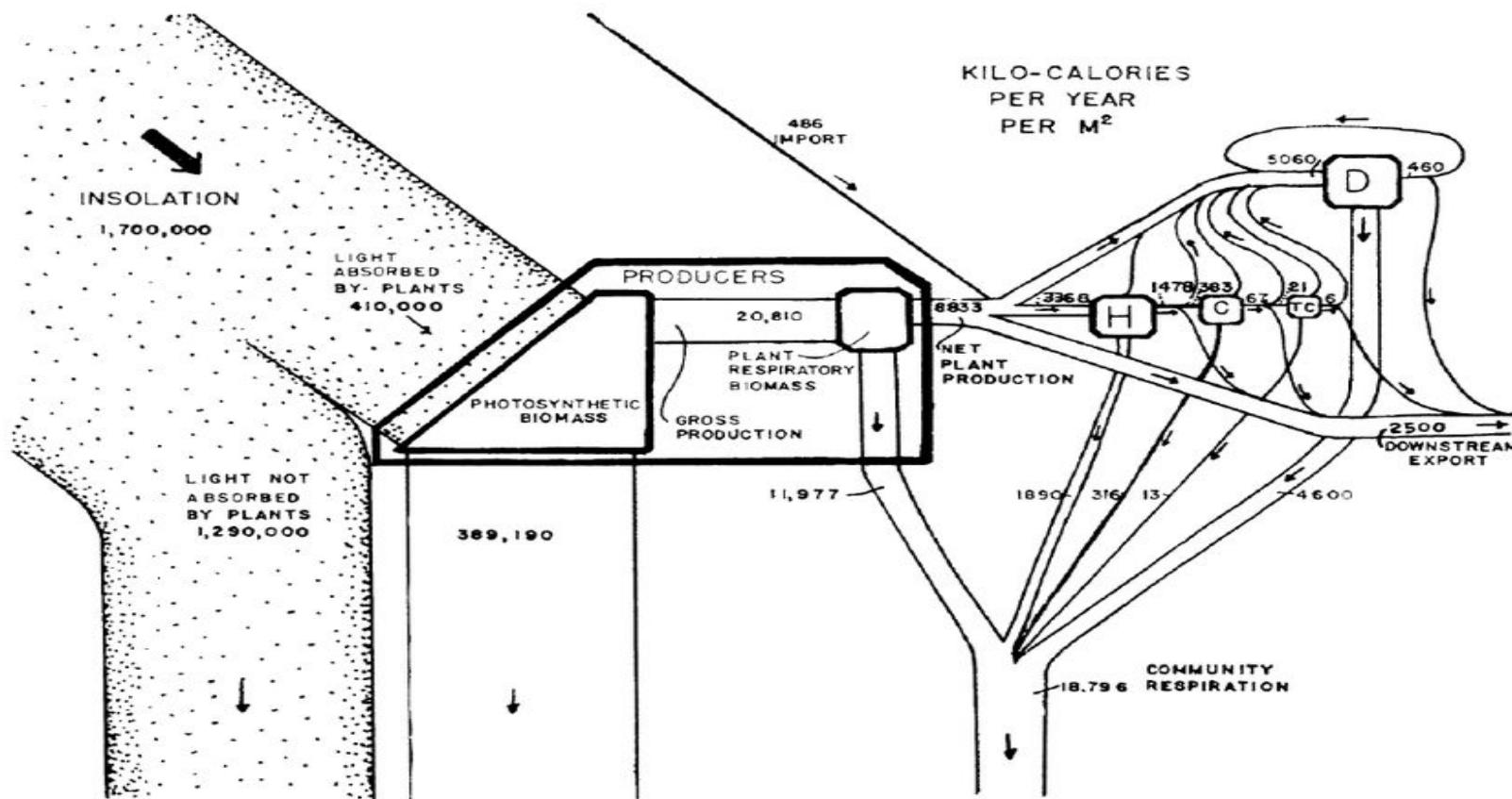


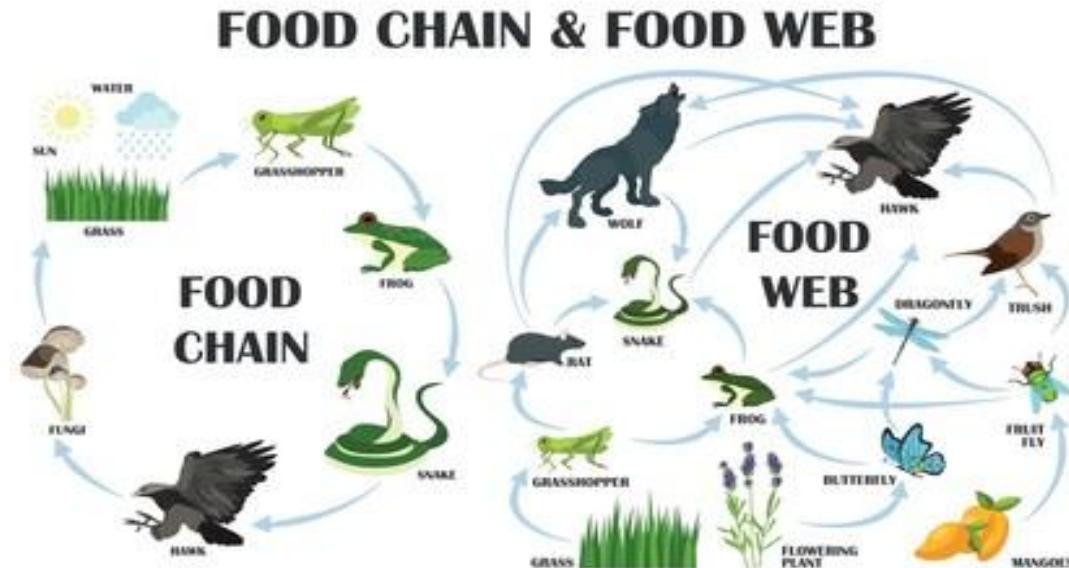
Universal Energy Flow Model (E.P. Odum, 1959)

