### **AIR POLLUTION**

# Abating ammonia is more cost-effective than nitrogen oxides for mitigating PM<sub>2.5</sub> air pollution

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Fine particulate matter ( $PM_{2.5}$ , particles with a mass median aerodynamic diameter of less than 2.5 micrometers) in the atmosphere is associated with severe negative impacts on human health, and the gases sulfur dioxide, nitrogen oxides, and ammonia are the main  $PM_{2.5}$  precursors. However, their contribution to global health impacts has not yet been analyzed. Here, we show that nitrogen accounted for 39% of global  $PM_{2.5}$  exposure in 2013, increasing from 30% in 1990 with rising reactive nitrogen emissions and successful controls on sulfur dioxide. Nitrogen emissions to air caused an estimated 23.3 million years of life lost in 2013, corresponding to an annual welfare loss of 420 billion United States dollars for premature death. The marginal abatement cost of ammonia emission is only 10% that of nitrogen oxides emission globally, highlighting the priority for ammonia reduction.

ir pollution from PM<sub>2.5</sub> (fine particulate matter with a mass median aerodynamic diameter <2.5 µm) has been estimated to cause millions of premature deaths annually in recent years (1, 2). Therefore, mitigating PM<sub>2.5</sub> pollution is a high priority for environmental protection in many countries such as China (3), India (4), the United States (5), and the member states of the European Union (EU) (6). The most costeffective abatement measures need to be identified to balance environmental protection and economic development. This is particularly relevant for atmospheric emissions of reactive nitrogen (N<sub>r</sub>), which are generally driven by fossil fuel combustion in power plants and transport and by the production of food and energy (7-9). Previous quantifications of the health impacts of nitrogen oxides (NO<sub>x</sub>) and ammonia (NH<sub>3</sub>) emissions from PM<sub>2.5</sub> pollution have not been conducted on a global scale because of differences in atmospheric chemistry, population density,

and regions and reflective of local conditions. Here, we developed and applied a metric that we call the "N-share" of PM<sub>2.5</sub> pollution, which is the contribution of N<sub>r</sub> compounds to total PM<sub>2.5</sub> concentration determined by modeling with and without N<sub>r</sub> emission. We applied three atmospheric chemistry transport models, EMEP-WRF (12), TM5-FASST (13), and GEOS-Chem (14), which include emissions of sulfur dioxide (SO<sub>2</sub>), NO<sub>x</sub>, NH<sub>3</sub>, volatile organic compounds, and primary PM2.5 (see the materials and methods). The N-share is different from the mass fraction of N<sub>r</sub> within  $PM_{2.5}$  because  $N_r$  emissions are not only the precursors of PM2.5 but also affect the chemical reactions that lead to  $PM_{2.5}$  formation (15). By combining the calculated values of N-share with the estimated global burden of disease derived from PM<sub>2.5</sub> pollution (1), we were able to estimate welfare loss associated with N<sub>r</sub> emissions. Finally, we applied the GAINS

model (16) to estimate the implementation

costs of N<sub>r</sub> air pollution abatement (see the

materials and methods), which has the advan-

tage of being able to quantify the abatement

cost of each measure in different countries,

allowing us to compare the relative welfare

gains of reducing different forms of N<sub>r</sub> emis-

sion. NO<sub>x</sub> is an important precursor to ozone

that also has a negative impact on human

health (9), so we also considered ozone for-

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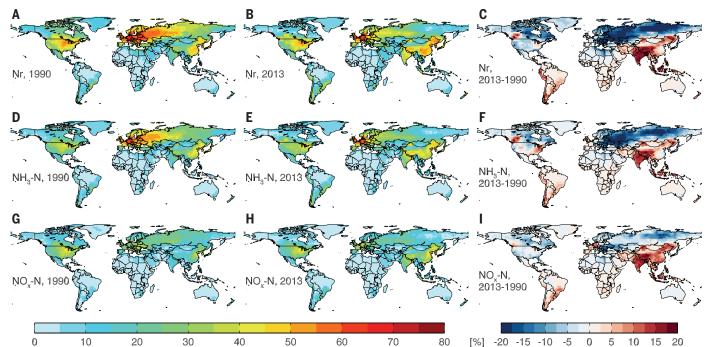
and exposure characteristics relative to the pollution consequences of  $N_r$  emissions in different countries (7, 9–11). In addition, differences in gross domestic product (GDP) and social preferences affect economic values of mortality (1). Economic growth and higher living standards increase healthy life expectancy and also increase the willingness to pay for actions that ultimately reduce the health risks of air pollution (1). Thus, a generic method is needed to ensure that the costs of mortality from  $N_r$  emissions are comparable across countries and regions and reflective of local conditions.

mation in the analysis. By contrast, we have not separately analyzed the direct health impacts of  $NO_2$ .

