

# TMSCA MIDDLE SCHOOL SCIENCE REGIONAL TEST© MARCH 27, 2021

### **GENERAL DIRECTIONS**

- 1. About this test:
- A. You will be given 40 minutes to take this test.
- B. There are 50 problems on this test.
- 2. All answers must be written on the answer sheet/Scantron form/Chatsworth card provided. If you are using an answer sheet be sure to use **BLOCK CAPITAL LETTERS**. Clean erasures are necessary for accurate grading.
- 3. If using a Scantron answer form, be sure to correctly denote the number of problems not attempted.
- 4. You may write anywhere on the test itself. You must write only answers on the answer sheet.
- 5. You may use additional scratch paper provided by the contest director.
- 6. All problems have **ONE** and **ONLY ONE** correct [BEST] answer. There is a penalty for all incorrect answers.
- 7. On the back of this page is a copy of the periodic table of the elements as well as a list of some potentially useful information in answering the questions.
- 8. A simple scientific calculator with the following formulas is sufficient for the science contest: +, -, %,  $^{\wedge}$ , log x,  $e^{x}$ , lnx,  $y^{x}$ , sin x, sin  $^{-x}$ , cos x, cos  $^{-x}$ , tan x, tan  $^{-x}$ , with scientific notation and degree/radian capability.

The calculator must be silent, hand-held and battery operated. The calculator cannot be a computer or cannot have built-in or stored functionality that provides scientific information and cannot have communication capability. If the calculator has memory, it must be cleared. Each student may bring one spare calculator. **NO GRAPHING CALCULATORS ARE PERMITTED.** 

- 9. All answers within  $\pm$  5% will be considered correct.
- 10. All problems answered correctly are worth **FIVE** points. **TWO** points will be deducted for all problems answered incorrectly. No points will be added or subtracted for problems not answered.
- 11. In case of ties, percent accuracy will be used as a tie breaker.

1A 1			Pe	erio	dic	Ta	ble	of	the	e El	em	ent	ts				8A 18
1 H	2A 2											за <b>13</b>	4A <b>14</b>	<sup>5A</sup> <b>15</b>	6A <b>16</b>	<sup>7А</sup> 17	2 He
3 Li 6.94	4 Be <sub>9.01</sub>											5 B 10.81	6 C 12.01	7 N 14.01	8 O 16.00	9 F 19.00	10 Ne 20.18
11 Na 22.99	12 Mg <sub>24.31</sub>	3B <b>3</b>	4B <b>4</b>	5B <b>5</b>	6B <b>6</b>	7В 7	8	—8B—	10	1B <b>11</b>	2B 12	13 Al <sub>26.98</sub>	14 Si <sub>28.09</sub>	15 P 30.97	16 S 32.07	17 Cl 35.45	18 Ar 39.95
19 K 39.10	20 Ca 40.08	21 Sc 44.96	22 Ti 47.87	23 V 50.94	24 Cr 52.00	25 Mn 54.94	26 Fe 55.85	27 Co 58.93	28 Ni 58.69	29 Cu 63.55	30 Zn 65.38	31 Ga <sub>69.72</sub>	32 Ge 72.64	33 As 74.92	34 Se 78.96	35 Br 79.90	36 Kr 83.80
37 Rb 85.47	38 Sr 87.62	39 Y 88.91	40 Zr 91.22	41 Nb <sub>92.91</sub>	42 Mo <sub>95.94</sub>	43 Tc (98)	44 Ru 101.07	45 Rh 102.91	46 Pd 106.42	47 Ag 107.87	48 Cd 112.41	49 In 114.82	50 Sn 118.71	51 Sb 121.76	52 Te 127.60	53     126.90	54 Xe 131.29
55 Cs 132.91	56 Ba 137.33	57 La 138.9	72 Hf 178.49	73 Ta 180.95	74 W 183.84	75 Re 186.21	76 Os 190.23	77  r   192.22	78 Pt 195.08	79 Au 196.97	80 Hg <sub>200.59</sub>	81 TI 204.38	82 Pb 207.20	83 Bi <sub>208.98</sub>	Po (209)	85 At (210)	86 Rn (222)
87 Fr (223)	88 Ra (226)	89 Ac (227)	104 Rf (261)	105 Db (262)	106 Sg (266)	107 Bh (264)	108 Hs (277)	109 Mt (268)	110 Ds (281)	111 Rg (281)	112 Cn (285)	113 Nh (286)	114 FI (289)	115 Mc (289)	116 Lv (293)	117 Ts (293)	118 Og (294)

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dν	Но	l Er	Tm	Yb	Lu
140.1	140.9	144.2	(145)	150.4	152.0	157.3	158.9	162.5	164.9	167.3	168.9	173.0	175.0
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
232.0	231.0	238.0	(237)	(244)	(243)	(247)	(247)	(251)	(252)	(257)	(258)	(259)	(262)

### OTHER USEFUL INFORMATION

Acceleration of gravity at Earth's surface,  $g = 9.81 \text{ m/s}^2$ 

Avogadro's Number, N = 6.02 x 10<sup>23</sup> molecules/mole

Planck's constant,  $h = 6.63 \times 10^{-34} \text{ J} \cdot \text{s}$ 

Planck's reduced constant,  $\hbar = h/2\pi = 1.05 \text{ X } 10^{-34} \text{ J} \bullet \text{s}$ 

Standard temperature and pressure (STP) is 0°C and I atmosphere

Gram molecular volume al STP = 22.4 liters

Velocity of light,  $c = 3.0 \times 10^8 \text{ m/sec}$ 

Absolute zero= 0 K = -273.15°C

Gas constant, R = 1.986 col/K•mole = 0.082 liter•otm/K•mole

One Faraday= 96,500 coulombs (9 .65 x 10<sup>4</sup> C)

Dulong and Pelil's constant= 6.0 amu•cal/gram•K

Electron rest mass,  $m_e = 9.11 \times 10^{-31} \text{ kg}$ 

Atomic mass unit,  $m_u = 1.66 \times 10^{-21} \text{ kg}$ 

Boltzmann constant,  $k_B = 1.38 \times 10^{-23} \text{ J/K}$ 

Permittivity of free space  $\varepsilon_0$  = 8.85 x  $10^{-12}$  C<sup>2</sup>/N•m<sup>2</sup>

Permeability of free space  $\mu_0 = 4\pi \times 10^{-7} \text{ T} \cdot \text{m/A}$ 

1 Atmosphere=  $1.02 \times 10^5 \text{ N/m}^2 = 760 \text{ Torr} = 760 \text{ mmHg}$ 

1 Electron Volt - 1.6 x 10<sup>-19</sup> Joules

Charge of on electron" -1.6 x  $10^{-19}$  coulombs (C)

1 horsepower (hp) = 746 W = 550 ft•lb/s

Neutron Moss= 1.008665 au

Proton Mass= 1.007277 au

1 au= 931.5 MeV

1 calorie= 4.184 Joules (J)

Specific heal of water= 4.18 J/g• °C

## 2020-2021 TMSCA Middle School Science Test – Regional

1. When discussing weather phenomenon, what is a "derecho"?

B. a huge wave o C. a line of intens	f devasting water that se, long-lived, fast m	ms that cause major in at occurs in a bay or loving windstorms the and composed of ice	agoon area at can cause major damage
2. Which of three bird fe a woodpecker that climbs bark?  A. A. B. B. C. C. D. none of these	• •	g to A.	B. C.
3. Owls cannot chew the feathers. Because of this. These little packets are ca	, they expel little pac		d cannot digest bones, fur, and ones, fur, and feathers.
A. coprolites	B. regurgitate	C. pellets	D. vomit
the water?			ows it to move up or down in
A. pharynx	B. air bladder	C. operculum	D. gills
5. Jennifer was working grandmother has a funny Grandmother said that wa longer give this vaccine to grandmother get vaccinate collaborative global vaccinate.  A. polio	looking scar on her as from a vaccination o kids because the died for? (This disease	arm so she asked her she got when she w sease was eradicated	grandmother about it. as a kid, but now they no . What disease did Jennifer's
A. pono	B. Tables	c. ellekeli pox	D. Smanpox
6. If an insect belongs to A. grasshoppers	the order Mantodea, B. praying mantis	•	at? D. wasps
7. A carnivorous animal A. piscivore	that primarily eats fi B. herbivores	sh is called a what? C. omnivores	D. detritivores

- 13. Students were conducting an investigation on the properties of water. They poured a cup of water down a string into a bowl. The water appeared to "cling" to the string as it went into the bowl. What does this investigation demonstrate?
  - A. water has the property of static
  - B. water is composed of hydrogen and oxygen molecules
  - C. water is the universal solvent
  - D. water has the property of adhesion
- 14. Thunder and lightning frightened a herd of bison and started a stampede. The bison covered a distance of 1.7 miles in 6 minutes. What was the average speed of the bison?
  - A. 28 miles per hour
  - B. 35 miles per hour
  - C. 10 miles per hour
  - D. 17 miles per hour



