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Treatment of recalcitrant crystalline polysaccharides with lytic polysaccharide monooxygenase relieves the need for glycoside hydrolase processivity

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ABSTRACT

Processive glycoside hydrolases associate with recalcitrant polysaccharides such as cellulose and chitin and repeatedly cleave glycosidic linkages without fully dissociating from the crystalline surface. The processive mechanism is efficient in the degradation of insoluble substrates, but comes at the cost of reduced enzyme speed. We show that less processive chitinase variants with reduced ability to degrade crystalline chitin, regain much of this ability when combined with a lytic polysaccharide monooxygenase (LPMO). When combined with an LPMO, several less processive chitinase mutants showed equal or even increased activity on chitin compared to the wild-type. Thus, LPMOs affect the need for processivity in polysaccharide degrading enzyme cocktails, which implies that the composition of such cocktails may need reconsideration.

1. Introduction

Chitin and cellulose represent some of nature's largest reservoirs of organic carbon in the form of monomeric hexose sugars (*N*-acetyl-glucosamine and glucose, respectively) linearly linked by β-1,4 glycosidic bonds. In their natural form, both polysaccharides are organized in crystalline arrangements that make up robust biological structures including crustacean cuticles (chitin) and plant cell walls (cellulose). Although these crystalline arrangements are crucial for biological function, they present a significant challenge in industrial utilization of biomass, where efficient enzymatic depolymerization is a critical step.

Enzymatic degradation of recalcitrant polysaccharides is thought to occur primarily through the synergistic action of glycoside hydrolases (GHs) that have complementary activities [1,2]. Endo-acting GHs make random scissions on the polysaccharide chains, whereas exo-acting GHs mainly target reducing and non-reducing chain ends. Both endo- and exo-acting GHs may be processive, which implies that they repeatedly cleave glycosidic linkages without fully dissociating from the crystalline surface. Lytic polysaccharide monooxygenases (LPMO) add to the degradation process by using a reduced oxygen species to cleave glycosidic bonds in crystalline regions, thus creating new chain ends for exo-acting GHs [3–6]. Processive GHs are typically the most abundant enzymes in both natural secretomes and industrial enzyme cocktails by virtue of their significant hydrolytic potential on crystalline substrates [7]. LPMOs also make considerable contributions to the overall efficiency of cellulolytic enzyme cocktails [8], but the interplay between LPMOs and individual (processive) GHs is not well understood.

Processive GHs have active site tunnels or deep clefts aligned with surface exposed aromatic amino acids that provide strong binding to the substrate [9–14]. Aromate-mediated carbohydrate recognition facilitates the necessary fluid binding to the crystalline polymers, increases the efficiency of substrate degradation, and may provide a necessary pushing potential to overcome substrate obstacles that otherwise could have led to stalling of the enzyme [14–20]. The cost of the processive mechanism is loss of enzyme speed as the required strong interactions make the enzymes intrinsically slow. Indeed, studies of processive chitinases [16,20] have shown that replacement of aromatic amino acids by alanine led to reduced processivity and decreased activity on crystalline chitin, while yielding an up to 30-fold increase in the rate of hydrolysis of water soluble chitosan. A consequence is that, in some cases, it might be better to focus strategies for enzymatic depolymerization of polysaccharide biomass on improving substrate accessibility for non-processive enzymes rather than on improving the properties of processive enzymes [16]. Such strategies could include the use of LPMOs, which are known to disrupt the crystalline structure of chitin and cellulose [6,21–23].

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In this work, we have investigated how treatment with a chitin-active LPMO affects the initial rate of chitin hydrolysis by engineered variants of two exo-processive chitinases from *Serratia marcescens* with different degrees of processivity. The results show that the LPMO makes the crystalline material more amenable to the action of chitinase variants with reduced processivity, and have potential implications for the future design of enzyme cocktails for conversion of recalcitrant polysaccharides.

2. Materials and methods

2.1. Chemicals

Chito-oligosaccharides (CHOS) were purchased from Megazyme (Wicklow, Ireland). Squid pen β-chitin was purchased from France Chitine (180 μm microparticulate, Marseille, France). All other chemicals were of analytical grade and purchased from standard manufacturers.

2.2. Site-directed mutagenesis

Most chitinase variants used in this study have been described previously. Variants *Sm*ChiB-F190A and *Sm*ChiB-F190A/W220A (using *Sm*ChiB-WT and *Sm*ChiB-F190A as template, respectively) were generated using the QuikChange site directed mutagenesis kit from Stratagene (La Jolla, CA, USA), mainly as described by the manufacturer. The primers used for the mutagenesis are listed in Table 1 and were purchased from Life Technologies (Carlsbad, CA, USA). To confirm that the gene contained the desired mutation and to check for the occurrence of non-desirable mutations, the mutated genes were sequenced using the LIGHTrun sequencing service of GATC Biotech (Constance, Germany), before they were transformed into *Escherichia coli* BL21Star (DE3) cells (Life Technologies).

2.3. Protein expression and purification

The chitinases were from *S. marcescens* strain BJL200 [24,25]. *Sm*ChiA-WT, *Sm*ChiB-WT, *Sm*ChiA-W167A, *Sm*ChiA-W275A, *Sm*ChiB-W97A, *Sm*ChiB-W220A, *Sm*ChiB-W97A/W220A, and *Sm*ChiB-F190A were expressed in *E. coli* as described previously [16,20,26]. The same applies for *Sm*LPMO10A (previously known as CBP21) [27]. The periplasmic extracts were loaded on a column packed with chitin beads (New England Biolabs, Ipswich, MA, USA) equilibrated in 50 mM Tris-HCl pH 8.0 (chitinases) or 50 Mm Tris-HCl pH 8.0, 1 M ammonium sulfate (*Sm*LPMO10A)After washing the column with the same buffer, the enzymes were eluted with 20 mM acetic acid. Purified proteins were further concentrated by use of Macrosep® Advances Centrifugal Devices (PALL Corporation, New York, NY) with a molecular mass cutoff of 30000 Da (chitinases) or Amicon Ultra Centrifugal filters (Merck Millipore, Darmstadt, Germany) with a molecular mass cutoff of 10000 Da (*Sm*LPMO10A). The chitinases and *Sm*LPMO10A were stored at 4 °C in 100 mM Tris-HCl pH 8.0 or 50 mM sodium acetate buffer pH 6.1, respectively.

The *Sm*ChiB-F190A/W220A gene was expressed as described previously for wild type *Sm*ChiB and its mutants [16,24]. The periplasmic extract was adjusted to Buffer A (50 mM Tris-HCl pH 8.0, 1 M $(NH_4)_2SO_4$) with 3 M $(NH_4)_2SO_4$ and loaded onto a HiTrap PhenyL HP column (5 mL) (GE Healthcare, Little Chalfont, Great Britain) connected to a BioLogic low-pressure protein purification system (Bio-Rad, Hercules, CA, USA). The chitinase was eluted by applying a two-step gradient where the first step was a linear salt gradient from 100% buffer A to 70% buffer B (50 mM Tris-HCl pH 8.0) over 10 column volumes at a flow rate of 4 ml/min. The second step was a new linear salt gradient from 70 to 100% buffer B over 15 column volumes at a flow rate of 4 ml/min. Finally, buffer B was applied for 5 column volumes. The chitinase containing fractions were detected by SDS-PAGE, pooled and concentrated to 1 ml, using Macrosep® Advances Centrifugal Devices (PALL Corporation) with a molecular mass cutoff of 30000 Da. Subsequently, samples were loaded onto a HiLoad 16/600 Superdex 75 Prepgrade column (GE Healthcare), with a running buffer consisting of 20 mM Tris-HCl pH 8.0 and 0.2 M NaCl, using a flow rate of 1 ml/min. The chitinase eluted approximately 60 min after injection. The buffer was changed to 100 mM Tris-HCl pH 8.0 using Macrosep® Advances Centrifugal Devices (30000 Da cutoff) (PALL Corporation).

For all enzymes, chitinases and *Sm*LPMO10A, enzyme purity was verified by SDS-PAGE and the enzyme concentration was estimated by using the Bradford Protein Assay from Bio-Rad.

Table 1
Primes used for site-directed mutagenesis.

