

# Investigating a removable capping layer for Fe<sub>3</sub>GeTe<sub>2</sub> films

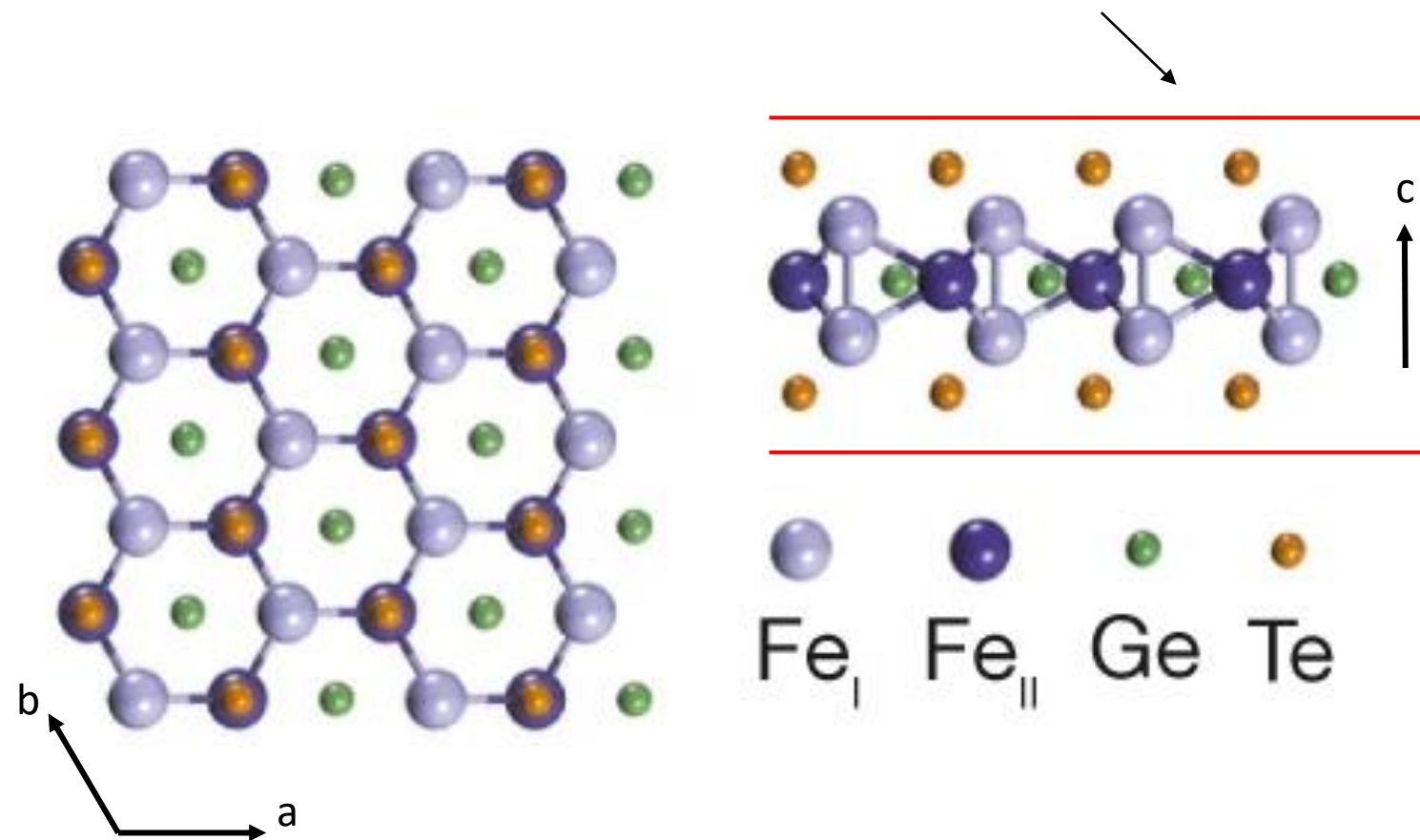
Dante Prins 2021



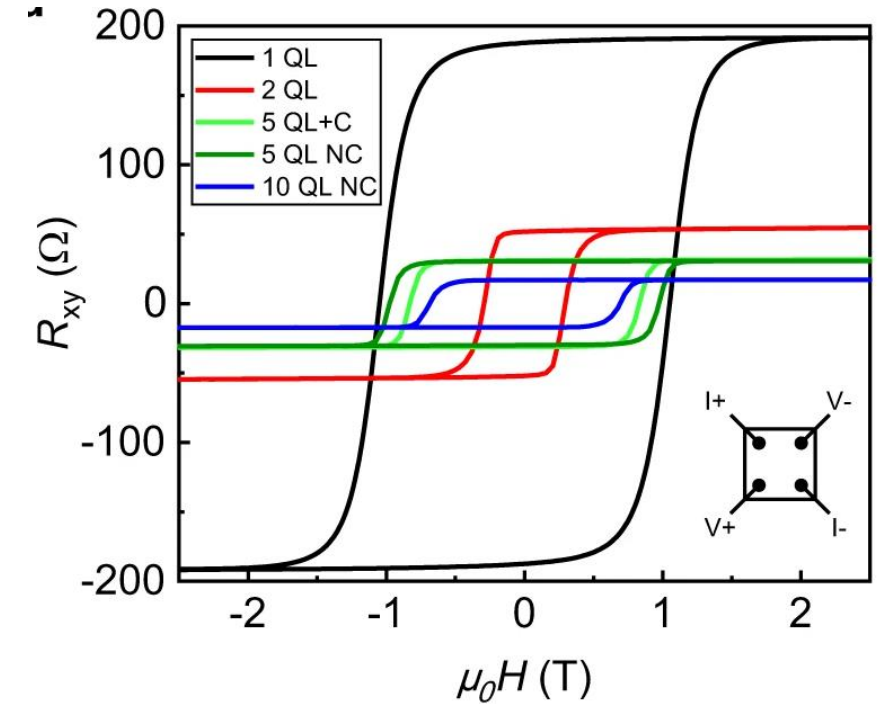
# What is Fe<sub>3</sub>GeTe<sub>2</sub>?

Atom structure of Fe<sub>3</sub>GeTe<sub>2</sub>.

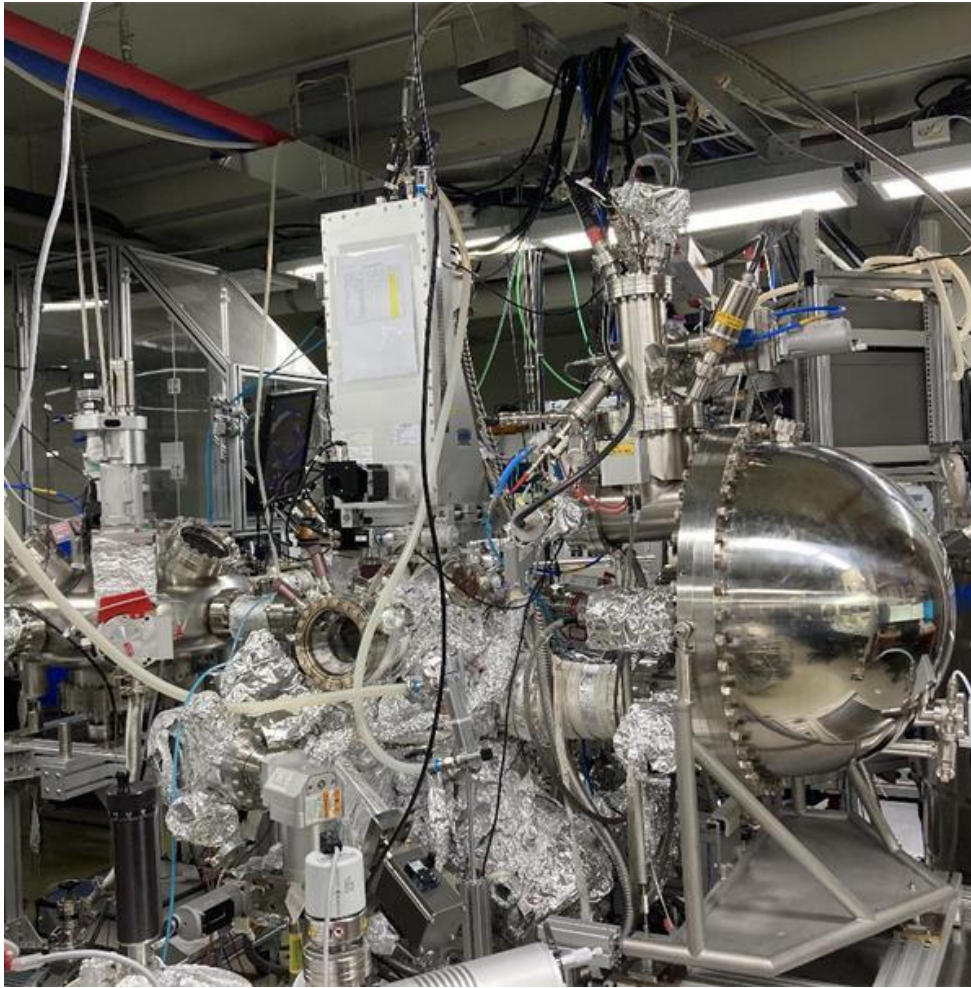
Van Der Waals gap between layers



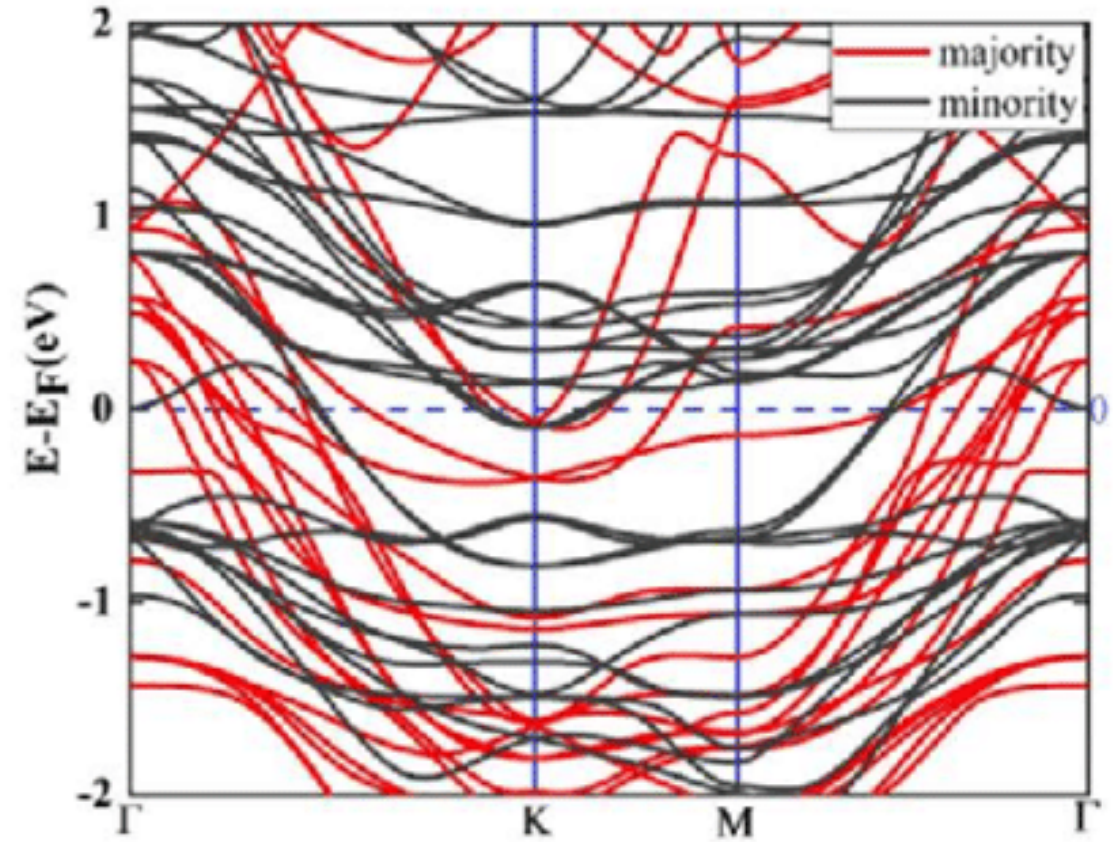
Transverse resistance hysteresis showing Anomalous Hall Effect behavior in different thickness samples.



Roemer, R., Liu, C. & Zou, K. Robust ferromagnetism in wafer-scale monolayer and multilayer Fe<sub>3</sub>GeTe<sub>2</sub>.



Spin-resolved Angle-resolved  
Photoemission Spectroscopy at the  
Electronic Structures Group in Korea.



Li, xinlu et al. Spin-dependent transport in van der  
Waals magnetic tunnel junctions with  $\text{Fe}_3\text{GeTe}_2$   
electrodes.

# Protective cap specifications



Growable and removable at under 360°C.

$O_2$

Protects sample from oxidation in air.



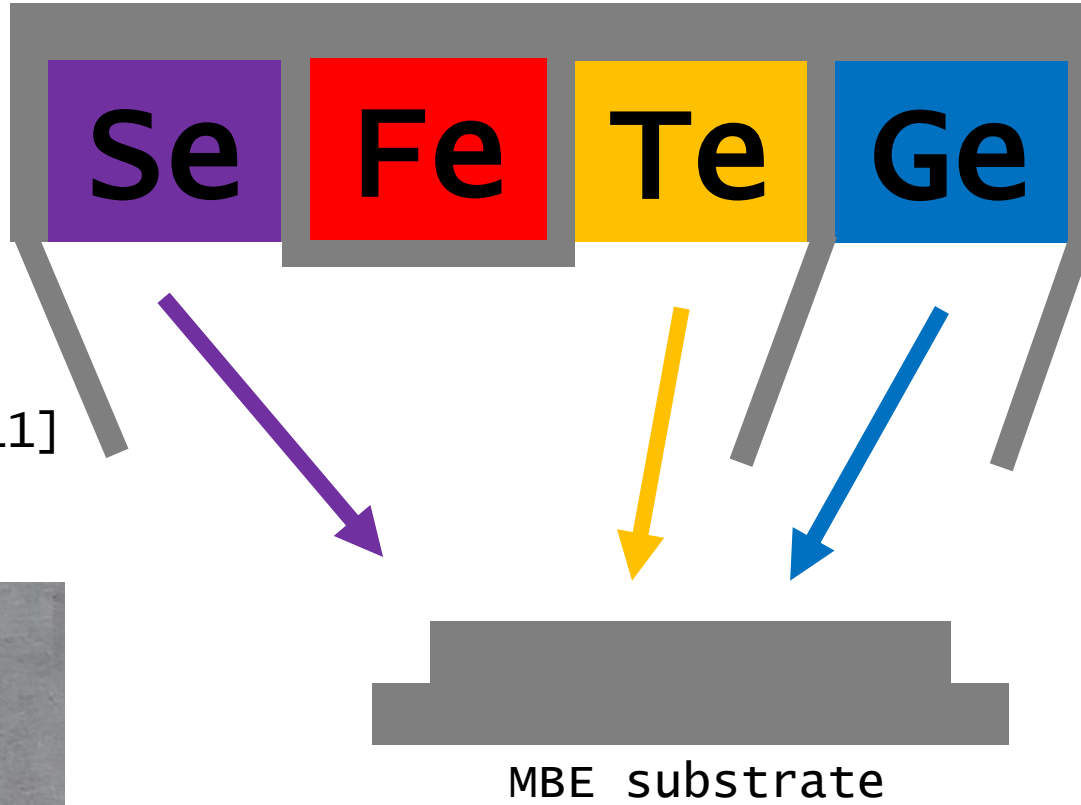
FGT Surface remains flat and unchanged after cap removal.



Polished Ge[111] substrate



# Molecular Beam Epitaxy film growth



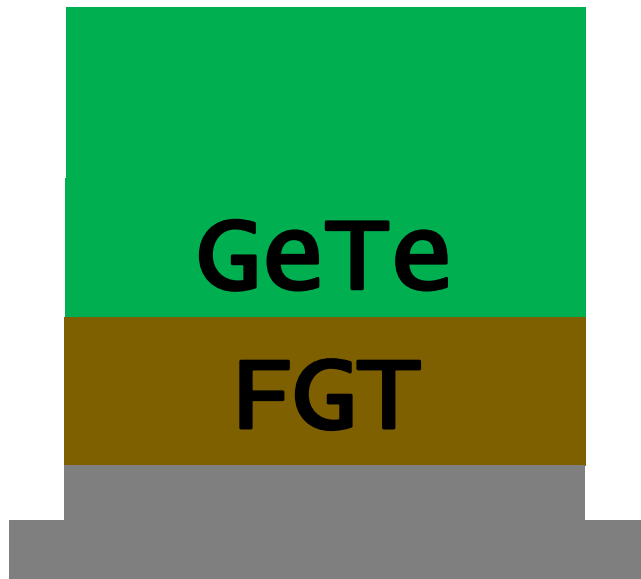
A Tellurium cell with power and cooling cables seen below.

Temperature at which each atom has  
vapor pressure of  $10^{-8}$  torr.

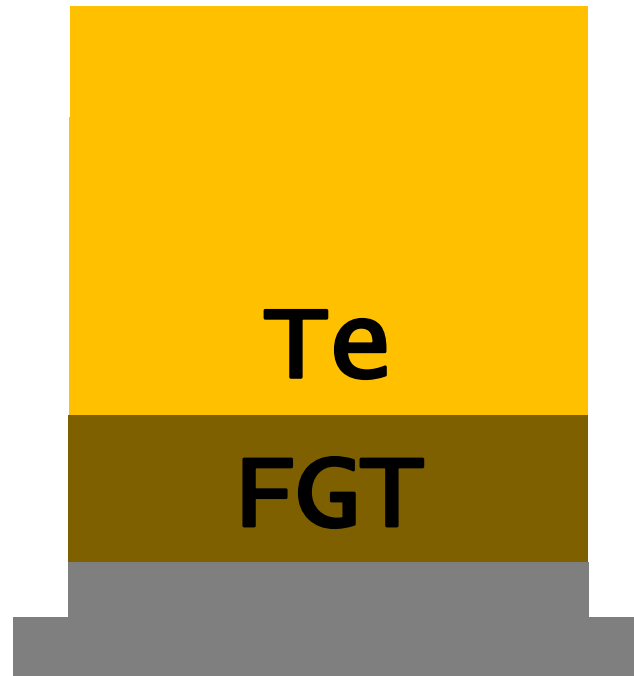
Se	Fe	Te	Ge
63°C	858°C	155°C	812°C

## Candidate materials

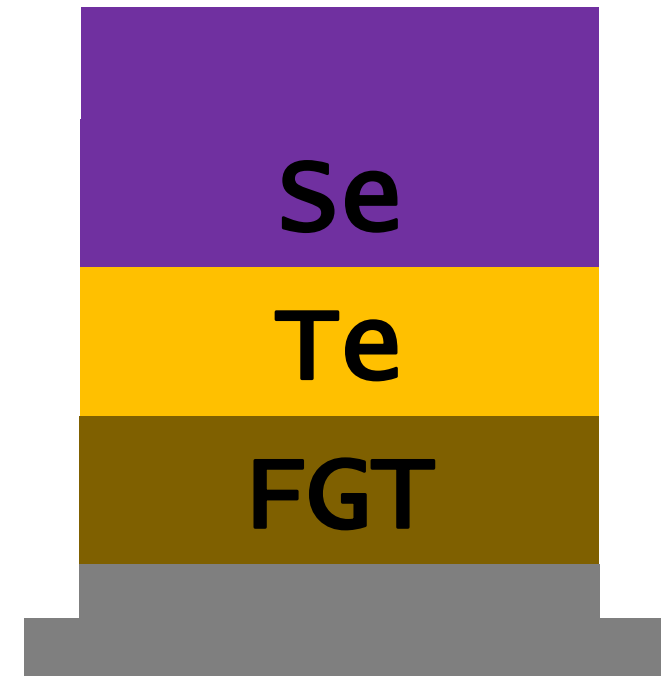
Thick germanium  
telluride cap



Thick tellurium cap

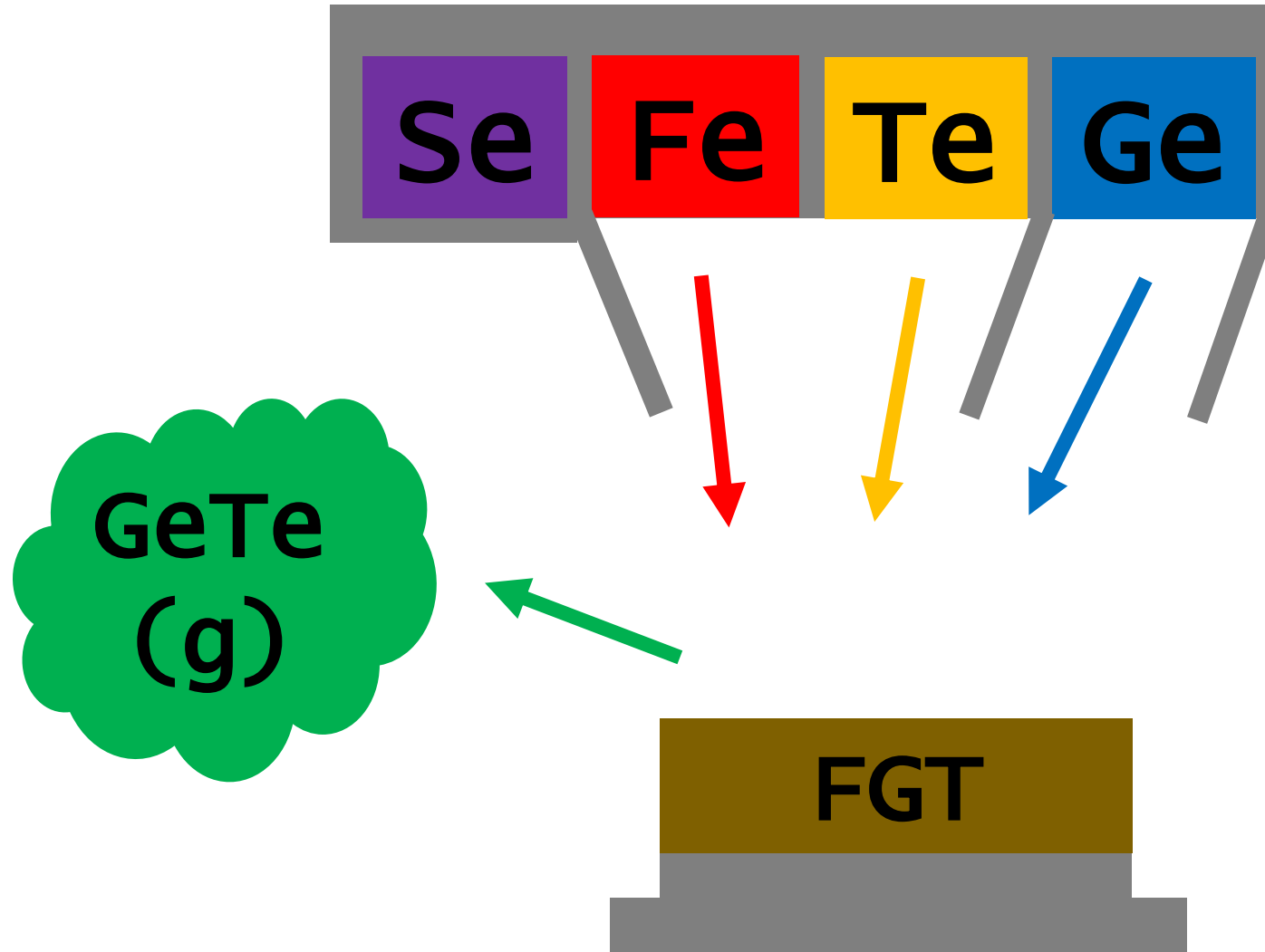


Tellurium buffer layer(s)  
with thick selenium cap

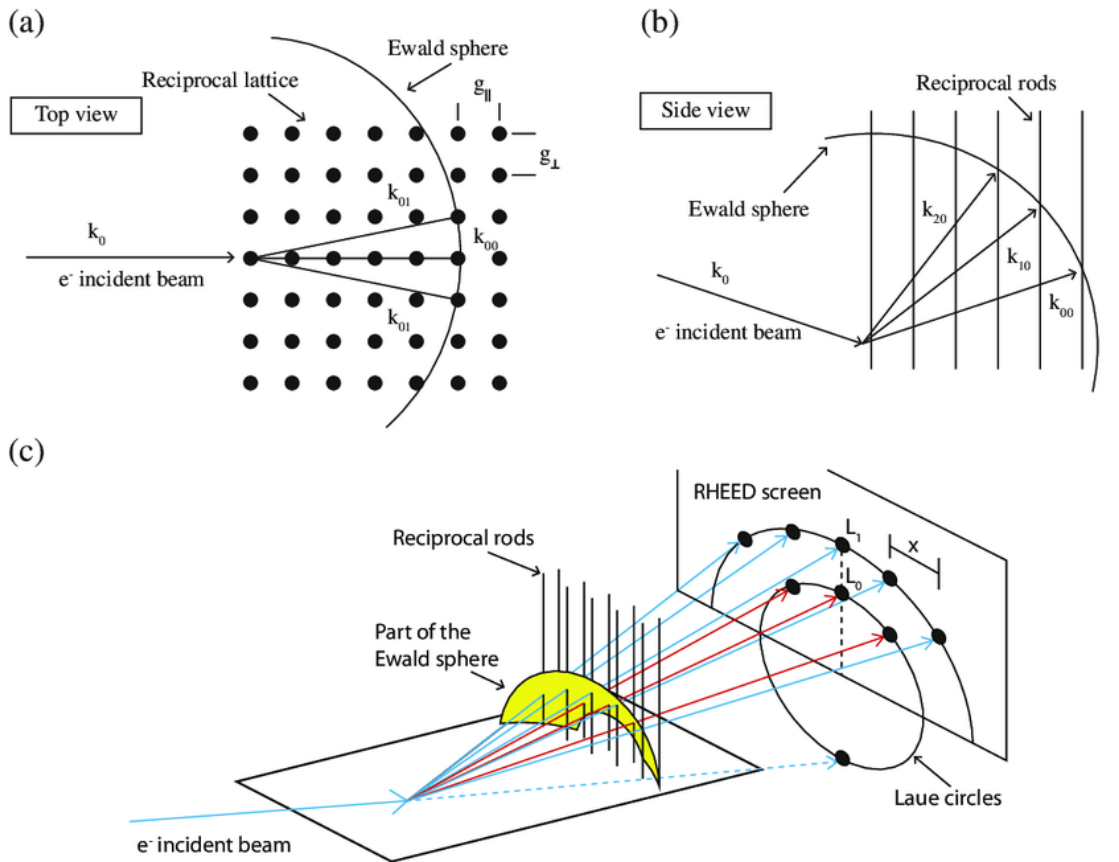


# Germanium telluride

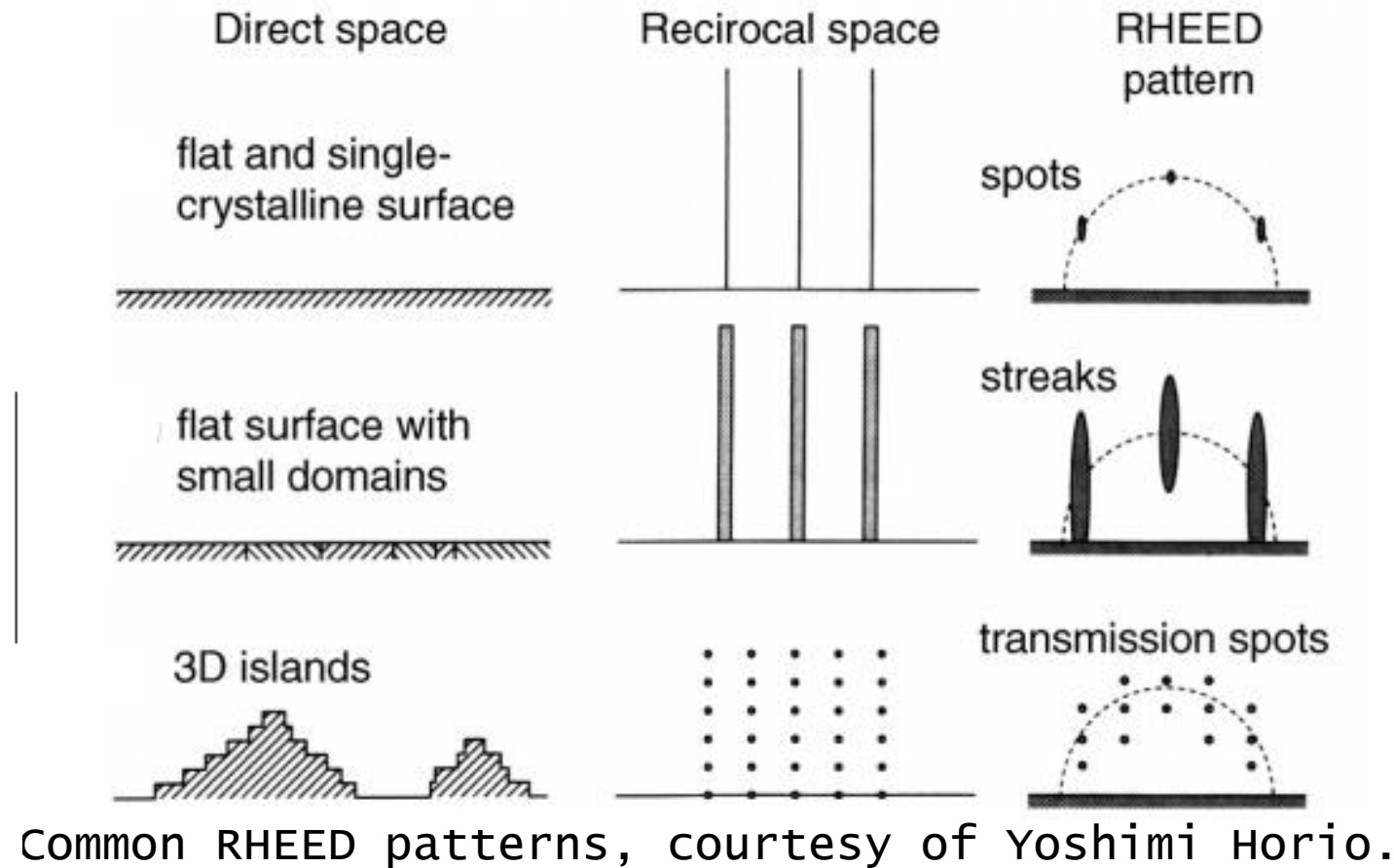
GeTe is known to vaporize at 360°C during FGT growth.



# Reflective High-Energy Electron Diffraction



Landgraf, Boris. (2014). Structural, magnetic and electrical investigation of Iron-based III/V-semiconductor hybrid structures.





Polished Ge[111]

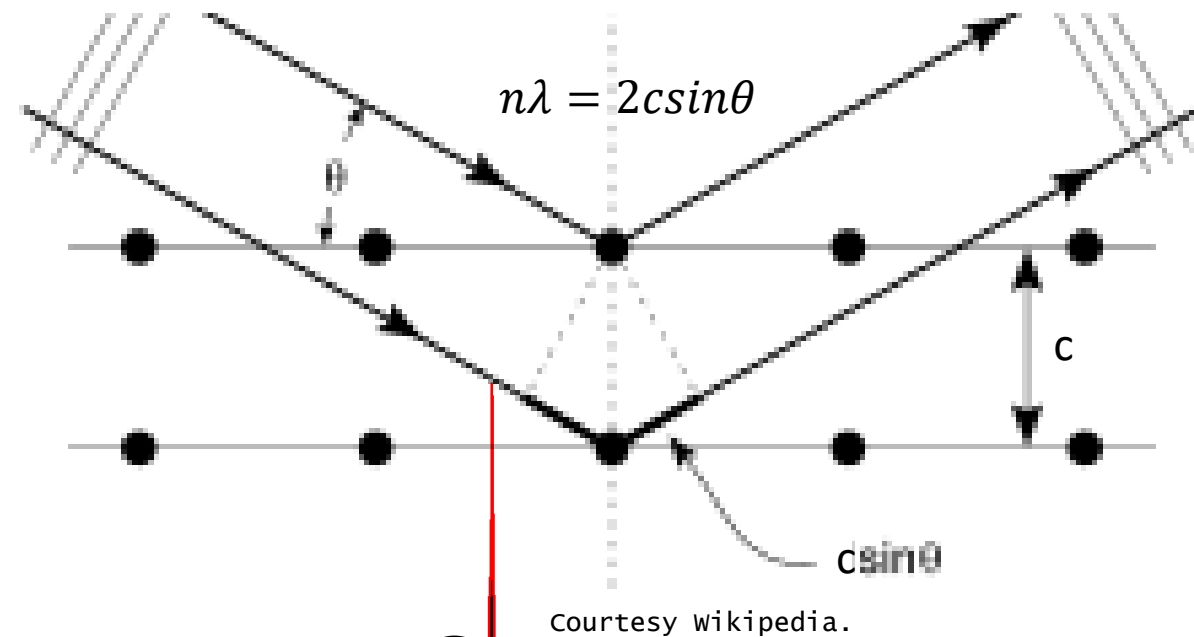
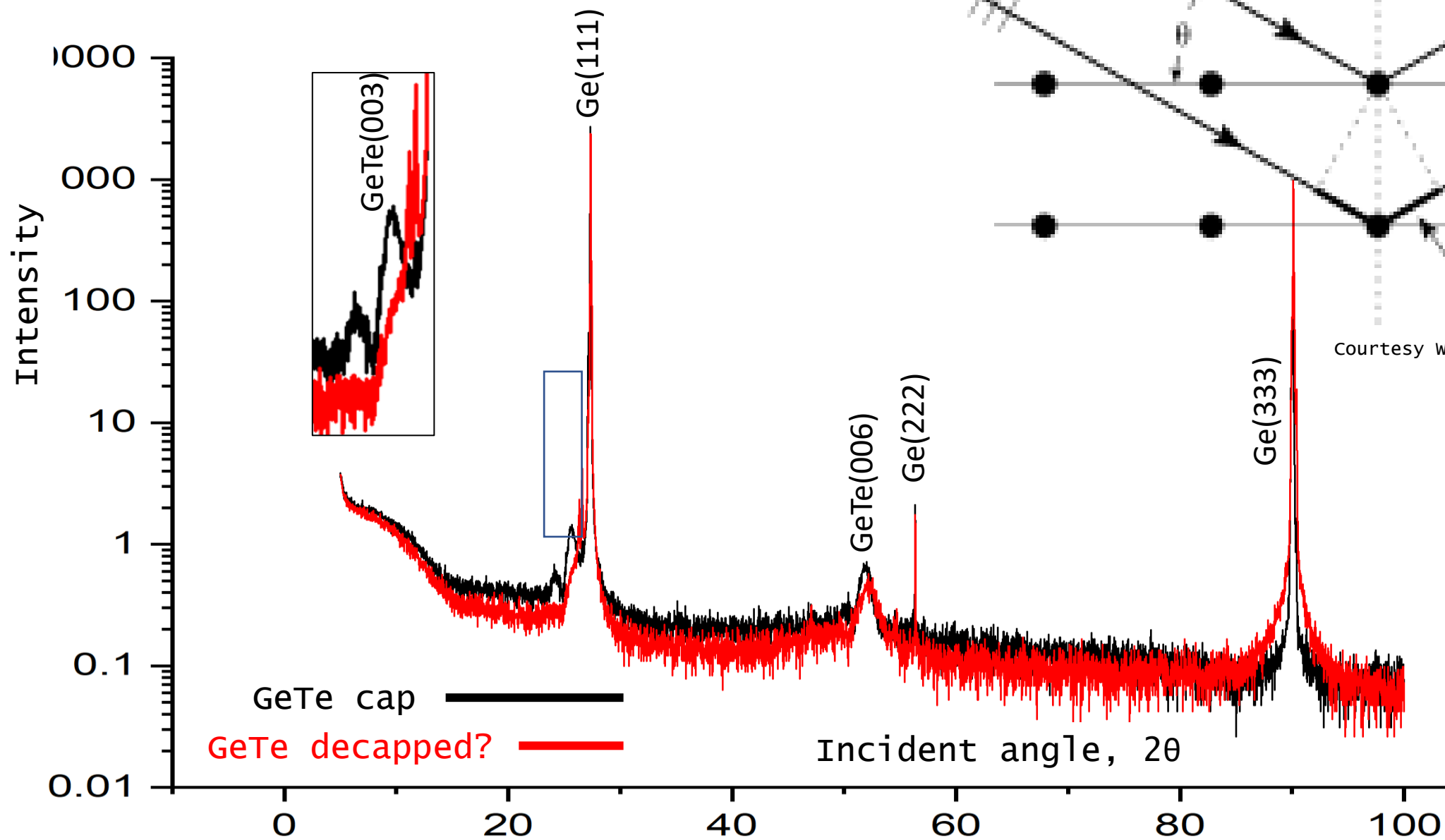
FGT

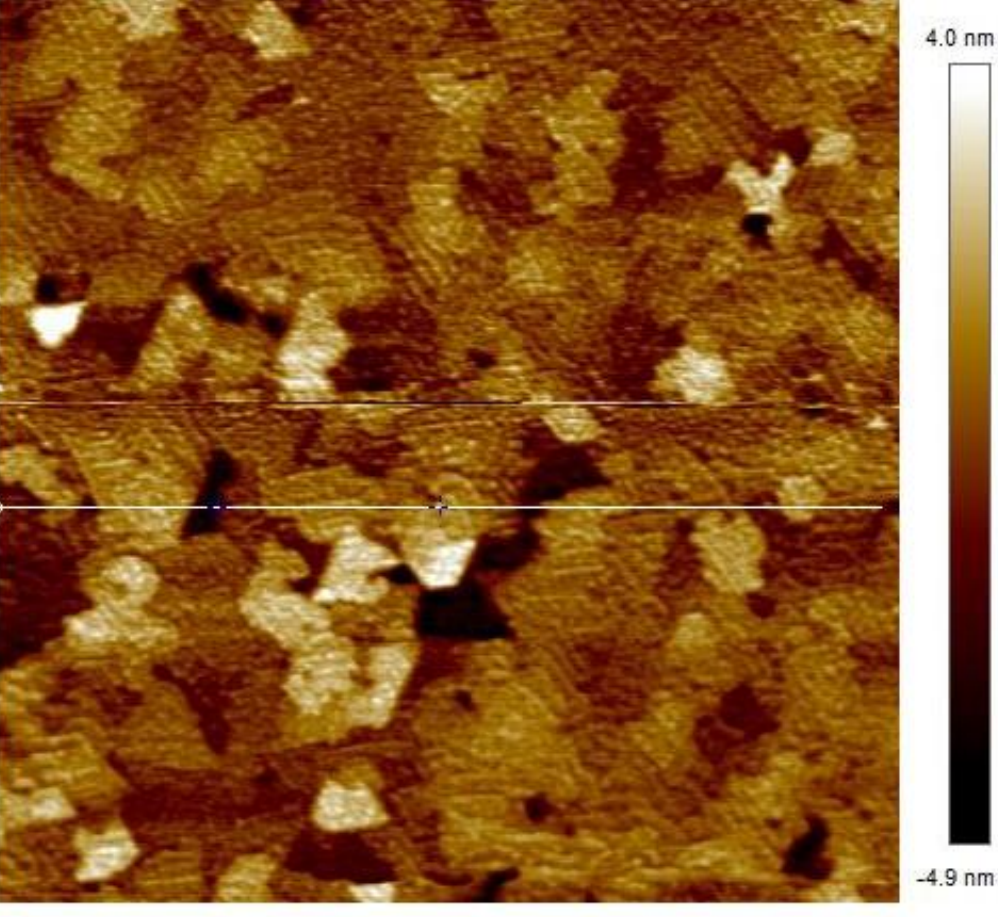
GeTe cap

Sample exposed to air and reheated. Has it been removed?

