malpathak, S.; Hazra, A.; Gnanaprakasam, B. Org. Lett. 2017, 19, 3628-3631.

Synthesis of 3-Hydroxy-3-(4-Methoxybenzyl)Indolin-2-One (**3b**). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3b; 3b was isolated as a yellow solid (73 mg, 54% yield); Mp.122-124 °C. ¹H NMR (400 MHz, DMSO d_6) δ [ppm]: 10.03 (s, 1H), 7.1–7.08 (m, 2H), 6.91 (m, 1H), 6.83–

6.77 (m, 2H), 6.66-6.63 (m, 2H), 6.61(d, J = 7.5 Hz, 1H), 6.07 (s, 1H), 3.63 (s, 3H), 3.09 (d, J = 12.8 Hz, 1H), 2.94 (d, J = 12.8 Hz, 1H). $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.8, 157.7, 141.6, 131.1, 131.0, 128.8, 126.8, 124.5, 121.2, 112.9, 109.2, 76.6, 54.8, 42.5. This compound was known: Chaudhari, M. B.; Sutar, Y.; Malpathak, S.; Hazra, A.; Gnanaprakasam, B. Org. Lett. 2017, 19, 3628-3631.

Synthesis of 3-Hydroxy-3-(4-(Trifluoromethyl)Benzyl)Indolin-2-One (3c). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3c; 3c was isolated as a yellow solid (131 mg, 85% yield); ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.15 (s, 1H), 7.48 (d, J = 8.0 Hz, 2H), 7.18-7.13 (m, 3H),7.13-7.10 (m, 1H), 6.93 (m, 1H), 6.66 (d, J = 7.7 Hz, 1H), 6.27(d, J = 1.3 Hz, 1H), 3.27 (d, J = 12.6 Hz, 1H), 3.10 (d, J = 12.6 Hz, 1H)1H). 13 C $\{^{1}$ H $\}$ NMR (100 MHz, DMSO- d_{6}) δ [ppm]: 178.5, 141.5, 140.0, 130.9, 130.6, 129.0, 127.1(q, J = 32.7 Hz), 124.5, 124.3(q, J = 32.7 Hz) 270.1 Hz), 124.2(q, *J* = 3.6 Hz), 121.3, 109.4, 76.3, 43.1. HRMS (EI) m/z calcd for C₁₆H₁₂F₃NO₂ (M⁺): 307.0820, found: 307.0818.

Synthesis of 3-(3-Bromobenzyl)-3-Hydroxyindolin-2-One (3d). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3d; 3d was isolated as a yellow solid (144 mg, 91% yield); Mp.135-136 °C. ¹H NMR (400 MHz, DMSO d_6) δ [ppm]: 10.18 (s, 1H), 7.33–7.28 (m, 1H), 7.18–7.11 (m, 2H), 7.10 (s, 1H), 7.07 (t, J = 7.9 Hz,1H), 6.96-6.90 (m, 2H), 6.67 (m, 1H), 6.24 (s, 1H), 3.19 (d, J = 12.7 Hz, 1H), 3.01 (d, J = 12.7 Hz, 1H). ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.6, 141.5, 137.9, 132.8, 130.6, 129.5, 129.3, 129.2, 129.0, 124.5, 121.3, 120.7,

109.4, 76.4, 42.8. This compound was known: Wu, Q.; Pan, L.; Du, G. M.; Zhang, C.; Wang, D. W. Org. Chem. Front. **2018**, 5, 2668–2675.

Synthesis of 3-Hydroxy-3-(2-Methoxybenzyl)Indolin-2-One (3e). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3e; 3e was isolated as a yellow solid (51 mg, 38% yield); Mp.136-137 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.08 (s, 1H), 7.12–7.02 (m, 3H), 6.92 (m, 1H), 6.81 (m, 1H), 6.76–6.70 (m, 2H), 6.59 (d, J = 7.5 Hz, 1H), 6.00 (s, 1H), 3.44 (s, 3H), 3.22 (d, J = 12.9 Hz, 1H), 2.97 (d, J = 12.9 Hz, 1H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 179.2, 157.1, 141.4, 131.1, 131.0, 128.4, 127.7, 124.7, 123.3, 120.6, 119.5, 110.4, 108.9, 76.3, 54.7, 35.9. This compound was known: Bisht, G. S.; Chaudhari, M. B.; Gupte, V. S.; Gnanaprakasam, B. ACS Omega. 2017, 2, 8234–8252.