Enzyme	Primer	Sequence
ChiB-F190A	forward	5′GCCGGCGGCGCCGCCTTCCTGTCGCG′3
	reverse	5′CGCGACAGGAAGGCGGCGCCGCCGGC′3
ChiB-F190A/W220A	forward	5′-TGGCCGGCCCCGCGGAGAAG-3′
	reverse	5′-CTTCTCCGCGGGGCCGGCCA-3′

2.4. Time course of chitin degradation

Hydrolysis of chitin (20 mg/ml) was carried out in 50 mM sodium acetate at pH 6.1. Previously, we have shown that 20 mg/ml chitin gives substrate saturating conditions [3]. The chitin samples were sonicated for 20 min in a sonication bath (Transonic Elma) to increase the surface of the substrate and thereby increase the availability of chitin ends for the enzymes [28]. The reaction tubes were incubated at 37 °C in an Eppendorf thermomixer at 800 rpm. The chitinase concentration was 100 nM in all experiments. Aliquots of 75 μl were withdrawn every hour for 7 h, and the enzymes were inactivated by adding 75 μl 20 mM H_2SO_4. Prior to further HPLC analysis all samples were filtrated through a 0.45 μm Duapore membrane (Merck Millipore) to remove denatured protein and chitin particles. All reactions were run in, at least, triplicate, and all samples were stored at −20 °C until HPLC analysis.

2.5. Time course of chitin degradation after treatment with SmLPMO10A

The degradation was carried out as described in section 2.4 with the following exception: After sonication, *Sm*LPMO10A and ascorbic acid were added to an end concentration of 1 μM and 2 mM, respectively. The samples were incubated at 37 °C in an Eppendorf thermomixer at 800 rpm for 2.5 h before the addition of chitinase.

2.6. Determination of apparent degree of processivity

In this work, apparent degree of processivity was assessed on the basis of the initial $[(GlcNAc)_2]/[GlcNAc]$ product ratio from chitin hydrolysis as described in Hamre et al. [15]. Hydrolysis of chitin (2.0 mg/ml) was carried out in 50 mM sodium acetate buffer at pH 6.1. The chitin samples were sonicated for 20 min in a sonication bath (Transsonic, Elma). The reaction tubes were then incubated at 37 °C in an Eppendorf thermomixer at 800 rpm to avoid settling of the chitin particles. The chitinase concentration was 2.5 μM. Aliquots of 75 μL were withdrawn at regular time intervals, and the enzymes were inactivated by adding 75 μL 20 mM H_2SO_4. Prior to further HPLC analysis all samples were filtrated through a 0.45 μm Duapore membrane (Merck Millipore) to remove denatured protein and chitin particles. All reactions were performed in duplicate, and all samples were stored at −20 °C until HPLC analysis.

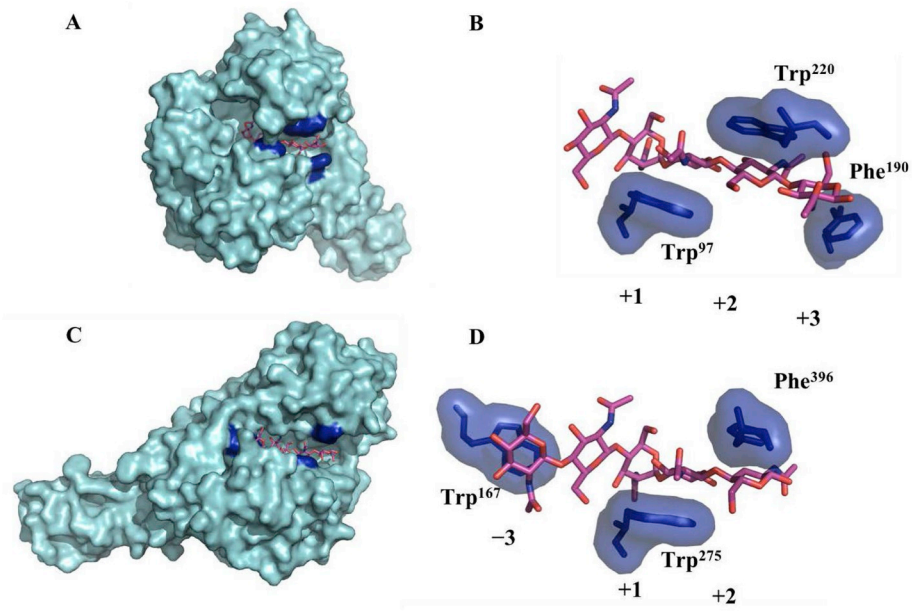

Fig. 1. Substrate binding in *Sm*ChiA and *Sm*ChiB. The panels show structurally aligned crystal structures of *Sm*ChiB (A&B; pdb 1e6n [31]) and *Sm*ChiA (C&D; pdb 1ehn [32]) with substrates. Selected surface exposed aromatic amino acids are highlighted in blue. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

