

This chart provides an overview of the program interface displayed by the command line.

| Interface | Description | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
|---|---|------------------------|-------|-----------|--------|------|------------|--------|------|---------|--------|------|----------|--------|------|----------|----------|-------|-----------|--------|------|------------|--------|------|---------|--------|------|----------|--------|------|----------|-----------------------------|------------------------|-----------|--------|------|------------|--------|------|---------|--------|------|----------|--------|------|---|
| <p>Step 1: Input Feed Chemical Composition</p> <p>Enter Chemical Component 1: methanol</p> <p>Would You Like To Add Another Chemical Component? yes</p> <p>Enter Chemical Component 2: ethanol</p> <p>Would You Like To Add Another Chemical Component? yes</p> <p>Enter Chemical Component 3: 1-propanol</p> <p>Would You Like To Add Another Chemical Component? yes</p> <p>Enter Chemical Component 4: 1-butanol</p> <p>Would You Like To Add Another Chemical Component? no</p> | <p>Enter the names of each component in the mixture. The name must be the component’s IUPAC name. Misspelled or incorrect names will result in the program being unable to find antoine coefficients for that component.</p> | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| <p>How Many Moles Are Fed To The Column: 100</p> <p>What Fraction Of The Feed Is 1-butanol : .35</p> <p>What Fraction Of The Feed Is 1-propanol : .15</p> <p>What Fraction Of The Feed Is ethanol : .20</p> <p>What Fraction Of The Feed Is methanol : .30</p> <p>Is this a flash system: yes</p> <p>What Is The Total Pressure Of The System: 760</p> <p>Is This Pressure In mmHg? yes</p> <p>What Fraction Of The Feed Vaporized : .4</p> <p>Would You Like To Analysis This System Using Another Model? yes</p> <p>Would You Like To Remodel The System As A Fractional Distillation System? yes</p> <p>What Is The Light Key? methanol</p> <p>What Fraction Of methanol Is Recovered In The Distillate? .95</p> <p>What Is The Heavy Key? ethanol</p> <p>What Fraction Of ethanol Remains In The Bottoms? .95</p> <p>Would You Like To Use The Same Total System Pressure Used In The Flash Calculation? yes</p> <p>What Is The Value Of q? 1.10</p> <p>You Can Input A Reflux Ratio OR Use 1.4X The Minimum Reflux Ratio</p> <p>Would You Like To Input A Reflux Ratio? yes</p> <p>What Is Your Reflux Ratio? 3.0</p> <p>Would You Like To Analysis This System Using Another Model? no</p> | <p>The program will ask for:</p> <ol style="list-style-type: none">1) Each component’s molar fraction of the mixture2) The operating pressure and percentage of feed vaporized <p><i>If this is a flash system a Rachford-Rice Model will be run. Additional models can also be chosen.</i></p> <ol style="list-style-type: none">3) Desired product specification of the distillation4) A reflux ratio. <i>If no reflux ratio is provided, 1.4X the minimum reflux ratio will be used (distillation heuristic)</i> | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| <p>FLASH DISTILLATION RESULTS USING EQUILIBRIUM AND RACHFORD-RICE</p> <p>Equilibrium Results</p> <p>Temperature Results</p> <p>Dew Temperature : 99.65 Celsius</p> <p>Boiling Temperature: 82.79 Celsius</p> <table><tr><td>Chemical</td><td>: Liquid</td><td>Vapor</td></tr><tr><td>1-butanol</td><td>: 0.69</td><td>0.09</td></tr><tr><td>1-propanol</td><td>: 0.14</td><td>0.08</td></tr><tr><td>ethanol</td><td>: 0.09</td><td>0.24</td></tr><tr><td>methanol</td><td>: 0.09</td><td>0.59</td></tr></table> <p>Rachford-Rice Results</p> <p>Feed Fraction Vaporized : 0.40</p> <p>Temperature Results</p> <p>Vaporization Temperature: 89.37 Celsius</p> <table><tr><td>Chemical</td><td>: Liquid</td><td>Vapor</td></tr><tr><td>1-butanol</td><td>: 0.48</td><td>0.16</td></tr><tr><td>1-propanol</td><td>: 0.17</td><td>0.12</td></tr><tr><td>ethanol</td><td>: 0.17</td><td>0.25</td></tr><tr><td>methanol</td><td>: 0.19</td><td>0.47</td></tr></table> <p>MULTICOMPONENT FRACTIONAL DISTILLATION RESULTS USING F.U.G.K</p> <p>Temperature At Top Of Column : 65.17 Celsius</p> <p>Temperature At Bottom Of Column: 95.76 Celsius</p> <p>Light Key : methanol</p> <p>Heavy Key : ethanol</p> <p>Total Feed Rate: 100.00 (mol/hr)</p> <table><tr><td>Chemical</td><td>: Distillate 29.50 (mol/hr)</td><td>Bottoms 70.50 (mol/hr)</td></tr><tr><td>1-butanol</td><td>: 0.00</td><td>0.50</td></tr><tr><td>1-propanol</td><td>: 0.00</td><td>0.21</td></tr><tr><td>ethanol</td><td>: 0.03</td><td>0.27</td></tr><tr><td>methanol</td><td>: 0.97</td><td>0.02</td></tr></table> <p>The Minimum Reflux Ratio At Total Reflux : 2.27</p> <p>Minimum Number Of Stages At Total Reflux : 11.41</p> <p>Number Of Stages For A Reflux Ratio Of 3.00 : 17.39</p> <p>Feed Tray Location For A Reflux Ratio Of 3.00: 8.28</p> <p>Program ended with exit code: 0</p> | Chemical | : Liquid | Vapor | 1-butanol | : 0.69 | 0.09 | 1-propanol | : 0.14 | 0.08 | ethanol | : 0.09 | 0.24 | methanol | : 0.09 | 0.59 | Chemical | : Liquid | Vapor | 1-butanol | : 0.48 | 0.16 | 1-propanol | : 0.17 | 0.12 | ethanol | : 0.17 | 0.25 | methanol | : 0.19 | 0.47 | Chemical | : Distillate 29.50 (mol/hr) | Bottoms 70.50 (mol/hr) | 1-butanol | : 0.00 | 0.50 | 1-propanol | : 0.00 | 0.21 | ethanol | : 0.03 | 0.27 | methanol | : 0.97 | 0.02 | <p>The results of each model run by the program will print out.</p> |
| Chemical | : Liquid | Vapor | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| 1-butanol | : 0.69 | 0.09 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| 1-propanol | : 0.14 | 0.08 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| ethanol | : 0.09 | 0.24 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| methanol | : 0.09 | 0.59 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Chemical | : Liquid | Vapor | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| 1-butanol | : 0.48 | 0.16 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| 1-propanol | : 0.17 | 0.12 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| ethanol | : 0.17 | 0.25 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| methanol | : 0.19 | 0.47 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Chemical | : Distillate 29.50 (mol/hr) | Bottoms 70.50 (mol/hr) | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| 1-butanol | : 0.00 | 0.50 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| 1-propanol | : 0.00 | 0.21 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| ethanol | : 0.03 | 0.27 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| methanol | : 0.97 | 0.02 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |