The difference between **SCIENCE** and **ART** is **TESTING**

why is physics related to all other sciences?

A/ Because all science originated from PHILOSOPHY

Physics is a branch of sciencece that aims to understand natural phenomenons and tries to get accurate and precise measurements and explains the properties of matter like mass, density, and temperature

Physics in itself, branches a lot, for example

- **Thermodynamics:** it deals with heat
- **Optics:** it deals with lights
- **Electromagnetism**: it deals with electricity and magnets
- **Relativity:** it deals with moving atoms and their collisions
- **Mechanics:** it deals with motion, friction

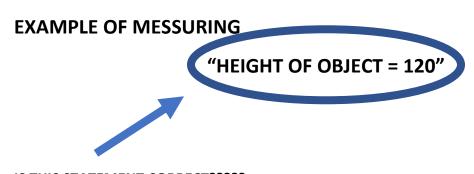
Testing in PHYSICS is mostly EXPERIMENTS & MEASUREMENTS

How?

You stumble across a behavior in nature



- You start asking questions
- You come up with a hypotheses
- You test the hypotheses using other measurements
- You conclude a law in nature



IS THIS STATEMENT CORRECT?????

A/ It's not, but why?

Because Measurement is

The process of comparing an UNKOWN QUANTITY with another KNOWN QUANTITY to know how many times does the first contain the second

THE ELEMENTS OF MEASUREMENT ARE

- 1. The value of the quantity that is measured
 - 2. The unit of the measurement
- 3. The tools that are used for measurement
 - 4. The measurement errors

UNITS AND DIMENSIONS ARE LINKED BUT DIFFERENT



GRAM

It has 2 meanings

- Dimension -> it means the type of the quantity, which means that we are measuring weight
- Unit -> it means how much of the quantity, which means that are measuring one gram worth of weight

UNITS IN DEPTH -> it's a known value of a quantity that is used to measure another quantity from the same **DIMENSION**

WEIGHT = 6 KILO GRAM

Sign magnitude prefix unit

THE UNIT SYSTEM

Is a system made to set the default unit for each quantity

We will use Length, mass, and time as examples

| Quantity | M.K.S | C.G.S | F.P.S |
|----------|---------------|--------------------|------------|
| Length | Meter (m) | Centimeter (cm) | Foot (ft) |
| Mass | Kilogram (kg) | Gram (g) | Pound (lb) |
| Time | Second (s) | second (s) | second (s) |



In 1960, the International System of Units (SI) was created It was based on the M.K.S (metric system)

Fundamental Units

Derived Units

| mass | Kilogram (kg) |
|---------------------|--------------------------------------|
| length | Meter (m) |
| time | Second (s) |
| temperature | Kelvin (k) |
| Electric current | Ampere (a) |
| Amount of substance | Mole (mol) |
| Luminous intensity | Candela (cd) |
| Plane angle | Radian (rad) |
| Solid angle | Steradian (sterad) |
| area | Square meter (m²) |
| volume | Cubic meter (m³) |
| speed | Meter per second (m/s) |
| acceleration | Meter per second square (m/s²) |
| Weight, force | Newton (N) |
| pressure | Pascal (Pa) |
| Energy, work | Joule (J) |



A standard unit system was helpful because It allowed scientists from all across the world to work together without having to translate each system

The standard length (meter)

It was the length between two points on a rod made from a platinum/iridium alloy and saved at 0°, it was redesigned as the distance light travels in a vacuum in the span of 1/299792458 seconds

The standard mass (kilogram)

It was the mass of a cylinder made from a platinum/iridium alloy and saved at 0°

The standard meter

It is now defined as the 9192631700 times the period of oscillation of radiation from the cesium atom

Poyle's law

Volume and pressure are in a negative correlation

There are **2 TYPES** of quantities

Fundamental Quantities (Basic)

 They are the most basic quantities that can't be defined by other quantities



 Like Mass, Length and Time, you can't define them using speed or something, it's a central quantity that works with other quantities to define stuff like force, or density

Derivative Quantities

- They are quantities that can be defined by other quantities (fundamental ones)
- Like **Speed**, or **force**, they are just fundamental quantities like mass, length and time working together

