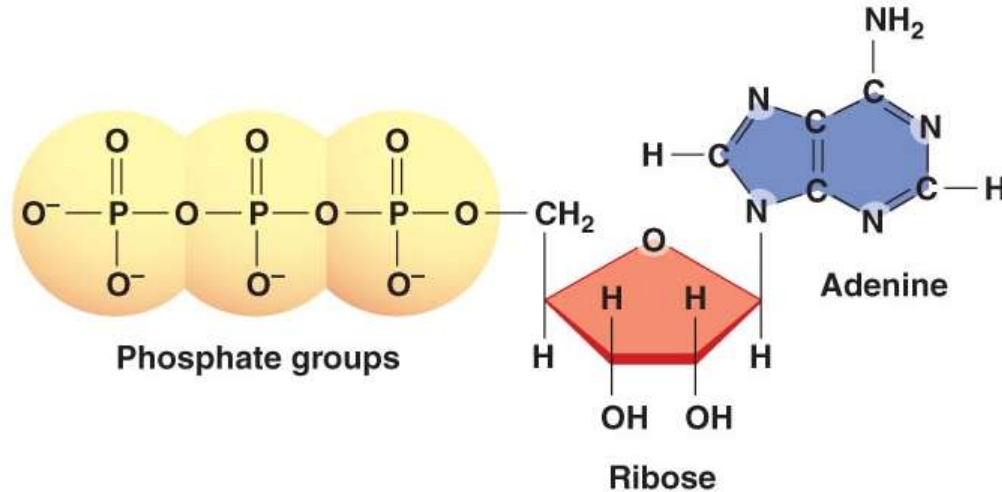


ATP: Energy Source of the cell

ATP:

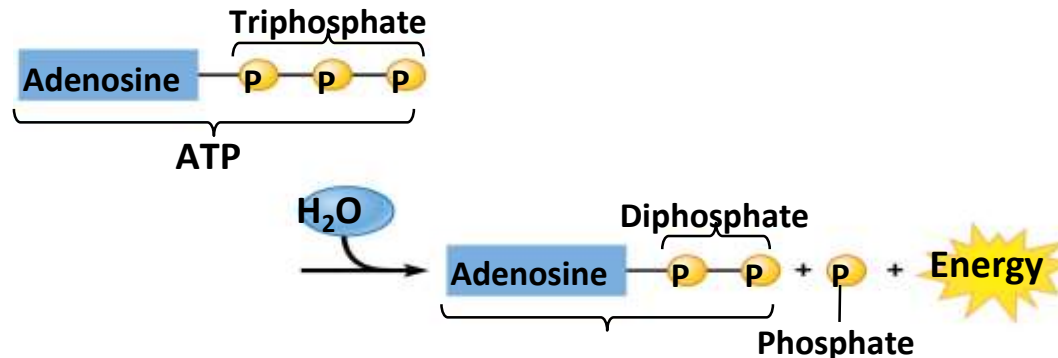
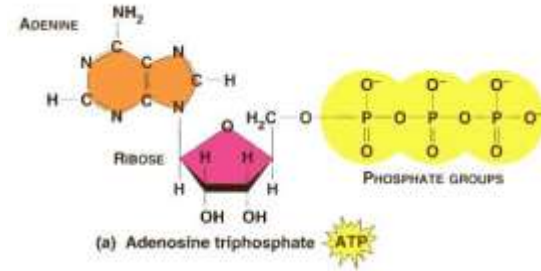
Energy Source of the cell

- What is ATP?



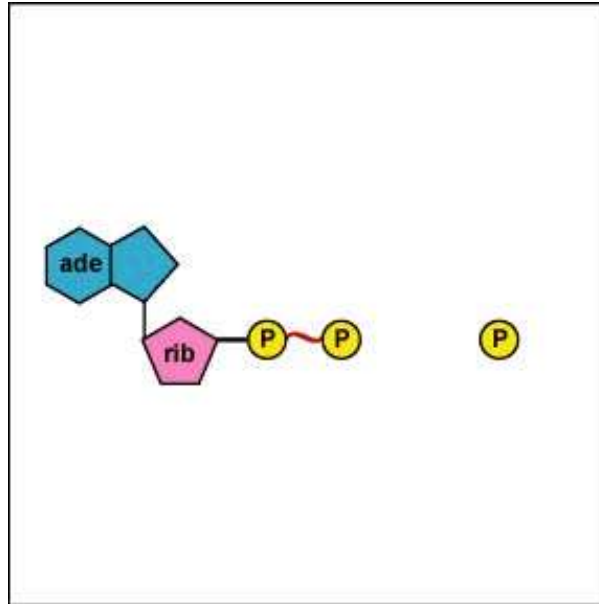
ATP drives cellular work by coupling exergonic and endergonic reactions

- **ATP**, adenosine triphosphate, powers nearly all forms of cellular work.
- ATP consists of adenosine and a triphosphate tail (3 phosphates)
- hydrolysis of ATP releases energy by transferring its third phosphate from ATP to some other molecule in a process called phosphorylation.



ATP drives cellular work by coupling exergonic and endergonic reactions

- A cell uses and regenerates ATP continuously.
- In the ATP cycle, energy released in an exergonic reaction, such as the breakdown of glucose during cellular respiration, is used in an endergonic reaction to generate ATP from ADP.



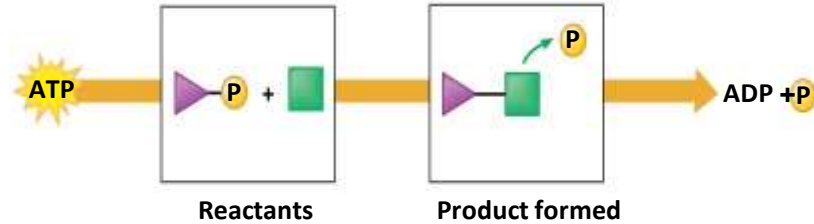
ATP drives cellular work by coupling exergonic and endergonic reactions

There are three main types of cellular work:

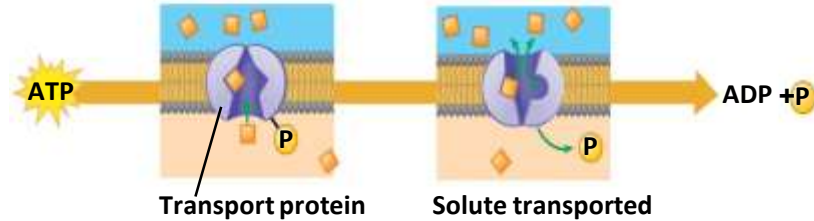
1. chemical
2. transport
3. mechanical

- ATP drives all three of these types of work.

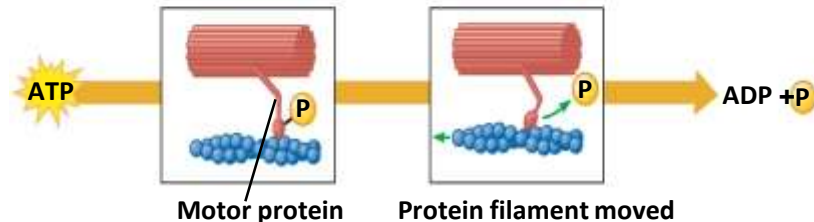
Chemical work



Transport work



Mechanical work



Adenosine triphosphate [ATP]

Phosphate Groups



Ribose

Adenine

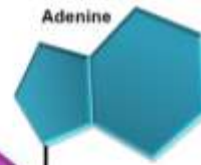


Phosphate Groups



Ribose

Adenine

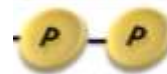


Energy Released for cell metabolism

P

Adenosine diphosphate
(ADP)

