

TMSCA MIDDLE SCHOOL SCIENCE TEST#9 © JANUARY 30, 2021

GENERAL DIRECTIONS

- 1. About this test:
- A. You will be given 40 minutes to take this test.
- B. There are 50 problems on this test.
- 2. All answers must be written on the answer sheet/Scantron form/Chatsworth card provided. If you are using an answer sheet be sure to use **BLOCK CAPITAL LETTERS**. Clean erasures are necessary for accurate grading.
- 3. If using a Scantron answer form, be sure to correctly denote the number of problems not attempted.
- 4. You may write anywhere on the test itself. You must write only answers on the answer sheet.
- 5. You may use additional scratch paper provided by the contest director.
- 6. All problems have **ONE** and **ONLY ONE** correct [BEST] answer. There is a penalty for all incorrect answers.
- 7. On the back of this page is a copy of the periodic table of the elements as well as a list of some potentially useful information in answering the questions.
- 8. A simple scientific calculator with the following formulas is sufficient for the science contest: +, -, %, $^{\wedge}$, $\log x$, e^{x} , $\ln x$, y^{x} , $\sin x$, \sin^{-x} , $\cos x$, \cos^{-x} , $\tan x$, \tan^{-x} , with scientific notation and degree/radian capability.

The calculator must be silent, hand-held and battery operated. The calculator cannot be a computer or cannot have built-in or stored functionality that provides scientific information and cannot have communication capability. If the calculator has memory, it must be cleared. Each student may bring one spare calculator. **NO GRAPHING CALCULATORS ARE PERMITTED.**

- 9. All answers within \pm 5% will be considered correct.
- 10. All problems answered correctly are worth **FIVE** points. **TWO** points will be deducted for all problems answered incorrectly. No points will be added or subtracted for problems not answered.
- 11. In case of ties, percent accuracy will be used as a tie breaker.

1A 1	Periodic Table of the Elements																
1 H	2A 2											за 13	4A 14	^{5A} 15	6A 16	^{7А} 17	2 He
3 Li 6.94	4 Be _{9.01}											5 B 10.81	6 C 12.01	7 N 14.01	8 O 16.00	9 F 19.00	10 Ne 20.18
11 Na 22.99	12 Mg _{24.31}	3B 3	4B 4	5B 5	6B 6	7В 7	8	—8B—	10	1B 11	2B 12	13 Al _{26.98}	14 Si _{28.09}	15 P 30.97	16 S 32.07	17 Cl 35.45	18 Ar 39.95
19 K 39.10	20 Ca 40.08	21 Sc 44.96	22 Ti 47.87	23 V 50.94	24 Cr 52.00	25 Mn 54.94	26 Fe 55.85	27 Co 58.93	28 Ni 58.69	29 Cu 63.55	30 Zn 65.38	31 Ga _{69.72}	32 Ge 72.64	33 As 74.92	34 Se _{78.96}	35 Br 79.90	36 Kr 83.80
37 Rb 85.47	38 Sr 87.62	39 Y 88.91	40 Zr 91.22	41 Nb 92.91	42 Mo _{95.94}	43 Tc (98)	44 Ru 101.07	45 Rh 102.91	46 Pd 106.42	47 Ag 107.87	48 Cd 112.41	49 In 114.82	50 Sn 118.71	51 Sb 121.76	52 Te 127.60	53 126.90	54 Xe 131.29
55 Cs 132.91	56 Ba 137.33	57 La 138.9	72 Hf 178.49	73 Ta 180.95	74 W 183.84	75 Re 186.21	76 Os 190.23	77 r 192.22	78 Pt 195.08	79 Au 196.97	80 Hg _{200.59}	81 TI 204.38	82 Pb 207.20	83 Bi _{208.98}	Po (209)	85 At (210)	86 Rn (222)
87 Fr (223)	88 Ra (226)	89 Ac (227)	104 Rf (261)	105 Db (262)	106 Sg (266)	107 Bh (264)	108 Hs (277)	109 Mt (268)	110 Ds (281)	111 Rg (281)	112 Cn (285)	113 Nh (286)	114 FI (289)	115 Mc (289)	116 Lv (293)	117 Ts (293)	118 Og (294)