Prefixes

They are used as multipliers for units

Like 1 kilometer = 1000 meters

And 1 millimeter = .001 meters



| Prefix | Symbol | Numerical Multiplier | Exponential Multiplier |
|-----------|--------|---|---------------------------|
| yotta | Υ | 1,000,000,000,000,000,000,000,000 | 10 ²⁴ |
| zetta | Z | 1,000,000,000,000,000,000,000 | 10 ²¹ |
| exa | E | 1,000,000,000,000,000,000 | 10 ¹⁸ |
| peta | Р | 1,000,000,000,000,000 | 10 ¹⁵ |
| tera | Т | 1,000,000,000,000 | 10 ¹² |
| giga | G | 1,000,000,000 | 10 ⁹ |
| mega | М | 1,000,000 | 10 ⁶ |
| kilo | k | 1,000 | 10 ³ |
| hecto | h | 100 | 10 ² |
| deca | da | 10 | 10¹ |
| No prefix | | 1 | 10 ⁰ |
| deci | d | 0.1 | 10 ⁻¹ |
| Centi | С | 0.01 | 10 ⁻² |
| Milli | m | 0.001 | 10 ⁻³ |
| Micro | μ | 0.000001 | 10 ⁻⁶ |
| Nano | n | 0.00000001 | 10 ⁻⁹ |
| Pico | р | 0.00000000001 | 10 ⁻¹² |
| Femto | f | 0.00000000000001 | 10 ⁻¹⁵ |
| Atto | а | 0.00000000000000001 | 10 ⁻¹⁸ |
| Zepto | Z | 0.000000000000000000001 | 10 ⁻²¹ |
| yocto | У | 0.0000000000000000000000000000000000000 | 10 ⁻²⁴ |

NOTES!

When we go to square or cubic lengths

We multiply the multiplier's power by 2 (for square) and 3 (for cubic)

So 1 cubic meter = 10⁶ cubic centimeters instead of 10²

And 1 square meter = 1^6 square millimeters instead of 1^3

As for liters

1 cubic meter = 10^3 liters



HOW TO CONVERT BETWEEN QUANTITIES

(1) Convert between prefixes

To convert from a prefix to a prefix we can use the exponential multiplier and put it in this formula

 $\frac{quantity\ number\ x\ current\ multiplier}{target\ multiplier}$

So to convert from let's say 10 micrometers to nanometers we do this

$$\frac{10 \times 10^{4} - 6}{10^{4} - 9} = 10000 \text{ nm}$$

(2) Convert between Units

Length

- 1 mile (mi) = 1609 m = 5280 ft
- 1 yard = 3 ft
- 1 ft = 30.48 cm = 12 in
- 1 in = 2.54 cm

Mass

- 1 Pound (lb) = 0.465 kg
- 1 ounces (oz) = 28.35 g
- 1 day = 86400s

Time

• 1 h = 3600 s



Temperature Conversions

| From | То | Formula |
|-----------------|--------------------|-----------------------------|
| Celsius (°C) | Fahrenheit (°F) | $F = (C \times 9/5) + 32$ |
| Fahrenheit (°F) | Celsius (°C) | $C = (F - 32) \times 5/9$ |
| Kelvin (K) | Celsius (°C) | C = K - 273.15 |
| Celsius (°C) | Kelvin (K) | K = C + 273.15 |
| Fahrenheit (°F) | Kelvin (K) | K = (F - 32) × 5/9 + 273.15 |
| Kelvin (K) | Fahrenheit (°F) | F = (K - 273.15) × 9/5 + 32 |

MEASUREMENT TOOLS

Length

- Vernier caliper
- Micrometer
- Ruler

Weight

• Sensitive balance



Time

Stop watch

Electric Current

- Ammeter
- Multimeter

Potential Difference

- Voltmeter
- Multimeter

Resistance

- Ohmmeter
- Multimeter

How to measure using micrometer



How to measure using vernier caliper



Measurement Errors

How do they happen?

- Wrong device -> using a ruler to measure a grain of sand
- Faulty Device -> weak magnet in an ammeter
- Environmental error -> air flow changing the reading of the sensitive balance
- Human error ->
 - o Reading error

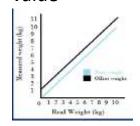


- o Angle error
- Mathematical error

There are 2 types of them

Systematic Errors

- They are a permanent deflection in the same direction from a true value
- They are usually caused by measuring tools that are incorrectly calibrated or used incorrectly
- They can be corrected, why?
 - Because they are just offsetted from the true value
 - It's like when you try to weigh multiple things but there is an additional object on the balance, so you just subtract that weight from everything and now you have the true value

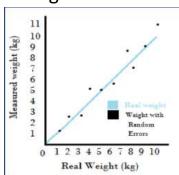


Random Errors (noisy errors)

- They are a short-term scattering of values around the true value
- They are caused randomally by multiple forces (usually environmental)
- They can't be corrected, why?
 - Because they are unpredictable



- It's like when you try to weigh something but every time you get a new higher or lower result
- You can't get the real value from it you just get an average

