

The Restricted Hartree Fock (RHF) Algorithm is a way of approaching the lowest energy of an electronic system. The restriction comes from the property that only even numbers of electrons (N_{elec}) can be treated. The RHF equations solve the eigenvalue problem of

$$FC = C\epsilon. \quad (1)$$

In (1) F is the Fock matrix, C is the matrix of eigenvectors and ϵ is the diagonal matrix of eigenvalues. The Fock matrix is given by

$$F_{\mu\nu} = H_{\mu\nu} + P_{\lambda\sigma} * [2(\mu\nu|\sigma\lambda) - (\mu\lambda|\sigma\nu)]. \quad (2)$$

H is the one-electron matrix, P the density matrix and $(\mu\nu|\sigma\lambda)$ is the two-electron matrix given in chemists notation. In (2) and below we use the Einstein convention that is summing over every index which occurs more than once. The one-electron-matrix and the two-electron-matrix are constant inputs for the RHF algorithm. Therefore we can treat them as given. The density matrix is given by

$$P_{\mu\nu} = \sum_a^{N_{occ}} C_{\mu a} C_{\nu a}^*. \quad (3)$$

N_{occ} stands for the number of occupations and is given as $\frac{N_{elec}}{2}$. We obtain C from decomposing F into its eigendecomposition.

$$C = \text{matrix of eigenvectors}(F) \quad (4)$$

The eigenvectors are sorted in ascending order based on the values of the eigenvalues. The density is built from the N_{occ} eigenvectors corresponding to the lowest eigenvalues. The resulting electronic energy (E) is given by

$$E = P_{\nu\mu}(H_{\mu\nu} + F_{\mu\nu}). \quad (5)$$

To obtain the total energy, the nuclear energy is added to the electronic energy. The RHF-algorithm is solved by iteratively computing (3), (2) and (4) while checking for convergence with (5).