Although the overall N-share of PM<sub>2.5</sub> pollution generally increased from 1990 to 2013, we note substantial regional variation, with increases in Asia, South America, and South Africa and a decrease in Europe (Fig. 1 and Table 1). We found that NH<sub>3</sub> emission made a larger contribution to PM2.5 than NOx emission globally and in most countries, indicating that PM<sub>2.5</sub> is more strongly NH<sub>3</sub> limited than NO<sub>x</sub> limited (17, 18). The N-share caused by NH<sub>3</sub> emissions contributed an estimated 25% (range, 20 to 31%) to PM<sub>2.5</sub> pollution in 1990, increasing to 32% (25 to 39%) in 2013, whereas NO<sub>x</sub> emission contributed 17% (14 to 20%) in 1990, increasing to 28% (23 to 33%) in 2013. These changes agree well with the widespread increase of N<sub>r</sub> emissions and the decrease of SO<sub>2</sub> emission in many areas of the world from 1990 to 2013 (fig. S5) (19).

The N-shares of total N<sub>r</sub> emissions are much smaller than the sum of N-shares from NH<sub>3</sub> and NO<sub>x</sub> separately (Fig. 1) because of the interactions between NH3 and NOx during secondary PM<sub>2.5</sub> formation. Alkaline NH<sub>3</sub> can form aerosols with the acidic products of NO<sub>v</sub> and SO<sub>2</sub> emission, so that reduction of NH<sub>3</sub> emission also tends to reduce the contribution of  $NO_x$  and  $SO_2$  to  $PM_{2.5}$  formation (15, 18, 20), which also explains the larger N-share of  $PM_{2.5}$ pollution arising from NH3 compared with NO<sub>x</sub>. In this study, we treated the health effects of NH3-derived PM2.5 the same as we did those of NOx-derived PM2.5 because evidence of differential harm between chemical species is lacking (see the supplementary materials) (21).

For the period 1990-2013, we estimate that total years of life lost (YLL) caused by PM<sub>2.5</sub> pollution derived from N<sub>r</sub> emissions increased from 19.5 to 23.3 million globally, with the NH<sub>3</sub> and NO<sub>x</sub> contributions increasing from 16.3 to 19.3 and 11.4 to 16.2 million, respectively (table S1 and fig. S6). We used YLL to quantify premature mortality because this metric is considered robust and versatile in capturing health impacts (22). Every premature death represents an individual story, with variation between age at death and expected lifespan, which can be well captured by the YLL (23). When expressed as YLL per gigogram of N (where  $1 \text{ Gg} = 10^9 \text{ g}$ ) emission (YLL/N), higher values were found in developing or transition economies such as Asia, East Europe, and some African countries, which have high PM<sub>2.5</sub> pollution and lower GDP per capita (Fig. 2 and fig. S6). Higher  $PM_{2.5}$ pollution increases the YLL, whereas lower GDP per capita normally indicates lower access to and quality of medical care (table S2) (24), as well as greater insecurity in access to food and water that can also increase the YLL/N. By contrast, lower YLL/N is found in



**Fig. 1. N-shares of PM**<sub>2.5</sub> **pollution and their changes between 1990 and 2013.** Values represent the percentage contribution of each form of N<sub>r</sub> emission to PM<sub>2.5</sub> pollution. (**A**) Total N<sub>r</sub>-share in 1990. (**B**) Total N<sub>r</sub>-share in 2013. (**C**) Change in total N<sub>r</sub>-share between 2013 and 1990. (**D**) NH<sub>3</sub>-N share in 1990. (**E**) NH<sub>3</sub>-N share in 2013. (**F**) Change in NH<sub>3</sub>-N share between 2013 and 1990. (**G**) NO<sub>x</sub>-N share in 1990.

(H)  $NO_x$ -N share in 2013. (I) Change in  $NO_x$ -N share between 2013 and 1990. (J) Change in  $NO_x$ -N share between 2013 and 1990. Results are based on the average value of simulations from GEOS-Chem and EMEP-WRF models using 100%  $N_r$  emission reduction with uncertainty analysis from figs. S2 to S5. Base map is applied without endorsement from Natural Earth (https://www.naturalearthdata.com/).