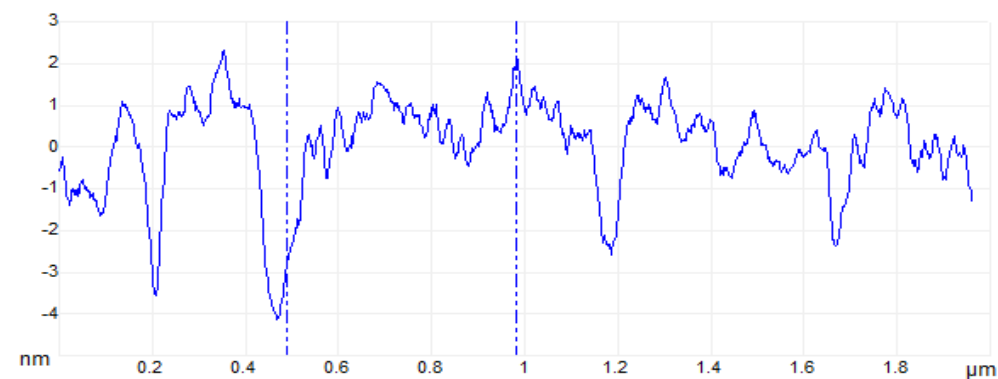
# X-ray Diffraction

## X-Ray Diffraction with Bragg Model

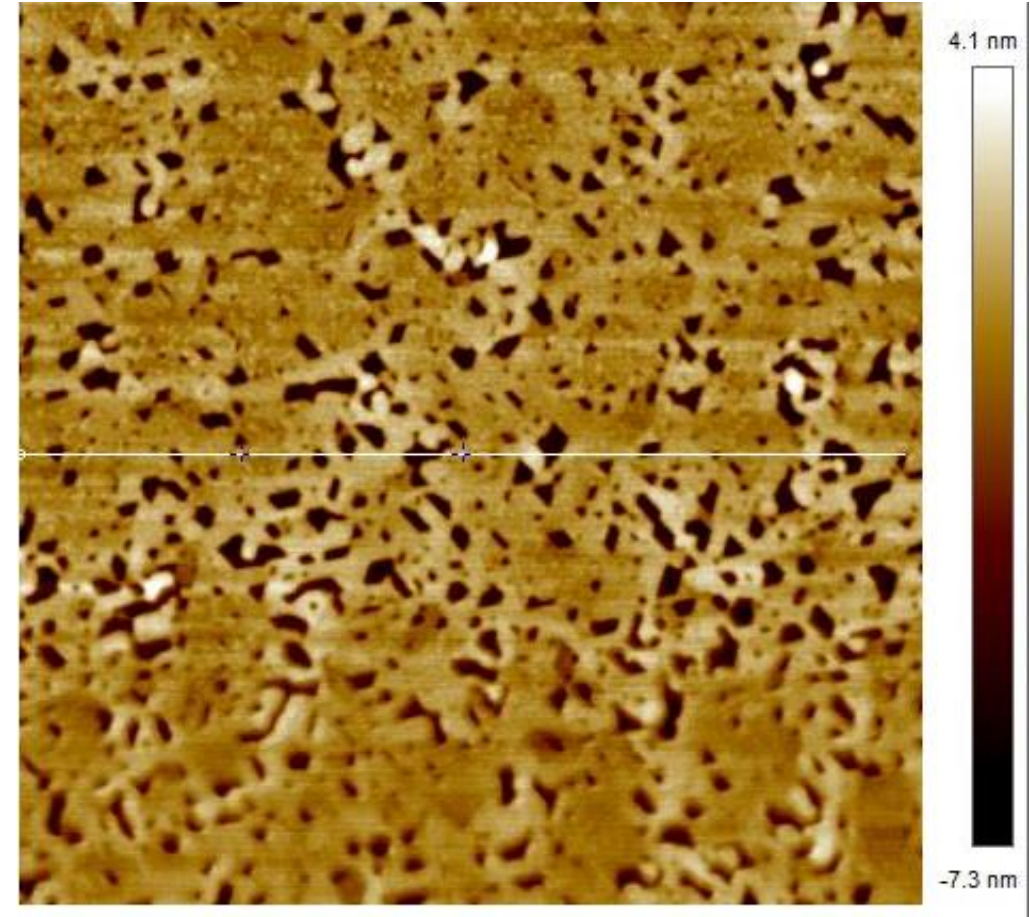




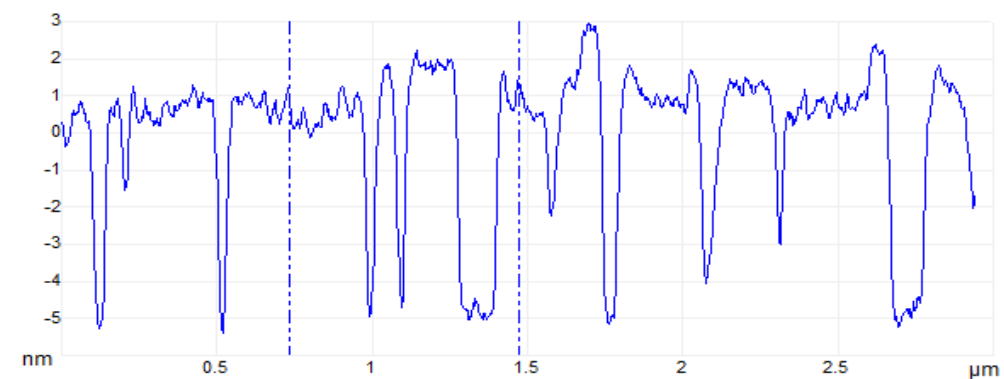
GeTe cap



# Atomic Force Microscopy

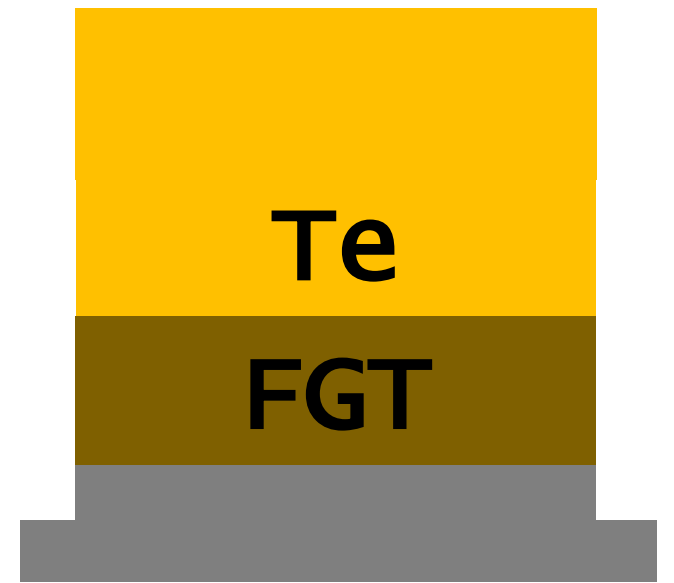


GeTe decapped?



# Thick Tellurium cap

Tellurium cap RHEED





Iron germanide

FGT

Te cap heated to 310°C.  
New phase:  $a = 3.8\text{\AA}$  2x2 reconstruction

Te cap heated to 410°C.  
FeGe?

FGT

Te cap heated to 310°C.  
New phase:  $a = 3.8\text{\AA}$  2x2 reconstruction

Te cap heated to 410°C.  
FeGe?

FeGe.

