### **Electricity and Magnetism**

Two fundamental properties of matter determine interactions between objects in classical physics **mass** and **electric charge**.

1. **Gravitational force** attracts objects together and is proportional to masses of the objects.

 $\tilde{\mathbf{F}}_{\mathbf{g}} = G \frac{Mm}{r_{12}^2} \hat{r}$ 

2. **Electromagnetic interaction** causes like charges to repel while opposite charges attract to each other, keeps electrons bound to nuclei, and causes chemical binding to keep atoms together in molecules or solid materials.

# **Electric Charge**

- Electric charge can't be created from nothing and can't be destroyed
- The process of removing an electron from the electron cloud of an atom, or adding an electron to it, is called **ionization**
- The electrons in an **insulator** (glass, ceramic, plastic) are all tightly bound to the positive nuclei and do not move around so they are immobile
- The outer electrons in a **conductor** (copper, silver, aluminum) are only weakly bound to the nuclei so they can easily become detached

Type	Charge	Mass	Mass in SI
electron	-e	1	$9.11 * 10^{-31} \text{ kg}$
proton	+e	1836	$1.673 * 10^{-27} \text{ kg}$
neutron	0	1839	$1.675 * 10^{-27} \text{ kg}$
positron	+e	1	$9.11 * 10^{-31} \text{ kg}$

One elementary charge, e is equal to  $1.602 * 10^{-19} C$ 

### **Electrostatics**

If charges move slow with small acceleration the effects of magnetic field and electromagnetic waves can be neglected

#### Coulomb's Law

$$m{F_{12}} = krac{q_1q_2}{r_{12}^2} m{\hat{r}_{12}}$$

where:

 $k = (4\pi\varepsilon_0)^{-1} = 9.0 * 10^9 Nm^2/C^2$ 

 $\varepsilon_0 = \text{permittivity of free space} = 8.86 * 10^{-12} \, C^2 / Nm^2$ 

## Electric Field

Electric field is the force exerted by source charges on a positive unit test charge

$$\boldsymbol{E(r)} = k \frac{Q}{r^2} \boldsymbol{\hat{r}}$$

If the source charge Q is positive, the direction of electric field vector is away from Q, otherwise the electric field vector is directed toward Q. Direction of E is the same as direction of force acting on positive charge.

If we know strength of the electric field, E, and the test charge value  $q_t$ , we can obtain the force acting on  $q_t$  with the formula,  $\mathbf{F} = q_t \mathbf{E}(\mathbf{r})$