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dν	Но	l Er	Tm	Yb	Lu
140.1	140.9	144.2	(145)	150.4	152.0	157.3	158.9	162.5	164.9	167.3	168.9	173.0	175.0
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
232.0	231.0	238.0	(237)	(244)	(243)	(247)	(247)	(251)	(252)	(257)	(258)	(259)	(262)

OTHER USEFUL INFORMATION

Acceleration of gravity at Earth's surface, $g = 9.81 \text{ m/s}^2$

Avogadro's Number, N = 6.02 x 10²³ molecules/mole

Planck's constant, $h = 6.63 \times 10^{-34} \text{ J} \cdot \text{s}$

Planck's reduced constant, $\hbar = h/2\pi = 1.05 \text{ X } 10^{-34} \text{ J} \bullet \text{s}$

Standard temperature and pressure (STP) is 0°C and I atmosphere

Gram molecular volume al STP = 22.4 liters

Velocity of light, $c = 3.0 \times 10^8 \text{ m/sec}$

Absolute zero= 0 K = -273.15°C

Gas constant, R = 1.986 col/K•mole = 0.082 liter•otm/K•mole

One Faraday= 96,500 coulombs (9 .65 x 10⁴ C)

Dulong and Pelil's constant= 6.0 amu•cal/gram•K

Electron rest mass, $m_e = 9.11 \times 10^{-31} \text{ kg}$

Atomic mass unit, $m_u = 1.66 \times 10^{-21} \text{ kg}$

Boltzmann constant, $k_B = 1.38 \times 10^{-23} \text{ J/K}$

Permittivity of free space ε_0 = 8.85 x 10^{-12} C²/N•m²

Permeability of free space $\mu_0 = 4\pi \times 10^{-7} \text{ T} \cdot \text{m/A}$

1 Atmosphere= $1.02 \times 10^5 \text{ N/m}^2 = 760 \text{ Torr} = 760 \text{ mmHg}$

1 Electron Volt - 1.6 x 10⁻¹⁹ Joules

Charge of on electron" -1.6 x 10^{-19} coulombs (C)

1 horsepower (hp) = 746 W = 550 ft•lb/s

Neutron Moss= 1.008665 au

Proton Mass= 1.007277 au

1 au= 931.5 MeV

1 calorie= 4.184 Joules (J)

Specific heal of water= 4.18 J/g• °C

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1.	Which of th	ne drawings	below is consi	dered to have pal	mate venation?	
	ONE		TWO	THE	ZEE	FOUR
			33		3	
	A. 1		B. 2	C.	3	D. 4
	A. was B. woo C. was	"apex preda s a skilled h uld only hu s eaten by a	tor." What nic unter and steal nt during the ni Il the other anir	he did this creature thy at finding food	re have in this end	dimetrodon that they nvironment?
3.	The commo		mass that stars	and planets actual C. apogee	ally orbit around D. barycen	is called the what?
4.	_			t" is to hemoglobi		D. inco
	A. blo	oa	B. oxyger	1	C. plants	D. iron
	• •		ction in which ectly joined is	two or more simp	le substances, e	ither elements or
511	A. dec B. con C. rep	composition nbination or lacement dation/redu	synthesis	caned what:		
Th	e area affec	ted looked l what type o		-	_	e landscape drastically. I to the area. This is
7.	The prefix A. all	-	ns what in medi B. far	cal terminology? C. under	D. both A and	i B

- 8. In Texas, some streams flow all year continuously (Type A), some at only certain times of year when springs or surface water provide water for it (Type B), and some flow only after precipitation, but don't have a spring or continuous surface water source (Type C). Which list below correctly matches these stream types to the flow type?
 - A. Type A perennial, Type B- intermittent, Type C-ephemeral
 - B. Type A ephemeral, Type B- intermittent, Type C- perennial
 - C. Type A intermittent, Type B- perennial, Type C-ephemeral
 - D. Type A perennial, Type B- ephemeral, Type C-intermittent

Shower	Approximate Dates	Associated Comet
Quadrantids	Jan. 4-6	
Lyrids	April 20-23	Comet 1861 I
Perseids	August 12	Comet 1862 I
Leonids	November 18	Comet 1866 I
Geminids	Dec. 4-16	

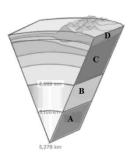
9. This table shows some of the major meteor showers.

