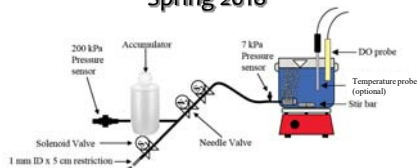


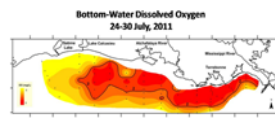
CEE 4530: Laboratory Research in Environmental Engineering

Spring 2018



Monroe Weber-Shirk
265 Hollister Hall
mw24@cornell.edu

CEE 4530: Laboratory Research in Environmental Engineering

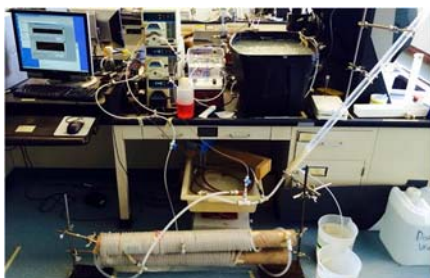


hypoxia



environmental
contaminants

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Agenda for our First Meeting

- Course overview
- Introductions
- Working in teams
- Chemical safety

Course Goals To Help Students Develop:

- An intuition for concepts seen in other classes
- Laboratory skills
- Research skills
 - Data analysis
 - Model development
- Engineering skills
 - Designing systems to meet an objective
 - Troubleshooting and problem solving
- Writing skills (Technical Writing Course)
- Teamwork skills

Course Info

- [Web site](#)
- Schedule
- Grades
- Text: "**Laboratory Research in Environmental Engineering: Laboratory Manual**" 2018.

Course Elements

- Laboratories – from canned to open-ended
- Lectures/discussions
- Prelab questions – done individually
- Laboratory reports – done in teams
- Project – done in teams

Projects

- A chance to develop your engineering skills
- Learn about process control, automation and application of lots of things that you've learned thus far!
- Nutrient Removal Project
- Research Project

Introductions...

- My name
- About me!
 - a hobby, or
 - what I did over break, or
 - what I like to do when I'm not doing coursework

Team Preferences?

- Teams of 3
- Fill out form at website by Monday! (link on syllabus page)
- We will assign you to a team in time for the Acid Rain laboratory

Your To Do list for 4530

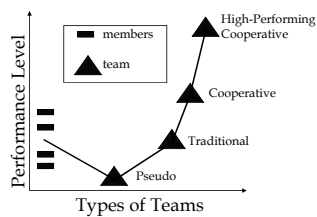
- Fill out team formation survey by Monday
- Continue working on Atom and Python Tutorial (due noon on Wednesday)
- Do prelab assignment for Fundamentals Lab **AND** for laboratory Safety before lab on Wednesday



Team Management Reflections

- Think about your most successful/effective team project experience
 - What were the characteristics of the team?
 - What were the conditions?
- Have you had experiences with teams that were ineffective or frustrating?
 - What made the team ineffective?
 - What could you have done to improve the team?

Types of Teams



Traditional Classroom Learning Teams

Team members:

- Believe they will be evaluated and rewarded as individuals, not as a team
- Interact primarily to clarify how assignments are to be done
- Have no motivation to teach what they know to their teammates
- May seek a free ride on the efforts of teammates
- Or may feel exploited and do less

Cooperative Learning Teams

- Have a goal of maximizing all members' learning
- Hold themselves accountable for doing high quality work
- Work face-to-face for efficient communication to produce joint work-products
- Help and encourage each other (academic and personal support)
- Share responsibility for providing leadership
- Analyze how effectively they are achieving their goals
- Emphasize continual improvement of the teamwork process

High-performance Cooperative Learning Team

- Meets all the expectations of the cooperative learning team
- Outperforms all reasonable expectations, given its membership
- High level of commitment
- Mutual concern for each other's personal growth
- Achieve high-performance and also have lots of fun

What Makes Cooperative Learning Work?

- Positive interdependence
- Face-to-face interaction
- Individual accountability/personal responsibility
- Teamwork skills
- Group processing
 - How are we doing as a team?
 - What could we do better?
 - What challenges are we facing?

Danger signs

- High achiever paired with procrastinator
- Procrastinator becomes less and less involved
- High achiever becomes frustrated
- Procrastinator learns little and gets a low final grade (team participation rating is a multiplier)

Antidote to Disengagement

- Engagement!
- Insist on being a full member of the team
- Real time collaboration

Course Grading

- Not on a curve
- You are not competing with your classmates
- If you all do well, you will all get good grades
- Cooperation is rewarded, not penalized

A Few Safety Rules

- Think through each step before beginning an experiment
- Identify possible hazards and take appropriate measures to contain or minimize hazards (read labels)
- If you don't know what is going to happen and think there may be a potential for danger
 - Stop!
 - Get help before proceeding or modify procedure to minimize risk.
- When using hazardous chemicals use secondary containment and appropriate personal protection

Unattended Experiment Guidelines

- Prepare for power failure, _____ failure, or a leak
 - _____ containment of hazardous chemicals
 - Prepare for flooded and _____ conditions
 - Do not let equipment such as power stirrers, hot plates, heating mantles, and water condensers run overnight without _____ provisions and the instructor's consent

Unattended Experiment Guidelines

- Everything well secured
- Check unattended reactions periodically
- Post contact information
- Accurate instructions and _____ for emergency personnel

Routes of Chemical Exposure

- skin
 - protective clothing
 - gloves
- ingestion
 - don't pipette by mouth!
 - no food in the lab
 - wash before leaving the lab
- inhalation
 - dust and vapors (use fume hood when either are present)
- eyes
 - eye protection against splashing and vapors



Safety Data Sheets: an information source

- Emergency information
 - chemical product and company identification
 - composition
 - hazards identification
- Appropriate response to hazardous situations
 - first aid measures
 - fire fighting measures
 - accidental release measures

<http://chemicalsafety.com/sds-search/>

Safety Data Sheets: an information source

- Prevention of hazardous situations
 - handling and storage
 - exposure controls and personal protection
 - physical and chemical properties
 - stability and reactivity
- Miscellaneous
 - toxicological information
 - ecological information
 - disposal considerations
 - transportation information
 - regulatory information

Safety Data Sheets: an example - Hydrochloric Acid

- You've been asked to mix up a solution of 1N HCl from concentrated hydrochloric acid. You would like to know of any hazards associated with this procedure.
- You remember the safety rule and you think through the steps of preparing the solution. You realize that you've never opened a container of concentrated HCl before. So you STOP! and get some help.

HYDROCHLORIC ACID

DANGER

Causes severe skin burns and eye damage
May be corrosive to metals
May cause damage to organs through prolonged or repeated exposure
May cause respiratory irritation

Wear protective gloves/protective clothing/eye protection/face protection
Use only outdoors or in a well-ventilated area
Keep only in original container
Immediately call a POISON CENTER or doctor/physician
Wash contaminated clothing before reuse
Store locked up
Store in a dry place

Hydrochloric Acid: My Solution to the solution

- Ventilation is required!
- Solution preparation should be done in a fume hood with gloves, protective clothing, and eye protection

SDS: Summary

- A good source of information on safe handling
- May provide useful information on physical-chemical properties
- Provide first aid, fire fighting, and accidental release measures
- Can be used to obtain concentrations of ingredients in proprietary solutions