Table 1. Estimated mortalit	y cost caused by	PM <sub>2.5</sub> air pollution	from total N <sub>r</sub> emissions
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	$N_r$ shares in PM <sub>2.5</sub> formation (%)		YLL attributable to $PM_{2.5}$ from $N_r$ emission (million years)		Total mortality cost of N <sub>r</sub> in PM <sub>2.5</sub> pollution (billion USD)		Marginal mortality cost (USD/kg N)	
	1990	2013	1990	2013	1990	2013	1990	2013
Africa	5.9	7.3	0.9	0.9	6.3	9.3	1.3	1.1
Asia	31.0	42.3	13.4	18.1	49.4	212.4	1.6	4.1
Europe	57.9	47.4	3.8	2.9	148.6	124.6	8.4	13.1
Latin America	20.9	23.5	0.6	0.7	8.6	19.0	1.5	2.1
North America	46.3	44.2	0.8	0.8	47.6	53.6	4.3	6.8
Oceania	7.8	11.2	0.01	0.01	0.5	0.6	0.6	0.5
World	30.2	38.6	19.5	23.3	261.1	419.5	3.6	4.8

America, western Europe, and Oceania (fig. S6). In these areas, populations are generally exposed to lower levels of  $PM_{2.5}$  pollution compared with Asia, and higher GDP per capita is associated with access to better medical care (table S2).

Quantifying the global welfare loss per YLL [in 2011 U.S. dollars (USD) converted at purchasing power parity rates], we found that the willingness to pay to reduce the risk of a YLL caused by  $N_{\rm r}$  emission represented a total cost of 261 (range, 156 to 398) billion USD in 1990, increasing by 60% to 420 (247 to

640) billion USD in 2013 (Table 1). Of this, by far the largest regional change was estimated for Asia (+330%, mainly linked to increased  $N_{\rm r}$  emissions), which may be contrasted with reduction in Europe (–16%, mainly linked to  $NO_x$  control). Ammonia contributes most to the estimated total cost of mortality from  $N_{\rm r}$  emissions both because it is found in large amounts (table S1) and because it is the limiting (alkaline) compound for aerosol formation (17, 18).

Globally, the average marginal cost of premature mortality caused by  $N_{\rm r}\, {\rm emission}$  in

2013 was estimated at 4.8 (range, 2.8 to 7.3) USD/kg  $N_r$  emission, compared with 3.6 (2.2 to 5.5) USD/kg  $N_r$  emission in 1990 (Table 1, Fig. 2, and fig. S7). This 31% increase can be mainly attributed to increase of  $N_r$  air pollution and higher costs per YLL caused by income growth. The marginal cost from NH<sub>3</sub> emission was ~44% higher than that of total  $N_r$  emission (table S1), affecting the overall pattern of  $N_r$  values (Fig. 2).

We found that the total social benefit of reduced mortality from the abatement of  $\mathrm{NH}_3$  emission was substantially larger than the

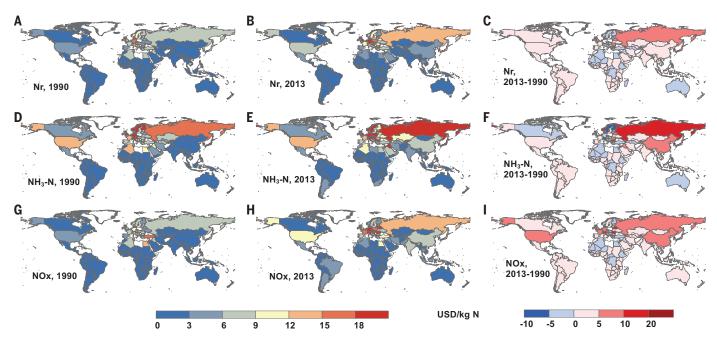
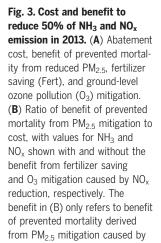
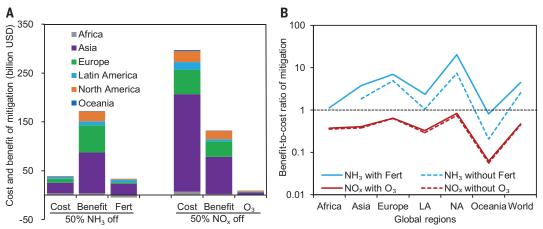


Fig. 2. Changes in mortality cost per kilogram  $N_r$  emission between 1990 and 2013 caused by  $PM_{2.5}$  pollution. Values are based on the control of different  $N_r$  components. Shown are control of total  $N_r$  emission for 1990 (**A**), 2013 (**B**), and differences between 2013 and 1990 (**C**); control of  $NH_3$  emission for 1990 (**D**), 2013 (**E**), and differences between 2013 and 1990 (**F**); and control of  $NO_x$  emission for 1990 (**G**), 2013 (**H**), and differences between 2013 and 1990 (**I**). Positive values for (C), (F), and (I) indicate an increase over time. Uncertainties for these figures can be found in fig. S7. White areas of some countries indicate a lack of data. Base map is applied without endorsement from Natural Earth (https://www.naturalearthdata.com/).





