

Electronic supplementary information

SYNTHESIS AND CRYSTAL STRUCTURE OF A NEW RHOMBOID $\{\text{Mn}^{\text{II}}_2\text{Mn}^{\text{III}}_2\}$ CLUSTER BASED ON CALIX[4]ARENE AND BATHOPHENANTHROLINE LIGANDS

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Single crystal X-ray diffraction study

Diffraction data for single-crystals of **3₂-Mn₄(bathophen)₂** were collected at 100 K on a Belok/XSA beamline ($\lambda = 0.745 \text{ \AA}$, ϕ -scans) of the Kurchatov Synchrotron Radiation Source (Moscow, Russian Federation) [S1, S2]. Diffraction patterns were collected using Mardtb goniometer equipped with Rayonix SX165 2D positional sensitive CCD detector at 100 K. In total, 450 frames were collected with oscillation range of 1° . The data were indexed, integrated and scaled; absorption correction was applied using the XDS program package [S3]. The structures were solved by direct methods with software SHELXT [S4]. The structural model was investigated and refined by using Olex2 software [S5] by a full-matrix least-squares method on F^2 with anisotropic displacement. Non-hydrogen atoms were refined anisotropically. All hydrogen atoms were placed in calculated positions and included in the refinement within the riding model with fixed isotropic displacement parameters $\text{Uiso}(\text{H}) = 1.5\text{Ueq}(\text{O})$, $1.2\text{Ueq}(\text{N})$, and $1.2\text{Ueq}(\text{C})$.

The crystallographic data and experimental details for **3₂-Mn₄(bathophen)₂** are presented in Table S1. CCDC 2383835 contains the supplementary crystallographic information for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via <https://www.ccdc.cam.ac.uk>.

Table S1. Crystallographic data and structure refinement parameters for the coordination compound obtained

Compound name	3₂-Mn₄(batophen)₂
Empirical formula	$\text{C}_{110}\text{H}_{86}\text{Mn}_4\text{N}_6\text{O}_{12}, 2(\text{CH}_4\text{O})$
Formula weight, $\text{g}\cdot\text{mol}^{-1}$	1967.69
Wavelength, \AA	0.745
Temperature	100(1)
Crystal system	triclinic
Space group	<i>P</i> -1
<i>a</i> , \AA	11.700(2)
<i>b</i> , \AA	15.170(3)
<i>c</i> , \AA	15.380(3)
α , $^\circ$	108.65(3)
β , $^\circ$	107.11(3)
γ , $^\circ$	103.43(3)
<i>V</i> , \AA^3	2305.8(10)
<i>Z</i> / <i>Z'</i>	1/0.5
ρ_{calc} , $\text{g}\cdot\text{cm}^3$	1.417

μ, cm^{-1}	0.684
F(000)	1020.0
Crystal size, mm^3	$0.3 \times 0.2 \times 0.2$
2 θ range for data collection, $^\circ$	$5.076 \leq 2\theta \leq 62.1$
Index ranges	$-16 \leq h \leq 16,$ $-20 \leq k \leq 21,$ $-21 \leq l \leq 21$
Reflections collected/independ.	57179/12278
R_{int}	0.0385
Completeness to $\theta_{\text{max}}/\%$	95.7
Data/restraints/parameters	12278/0/617
Goodness-of-fit on F^2	1.031
Final R indexes [$I > 2\sigma(I)$] R_1/wR_2	0.0415/0.1056
Final R indexes [all data]	0.0470/0.1090
R_1/wR_2	
$\rho_{\text{max}}/\rho_{\text{min}}, \text{e}\text{\AA}^{-3}$	0.71/−0.54

