

a. Asociación del ligando auxiliar

Amoniaco $\log \beta = 9,37$

b. Disociación de los quelones

EDTA $pQ_{\alpha 1} = 1,91$; $pQ_{\alpha 2} = 2,68$; $pQ_{\alpha 3} = 6,11$; $pQ_{\alpha 4} = 10,17$

Calmagita $pQ_{\text{Ind}1} = 8,10$; $pQ_{\text{Ind}2} = 12,4$

Calcón $pQ_{\text{Ind}1} = 7,00$; $pQ_{\text{Ind}2} = 13,5$

c. Constantes de estabilidad de los diferentes complejos

Calcio	EDTA	$\log Q_{\text{CaY}} = 10,7$
	Amoniaco	$\log \beta_1 = -0,2$; $\log \beta_2 = -0,8$; $\log \beta_3 = -1,6$; $\log \beta_4 = -2,7$
	Calcón	$\log C_{\text{Ind}} = 5,3$
	Calmagita	$\log C_{\text{Ind}} = 6,1$

Magnesio	EDTA	$\log Q_{\text{MgY}} = 8,70$
	Amoniaco	$\log \beta_1 = 0,23$; $\log \beta_2 = 0,08$; $\log \beta_3 = -0,36$; $\log \beta_4 = -1,1$
	Calcón	$\log M_{\text{gInd}} = 7,6$
	Calmagita	$\log M_{\text{gInd}} = 8,10$

d. Constantes de estabilidad de los quelatos protonados

Calcio $Q_{\text{CaHY}} = 10^{3,4}$

Magnesio $Q_{\text{MgHY}} = 10^{3,9}$

Factores

$$F_{HY} = 2,47$$

$$F_{LM} \quad \text{Calcio} = 1,41 \\ \text{Magnesio} = 2,63$$

$$F_{MHY} \quad \text{Calcio} = 1,00 \\ \text{Magnesio} = 1,00$$

$$F_{HInd} \quad \text{Calcio} = 10^{2,45} \\ \text{Magnesio} = 10^{1,32}$$

Constantes condicionales

$$Q'_{CaY} = 10^{10,49} ; Q_{MgY} = 10^{8,49}$$

Cálculos dependientes del P.E

$$[Ca^{2+}]_{P.E} = 2,96 \times 10^{-7} M \therefore pCa_{P.E} = 6,52$$

$$[Mg^{2+}]_{P.E} = 2,16 \times 10^{-6} M \therefore pMg_{P.E} = 5,66$$

Cálculo de porcentaje de error

Magnesio

$$[Mg^{2+}]_{P.F} = \frac{F_{HInd}}{10^{Q_{MgInd}}} = \frac{10^{2,46}}{10^{8,1}} = 2,24 \times 10^{-6} M \therefore pMg_{P.F} = 5,65$$

$$\Delta pM = pMg_{P.F} - pMg_{P.E} = 5,65 - 5,66 = -0,01$$

$$\% \text{ Error} = \frac{10^{\Delta pM} - 10^{-\Delta pM}}{\sqrt{Q'_{MY} \cdot C_{MY}}} \cdot 100 = \frac{10^{-0,01} - 10^{0,01}}{\sqrt{10^{8,49} \cdot 2,5 \times 10^{-3} M}} \cdot 100 = +0,22$$

Calcio

$$Q'_{MgY} = \frac{[MgY^{2-}] F_{MY}}{[Mg^{2+}]_{P.F} \cdot [Y^{4-}] \cdot F_{LM} \cdot F_{HY}} \therefore [Y^{4-}] = \frac{[MgY^{2-}] F_{MY}}{[Mg^{2+}]_{P.F} \cdot Q'_{MgY} \cdot F_{LM} \cdot F_{HY}} = 3,42 \times 10^{-7} M$$

$$Q'_{CaY} = \frac{[CaY^{2-}] F_{MY}}{[Ca^{2+}]_{P.F} \cdot [Y^{4-}] \cdot F_{LM} \cdot F_{HY}} \therefore [Ca^{2+}]_{P.F} = \frac{[CaY^{2-}] F_{MY}}{Q'_{CaY} \cdot [Y^{4-}] \cdot F_{LM} \cdot F_{HY}} = 4,17 \times 10^{-8} M$$

$$pCa_{P.F} = \log\left(\frac{1}{[Ca^{2+}]_{P.F}}\right) = 7,34$$

$$\Delta pM = pCa_{P.F} - pCa_{P.E} = 7,34 - 6,52 = 0,82$$

$$\% \text{ Error} = \frac{10^{\Delta pM} - 10^{-\Delta pM}}{\sqrt{Q'_{MY} \cdot C_{MY}}} \cdot 100 = \frac{10^{0,82} - 10^{-0,82}}{\sqrt{10^{10,49} \cdot 2,5 \times 10^{-3} M}} \cdot 100 = +0,07$$

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d. Constantes de estabilidad de los quelatos protonados

Calcio $K_{CaHY} = 10^{3,1}$

Magnesio $K_{MgHY} = 10^{3,9}$

Factores

FHY = 2,47

FLM Calcio = 1,41
Magnesio = 2,63

FMHY Calcio = 1,00
Magnesio = 1,00

FHInd Calcio = $10^{2,45}$
Magnesio = $10^{1,32}$

Constantes condicionales

$K'_{CaY} = 10^{10,49}$; $K_{MgY} = 10^{8,49}$

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Calcio

$K'_{MgY} = \frac{[MgY^{2-}] F_{MHY}}{[Mg^{2+}]_{p.F} \cdot [Y^{4-}] \cdot F_{LM} \cdot F_{HY}} \therefore [Y^{4-}] = \frac{[MgY^{2-}] F_{MHY}}{[Mg^{2+}]_{p.F} \cdot K'_{MgY} \cdot F_{LM} \cdot F_{HY}} = 3,42 \times 10^{-7} M$

$K'_{CaY} = \frac{[CaY^{2-}] F_{MHY}}{[Ca^{2+}]_{p.F} \cdot [Y^{4-}] \cdot F_{LM} \cdot F_{HY}} \therefore [Ca^{2+}]_{p.F} = \frac{[CaY^{2-}] F_{MHY}}{K'_{CaY} \cdot [Y^{4-}] \cdot F_{LM} \cdot F_{HY}} = 4,17 \times 10^{-8} M$

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