Synthesis of 3-Hydroxy-3-(Naphthalen-2-Ylmethyl)Indolin-2-One (3f). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3f; 3f was isolated as a yellow solid (117 mg, 81% yield); Mp.145-146 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.06 (s, 1H), 7.77–7.67 (m, 1H), 7.63–7.60 (m, 1H), 7.57 (d, J = 8.4 Hz, 1H), 7.41 (d, J = 1.6 Hz, 1H), 7.37–7.32 (m, 2H), 7.15 (m, 1H), 7.05 (m, 2H), 6.87 (m, 1H), 6.54 (d, J = 7.6 Hz, 1H), 6.24 (s, 1H), 3.34 (d, J = 12.6 Hz, 1H), 3.19 (d, J = 12.6 Hz, 1H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.8, 141.6, 132.9, 132.5, 131.7, 131.0, 128.9, 128.7, 128.6, 127.4, 127.3, 126.7, 125.8, 125.4, 124.6, 121.3, 109.3, 76.8, 43.7. HRMS (EI) m/z calcd for C_{19} H₁₅NO₂ (M^+): 289.1103, found: 289.1100.

Synthesis of 3-Hydroxy-3-(2-Methylbenzyl)Indolin-2-One (**3g**). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product **3g**; **3g** was isolated as a white solid (57 mg, 45% yield); ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.18 (s, 1H), 7.14 (m, 1H), 7.07–7.00 (m, 3H), 6.97–6.94(m, 1H), 6.84 (m, 1H), 6.79 (m, 1H), 6.71 (d, J = 7.6 Hz, 1H), 6.09 (s, 1H), 3.15 (d, J = 13.5 Hz, 1H), 2.97 (d, J = 13.5 Hz, 1H), 1.96 (s, 3H). ¹³C{¹H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 179.5, 141.5, 136.9, 133.7, 131.1, 130.8, 129.7, 128.8, 126.4, 124.9, 124.6, 121.0, 109.3, 76.4, 39.5, 19.5. This compound was known: Chaudhari, M. B.; Sutar, Y.; Malpathak, S.; Hazra, A.; Gnanaprakasam, B. *Org. Lett.* **2017**, 19, 3628–3631.

Synthesis of 3-Hydroxy-3-(4-Methylbenzyl)Indolin-2-One (3h). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3h; 3h was isolated as a white solid (71 mg, 56% yield); Mp.185-188 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.03 (s, 1H), 7.15 (m, 1H), 7.10 (m, 1H), 6.94–6.86 (m, 3H), 6.77 (d, J = 7.9 Hz, 2H), 6.60 (d, J = 7.6 Hz, 1H), 6.10 (s, 1H), 3.12 (d, J = 12.6 Hz, 1H), 2.98 (d, J = 12.6 Hz, 1H), 2.16 (s, 3H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.7, 141.6, 135.1, 131.9, 131.0, 129.9, 128.8, 128.1, 124.5, 121.1, 109.2, 76.6, 43.1, 20.5. This compound was known: Chaudhari, M. B.; Sutar, Y.; Malpathak, S.; Hazra, A.; Gnanaprakasam, B. *Org. Lett.* **201**7, 19, 3628–3631.

Synthesis of 3-(4-Bromobenzyl)-3-Hydroxyindolin-2-One (3i). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3i; 3i was isolated as a yellow solid (128 mg, 81% yield); Mp.161-162 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.14 (s, 1H), 7.33–7.26 (m, 2H), 7.17–7.11 (m, 2H), 6.95–6.91 (m, 1H), 6.90–6.84 (m, 2H), 6.69–6.64 (m, 1H), 6.22 (s, 1H), 3.16 (d, J = 12.7 Hz, 1H), 3.00 (d, J = 12.7 Hz, 1H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.6, 141.5, 134.5, 132.3, 130.7, 130.3, 129.0, 124.5, 121.3, 119.7, 109.4, 76.3, 42.8. This compound was known: Wu, Q.; Pan, L.; Du, G. M.; Zhang, C.; Wang, D. W. *Org. Chem. Front.* 2018, 5, 2668–2675.

