## **Electronic supplementary information**

## SYNTHESIS AND CRYSTAL STRUCTURE OF A NEW RHOMBOID {Mn<sup>II</sup><sub>2</sub>Mn<sup>III</sup><sub>2</sub>} CLUSTER BASED ON CALIX[4]ARENE AND BATHOPHENANTHROLINE LIGANDS

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## Single crystal X-ray diffraction study

Diffraction data for single-crystals of  $3_2$ -Mn<sub>4</sub>(bathophen)<sub>2</sub> were collected at 100 K on a Belok/XSA beamline ( $\lambda = 0.745$  Å,  $\phi$ -scans) of the Kurchatov Synchrotron Radiation Source (Moscow, Russian Federation) [S1, S2]. Diffraction patterns were collected using Mardtb goniometer equipped with Rayonix SX165 2D positional sensitive CCD detector at 100 K. In total, 450 frames were collected with oscillation range of 1°. The data were indexed, integrated and scaled; absorption correction was applied using the XDS program package [S3]. The structures were solved by direct methods with software SHELXT [S4]. The structural model was investigated and refined by using Olex2 software [S5] by a full-matrix least-squares method on F² with anisotropic displacement. Non-hydrogen atoms were refined anisotropically. All hydrogen atoms were placed in calculated positions and included in the refinement within the riding model with fixed isotropic displacement parameters Uiso(H) = 1.5Ueq(O), 1.2Ueq(N), and 1.2Ueq(C).

The crystallographic data and experimental details for 3<sub>2</sub>-Mn<sub>4</sub>(bathophen)<sub>2</sub> are presented in Table S1. CCDC 2383835 contains the supplementary crystallographic information for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre *via* https://www.ccdc.cam.ac.uk.

**Table S1**. Crystallographic data and structure refinement parameters for the coordination compound obtained

Compound name	32-Mn4(batophen)2
Empirical formula	$C_{110}H_{86}Mn_4N_6O_{12},$
Empiricar formula	2(CH <sub>4</sub> O)
Formula weight, g·mol <sup>-1</sup>	1967.69
Wavelength, Å	0.745
Temperature	100(1)
Crystal system	triclinic
Space group	P-1
a, Å	11.700(2)
b, Å	15.170(3)
c, Å	15.380(3)
α, °	108.65(3)
<i>β</i> , °	107.11(3)
γ, °	103.43(3)
$V, A^3$	2305.8(10)
Z/Z'	1/0.5
$\rho_{calc}$ g, cm <sup>3</sup>	1.417

μ, cm <sup>-1</sup>	0.684
F(000)	1020.0
Crystal size, mm <sup>3</sup>	$0.3\times0.2\times0.2$
2θ range for data collection, °	$5.076 \le 2\theta \le 62.1$
-	$-16 \le h \le 16$ ,
Index ranges	$-20 \le k \le 21$ ,
	$-21 \le 1 \le 21$
Reflections collected/independ.	57179/12278
R <sub>int</sub>	0.0385
Completeness to $\theta_{\rm max}$ /%	95.7
Data/restraints/parameters	12278/0/617
Goodness-of-fit on F <sup>2</sup>	1.031
Final R indexes [I > $2\sigma$ (I)] $R_1/wR_2$	0.0415/0.1056
Final R indexes [all data]	0.0470/0.1090
$R_1/wR_2$	
$ ho_{ m max}/ ho_{ m min},{ m e}{ m \AA}^{-3}$	0.71/0.54

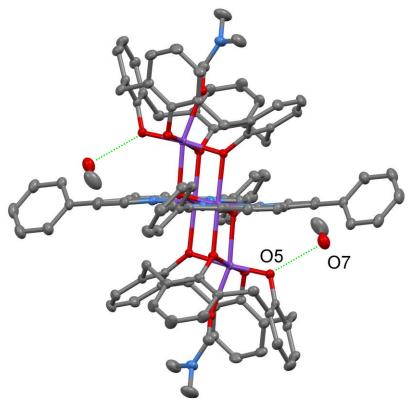
Table S2. Coordination bond angles for Mn ions of  $3_2$ -Mn<sub>4</sub>(bathophen)<sub>2</sub> (°)

	81.71(6)
$\mathbf{O}_{\mathbf{X}}$ -Mn(1)- $\mathbf{O}_{\mathbf{Y}}$	98.71(6)
	178.89(5)
	89.14(6)
	99.40(6)
	81.38(6)
	85.78(6)
	92.42(6)
	166.60(6)
O M. (1) N	96.69(6)
$\mathbf{O}_{\mathbf{X}}$ - $\mathbf{M}$ n(1)- $\mathbf{N}_{\mathbf{X}}$	97.53(6)
	168.09(6)
	93.11(6)
	87.41(6)
$\mathbf{N}_{\mathbf{X}}$ - $\mathbf{M}$ n(1)- $\mathbf{N}_{\mathbf{Y}}$	78.99(6)
	92.09(5)
	97.18(5)
	91.67(5)
	87.22(5)
	162.64(5)
	92.46(6)
	175.27(6)
$\mathbf{O}_{\mathbf{X}}$ - $\mathbf{M}$ n(2)- $\mathbf{O}_{\mathbf{Y}}$	89.50(6)
	97.93(5)
	89.92(6)
	175.11(6)
	96.52(5)
	87.82(6)
	77.73(5)
	78.77(5)

Table S3. CShM deviation values calculated by SHAPE program for Mn atoms in 32-Mn4(bathophen)2

Complex name	Mn	HP-6	PPY-6	OC-6	TPR-6	JPPY-6
2 Mr. (d., 6)	Mn(1)	29.386	24.558	1.319	13.669	27.791
$3_2$ -Mn <sub>4</sub> (dmf) <sub>4</sub>	Mn(2)	31.288	26.303	1.329	14.137	29.970

Label	Symmetry	Shape
HP-6	$\mathrm{D}_{6\mathrm{h}}$	Hexagon
PPY-6	$C_{5v}$	Pentagonal pyramid
OC-6	$\mathrm{O_{h}}$	Octahedron
TPR-6	$\mathrm{D}_{3\mathrm{h}}$	Trigonal prism
JPPY-6	$C_{5v}$	Johnson pentagonal
		pyramid (J2)



**Figure S1**. H-bonding between the solvate MeOH molecules with **32-Mn4(bathophen)2** cluster molecule within the crystal packing.

## References

- S1. V. A. Lazarenko, P. V. Dorovatovskii, Y. V. Zubavichus, A. S. Burlov, Y. V. Koshchienko, V. G. Vlasenko, V. N. Khrustalev, *Crystals*, **2017**, *7*, 325. DOI: 10.3390/cryst7110325
- S2. R. D. Svetogorov, P. V. Dorovatovskii, V. A. Lazarenko, *Cryst. Res. Technol.*, **2020**, *55*, 1900184. DOI: 10.1002/crat.201900184
- S3. W. Kabsch, *Acta Crystallogr.*, *Sect. D: Biol. Crystallogr.*, **2010**, *D66*, 125–132. DOI: 10.1107/S0907444909047337
- S4. G. M. Sheldrick, *Acta Crystallogr. A: Found. Adv.*, **2015**, *71*, 3–8. DOI: 10.1107/S2053273314026370
- S5. O. V. Dolomanov, L. J. Bourhis, R.J. Gildea, J. A. K. Howard, H. J. Puschmann, *J. Appl. Crystallogr.*, **2009**, *42*, 339–341. DOI: 10.1107/S0021889808042726