2.7. High performance liquid chromatography (HPLC)

Concentrations of mono- and disaccharides were determined using a Dionex Ultimate 3000 UHPLC system (Dionex Corp., Sunnyvale, CA, USA) equipped with a Rezex Fast fruit H^+ column (100 mm length and 7.8 mm inner diameter) (Phenomenex, Torrance, CA, USA). The sample size was 8 µL, and the mono- and disaccharides were eluted isocratically at 1 mL/min with 5 mM H_2SO_4 at 85 °C. The mono- and disaccharides were monitored by measuring absorbance at 195 nm, and the amounts were quantified by measuring peak areas. Peak areas were compared with peak areas obtained with standard samples with known concentrations of mono- and disaccharides. The degree of degradation (in percent) is defined as the amount of solubilized GlcNAc units divided by the amount of GlcNAc units in the chitin polymer at the start of the experiment.

2.8. Determination of A- and b-values for activity

According to Kostylev and Wilson, the following two-parameter kinetic model can be used to determine a time-dependent activity constant for polysaccharide degradation by a glygoside hydrolase:

$$X = At^b \quad (1)$$

where t is time (h), X is degree of degradation (in percent) is defined as the amount of solubilized GlcNAc units divided by the amount of GlcNAc units in the chitin polymer at the start of the experiment, A is the activity of the added enzyme, and b is an intrinsic constant that quantifies the curvature of the time course profile [29]. A and b values were determined after fitting the data from chitin degradation reactions to equation (1) by use of Origin Pro 7.5 Software.

3. Results and discussion

Serratia marcescens is a soil bacterium that produces four chitin-depolymerizing enzymes: *Sm*ChiC, an endo-nonprocessive chitinase, *Sm*ChiA and *Sm*ChiB, two exo-processive chitinases moving along chitin chains in opposite directions, and *Sm*LPMO10A, a surface-active lytic polysaccharide monooxygenase, also known as CBP21, that introduces chain breaks by oxidative cleavage [30]. Previous studies have shown that aromatic amino acids in so-called substrate binding subsites, i.e. subsites interacting with the polymeric part of the chitin chain during processive action, are crucial for the processive ability of *Sm*ChiA and *Sm*ChiB [16,20].

In *Sm*ChiB, residues important for processivity include Trp^{97} (subsite +1), Trp^{220} (subsite +2), and Phe^{190} (subsite +3) (Fig. 1). In *Sm*ChiA, Trp^{167} in the substrate-binding subsite −3 is crucial for processivity, while processivity is also affected, albeit to a lesser extent byTrp^{275} and Phe^{396} in product binding subsites +1 and + 2, respectively, where dimeric products are released during processive hydrolysis [20]. Processivity implies that upon hydrolysis and release of the dimeric product the enzyme has a larger change of rebinding compared to diffusing away from the remaining substrate and the impact of the product binding subsites in *Sm*ChiA is likely due to promoting rebinding [18,20].

Experimental assessment of processivity may be done using several methods but is not straightforward (see Horn et al. for further discussion [33]). The apparent processive ability (P^{app}) provides a quantitative parameter and is defined as the average number of catalytic acts that an enzyme performs per one initiation of a processive run, which may be derived from various types of experiments, each with potential limitations. In Fig. 2, we have illustated how P^{app} can be determined from determination of the ratio of dimeric products vs. monomeric products, which is a “classical” method for processive GH action and used to for *Sm*ChiB-F190A in this study [33]. Regardless of these limitations, especially when it comes to quantification, available previous and newly generated data (Table 2) clearly show that all single mutants used in this study exhibit reduced processivity compared to the wild types.