Phosphate Groups



Ribose

Adenine

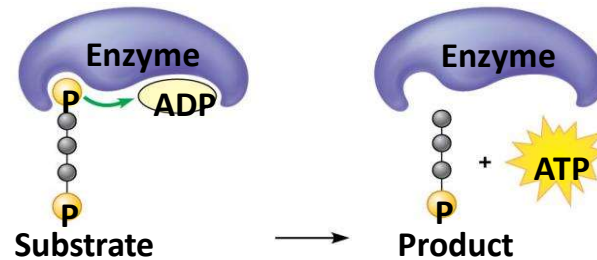


CELLULAR RESPIRATION
RECHARGES ADP

P

Substrate Level Phosphorylation

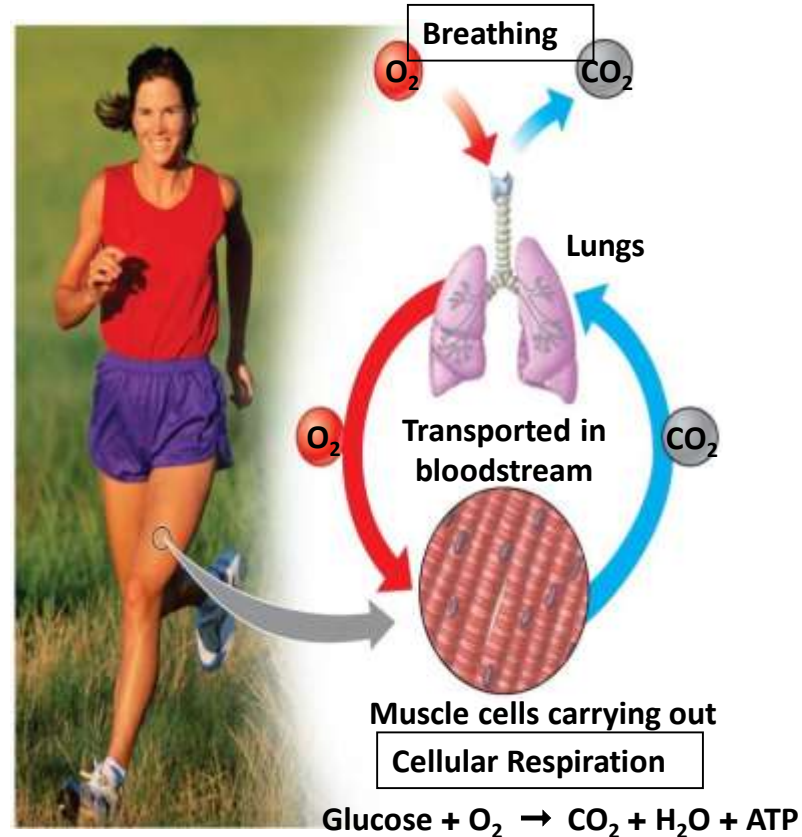
- ATP is formed during cellular respiration, but also can be formed by **substrate-level phosphorylation** during which an **enzyme** transfers a phosphate group from a **substrate** molecule to ADP and ATP is formed.



Cellular Respiration: Aerobic Harvesting of Energy

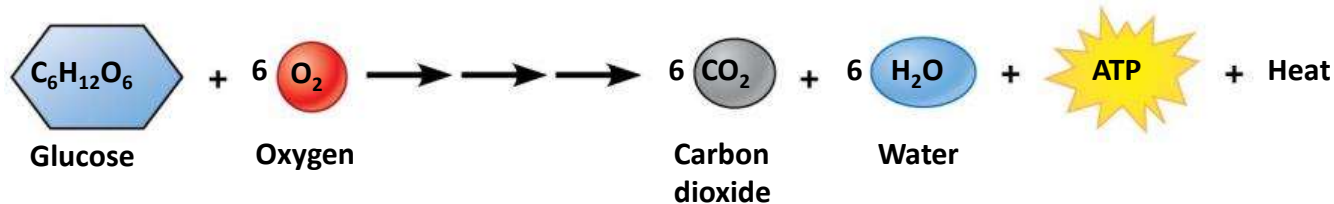
Breathing supplies O₂ and removes CO₂ in cellular respiration

- Respiration,(as it relates to breathing), and cellular respiration are not the same.
- Respiration, in the breathing sense, refers to an exchange of gases. Usually an organism brings in oxygen from the environment and releases waste CO₂.
- **Cellular respiration** is the aerobic (oxygen-requiring) harvesting of energy from food molecules by cells.



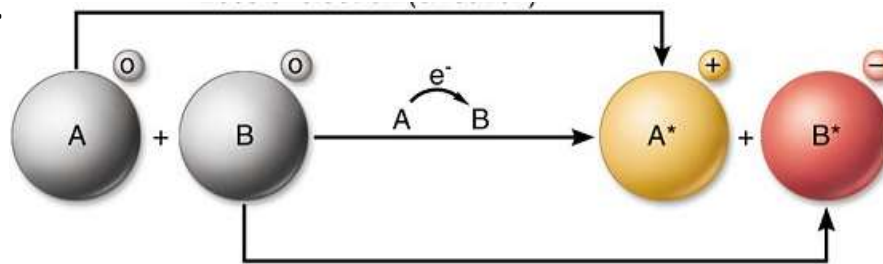
Cellular respiration banks energy in ATP molecules

- Cellular respiration is an exergonic (energy- releasing) process that transfers energy from the bonds in glucose to form ATP.
- Cellular respiration can produce up to 32 ATP molecules for one glucose molecule and use about 34% of the energy originally stored in glucose, releasing the other 66% as heat
- This energy conversion efficiency is better than most energy conversion systems compared to only **25% of the energy** in gasoline produces the kinetic energy of movement.