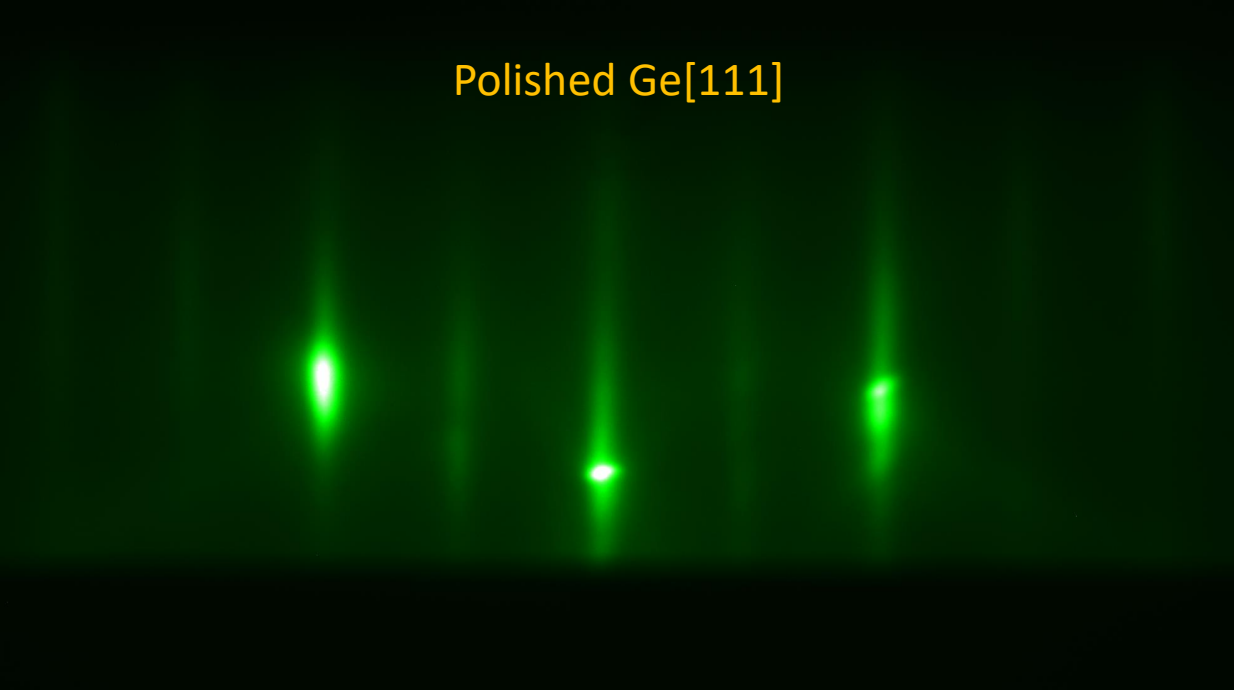
# Thin Tellurium cap

1-6 layers of Tellurium < 25Å.

Tellurium cap RHEED  
FGT signature still visible.



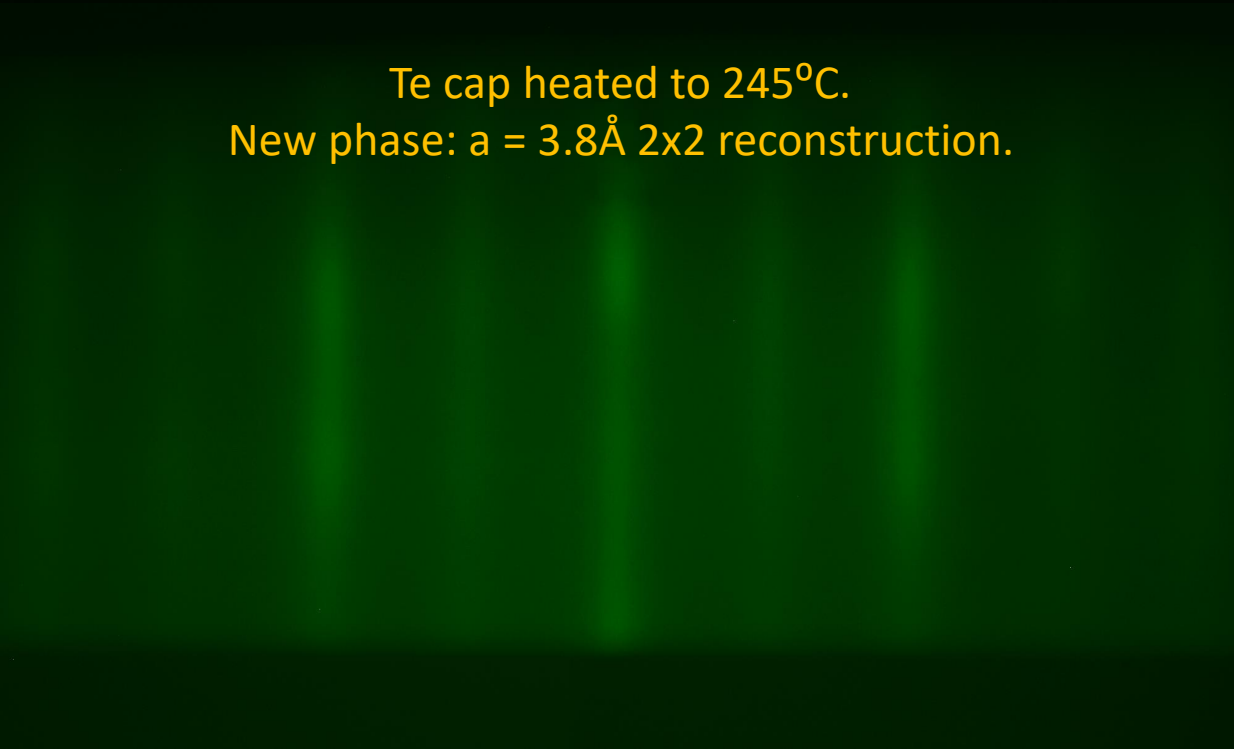
Polished Ge[111]



FGT



Te cap heated to 245°C.  
New phase:  $a = 3.8\text{\AA}$  2x2 reconstruction.

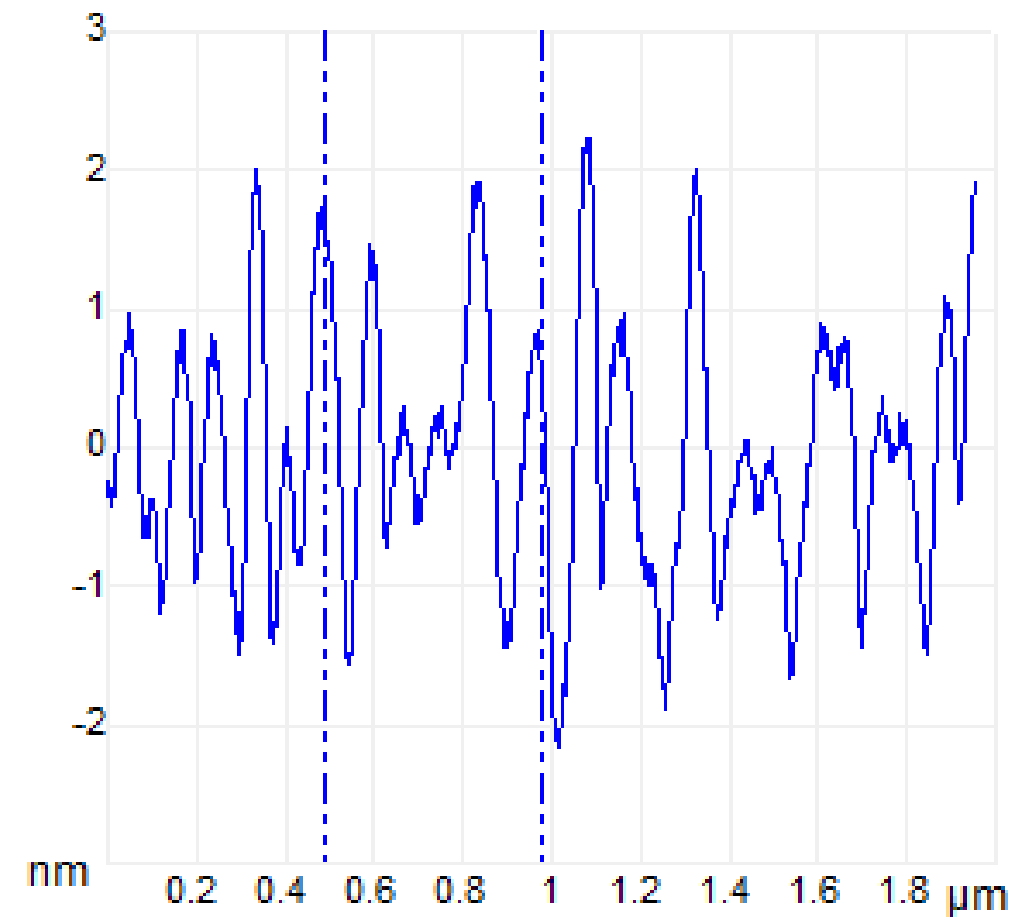
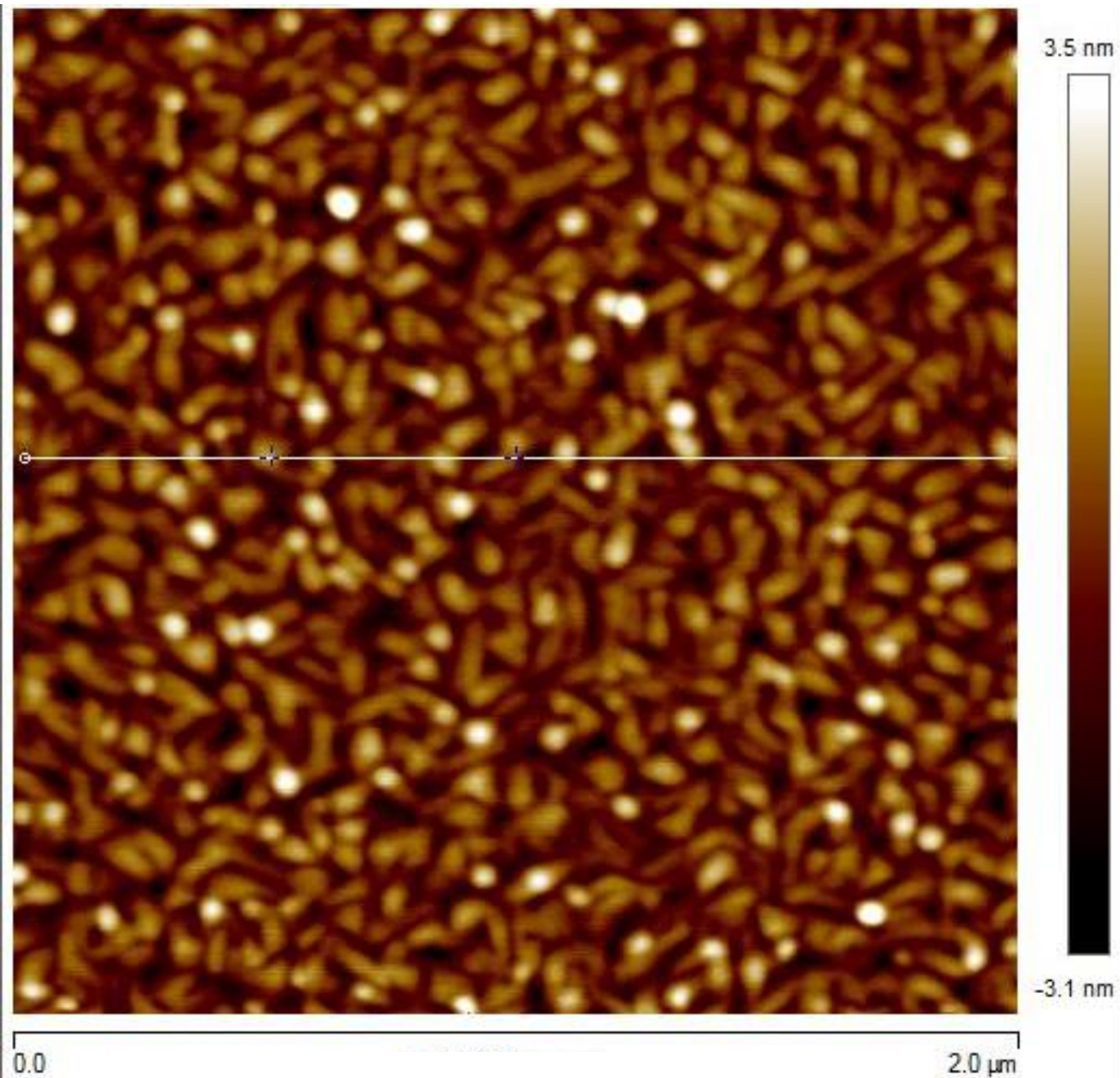