Assessing kinetics of GH catalyzed hydrolysis of recalcitrant polysaccharides are not trivial [34]. It is very common to observe nonlinear behavior at a low degree of substrate conversion [19,29]. Different enzyme- and substrate-related factors can be responsible for the rapid decrease in hydrolysis rates such as the substrate being heterogeneous resulting in different parts being degraded faster than others, and that GHs tend to get stuck on the surface [17,34,35]. Kostylev and Wilson have developed a simple two-parameter equation (Equation (1)) to tackle this. One of the parameters is a total activity coefficient and the other is an intrinsic constant that reflects the ability of the GHs to overcome the varying degree of substrate recalcitrance. This properly addresses the observed nonlinearities and allows for the direct comparison between different GH actions on the same substrate [29]. Since all wild types and mutated variants showed nonlinear rates in the presences of *Sm*LPMO10A, we opted to use Equation (1) in our kinetic analysis. It is proposed that LPMO action create new chain ends, and by

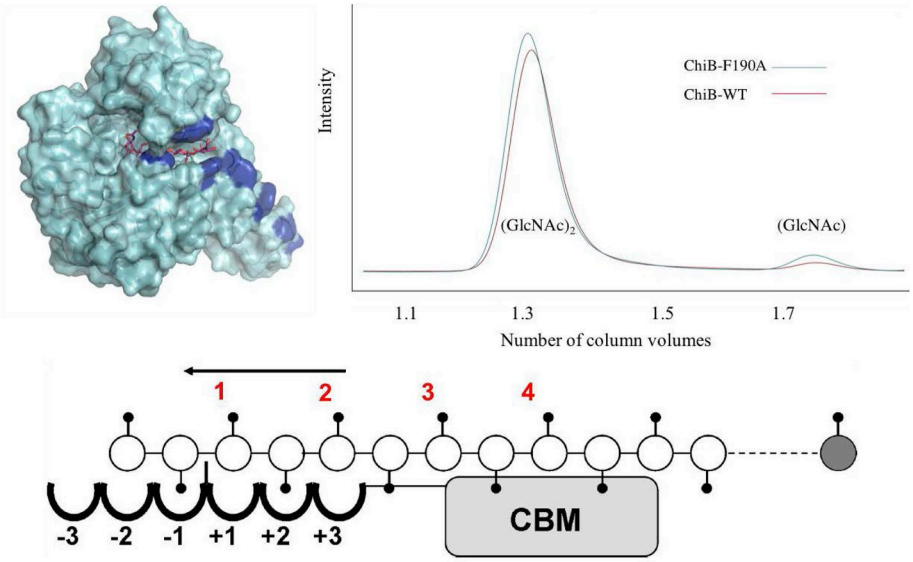

Fig. 2. Crystal structure of *Sm*ChiB (top left) and a schematic picture of *Sm*ChiB in complex with a single chitin chain (bottom). Surface exposed aromatic amino acids stacking with sugar moieties (being individual subsites) are highlighted in blue. The glycosidic bond between the sugar residues in subsite −1 and +1 is enzymatically cleaved. A correctly positioned *N*-acetyl group (shown as sticks) in the −1 subsite is essential for the substrate-assisted catalysis. Due to that the smallest structural unit of chitin is a disaccharide, the product of repeated processive enzymatic actions will be dimers, $(GlcNAc)_2$. Monomers, GlcNAc, originate from initial productive binding when the sugar in the non-reducing end occupies a subsite with an odd number. Hence, a high ratio of $[(GlcNAc)_2]/[GlcNAc]$ indicates a high degree of apparent processivity. A chromatogram (top right) of $(GlcNAc)_2$ and GlcNAc resulting from chitin degradation (4%) is shown for *Sm*ChiB-F190A (red line) and compared to that of *Sm*ChiB-WT (blue line). *Sm*ChiA-F190A is less processive (P^{app} = 11.6) than *Sm*ChiB-WT (P^{app} = 24.3) thus producing higher concentrations of GlcNAc than the wild type. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