- 15. The photo shows a dimetrodon skeleton. Sometimes fossil skeletons are found "articulated". What does articulated mean?
  - A. upside down
  - B. taken apart into pieces
  - C. scattered in no particular order
  - D. arranged in the correct order



- 16. A protozoan parasite that infects Monarch butterflies and causes them to fail to emerge properly from their pupal state is what?
  - A. Pseudogymnoascus destructans
  - B. Naegleria fowleri
  - C. Plasmodium falciparum
  - D. Ophryocystis elektroscirrha
- 17. The Hertzsprung-Russell diagram is used to show all of the following about stars except what?
  - A. intrinsic brightness and surface temperature
  - B. absolute magnitude and spectral type
  - C. luminosity and surface temperature or color
  - D. the photon absorption rates
- 18. Monarch butterflies go through which type of metamorphosis?
  - A. incomplete
- B. complete
- C. partial
- D. Both A and C

- 19. There is a reddish substance of a waste product that Monarch butterflies expel when they leave the chrysalis. What is this substance called?
  - A. pheromones
  - B. glycosides
  - C. meconium
  - D. alkaloids
- 20. Monarch butterflies migrate from Canada to Mexico during the fall to overwinter in warmer temperatures. Researchers want to know what the monarch butterflies use in order to orientate their flight direction. Maybe they use the position of the sun, features of the landscape, the magnetic field, or something that we don't know about yet to help them know which way to fly. What would be the most reasonable testable hypothesis for this investigation?
  - A. When monarch butterflies see a body of water, they will fly over it.
  - B. Every year, the monarch butterflies come to the same tree in my backyard.
  - C. When a monarch butterfly comes near a magnet, it flies in a sporadic manner 75% of the time.
  - D. When given several choices of sunlight angles, the monarch butterflies will choose to fly with the sun is at 57 degrees above the horizon at noon.
- 21. How many neutrons would a neutral atom of Iron have?
  - A. 26
- B. 56
- C. 30
- D. 82
- 22. Which of the following shows a pair of inorganic compounds?
  - A. NaCl CaCO<sub>3</sub>
  - B. C<sub>3</sub>H<sub>8</sub> C<sub>2</sub>H<sub>7</sub>O<sub>4</sub>P
  - C. C<sub>2</sub>H<sub>6</sub> CH<sub>4</sub> N<sub>2</sub>O
  - D.  $C_6H_{12}O_6$   $H_2O$
- 23. Why does Earth have a "leap year"?
  - A. because the sun has a slight wobble
  - B. to line up the planets with their revolutions around the sun
  - C. to celebrate an extra day as Earth Day in the calendar
  - D. to synchronize the human calendar year with the solar year
- 24. What is the axial tilt of planet Earth?
  - A. 23.5 degrees
- B. 25 degrees
- C. 90 degrees
- D. 28.2 degrees

- 25. What is the name of this moon phase?
  - A. waxing crescent
  - B. waning crescent
  - C. 1<sup>st</sup> quarter
  - D. waning gibbous



<ul> <li>26. The "far side" of the moon is what?</li> <li>A. changes depending on where you are located on Earth</li> <li>B. the side of the moon that never faces Earth because of the moon's rotation and revolution around the Earth.</li> <li>C. where the aliens live</li> <li>D. the side of the moon that never receives sunlight</li> </ul>
<ul> <li>27. Which statement below bests describes a cloud in weather?</li> <li>A. a large mass of water vapor suspended in the air formed by wind</li> <li>B. an invisible mass of air containing particles of matter including smoke</li> <li>C. a visible mass of water droplets, ice crystals, or a mixture suspended in the air</li> <li>D. a collection of ice crystals that is held aloft by wind currents in the sky</li> </ul>
<ul> <li>28. What person is given credit for developing the first true battery?</li> <li>A. Heinrich Hertz</li> <li>B. Nikola Tesla</li> <li>C. Thomas Edison</li> <li>D. Alessandro Volta</li> </ul>
<ul> <li>29. When observing properties of light waves of the electromagnetic spectrum, which of the following statements below is true?</li> <li>A. When the wavelength increases, the frequency will also increase.</li> <li>B. The wavelength and frequency have no connection to each other.</li> <li>C. When the wavelength increases, the frequency will decrease.</li> <li>D. The wavelength and frequency of waves are equal to each other.</li> </ul>
30. Which of the following is not an example of a man-made plasma?  A. lightning B. neon light C. welding arc D. fluorescent light
31. How many milliliters are there in 5 kilometers?  A. 1 million B. 5 million C. 500,000 D. there is no way to tell with this information
32. I was a pioneer of modern atomic theory. I also identified that red-green color blindness was hereditary. The law that states that the total pressure of a mixture of gases is equal to the partial pressures of the individual gases that make it up is named after me. Who am I?  A. John Dalton B. Democritus C. Niels Bohr D. Ernest Rutherford
33. The prefix "macro" means what? A. small B. large C. crisp D. taste