Te cap heated to 310°C.  
Is this FGT, or FeGe?





# Atomic Force Microscopy




# Thin Tellurium cap adding selenium

1-6 layers of Tellurium < 25Å  
And then 20nm of Se.


Amorphous Se cap RHEED signature.





Cap heated to 200°C.  
Se leaves revealing Te crystal.

This RHEED pattern shows a central bright spot surrounded by several distinct, sharp diffraction spots, indicating a crystalline structure.



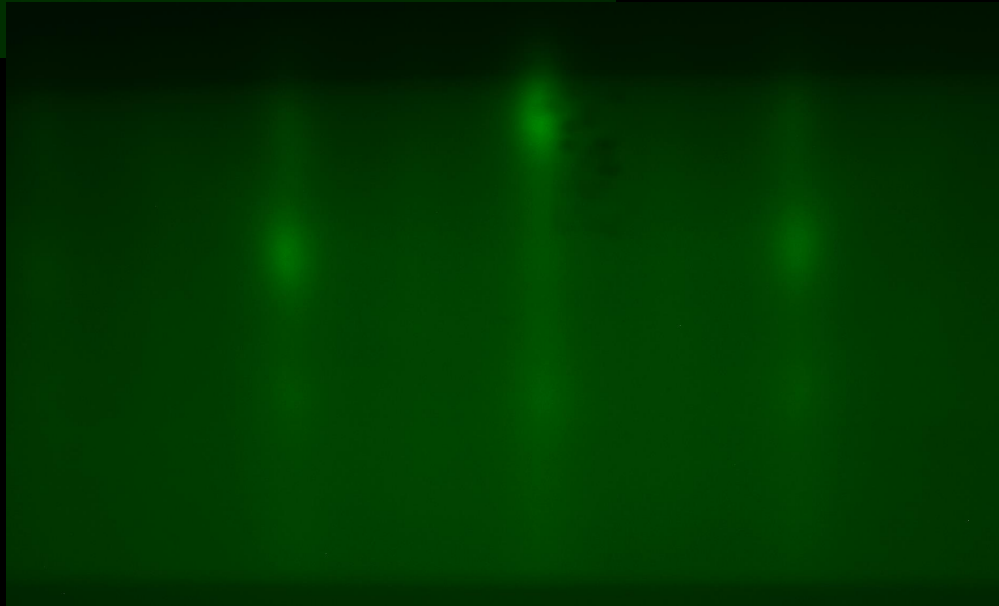
Cap heated to 235°C.  
Te crystal disappears and  
RHEED returns to amorphous.

This RHEED pattern shows a central bright spot with a diffuse, amorphous-like halo, indicating a loss of long-range crystalline order.



New 3D phase appears at 275°C.

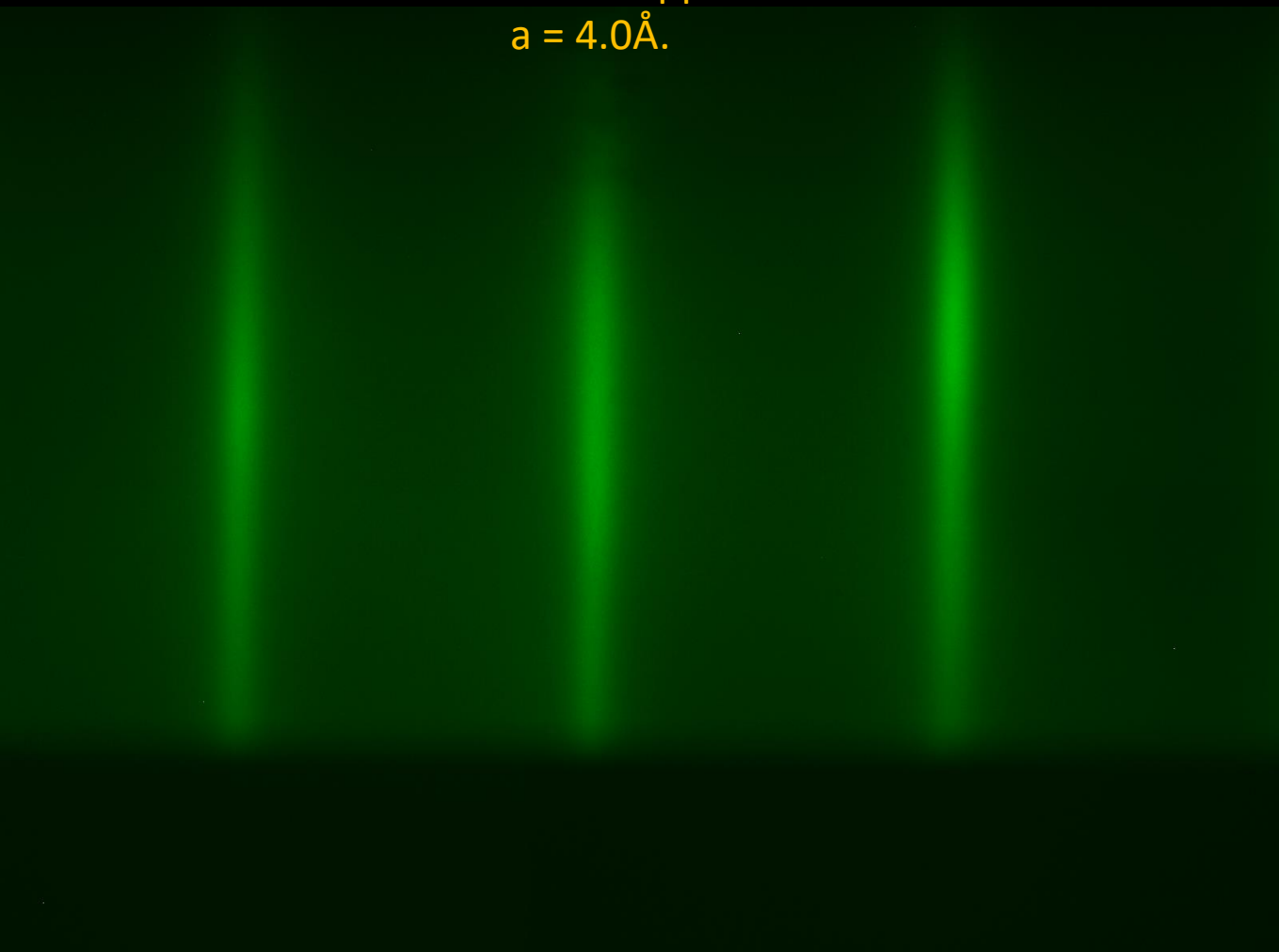
This RHEED pattern shows a central bright spot with several distinct, sharp diffraction spots, indicating a new crystalline phase.