Synthesis of 3-(4-Fluorobenzyl)-3-Hydroxyindolin-2-One (3j). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3j; 3j was isolated as a white

solid (95 mg, 74% yield); Mp.126-127 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.12 (s, 1H), 7.13 (m, 2H), 6.95 (d, J = 1.2 Hz, 2H), 6.94–6.92 (m, 2H), 6.91 (d, J = 1.0 Hz, 1H), 6.65 (d, J = 7.6 Hz, 1H), 6.19 (s, 1H), 3.18 (d, J = 12.8 Hz, 1H), 3.01 (d, J = 12.8 Hz, 1H). 13 C{¹H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.7, 162.9 (d, J = 240.4 Hz), 141.6, 131.8 (d, J = 10.6 Hz), 131.2, 130.8, 128.9, 124.5, 121.2, 114.2 (d, J = 21.4 Hz), 109.3, 76.5, 42.5. This compound was known: Wu, Q.; Pan, L.; Du, G. M.; Zhang, C.; Wang, D. W. *Org. Chem. Front.* **2018**, 5, 2668–2675.

Synthesis of 3-(2-Chlorobenzyl)-3-Hydroxyindolin-2-One (**3k**). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product **3k**; **3k** was isolated as a white solid (115 mg, 84% yield); Mp.168-171 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.28 (s, 1H), 7.39–7.36 (m, 1H), 7.29–7.24 (m, 1H), 7.23–7.17 (m, 2H), 7.13 (m, 1H), 6.85–6.80 (m, 2H), 6.71 (d, J = 7.7 Hz, 1H), 6.22 (s, 1H), 3.32 (d, J = 13.5 Hz, 1H), 3.15 (d, J = 13.5 Hz, 1H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 179.0, 141.3, 134.1, 133.1, 132.1, 130.5, 128.9, 128.8, 128.3, 126.3, 124.6, 121.1, 109.3, 75.9, 39.2. This compound was known: Wu, Q.; Pan, L.; Du, G. M.; Zhang, C.; Wang, D. W. *Org. Chem. Front.* **2018**, 5, 2668–2675.

Synthesis of 3-Hydroxy-3-(3-Methylbenzyl)Indolin-2-One (3I). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3I; 3I was isolated as a white solid (82 mg, 65% yield); Mp.155-156 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.08 (s, 1H), 7.17–7.08 (m, 2H), 6.97–6.88 (m, 3H), 6.73 (s, 1H), 6.68 (d, J = 7.5 Hz, 1H), 6.62 (d, J = 7.6 Hz, 1H), 6.13 (s, 1H), 3.14 (d, J = 12.6 Hz, 1H), 2.98 (d, J = 12.6 Hz, 1H), 2.11 (s, 3H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.8, 141.6, 136.2, 134.9, 131.0, 130.8, 128.8, 127.3, 127.2, 126.9, 124.6, 121.1, 109.2, 76.6, 43.4, 20.9. This compound was known: Chaudhari, M. B.; Sutar, Y.; Malpathak, S.; Hazra, A.; Gnanaprakasam, B. Org. Lett. 2017, 19, 3628–3631.

Synthesis of 3-(4-Chlorobenzyl)-3-Hydroxyindolin-2-One (3m). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3m; 3m was isolated as a white solid (127 mg, 93% yield); Mp.142-143 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.14 (s, 1H), 7.20–7.13 (m, 2H), 7.13 (d, J = 7.5 Hz, 2H), 6.94 (m, 3H), 6.66 (d, J = 7.4 Hz, 1H), 6.23 (s, 1H), 3.18 (d, J = 12.7 Hz, 1H), 3.02 (d, J = 12.7 Hz, 1H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.6, 141.5, 134.1, 131.9, 131.2, 130.7, 129.0, 127.4, 124.5, 121.3, 109.4, 76.4, 42.7. This compound was known: Wu, Q.; Pan, L.; Du, G. M.; Zhang, C.; Wang, D. W. Org. Chem. Front. 2018, 5, 2668–2675.

Synthesis of 3-Hydroxy-3-(Thiophen-2-Ylmethyl)Indolin-2-One (3n). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3n; 3n was isolated as a white solid (42 mg, 34% yield); Mp.115-116 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.16 (s, 1H), 7.20 (m, 1H), 7.16 (m, 1H), 7.11 (m, 1H), 6.92 (m, 1H), 6.79 (m, 1H), 6.68 (d, J = 7.7 Hz, 1H), 6.58 (m, 1H), 6.23 (s, 1H), 3.34 (d, J = 14.0 Hz, 1H), 3.25 (d, J = 14.0 Hz, 1H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.4, 142.0, 136.6, 130.8, 129.1, 127.1, 126.1, 124.9, 124.4, 121.4, 109.4, 75.9, 37.7. This compound was known: Wu, Q.; Pan, L.; Du, G. M.; Zhang, C.; Wang, D. W. Org. Chem. Front. 2018, 5, 2668–2675.