<ul> <li>34. Why do people add rock salt to the ice bucket when making homemade ice cream?</li> <li>A. it helps by lowering the freezing point of the ice-salt mixture</li> <li>B. it helps by raising the freezing point of the ice-salt mixture</li> <li>C. it helps by slowing down the freezing process for the ice cream</li> <li>D. it helps in no way, the rock salt companies just want to make money</li> </ul>
35. The root word "acanth" means what? A. spiny B. smooth C. red D. tall
<ul><li>36. Which element listed below has an extremely low boiling point?</li><li>A. Sulfur B. Iodine C. Tungsten D. Helium</li></ul>
37. An atom of hydrogen has 1 proton and 2 electrons. This means the net charge is what?  A. 1 + (cation) B. 1 - (anion) C. 2+ (cation) D. 2- (anion)
38. An atom or molecule with a net charge that is either positive or negative is called what?  A. isotope B. ion C. neutron D. electron
<ul> <li>39. Which of the following shows evidence that a chemical reaction took place?</li> <li>A. ice melting into a puddle of water</li> <li>B. a poster changing color after exposure to the sun for a week</li> <li>C. salt dissolving in a cup of water</li> <li>D. dry ice changing to gas after setting on counter for a few hours</li> </ul>
<ul> <li>40. If 1 kilometer is equal to 3,281 feet, how many feet are there in 5 kilometers?</li> <li>A. 16,405</li> <li>B. 656.2</li> <li>C. 15,300</li> <li>D. there is no way to tell with this information</li> </ul>
<ul> <li>41. I discovered that mass is conserved when a chemical reaction takes place. I was a French chemist who worked during the late 18<sup>th</sup> century. Who am I?</li> <li>A. Niels Bohr</li> <li>B. Louis Pasteur</li> <li>C. Joseph Proust</li> <li>D. Antonine Lavoisier</li> </ul>
42. Jamie applied 100 N of force on the piano to push it to the east side of the room. Her friend pushed with 250 N of force. The piano did not move. How much work was done on the piano?
A. 0 J B. 100 J C. 250 J D. 350 J
43. Which of the following is not attracted to a magnet?  A. iron B. cobalt C. aluminum D. nickel

44. \	A. B. C.	very few life form	tation, less than 25		system?			
45. 7		eutral atom has an Platinum	atomic mass of 133 B. Neptunium	and 78 neutr C. Argon		element is t D. Cesiun		
	A. B. C.	throw away this so send it back to the turn it on/off agair	as on the lab table, less to be done before cale and get a new of place you bought it in or push the tare but on the scale and push	using this scane ne from tton dependir	ng on the ty	pe of scale	14	
47. <b>V</b>	A. B. C.	moon phase would full moon new moon waning crescent waning gibbous	come next (#4)?		2.	3.	?	
	found A. B. C.	animal would most d under a highway yellow-jacket was cliff swallows flying squirrels burrowing owls	•	these structure	es that			
49. V	A. B.	G, U, C A, T, A G, A, C	in this DNA sequen	TICIAITIA	from left to	Па		
	A. cai B. cai	n be conserved if a n be created by tak	n of Energy states the ll types of energy are ing mass and conve- en the mass is more	e utilized. rting it to ene				

D. can neither be created or destroyed, it just changes form.

# 2020 - 2021 TMSCA Middle School Science Test- Regional - Key

1. C	18. B	35. A
2. A	19. C	36. D
3. C	20. D	37. B
4. B	21. C	38. B
5. D	22. A	39. B
6. B	23. D	40. A
7. A	24. A	41. D
8. D	25. A	42. A
9. C	26. B	43. C
10. C	27. C	44. A
11. D	28. D	45. D
12. C	29. C	46. C
13. D	30. A	47. B
14. D	31. D	48. B
15. D	32. A	49. D
16. D	33. B	50. D

17. D 34. A