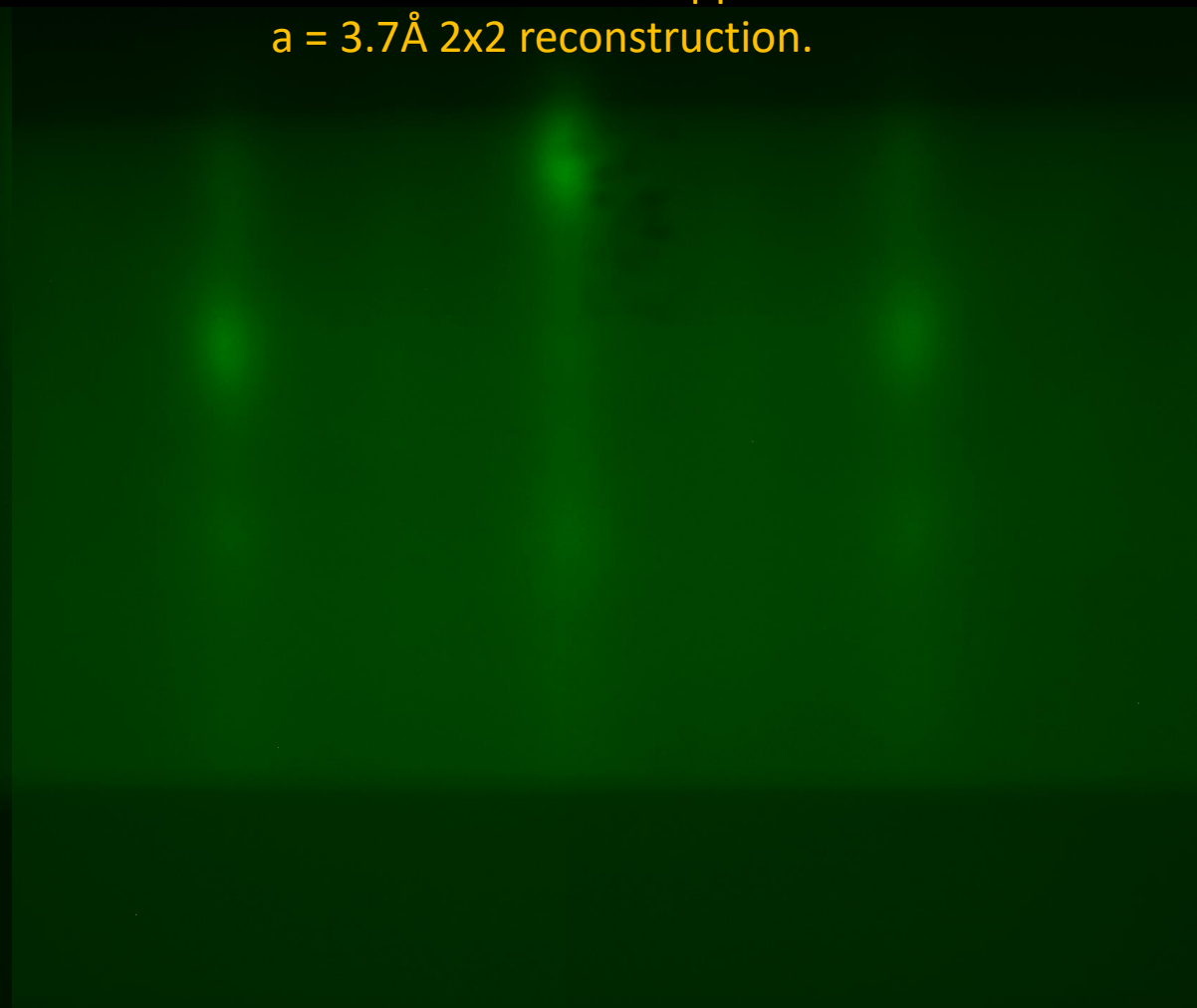
New phase has:  $a = 3.7\text{\AA}$ , is 3D,  
slight reconstruction.

This RHEED pattern shows a central bright spot with several distinct, sharp diffraction spots, indicating a new crystalline phase.

Thin Te decapped  
 $a = 4.0\text{\AA}$ .



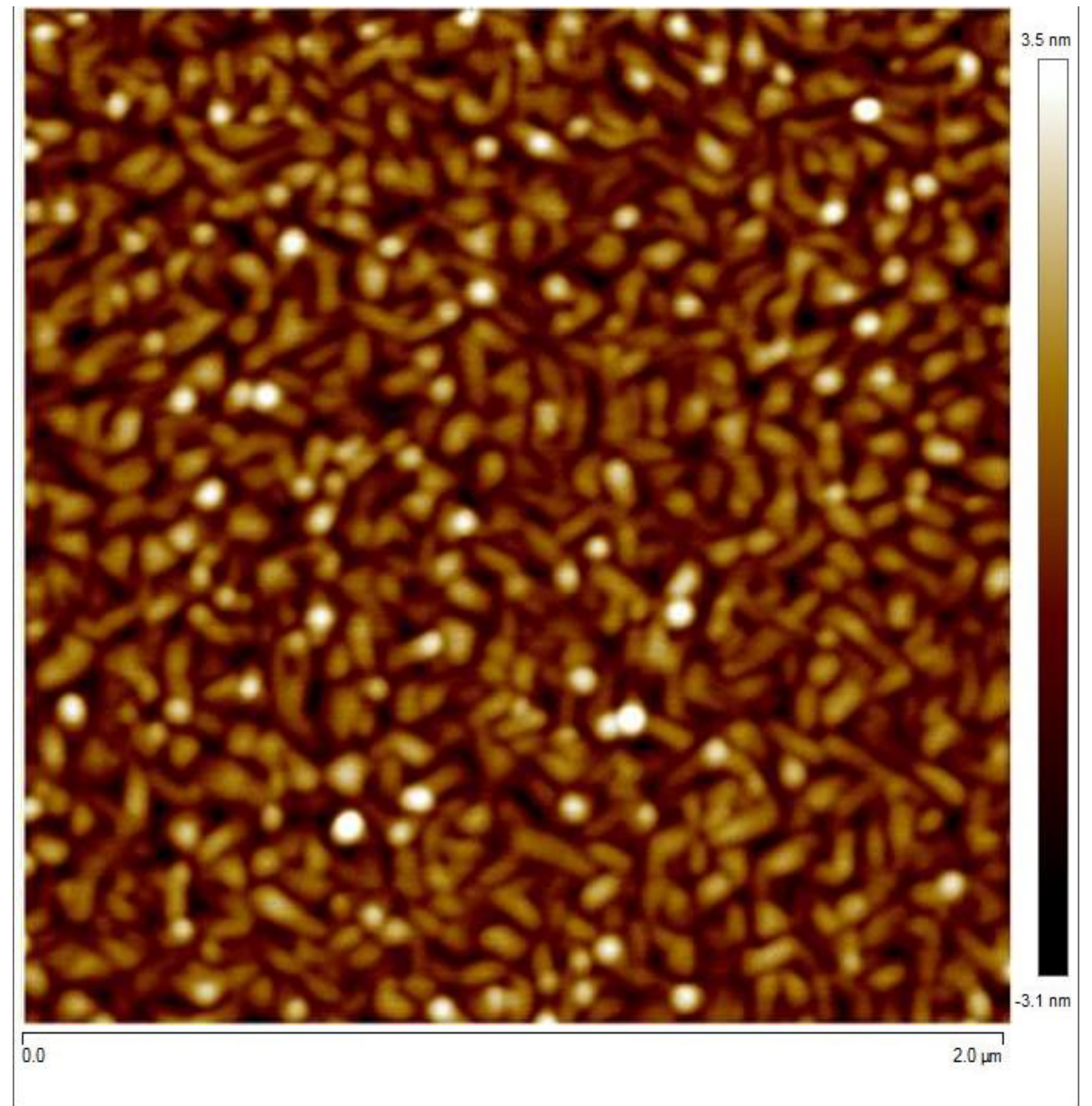
Thin Te with Se decapped  
 $a = 3.7\text{\AA}$  2x2 reconstruction.





# Selenium penetration

At low temperatures tellurium does not have energy to reorganize to a single crystal, resulting in random distributions.



AFM of a flat crystal capped with Tellurium at room temperature.

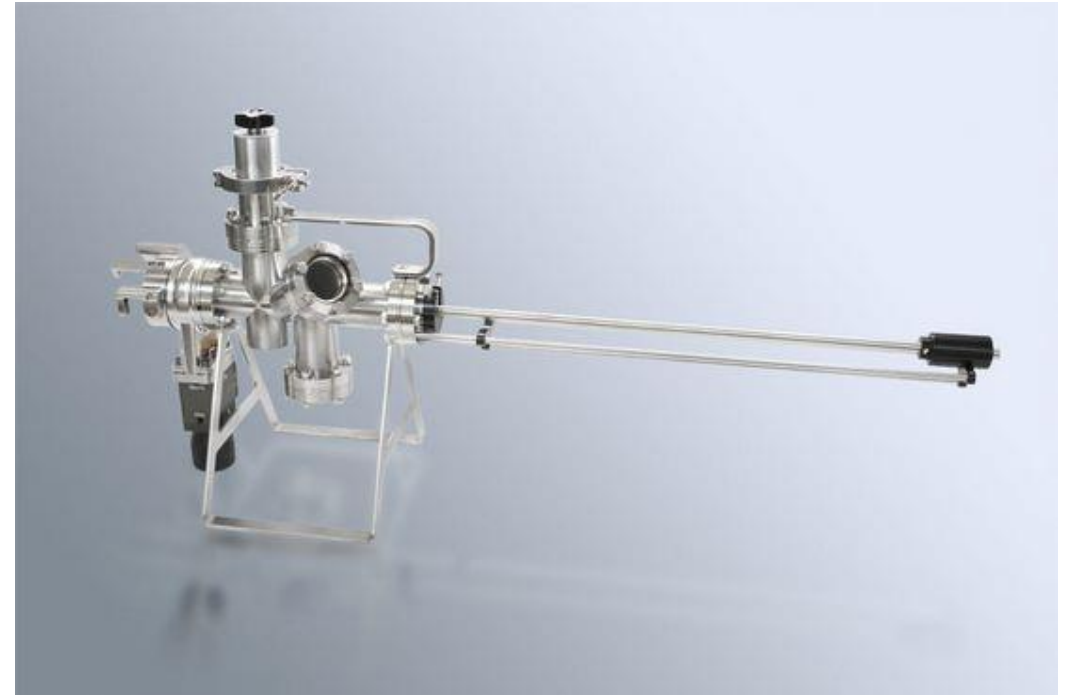
## Next steps for the project

## New elements for the chalcogen MBE chamber?

The periodic table is color-coded to show groups of elements with similar properties. The colors are as follows:

- Red:** Group 1 (Alkali Metals)
- Orange:** Group 2 (Alkaline Earth Metals)
- Purple:** Groups 3-10 (Transition Metals)
- Green:** Groups 11-12 (Post-Transition Metals)
- Light Green:** Groups 13-15 (Metalloids)
- Blue:** Groups 16-17 (Nonmetals)
- Yellow:** Group 18 (Noble Gases)

The lanthanide and actinide series are shown at the bottom in a separate row, color-coded by group.



## Long range Ultra-High Vacuum suitcase?