Synthesis of 3-Hydroxy-3-(Pyridin-2-Ylmethyl)Indolin-2-One (**30**). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product **30**; **30** was isolated as a white solid (76 mg, 63% yield); Mp.167-169 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.16 (s, 1H), 8.30 (m, 1H), 7.58 (m, 1H), 7.16–7.06 (m, 3H), 6.93 (m, 1H), 6.83 (m, 1H), 6.66 (d, J = 7.7 Hz, 1H), 6.34 (s, 1H), 3.31 (d, J = 13.1 Hz, 1H), 3.19 (d, J = 13.1 Hz, 1H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.5, 156.0, 148.2, 141.6, 135.8, 130.9, 128.7, 124.5, 124.3, 121.6, 121.1, 109.2,

75.6, 45.1. This compound was known: Yaragorla, S.; Singh, G.; Dada, R. *Tetrahedron Lett.* **2015**, *56*, 5924–5929.

Synthesis of 3-Benzyl-6-Chloro-3-Hydroxyindolin-2-One (*3p*). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product **3p**; **3p** was isolated as a white solid (98 mg, 72% yield); Mp.119–121 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.27 (s, 1H), 7.14 (s, 1H), 7.13–7.09 (m, 3H), 6.95 (m, 3H), 6.66 (d, J = 1.9 Hz, 1H), 6.29 (s, 1H), 3.20 (d, J = 12.7 Hz, 1H), 3.04 (d, J = 12.7 Hz, 1H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.7, 143.1, 134.7, 133.0, 130.1, 129.8, 127.6, 126.4, 126.0, 120.9, 109.4, 76.3, 43.2. This compound was known: Bisht, G. S.; Chaudhari, M. B.; Gupte, V. S.; Gnanaprakasam, B. *ACS Omega* **2017**, 2, 8234–8252.

Synthesis of 3-Benzyl-6-Bromo-3-Hydroxyindolin-2-One (3q). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3q; 3q was isolated as a white solid (106 mg, 67% yield); Mp.135-136 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.22 (s, 1H), 7.14–7.10 (m, 3H), 7.09 (d, J = 1.8 Hz, 1H), 7.04 (d, J = 7.9 Hz, 1H), 6.92 (m, 2H), 6.76 (d, J = 1.7 Hz, 1H), 6.24 (s, 1H), 3.17 (d, J = 12.7 Hz, 1H), 3.01 (d, J = 12.7 Hz, 1H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.6, 143.3, 134.7, 130.3, 130.1, 127.6, 126.5, 126.4, 123.8, 121.4, 112.1, 76.3, 43.1. HRMS (EI) m/z calcd for C₁₅H₁₂BrNO₂ (M⁺):319.0031, found: 319.0030.

Synthesis of 3-Hydroxy-3-Pentylindolin-2-One (3r). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3r; 3r was isolated as a white solid (32 mg, 29% yield); 1 H NMR (400 MHz, DMSO- 4 6) δ [ppm]: 10.21 (s, 1H), 7.24 (m, 1H), 7.19 (m, 1H), 6.96 (m, 1H), 6.80 (d, 4 J = 7.6 Hz, 1H), 5.82 (s, 1H), 1.75 (m, 2H), 1.15 (m, 4H), 1.08–0.84 (m, 2H), 0.86–0.72 (m, 3H). 13 C{ 1 H} NMR (100 MHz, DMSO- 4 6) δ [ppm]: 179.4, 141.7, 132.1, 128.7, 123.7, 121.5, 109.4, 75.6, 37.6, 31.3, 22.3, 21.8, 13.7. This compound was known: Wu, Q.; Pan, L.; Du, G. M.; Zhang, C.; Wang, D. W. *Org. Chem. Front.* 2018, 5, 2668–2675.