 $NH_3$  and  $NO_x$  abatement. LA, Latin America, NA, North America. For Africa, fertilizer savings are negative because of too little  $N_r$  input compared with crop need, and the best management practices would increase  $N_r$  fertilizer use (see the supplementary materials).

abatement costs. We estimated the global average cost of reducing  $\rm NH_3$  emission at 1.5 USD/kg  $\rm NH_3$ -N (weighted mean of measures) using the GAINS method (see the materials and methods), which is substantially less than the welfare benefit of the associated reduction of mortality at 6.9 (range, 3.8 to 10.9) USD/kg  $\rm NH_3$ -N globally (Table 1 and fig. S8). By contrast, we estimated the abatement cost of  $\rm NO_x$ -m which is larger than the welfare ben-

efits of reduced mortality at 7.3 (4.0 to 11.8) USD/kg NO<sub>x</sub>-N (table S1).

North America has the largest benefit-to-cost ratio for  $\mathrm{NH}_3$  mitigation, followed by Europe and Asia, suggesting reduction of  $\mathrm{NH}_3$  emission as a favorable option to increase social benefit (Fig. 3B). The benefit-to-cost ratio for  $\mathrm{NO}_x$  mitigation is also the largest in North America compared with other world regions. However, globally, this ratio is <1 even when including both  $\mathrm{PM}_{2.5}$  and ozone effects, suggest-

ing a negative net benefit to further reducing  $\mathrm{NO_x}$  emissions (Fig. 3B). The benefit-to-cost ratios to mitigate  $\mathrm{NH_3}$  and  $\mathrm{NO_x}$  emission are both <1 in Oceania (Fig. 3B). This indicates that the costs of  $\mathrm{N_r}$  mitigation are not justified by the expected regional benefits for  $\mathrm{PM}_{2.5}$ , although other considerations may still justify  $\mathrm{N_r}$  mitigation in this region [e.g., ecosystem benefits (25)]. In Africa, food security is still a challenge, and improved management practices may need to increase nitrogen inputs (whether

by fertilizer or biological nitrogen fixation) and reduce wasteful nitrogen losses to avoid soil degradation (26).

The main opportunities for NH<sub>3</sub> abatement concern agricultural sources, for which abatement measures are relatively easy and inexpensive. For instance, optimizing nitrogen fertilization will not only abate NH3 emission but will also reduce nitrogen fertilizer use, which can save the implementation cost while offering opportunities for net cost savings (27-29). For nonagricultural sources of NH<sub>3</sub>, which account for ~25% of emissions based on the Community Emissions Data System (CEDS) inventory, the reduction is more related to fossil fuel combustion, biomass burning, and waste treatment, which have received much less attention in the past (30). If global NH<sub>3</sub> emissions (from all sources, including both agricultural and nonagricultural) were to be reduced by 50%, then the total implementation cost is estimated at 38 billion USD (Fig. 3A). This is smaller than the social benefit of prevented mortality (172 billion USD) derived from mitigation of  $PM_{2.5}$  pollution for NH<sub>3</sub> emission (Fig. 3A). Additionally, this would save ~20% of global nitrogen fertilizer use with a net value of ~28 billion USD (see the materials and methods for the calculation). In practice, this means that many measures to control NH3 emission can have a zero cost implementation or represent a net economic benefit for farmers. This is in addition to the substantial societal co-benefits for natural ecosystems from reducing N<sub>r</sub> air pollution (8, 31, 32), such as improved water quality, biodiversity conservation, and reduced nitrous oxide emissions (33).