What meteor shower would be happening during the spring season? (in northern hemisphere)

- A. Lyrids
- B. Perseids
- C. Leonids
- D. Geminids
- 10. Which of the following statements about the Coriolis force is untrue?
 - A. The Coriolis force applies to movement on rotating objects.
 - B. The Coriolis force is zero at the equator.
 - C. The Coriolis force is weakest at the poles.
 - D. Storms in the north swirl counter-clockwise.
- 11. This moon phase would be called what?
 - A. 1st quarter
 - B. 3rd quarter
 - C. full moon
 - D. gibbous



- 12. Which list below shows the Galilean moons of Jupiter in order from farthest from Jupiter to closest to Jupiter?
 - A. Callisto, Ganymede, Europa, Io
 - B. Io, Europa, Ganymede, Callisto
 - C. Europa, Ganymede, Callisto, Io
 - D. Ganymede, Io, Callisto, Europa
- 13. In this model of a slice of the Earth, what letter best represents the layer called the mantle?
 - A. A
- B. B
- C. C
- D. D



	ny noticed some hig e of cloud was this		s that had a feathe	rlike appearance to them.
• •			rus D. nimbo	ostratus
	her on Earth takes stratosphere B.	place in what layer of ionosphere C	-	D. troposphere
balloon po	opped. Which law	explains why this ha	ppened?	e. After a short time, the
	-	A shows the elemen		the pentagon represents a ox B shows them bonded.
	0	0 00		
A. B. C.	no, because the eno, because the elyes, because they	A to Box B represent lements have not chall lements only changed combined to make a are in the same state	nged any l phase new substance	ge?
18. When what?	looking at the per	iodic table, the eleme	ents found on the	right side of the table are
A.	metals	B. nonmetals	C. metalloids	D. rare Earth metals
	vice that stores che transformer	mical energy and con B. generator	verts it to electric C. conduit	al energy is called a what? D. battery
	is the SI unit to mampere	easure power? B. watt	C. hertz	D. newton
	conditions below warm and wet	are more favorable for B. cold and dry	•	to be present? y D. cold and wet
	famous family of Leakey	scientists has won the B. Cousteau	e most Nobel prize C. Irwin	es? D. Curie

- 23. Simon bought a 2-liter bottle of soft drink. He had a recipe to use 3 quarts of soft drink. Does he have enough soft drink for this recipe?
 - A. yes, he will have just enough, 3.114 quarts
 - B. yes, he will have enough because 1 quart is half a liter
 - C. no, he will not have enough because 2 L is less than 3 quarts
 - D. no, he will not have enough because 3 quarts is less than the 2 L bottle
- 24. Allison was tasting a new type of candy. She put it on the tip of her tongue because she heard that the tip of the tongue was the section that tastes sweets the best. Which statement below about tastes and the tongue is true?
 - A. Allison is correct. The tip of the tongue is best for tasting sweets.
 - B. Any part of the tongue can test the sweets.
 - C. The back part of the tongue is for tasting sweets.
 - D. The tongue has sections for tasting different types of tastes and the middle is for sweets.
- 25. Scientists were trying to find out what causes locusts to gather in large groups (which can be devasting to crops). They know that insects emit pheromones that can communicate to others of their species. Knowing this, a group of scientists decided to try to locate the one chemical scent that these locusts use to aggregate. They isolated 6 scents that might be the possible one they use to get a group to gather together. They tested the scents on locusts separately to see if the locusts would gather in a large group.

What would be the dependent variable in this experiment?

- A. whether or not the locusts gathered together when the scent was used
- B. the six types of scents
- C. the time it takes for the locust to gather together
- D. the amount of scent used in the experiment
- 26. The prefix "micro" means what?
 - A. small B. large
 - large C. lens
- D. medium
- 27. Saltwater has a lower freezing point temperature than freshwater. Which statement below supports this?
 - A. When you mix salt with ice water, the water temperature increases.
 - B. The ocean freezes faster than a freshwater lake.
 - C. When making homemade ice cream, the salty water slush helps to slow down the freezing process.
 - D. When you put salt on ice on a sidewalk, the ice melts.
- 28. Which of the following organisms are not considered to be herbivores?
 - A. white-tailed deer
- B. monarch caterpillars
- C. jackrabbits
- D. raccoons
- 29. The act of an organism hunting, killing, and eating another organism is called what?
 - A. predation
- B. commensalism
- C. parasitism
- D. mutualism