this will increase the heterogeneity of the substrate. All *Sm*ChiB mutated variants showed activities (A values) amounting to approximately 15% of the wild type activity (Table 2 and Fig. 3). As expected, the addition of *Sm*LPMO10A resulted in an increase (13-fold) in *Sm*ChiB activity. Remarkably, the effect of *Sm*LPMO10A on less processive *Sm*ChiB variants was much more pronounced. For example, the activity of *Sm*ChiB-W97A was increased 55-fold and in the presence of *Sm*LPMO10A the activity was 65% of the wild type, as opposed to 16% in the absence of *Sm*LPMO10A. Likewise, addition of *Sm*LPMO10A yielded a 68-fold increase in the activity of *Sm*ChiB-F190A and the activity was equal to that of the wild type enzyme. Interestingly, chitin hydrolysis by *Sm*ChiB-W220A became 160-fold faster in the presence of *Sm*LPMO10A making this mutant 1.55-fold more active than the wild type enzyme. In the presence of *Sm*LPMO10A, each of the double mutants displayed activities similar to the wild type enzyme.

Interestingly, the presentence of *Sm*LPMO10A resulted in less activation of *Sm*ChiA compared to ChiB (7-fold vs. 13-fold), in line with previous results [3]. Also for *Sm*ChiA, the presence of *Sm*LPMO10A caused rate enhancements for the less processive mutants that were larger compared to the wild-type, but to a lesser degree than for the *Sm*ChiB mutants (30- and 12-fold increase for *Sm*ChiA-W167A and *Sm*ChiA-275A, respectively). Furthermore, in the presence of *Sm*LPMO10A the activities of the less processive mutants only reached 66% and 48%, respectively, of the wild type activity (as compared to up to 155% for *Sm*ChiB-W220A).

From these results, it is clear that the need for processivity in enzymatic degradation of polysaccharides may be reduced when the crystalline substrate is treated with an LPMO. The trends are the same for all less processive variants, but the clearest example is provided by *Sm*ChiB-W220A, which in reactions with *Sm*LPMO10A displayed a 55% increase in activity compared to the wild type under identical conditions. Crystallographic studies (Fig. 1) show that Trp^{220} interacts closely with the substrate, thus likely making a major contribution to keeping the enzyme closely associated to the substrate in between subsequent hydrolytic reactions and preventing once-detached single chains from re-associating with the insoluble material [16,36]. Having access to numerous new chain ends provided by the LPMO liekly reduces the beneficial effect of the "stickiness" provided by Trp220 (and other aromatic residues near the catalytic center). This reasoning is in line with the notion that the rate-determining step for chitinase catalyzed hydrolysis changes from substrate association, for chitin, to

Table 2
Activity on β-chitin in the absence and presence of *Sm*LPMO10A for *Sm*ChiA and *Sm*ChiB and mutants investigated in this study.

Enzyme	*A*	*b*	P^{app}, [a]	P^{app}, [b]
SmChiA				
WT	0.32 ± 0.06	0.65 ± 0.13	30.1[c]	36[d]
W167A*[e]	0.05 ± 0.02	0.90 ± 0.25	*n.a.*[f]	16[d]
W275A*	0.06 ± 0.02	0.62 ± 0.21	*n.a.*[f]	14[d]
SmChiA with SmLPMO10A				
WT	2.29 ± 0.28	0.51 ± 0.08		
W167A	1.51 ± 0.08	0.38 ± 0.04		
W275A	1.09 ± 0.14	0.75 ± 0.07		
SmChiB				
WT	0.31 ± 0.04	0.73 ± 0.1	24.3[c]	*n.d.*[g]
W97A*	0.05 ± 0.01	0.96 ± 0.1	11.0[c]	*n.d.*[g]
W220A*	0.04 ± 0.01	1.16 ± 0.2	9.8[h]	*n.d.*[g]
W97A/W220A	0.06 ± 0.01	0.54 ± 0.1	*n.d.*[g]	*n.d.*[g]
F190A	0.06 ± 0.01	0.84 ± 0.1	11.6[h]	*n.d.*[g]
F190A/W220A	0.04 ± 0.01	0.78 ± 0.1	*n.d.*[g]	*n.d.*[g]
SmChiB with SmLPMO10A				
WT	4.16 ± 0.31	0.45 ± 0.05		
W97A	2.71 ± 0.23	0.40 ± 0.06		
W220A	6.48 ± 0.74	0.30 ± 0.08		
W97A/W220A	4.43 ± 0.33	0.23 ± 0.10		
F190A	4.06 ± 0.60	0.54 ± 0.09		
F190A/W220A	4.46 ± 0.20	0.37 ± 0.03		

[a] As determined on the basis of the $[(GlcNAc)_2]/[GlcNAc]$ product ratio during the initial phase of degradation of β-chitin [15].