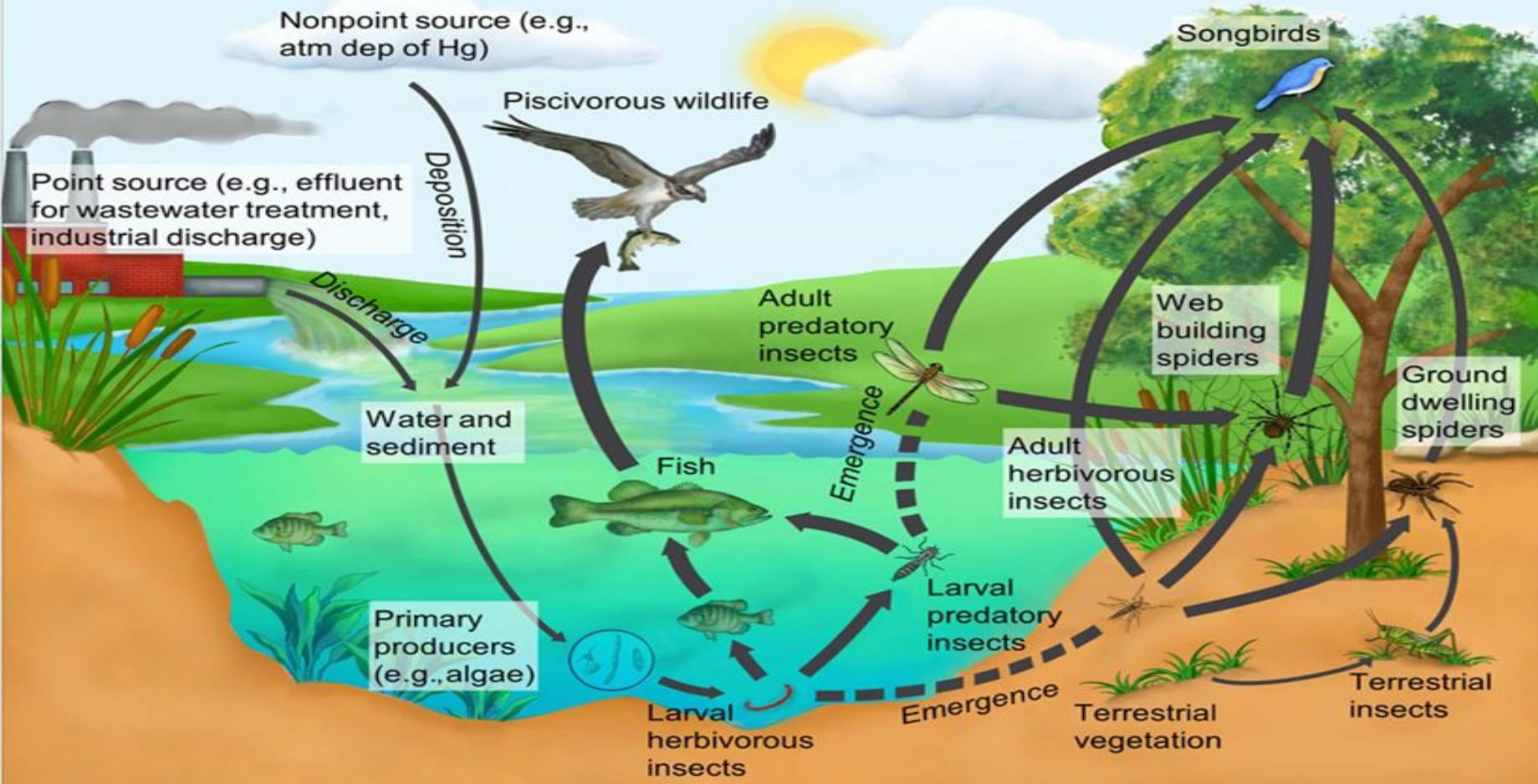


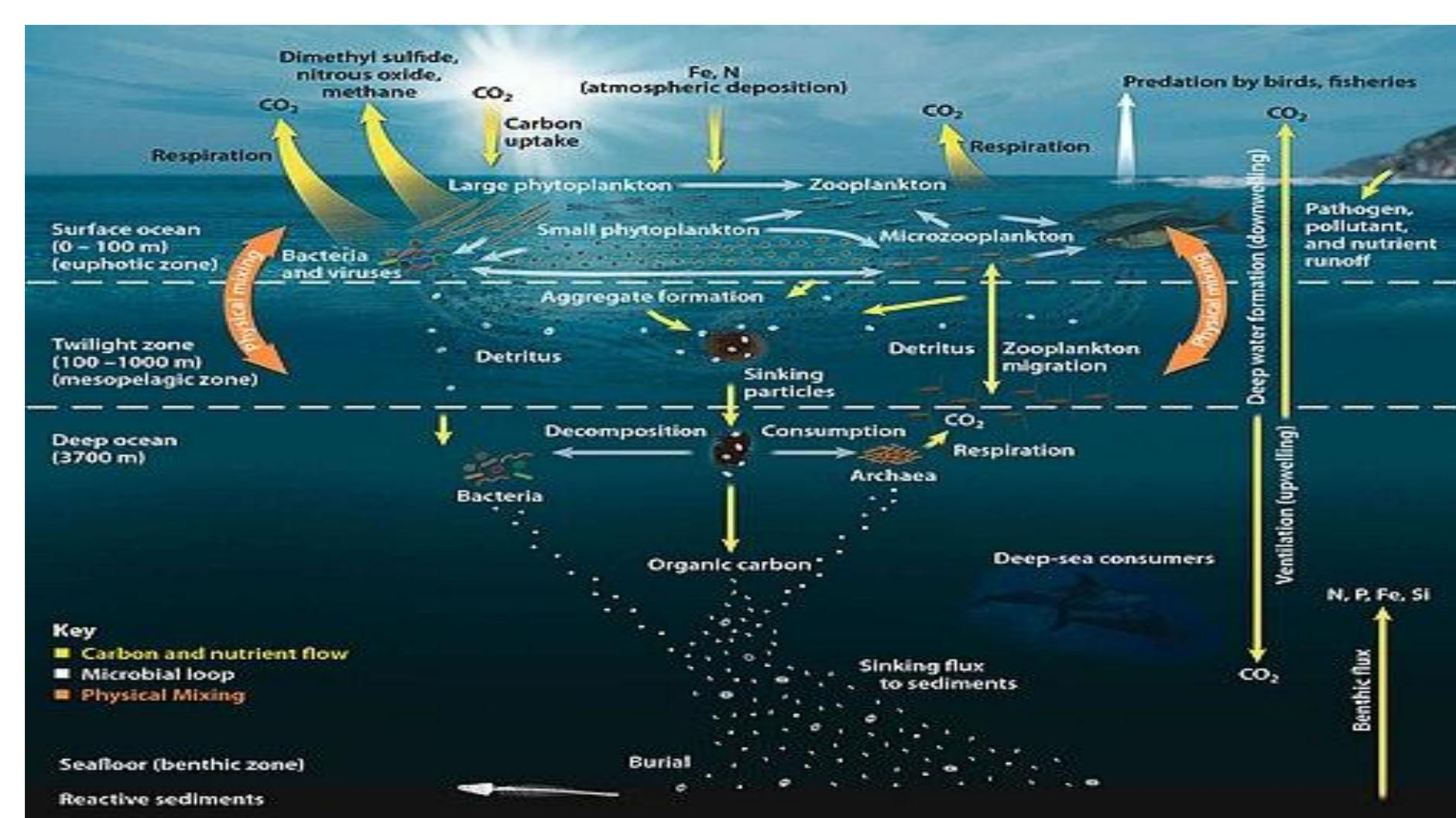
Food Chain: It is a energy sequence of links or **trophic level** that start with a species that eats no other species and ends with a species that is eaten by no other species.

Food Web: At each **trophic level**, number of different species are interconnected via multiple pathways resembling complex web, referred as Food web.

Trophic level is each step or position that an organism occupies in a food chain or food web, based on its source of energy. It represents how energy and nutrients flow through an ecosystem.







Energetic Hypothesis (Hutchinson, 1959)

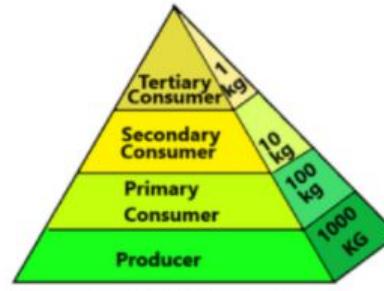
In case of energy flow through an ecosystem, the number of trophic levels is limited because of the decreasing availability of energy resulting from the inefficiencies in energy transfer from one trophic level to another. Thus, chains should not be longer in ecosystem with higher primary productivity.

Dynamical Hypothesis (Pimm and Lawton, 1978)

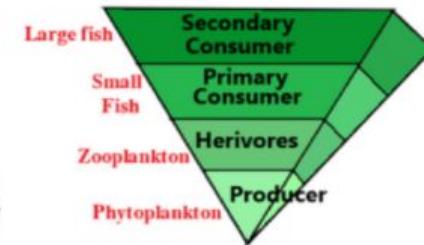
According to this hypothesis, food chains should be longer in ecosystems that are not subjected to disturbance by catastrophic events including fires, flooding, drought, widespread disease, etc. Ecosystems with longer chains take longer to return to equilibrium, once disturbed, thus they are less likely to persist in nature.