Synthesis of 3-Benzyl-5-Fluoro-3-Hydroxyindolin-2-One (*3s*). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product *3s*; *3s* was isolated as a white solid (112 mg, 87% yield); ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.13 (s, 1H), 7.17–7.07 (m, 3H), 7.00 (m, 1H), 6.97–6.91 (m, 3H), 6.61 (m, 1H), 6.31 (s, 1H), 3.21 (d, J = 12.7 Hz, 1H), 3.05 (d, J = 12.7 Hz, 1H). ¹³C{¹H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.7, 157.71(d, J = 235.2 Hz), 137.7, 134.7, 132.7(d, J = 8.7 Hz), 130.1, 127.5, 126.4, 115.0(d, J = 23.6 Hz), 112.3(d, J = 24.4 Hz), 110.0(d, J = 8.0 Hz), 77.0, 43.2. HRMS (EI) m/z calcd for $C_{15}H_{12}FNO_2$ (M⁺):257.0852, found: 257.0854.

Synthesis of 3-Benzyl-7-Chloro-3-Hydroxyindolin-2-One (*3t*). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product *3t*; *3t* was isolated as a white solid (123 mg, 90% yield); Mp.227-229 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.54 (s, 1H), 7.19 (m, 1H), 7.11 (m, 3H), 7.09 (d, J = 1.1 Hz, 1H), 7.00–6.89 (m, 3H), 6.33 (s, 1H), 3.21 (d, J = 12.7 Hz, 1H), 3.06 (d, J = 12.7 Hz, 1H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.7, 139.3, 134.6, 132.9, 130.1, 128.8, 127.6, 126.5, 123.2, 122.6, 113.5, 77.2, 43.3. HRMS (EI) m/z calcd for C $_{15}$ H $_{12}$ ClNO $_{2}$ (M*):273.0557, found: 273.0556.

Synthesis of 3-(Furan-2-Ylmethyl)-3-Hydroxyindolin-2-One (3u). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 3u; 3u was isolated as a white solid (64 mg, 56% yield); Mp.140-142 °C. ¹H NMR (400 MHz, DMSO- d_6) δ [ppm]: 10.20 (s, 1H), 7.35 (m, 1H), 7.16 (m, 1H), 7.11 (m, 1H), 6.92 (m, 1H), 6.72 (d, J = 7.7 Hz, 1H), 6.22 (m, 1H), 6.19 (s, 1H), 5.79 (m, 1H), 3.22 (d, J = 14.4 Hz, 1H), 3.08 (d, J = 14.4 Hz, 1H). 13 C{ 1 H} NMR (100 MHz, DMSO- d_6) δ [ppm]: 178.4, 149.8, 141.7, 141.5, 131.0, 129.0, 124.3, 121.4, 110.3, 109.3, 107.4, 74.9,

35.9. This compound was known: Wu, Q.; Pan, L.; Du, G. M.; Zhang, C.; Wang, D. W. Org. Chem. Front. **2018**, *5*, 2668–2675.

Synthesis of 3,4-Dihydro-2-Hydroxy-2-(Phenylmethyl)-1(2H)-Naphthalenone (4a). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 4a; 4a was isolated as a yellow solid (16 mg, 13% yield); Mp. 97-99 °C. ¹H NMR (400 MHz, CDCl₃) δ [ppm]: 8.03 (d, J = 8.0 Hz, 1H), 7.57 (dt, J = 1.2 Hz, 1H), 7.38 (m, 2H), 7.30 (m, 3H), 7.16 (m, 2H), 4.70 (bs, 1H), 3.28 (m, 1H), 3.07 (m, 1H), 3.00 (d, J = 16 Hz, 1H), 2.93 (d, J = 12 Hz, 1H), 2.29 (m, 1H), 2.21 (m, 1H). 13 C{ 1 H} NMR (150 MHz, CDCl₃) δ [ppm]: 201.0, 143.3, 135.4, 134.3, 130.5, 129.2, 128.7, 128.2, 128.1, 127.8, 127.2, 127.1, 127.0, 76.2, 42.1, 33.9, 26.5. This compound was known: Chaudhari, M. B.; Sutar, Y.; Malpathak, S.; Hazra, A.; Gnanaprakasam, B. Org. Lett. 2017, 19, 3628–3631.