Agricultural NH3 is distributed over many individual facilities, so control requires action by farmers in many different contexts (34). In the United States, agricultural NH<sub>3</sub> emission is not well controlled, and it is a cause of damage to both the environment and human health (18). For larger farms with access to advanced technologies, it is technically feasible to reduce NH<sub>3</sub> emission from agriculture without risk of production loss through measures such as adoption of enhanced-efficiency fertilizers and improved manure management practices (35, 36). With the revised EU National Emission Ceilings directive (adopted in 2016), reduction of NH<sub>3</sub> emission will be considered across sectors for the target year 2030 (37). Policies in the Netherlands and Denmark have required NH3 abatement, leading to reported emission reductions of 35 to 66% between 1990 and 2011 (38). China started to address reduction of NH<sub>3</sub> emission from agricultural sources during its 13th Five Year Plan (2016-2020) (39), and it has been estimated that agricultural  $\mathrm{NH}_3$  emission could be reduced by one-third and at a low cost through appropriate policies, such as subsidies for enhanced-efficiency fertilizers and fertilizer application machinery (40, 41). Taking reduction of  $\mathrm{NH}_3$  emission into consideration for  $\mathrm{PM}_{2.5}$  pollution control is therefore critical, presenting the opportunity for future legislation to mitigate  $\mathrm{NH}_3$  emissions at national to global scales (34).

The estimated implementation cost (297 billion USD) for reducing NO<sub>x</sub> emissions is larger than the benefit of reduced mortality (132 billion USD) derived from reduced PM2.5 pollution (fig. S8). However, reduction of NO<sub>x</sub> emission can also have other benefits, such as alleviation of ground-level ozone (9 billion USD, estimated based on the similar method with PM<sub>2.5</sub>; see the supplementary materials). The modest 7% additional benefit shows how the PM<sub>2.5</sub> costs dominate welfare loss derived from NO<sub>x</sub> emission (Fig. 3B), whereas reducing both NOx and NH3 emissions will also benefit terrestrial and aquatic ecosystems (9). NO<sub>v</sub> emissions have already been reduced in high-income regions such as North America and Europe between 1990 and 2013 (fig. S5), which means that the marginal implementation cost to reduce NO<sub>x</sub> emissions further is much higher in these countries because the most cost-effective measures are in place already. For other countries where NO<sub>x</sub> emissions are still increasing, implementation costs are smaller than those for Europe and North America, but even these costs are much higher than those of NH<sub>3</sub> mitigation (fig. S8 and Table 1). Innovation to recapture NO<sub>x</sub> as value-added N<sub>r</sub> products rather than the present focus on wasteful destruction to form di-nitrogen (N2) may have considerable future potential to reduce costs (33) but is still far from commercial availability. Considering the overall costs and benefits, our analysis highlights the priority for air pollution policies to give increased attention to controlling NH<sub>3</sub> emissions, complementing successful policies on NOx and SO2 (Fig. 3B) as part of implementing the ambition of the Colombo Declaration to "halve nitrogen waste" from all sources globally.

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29/Z/ST10/02986 (NCN, Poland), 71961137011 (NSFC, China), and 870234 (FFG, Austria). D.C. acknowledges support received from the Australian Research Council through the Industrial Transformation Research Hub (IH200100023). **Author contributions:** B.G. and H.JM.V.G. designed the study as part of the wider INMS framework designed by M.A.S. R.V.D., M.V., and L.Z. applied the global atmospheric models. B.G. conducted the health impact calculation and analysis with help from R.V.D. L.Z. conducted the N-share analysis with help from R.V.D. and M.V. S.Z. and X.Z. conducted the abatement cost analysis using the GAINS

model with help from W.W. S.W. and C.R. conducted the spatial and statistical analyses. B.G. analyzed all the data, interpreted the results, and wrote the first draft of the paper. All authors contributed to the discussion and revision of the paper, which was finalized under the lead of B.G. and M.A.S. **Competing interests:** The authors declare no competing interests. **Data and materials availability:** Data supporting the main findings can be found in the supplementary materials. Further data that support the findings of this study are collated from online open databases or literature sources as cited in the reference list.

#### SUPPLEMENTARY MATERIALS

science.org/doi/10.1126/science.abf8623 Materials and Methods Supplementary Text Figs. S1 to S9 Tables S1 to S6 References (42–64)

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# Abating ammonia is more cost-effective than nitrogen oxides for mitigating PM air pollution

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#### Little things matter

Particulate air pollution 2.5 micrometers or smaller in size (PM2.5) is a major cause of human mortality, and controlling its production is a health policy priority. Nitrogen oxides are an important precursor of PM2.5 and have been a focus of pollution control programs. However, Gu *et al.* now show that abating ammonia emissions is also an important component of PM2.5 reduction, and the societal benefits of abatement greatly outweigh the costs (see the Perspective by Erisman). Reducing ammonia emissions thus would be a cost-effective complement to nitrogen oxides and sulfur dioxide controls. —HJS

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