Pyramid of biomass

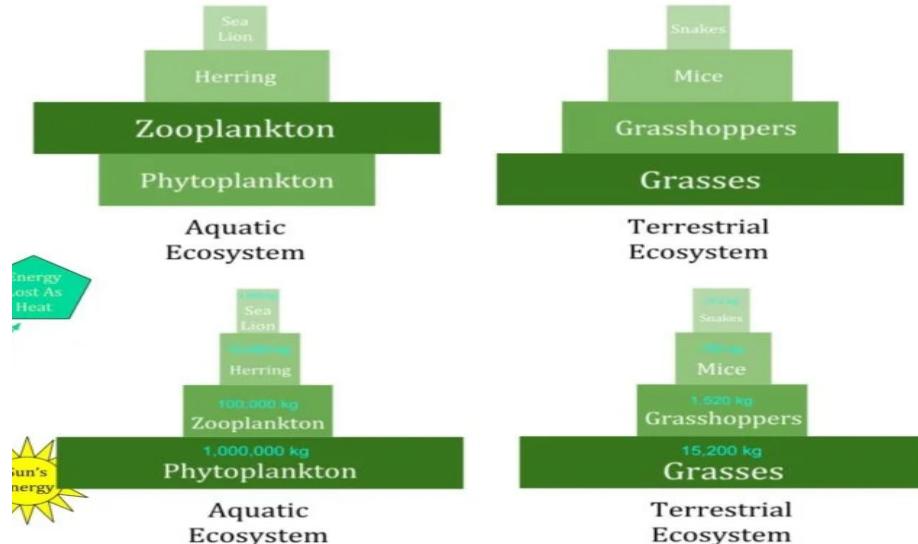
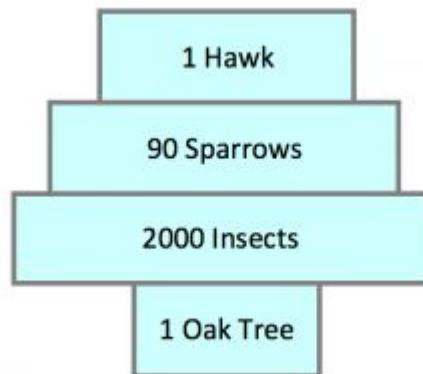
- Energy Pyramid
- Biomass Pyramid
- Number Pyramid