Synthesis of 3,4-Dihydro-2-Hydroxy-2-[(4-Methylphenyl)-Methyl]-1(2H)-Naphthalenone (4b). Following the general procedure, the crude product was purified using silica gel column chromatography (EtOAc/petroleum ether = 1:10) to produce the desired product 4b; 4b was isolated as a yellow solid (13 mg, 10% yield); Mp.103–105 °C. ¹H NMR (400 MHz, CDCl₃) δ [ppm]: 8.03 (d, J = 8.0 Hz, 1H), 7.57 (dt, J = 1.2 Hz, 1H), 7.38 (t, J = 8.0 Hz, 1H), 7.31 (d, J = 8.0 Hz, 1H), 7.09 (d, J = 8.0 Hz, 2H), 7.04 (d, J = 8.0 Hz, 2H), 3.76 (bs, 1H), 3.27 (m, 1H), 3.04 (m, 1H), 2.97 (d, J = 16 Hz, 1H), 2.89 (d, J = 16 Hz, 1H), 2.32 (s, 3H), 2.22 (m, 2H). 13 CC 1 H NMR (150 MHz, CDCl₃) δ [ppm]: 200.1, 142.33, 135.5, 133.2, 131.3, 129.6, 129.3, 128.2, 128.0, 127.1, 126.2, 75.2, 40.7, 33.0, 25.5, 20.2. This compound was known: Chaudhari, M. B.; Sutar, Y.; Malpathak, S.; Hazra, A.; Gnanaprakasam, B. Org. Lett. 2017, 19, 3628–3631.

ASSOCIATED CONTENT

Solution Supporting Information

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¹H-NMR, ¹³C-NMR copies of compounds (3a-4b), and additional spectroscopic data (PDF)

AUTHOR INFORMATION

Corresponding Authors

Xinhong Yu — Engineering Research Centre of Pharmaceutical Process Chemistry, Ministry of Education; Shanghai Key Laboratory of New Drug Design, School of Pharmacy, and State Key Laboratory of Bioengineering Reactors, East China University of Science & Technology, Shanghai 200237, People's Republic of China; orcid.org/0000-0003-0193-3702; Email: xhyu@ecust.edu.cn

Chengcai Xia — Institute of Pharmacology, School of Pharmaceutical Sciences, Taishan Medical University, Taian, Shandong 271016, People's Republic of China; oorcid.org/ 0000-0002-1943-5720; Email: xiachc@163.com

Jianhong Zhao — Engineering Research Centre of Pharmaceutical Process Chemistry, Ministry of Education; Shanghai Key Laboratory of New Drug Design, School of Pharmacy, and State Key Laboratory of Bioengineering Reactors, East China University of Science & Technology, Shanghai 200237, People's Republic of China; Email: zjhpharm@163.com

Authors

Siwei Wu — Engineering Research Centre of Pharmaceutical Process Chemistry, Ministry of Education; Shanghai Key Laboratory of New Drug Design, School of Pharmacy, and State Key Laboratory of Bioengineering Reactors, East China University of Science & Technology, Shanghai 200237, People's Republic of China

- Wei Song Engineering Research Centre of Pharmaceutical Process Chemistry, Ministry of Education; Shanghai Key Laboratory of New Drug Design, School of Pharmacy, and State Key Laboratory of Bioengineering Reactors, East China University of Science & Technology, Shanghai 200237, People's Republic of China
- Runyu Zhu Engineering Research Centre of Pharmaceutical Process Chemistry, Ministry of Education; Shanghai Key Laboratory of New Drug Design, School of Pharmacy, and State Key Laboratory of Bioengineering Reactors, East China University of Science & Technology, Shanghai 200237, People's Republic of China
- Jingwen Hu Engineering Research Centre of Pharmaceutical Process Chemistry, Ministry of Education; Shanghai Key Laboratory of New Drug Design, School of Pharmacy, and State Key Laboratory of Bioengineering Reactors, East China University of Science & Technology, Shanghai 200237, People's Republic of China
- Lin Zhao Engineering Research Centre of Pharmaceutical Process Chemistry, Ministry of Education; Shanghai Key Laboratory of New Drug Design, School of Pharmacy, and State Key Laboratory of Bioengineering Reactors, East China University of Science & Technology, Shanghai 200237, People's Republic of China
- Zheyao Li Engineering Research Centre of Pharmaceutical Process Chemistry, Ministry of Education; Shanghai Key Laboratory of New Drug Design, School of Pharmacy, and State Key Laboratory of Bioengineering Reactors, East China University of Science & Technology, Shanghai 200237, People's Republic of China

Complete contact information is available at: https://pubs.acs.org/10.1021/acs.joc.1c02185

Notes

The authors declare no competing financial interest.

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