[b] As determined from the plot of the concentration of enzyme-generated reducing groups *versus* insoluble reducing groups upon chitin degradation under single-hit conditions [18].

[c] Values obtained from Hamre et al. [15].

[d] Values obtained from Kurašin et al. [18].

[e] For mutants marked with a *, the loss of processivity has also been shown in experiments with chitosan; Se Horn et al. and Zakariassen et al. for more details [16,20].

[f] Not applicable. An assumption for the use of the $[(GlcNAc)_2]/[GlcNAc]$ product ratio upon degradation of β-chitin is that there should not be a significant different probability of endo-mode initiation between enzymes nor too high degree of endo-mode initiation. Since ChiA-W167A and ChiB-W275A display a high degree of endo-mode initiation that differs significantly from the wild type [18], this method is not used to assess P^{app}.

[g] Not determined.

[h] Determined in this work.

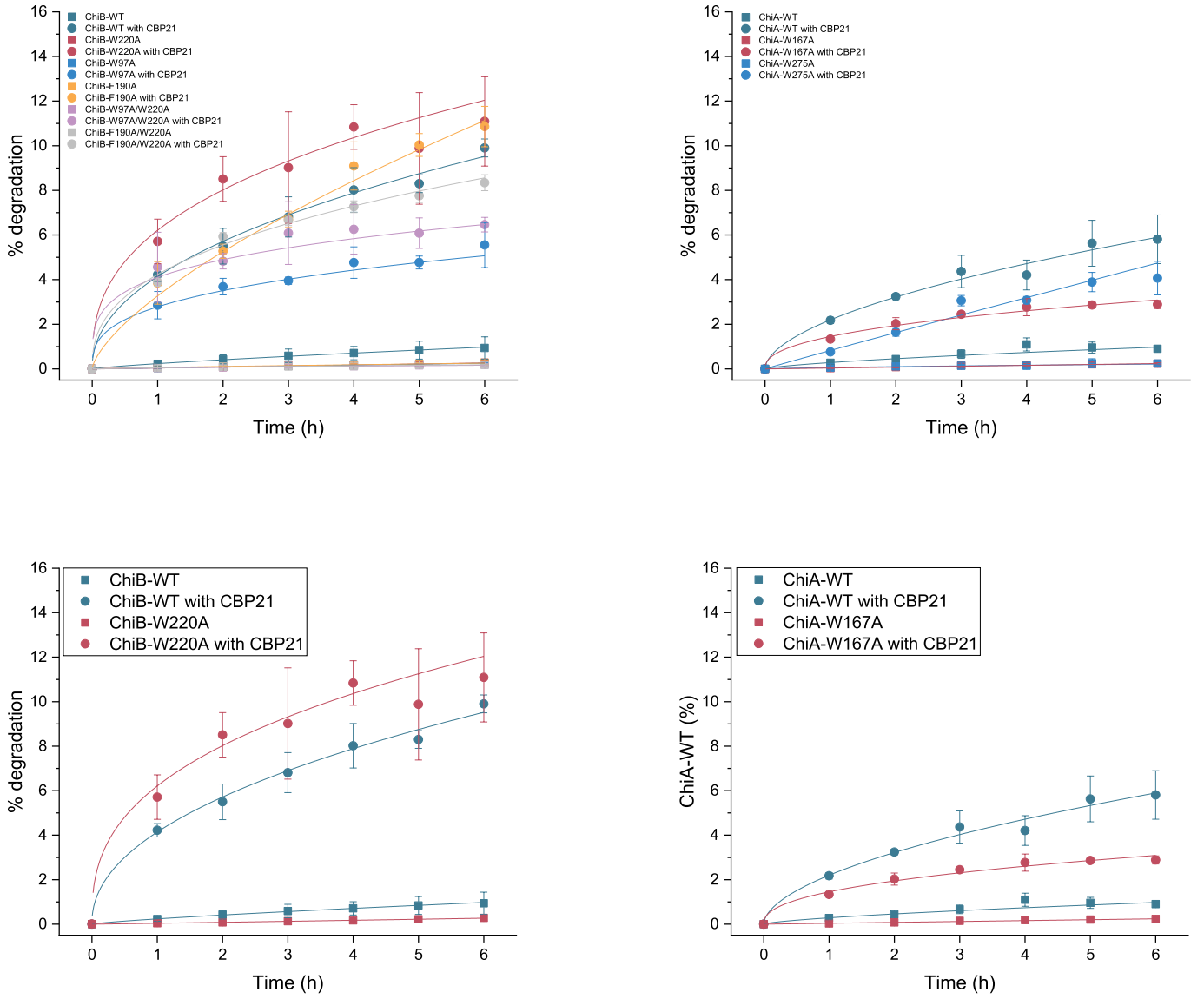

Fig. 3. Examples of progress curves fitted to Eq. (1). The top left panel shows data for *Sm*ChiB-WT (green), *Sm*ChiB-W220A (red), *Sm*ChiB-W97A (blue), *Sm*ChiB-F190A (orange), *Sm*ChiB-W97A/W220A (violet), and *Sm*ChiB-F190A/W220A (grey), with (circle) and without (square) *Sm*LPMO10A. The top right panel shows data for *Sm*ChiA-WT (green), *Sm*ChiB-W167A (red), and *Sm*ChiA-W275A (blue), with (circle) and without (square) *Sm*LPMO10A. The hydrolysis were performed with 100 nM enzyme in 50 mM sodium acetate pH 6.1 at 37 °C with 20 mg/ml chitin. The bottom left and right panels focus on the wild types of *Sm*ChiB and *Sm*ChiA, respectively, and the most active mutants, *Sm*ChiB-W220A and *Sm*ChiA-W167A. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

product displacement when the substrate becomes more available as in the soluble chitin variant chitosan [37].

It has been shown that Trp^{97} (subsite +1) and Phe^{190} (subsite +3) provide less binding free energy compared to Trp^{220} [11,38]. Moreover, While Trp^{220} seems only involved in substrate-binding and seems pre-ordered to do so in the apo-enzyme, Trp^{97} has an additional catalytic role since it takes part in stabilizing the energetically demanding $^4C_1$ to $^{1,4}B$ conformational change of the *N*-acetylglucosamine in the −1 subsite [31,39]. Phe^{190} shows a −91° rotation around χ_1 upon substrate binding [31], which may slightly offset the beneficial effect of the Phe-substrate interaction. These differences may underly the observed variations in the rate enhancements observed upon adding *Sm*LPMO10A.

Previous studies have shown that the initial rate of endo-non-processive *Sm*ChiC of *S. marcescens* is not enhanced by treatment with an LPMO, in line with the proposed main role of LPMOs to create new chain-ends for exo-acting GHs [3]. To this regard, it is interesting to observe that the rate of hydrolysis is more enhanced for *Sm*ChiB than for *Sm*ChiA. It has been suggested that, when acting on chitin, *Sm*ChiA is the most endo-active of the two GHs [26]. Moreover, the mutation of either Trp^{167} or Trp^{275} to Ala, changes the probability of endo mode-initiation of hydrolysis by *Sm*ChiA from 76% to 90% [18]. The increase in endo mode-initiation is one possible cause of the observed lower effect of *Sm*LPMO10A on the efficiency of the *Sm*ChiA mutants compared to the *Sm*ChiB mutants.

The advent of LPMOs has already led to considerable improvements of enzyme cocktails for biomass processing [8,40], while recent results related to the nature of the co-substrate (O_2 or H_2O_2) and enzyme stability suggest that the potential of LPMOs has not yet been fully harnessed [41–43]. So far, the interplay between LPMOs and individual GHs with varying functionalities (endo-, exo, processive) has not been studies in much detail. The present data show that the inclusion of LPMOs in enzyme cocktails may affect the need for certain GH types, in this case the highly “sticky” exoprocessive enzymes often referred to as

chito- or cello-biohydrolases. Thus, harnessing the full potential of LPMOs not only requires selection of the best LPMOs and LPMO-friendly processing conditions, but also a reconsideration of the glycoside hydrolase composition of polysaccharide degrading enzyme cocktails.

Acknowledgments

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.carres.2019.01.001.

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