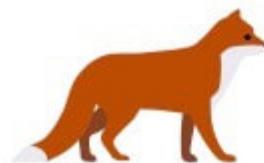
A. Upright pyramid



B. Inverted pyramid



Tertiary
Consumers



Secondary
Consumers



Primary
Consumers



Producers



10 J

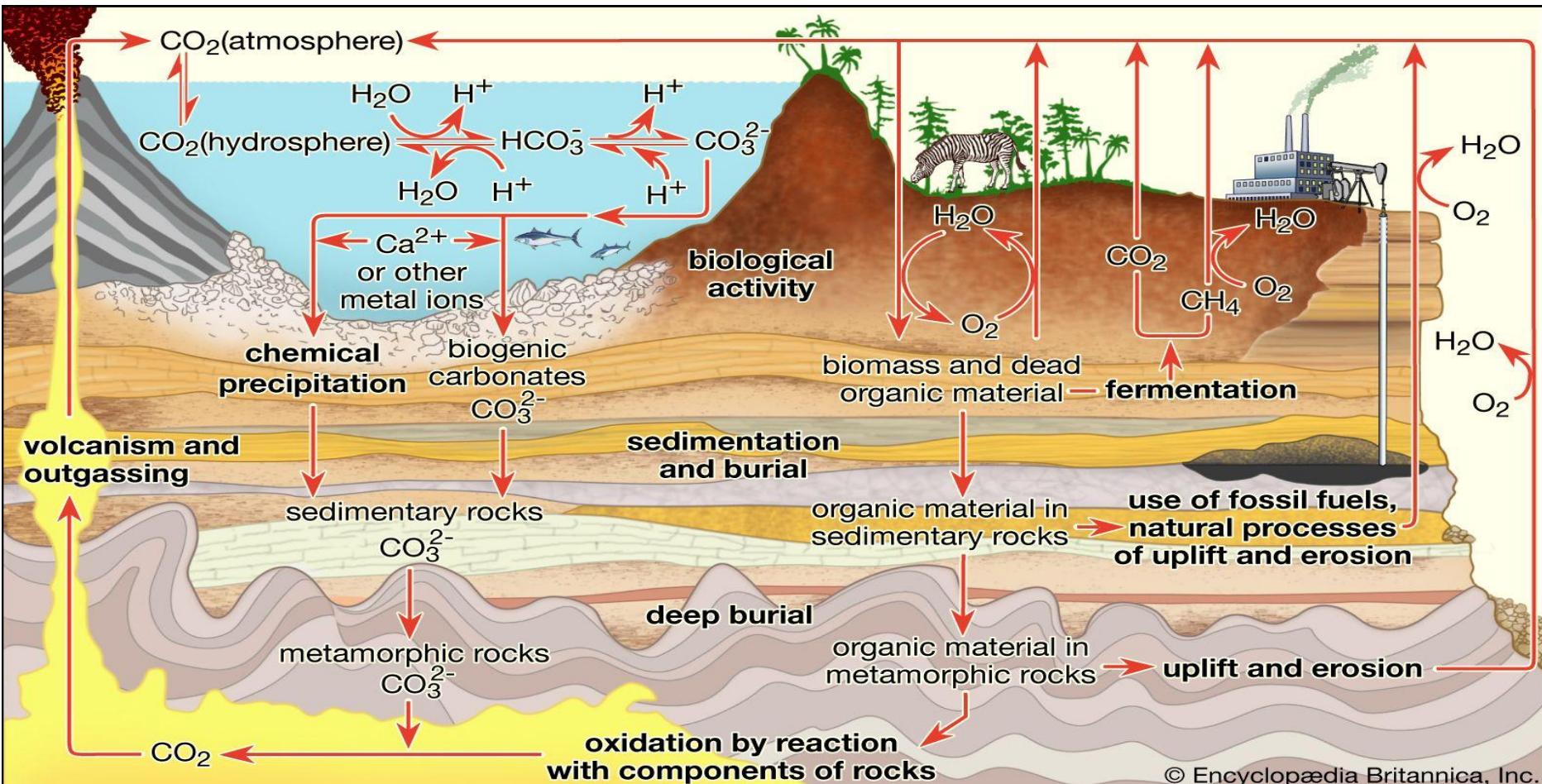
100 J

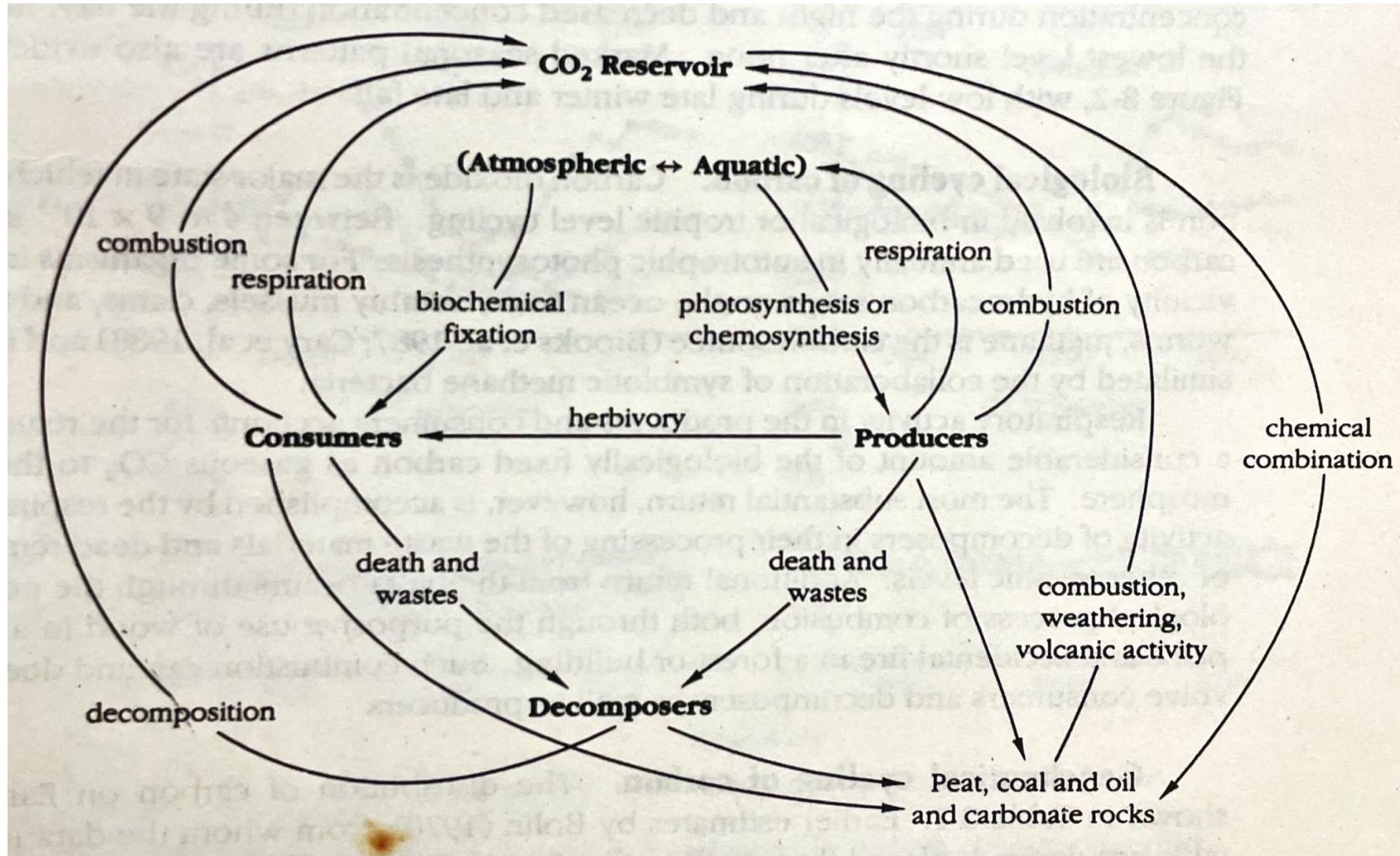
1,000 J

10,000 J

1,000,000 J of sunlight

Global Carbon Cycle





- Carbon constitute 49% of their dry weight.
- Carbon returned to the environment about as fast as it is removed.
- The basic movement of carbon is from the atmospheric reservoir to producers to consumers and from these groups to the decomposers and back to the reservoir.

Biological Recycling of Carbon

- Carbon dioxide is major state in which carbon is involved carbon is involved in biological trophic level cycling.
- For some organisms dwelling on the ocean floor (mussels, clams, worms, etc), methane can be the source of carbon.
- The methane in the ocean floor dwellers is assimilated by the collaboration of symbiotic methane bacteria.

- Respiratory activity in the producers and consumers, accounts for the return of a considerable amount of biologically fixed carbon as gaseous CO₂ to the atmosphere.
- Most substantial return is accomplished by the respiratory activity of decomposers in their processing of the waste materials and dead remains.
- Additional return from the biota occurs through the non biological process of combustion of wood and fossil fuels.