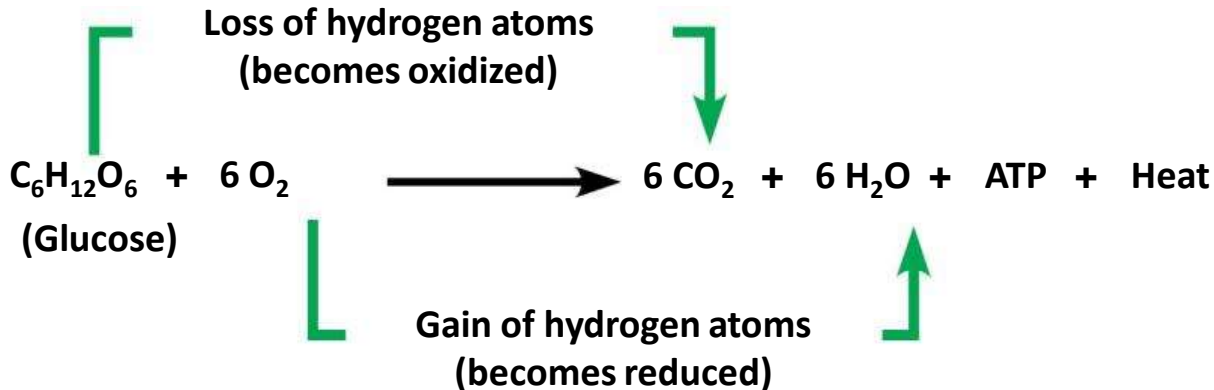
Cells capture energy from electrons “falling” from organic fuels to oxygen

- **Oxidation-reduction reaction, or redox reaction.**
 - The movement of **electrons** from one molecule to another.
 - the loss of electrons from one substance is called **oxidation** and the **addition** of electrons to another substance is called **reduction**
 - a molecule is **oxidized** when it loses one or more electrons, and a molecule is **reduced** when it gains one or more electrons.



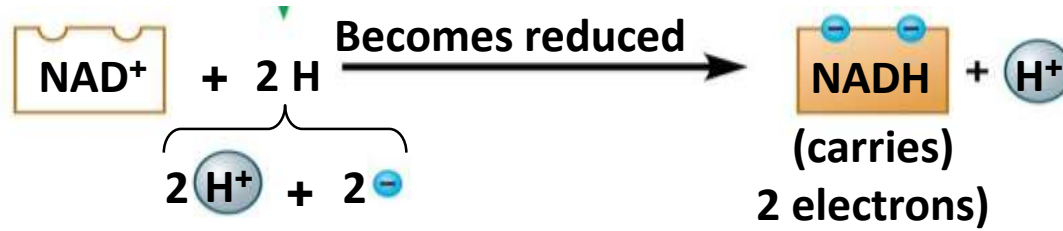
Cells capture energy from electrons “falling” from organic fuels to oxygen

- A cellular respiration equation is helpful to show the changes in hydrogen atom distribution.
 - Glucose **loses** its hydrogen atoms and becomes **oxidized** to CO_2 .
 - Oxygen **gains** hydrogen atoms and becomes **reduced** to H_2O .



Cells capture energy from electrons “falling” from organic fuels to oxygen

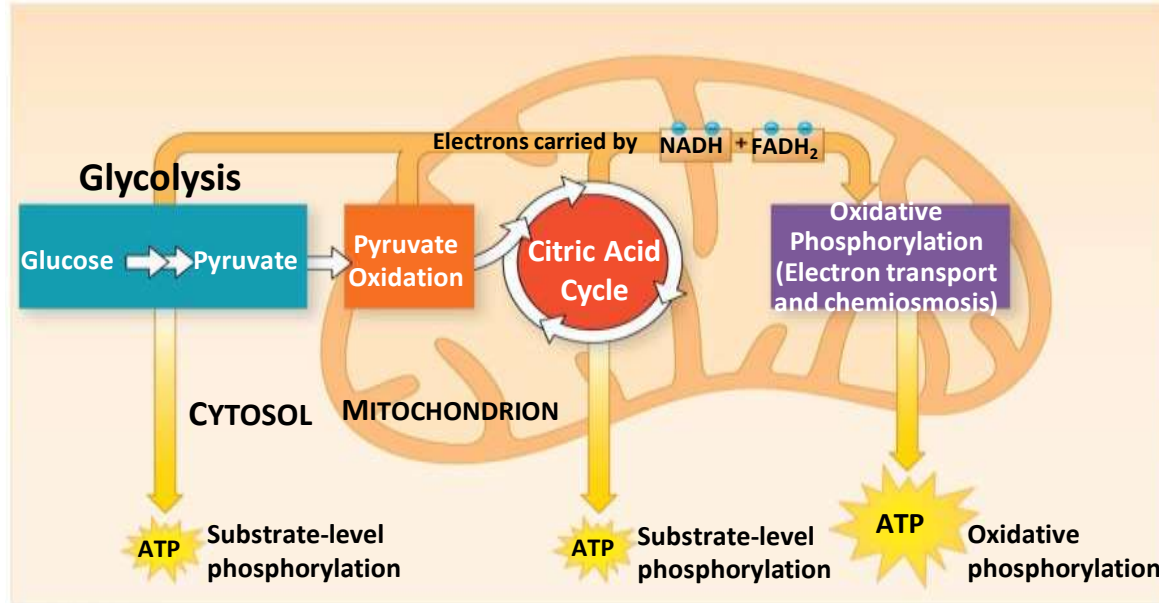
- An important player in the process of oxidizing glucose is a coenzyme called **NAD⁺**, which accepts electrons and becomes reduced to **NADH**. (electron acceptors)