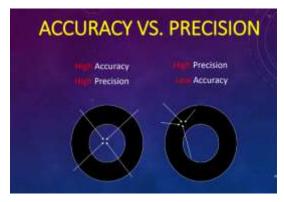


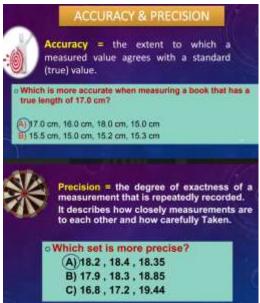
Of course, to do your testing/experimenting you will need to use tools, or in our case, they're called **Instruments**, when we do our measuring We have two types of results

- **Accurate results:** when we have the measured value be close to the true value
- Precise results: when multiple measurements give nearly identical values

aka: Omar Tar

The goal of science is to obtain measurements that are both precise and accurate, that will depend on the type of measurement





TYPES OF MEASUREMENTS

| direct | indirect |
|--------|----------|
|--------|----------|



| it happens with one procedure or one tool | it happens with multiple procedures or tools |
|--|--|
| it can only get 1 error | it gets many errors and they stack up |
| we do not use mathematical equations | we use mathematical equations |
| using a graduated cylinder to measure the size of a cube | measuring the x, y, z lengths and doing the xyz equation to get the size |

HOW TO CALCULATE ERRORS!?

Let's go over some terms first

Absolute Error (ΔX)

it is the difference between the real value (X_{O}) and the measured value (x)

$$\Delta X = |X_O - X|$$

Relative Error (r)



it is the ratio between the absolute value (ΔX) and the real value (X_0)

$$r = \frac{\Delta X}{X_0} * 100\%$$

you'll see that the relative error indicates better accuracy than the absolute because it tells you the exact percentage so when you get smaller, you're getting more accurate

when you type a measurement reading you do it like this

Quantity = (true value ± absolute error)

 $Q = (x \pm absolute error)$

 $Q = (x_0 \pm \Delta x)$

EXAMPLE PROBLEMS

A student measured the length of pen and they measured 9.9 cm

While the true length was 10 cm, measure the absolute error, relative error and write the measurement reading?

$$\Delta x = |x_0 - x| = |10 - 9.9| = .1 \text{ cm}$$

$$R = (\Delta x / x_o) * 100\% = (.1 / 10) * 100 = 1\%$$

When you add/subtract errors



The new $\Delta x = \Delta x_1 \pm \Delta x_2$

The new
$$r = \frac{\text{new } \Delta x}{\text{Xo1} \pm \text{Xo2}}$$

When you multiply/divide errors

The new
$$r = r_1 + r_2 = \frac{\Delta X1}{X01} + \frac{\Delta X2}{X02}$$

The new Δx = the new $X_0 * R$

Calculate the relative error and absolute error of the area of a rectangle, its length = (6 + 0.1) m and its width (5 + 0.2) m?

$$R_1 = .1 / 6 = .0166666 \sim .017$$

$$R_2 = .2 / 5 = .04$$

$$R = .017 + .04 = .057$$

$$\Delta X = (6*5) * .057 = 1.71 M^2$$

$$X_0 = 30 \text{ M}^2$$

$$Q = (30 \pm 1.71) M^2$$

If L_1 = (5.2 ± .1) cm and L_2 = (5.8 ± .2) cm and L = L_1 + L_2



Calculate L

$$\Delta x = .1 + .2 = .3 \text{ cm}$$

$$R = \Delta x / (x_{o1} + x_{o2}) = .3 / 11 = .027 = 2.7\%$$

$$L = (11 \pm 0.3)$$

We simplify **Fundamental Quantities** by using **Dimensions**"the power to which the fundamental quantities are raised"

So instead of Mass we have (M), Instead of Time we have (T), etc etc

And we use these symbols to simplify physical laws and derivative quantities

So instead of saying [velocity = distance / time]

We say [
$$V = L.T^{-1}$$
]

Calculations using Symbols

- M.L + M.L = M.L
- M.L + M.T = undefined
- M.L * M.T = M².L.T
- $L.T^2 / T = L.T$
- L/L=1
- $M^0 = 1$



| Dimensiona | l Formula of some F | Physical Quantities |
|-------------------|--------------------------------|---|
| Physical quantity | Relation with other quantities | Dimensional formula |
| Areas | Length x breadth | $L \times L = L^2 = [M^0 L^2 T^0]$ |
| Volume | Length x breadth x height | L x L x L= L ³ = [M ⁰ L ³ T ⁰] |
| Density | Mass volume | $\frac{M}{L^3} = \left[ML^{-3}T^0\right]$ |
| Speed or velocity | distance time | $\frac{L}{T} = [M^0L T^{-1}]$ |
| Acceleration | velocity time | $\frac{LT^{-1}}{T} = [M^0L T^{-2}]$ |