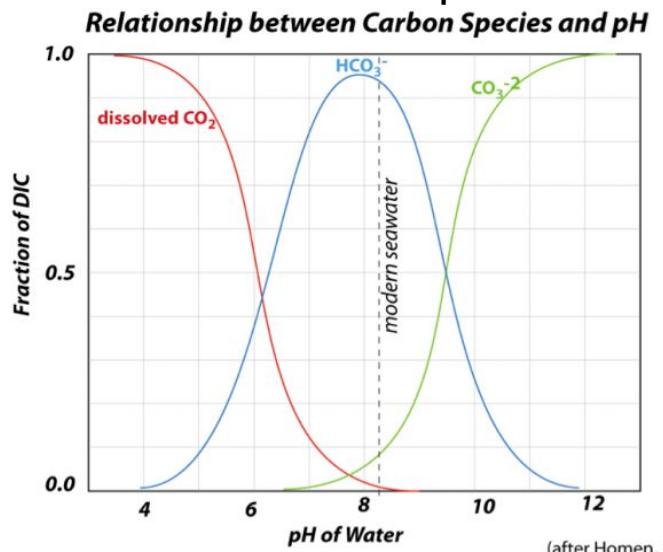
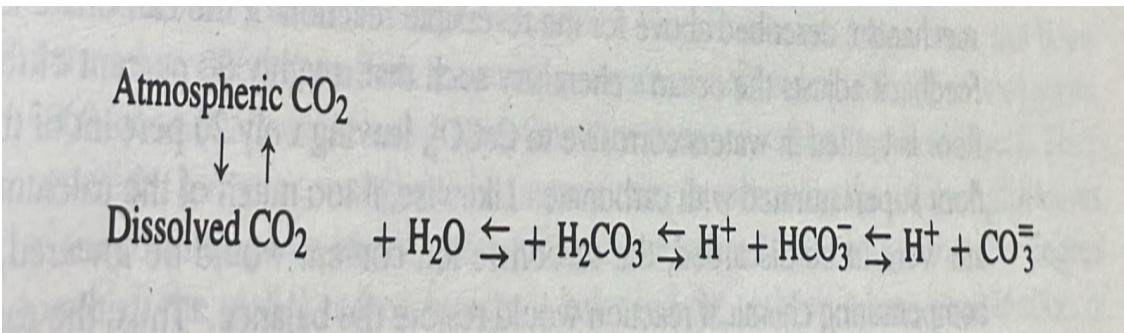
Geochemical cycling of carbon

- Major fraction of carbon is stored in the geologic component of the environment, out of which large portion is inorganic and in sedimentary form.
- It also indicates that major supply of carbon to the Earth's surface, where it is biologically involved is by erosion and metamorphism of sediments.
- Sedimentary carbon include animal remains such as protozoan remnants, coral & mollusks shells, echinoderm and vertebrate skeletal material, carbonate rocks.

- A number of aquatic plants occurring in alkaline waters release calcium carbonate as a byproduct of photosynthetic assimilation.
- On weathering and dissolution of carbonate rocks, the bound carbon returns to the atmospheric-aquatic reservoir.

Aquatic-atmospheric interchange of carbon

- The interplay between atmospheric and aquatic carbon is critical component of carbon recycling in the environment.



- Interchange between the phases occurs through diffusion. Passage into the aquatic phase also takes place through precipitation.
- Dissolved CO₂ combines with water in the soil or in aquatic ecosystem to form carbonic acid (H₂CO₃) in a reversible reaction. H₂CO₃ dissociates into H₂ and bicarbonate ions (HCO₃⁻) and further into carbonate ions CO₃²⁻.
- The direction of the reaction depends on the concentration of the components. The net movement of carbon between the different carbon species is much more complicated and also dependant on the pH of the water.
- Thus, manipulation of pH and CO₂ concentration affects the ability of the system to achieve equilibrium.

Inter-Species Relationships

Symbiosis

- Neutralism
- Competition
- Parasitism
- Predation
- Commensalism
- Protocooperation
- Mutualism