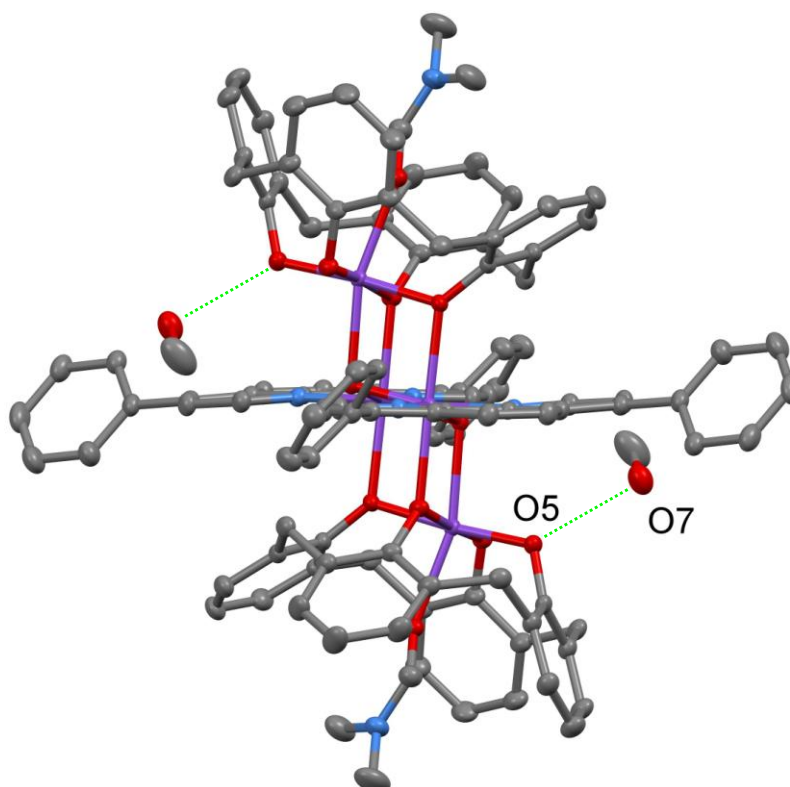
Table S2. Coordination bond angles for Mn ions of **3₂-Mn₄(bathophen)₂** ($^\circ$)

$\text{O}_X\text{--Mn(1)--O}_Y$	81.71(6)
	98.71(6)
	178.89(5)
	89.14(6)
	99.40(6)
	81.38(6)
$\text{O}_X\text{--Mn(1)--N}_X$	85.78(6)
	92.42(6)
	166.60(6)
	96.69(6)
	97.53(6)
	168.09(6)
$\text{N}_X\text{--Mn(1)--N}_Y$	93.11(6)
	87.41(6)
$\text{O}_X\text{--Mn(2)--O}_Y$	78.99(6)
	92.09(5)
	97.18(5)
	91.67(5)
	87.22(5)
	162.64(5)
	92.46(6)
	175.27(6)
	89.50(6)
	97.93(5)
	89.92(6)
	175.11(6)
	96.52(5)
	87.82(6)
	77.73(5)
	78.77(5)

Table S3. CShM deviation values calculated by *SHAPE* program for Mn atoms in $3_2\text{-Mn}_4(\text{bathophen})_2$

Complex name	Mn	HP-6	PPY-6	OC-6	TPR-6	JPPY-6
$3_2\text{-Mn}_4(\text{dmf})_4$	Mn(1)	29.386	24.558	1.319	13.669	27.791
	Mn(2)	31.288	26.303	1.329	14.137	29.970

Label	Symmetry	Shape
HP-6	D _{6h}	Hexagon
PPY-6	C _{5v}	Pentagonal pyramid
OC-6	O _h	Octahedron
TPR-6	D _{3h}	Trigonal prism
JPPY-6	C _{5v}	Johnson pentagonal pyramid (J2)

**Figure S1.** H-bonding between the solvate MeOH molecules with $3_2\text{-Mn}_4(\text{bathophen})_2$ cluster molecule within the crystal packing.

References

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- S3. W. Kabsch, *Acta Crystallogr., Sect. D: Biol. Crystallogr.*, **2010**, D66, 125–132. DOI: 10.1107/S0907444909047337
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