Mass x velocity

Force area

Mass x acceleration

Momentum

Force

Pressure

 $\mathsf{M}\times\mathsf{L}\mathsf{T}^{-1}=\left[\mathsf{M}\;\mathsf{L}\;\mathsf{T}^{-1}\right]$

 $M \times LT^{-2} = [M LT^{-2}]$

 $\frac{MLT^{-2}}{L^2} = [ML^{-1}T^{-2}]$

| Work | Force x distance | $M LT^{-2} \times L = [ML^2T^{-2}]$ |
|---------------------------|--|--|
| Energy | Work | M L ² T ⁻² |
| (mechanical, heat, | | |
| light etc) | | |
| Power | work time | $\frac{ML^2T^{-2}}{T} = ML^2T^{-3}$ |
| Stress | Force area | $\frac{MLT^{-2}}{L^2} = \left[ML^{-1}T^{-2}\right]$ |
| Strain | Change in dimension Original dimension | Dimensionless |
| Coefficient of elasticity | stress strain | $\frac{ML^{-1}T^{-2}}{1} = \left[ML^{-1}T^{-2}\right]$ |
| Moment of inertia | Mass x (radius of gyration) ² | $M \times L^2 = [ML^2T^0]$ |
| Angular | Moment of inertia x | $ML^2 \times T^{-1} = [ML^2T^{-1}]$ |
| momentum | angular velocity | |
| Torque or couple | Force x perpendicular distance | $MLT^{-2} \times L = [ML^2T^{-2}]$ |



Constants do not have dimensions like 2, $\frac{1}{2}$, and π Trigonometric functions like sin, cosine, tan do not have

We can use dimensional equations to verify an equation but that does not mean that it's always correct

EXAMPLE 1

dimensions

the equation for sphere area is $\pi r^2\, so$ its dimensions are L^2 the area of a square is L^2

dimensions say those 2 are EQUAL but they are NOT that's why we can only use them to verify

EXAMPLE 2

Verify the law [Kinetic energy = $.5 \text{ mass x velocity}^2$]

- (1) Energy = work = Force x distance = $M.L.T^{-2} x L = M.L^2.T^{-2}$
- (2) Kinetic Energy = .5 mass x velocity² = M x $(L.T^{-1})^2$ = M x $L^2.T^{-2}$ = M.L².T⁻²

so the law is correct, we have the same dimensions on both sides

because kinetic energy is taken from energy so it should have the same law



NANOSCALE

It's where objects become so small that we can't even see it with a microscope

The reason is that it's dimensions are smaller than even some electromagnetic waves

Its dimensions will range from 1 nano meter to 100 nanometer

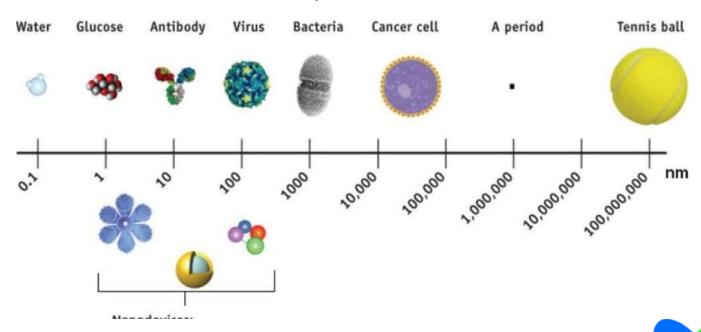
When an object enters the nano scale, it gets very new properties on the physical, chemical, and mechanical scale

These properties change within the nanoscale spectrum

For example

- Nano-scale copper will be a lot tougher
- Nanoscale gold will become a liquid and will be different colors from red to orange to green to blue
 - Nanoscale salt will react a lot quicker

All that is because when you chop an object to the nano scale, it's surface area will increase exponentially and the volume will stay the same leading to new effects like, very fast reaction speed





This learning out come can help a lot with

CAPSTONE

- We can use measurements to make the capstone's prototype, then we can make it more accurate and precise to achieve our required results
- Exponential population growth
 - We can use measurements and dimensional analysis to create a data set for the pobulation growth and detect similar trends, and we can use measurement errrors to make our measurements more precise
- Reducing Urban Congestion
 - We can use measurements to see the traffic flow and congestion levels and with physical principles we can create a model to fix the problem