	A. Some flowers may not have all four of the basic flower parts – sepals, petals, stamens and pistils. B. Ovules develop in the ovary of the flower. C. The stamens produce the pollen for the flower. D. The sepals produce the ovules of the flower that make the seeds.
plant m	nat do you call the muscular chamber found inside a bird's stomach where consumed naterial is ground into a pulp? A. keel B. gizzard C. crop D. pygostyle
	nen in the field of genetics, you may see the acronym CRISPR. What does this stand for? A. crimson regular international speedy progress repetition B. clustered regularly interspaced short palindromic repeats C. counting regular interval signs per ribonucleic acid D. none of these
	e way to tell the age of a freshwater fish is to do what? A. look at the formation of the gills B. measure the length of the caudal fin C. look at the growth rings on the scales D. measure the opening in the mouth
This all called?	me plants release protective chemicals to keep other plants from growing close to them. lows the plants to grow without other competition for resources. What is this process A. allelopathy B. germination C. tropisms D. auxin
through	is part found in the circulatory system makes sure that the blood or fluid that passes a the vessels and passageways continue to move in the right direction. What is it? A. valve B. capillaries C. node D. platelet
	an insect belongs to the order Lepidoptera, then it may be a what? A. moth B. bee C. butterfly D. Both A and C
	nich of the following would be found on an insect leg? A. tarsus B. spiracle C. antennae D. ganglion
	e suffix "emia" associates the root word with what? A. blood B. pain C. breathing D. inflammation

dev	Mammals can be classified into subclasses by how they reproduce and the embryonic elopment. Which of the following mammals are born very tiny and then climb into their ther's pouch? A. marsupials B. placental
	C. monotremes D. no mammals develop this way
40.	Which statement below about Monarch butterflies is not true? A. Monarchs sense most smells with their antennae. B. Monarchs use chemoreceptors on the back of the leg to taste dissolved sugar. C. Monarchs can see polarized light. D. Monarchs have exceptional hearing.
41.	The specialized tissue of a plant that transports water to the leaves is called what? A. phloem B. xylem C. cambium D. both A and C
42.	Which of the following organisms have a parasitic relationship with a host? A. helminths B. amoebas C. copepods D. all of these
43.	Which of the following cells are somatic cells in humans? A. germ cells B. sperm cells C. nerve cells D. egg cells
	A chemical reaction in which a compound (reactant) is broken down into simpler elements compounds (products) is called what? A. decomposition B. synthesis C. replacement D. oxidation/reduction
45.	What is the atomic number for Polonium? A. 84 B. 15 C. 78 D. 208
46.	Which of these planets in our solar system has the lowest amount of axial tilt? A. Venus B. Mercury C. Mars D. Jupiter
47.	What moon of Jupiter has been noted as the only moon with a magnetic field? A. Io B. Europa C. Titan D. Ganymede

48. Look at the chart below.

Gas in atmosphere	Earth without life	Earth with life
Carbon dioxide	98%	0.03%
Nitrogen	1.9 %	78%
Oxygen	trace	21%
Argon	0.1%	1%

According to this projection of what Earth's atmosphere would be without life present on Earth, what is the most affected gas?

- A. Argon
- B. Oxygen
- C. Nitrogen
- D. Carbon dioxide
- 49. The moon rotates as it revolves around the Earth just enough to do what?
 - A. rotate 8 times before the month is over
 - B. keep the same side of the moon always facing Earth
 - C. create enough gravity to keep moon in place
 - D. cause craters to form on both sides
- 50. What type of clouds are in this photo?
 - A. stratocumulus
 - B. cumulonimbus
 - C. stratus
 - D. cirrocumulus



2020 - 2021 TMSCA Middle School Science Test #9 - Key

18. B	35. A
19. D	36. D
20. B	37. A
21. B	38. A
22. D	39. A
23. C	40. D
24. B	41. B
25. A	42. D
26. A	43. C
27. D	44. A
28. D	45. A
29. A	46. B
30. D	47. D
31. B	48. D
32. B	49. B
	19. D 20. B 21. B 22. D 23. C 24. B 25. A 26. A 27. D 28. D 29. A 30. D 31. B

33. C

34. A

50. D

16. C

17. C