Stages of Cellular Respiration

Overview: Cellular respiration occurs in three main stages

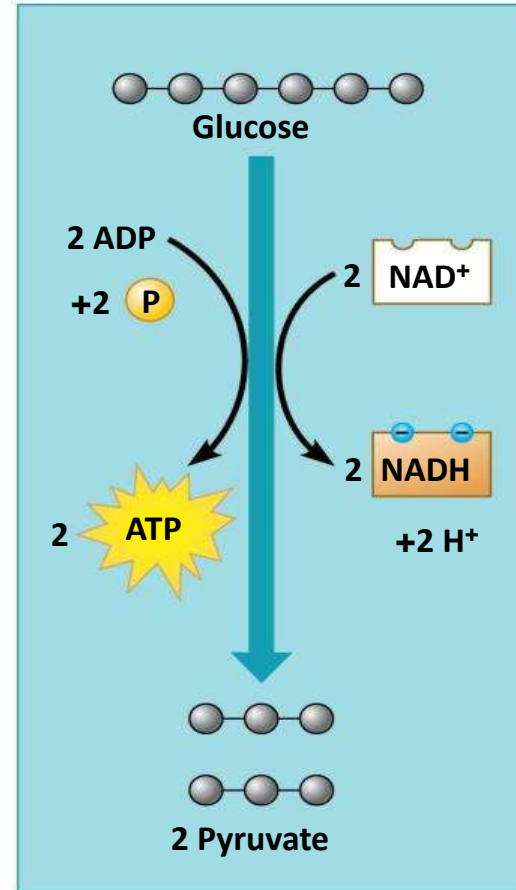
- Cellular respiration consists of a sequence of steps that can be divided into three stages.
 - Stage 1: Glycolysis (occurs in cytosol)
 - Stage 2: Pyruvate oxidation and the citric acid cycle
 - Stage 3: Oxidative phosphorylation or ETC
- } occurs in mitochondria



Glycolysis harvests chemical energy by oxidizing glucose to pyruvate

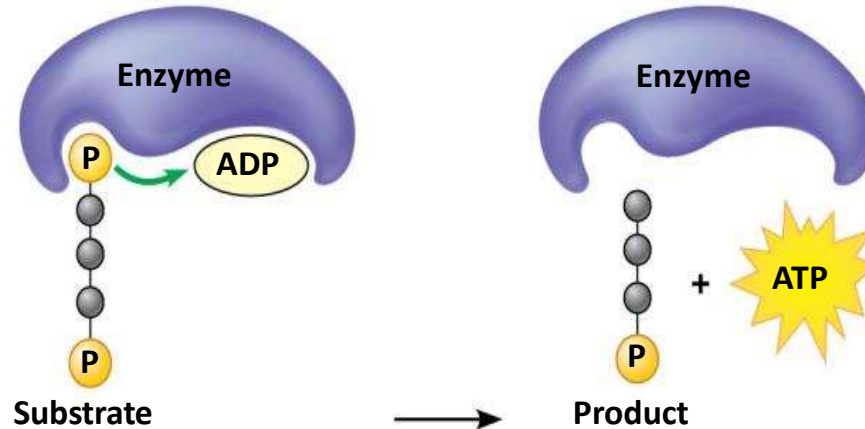
In **glycolysis**:

- a single molecule of glucose is enzymatically split in half through a series of steps
- two molecules of pyruvate are produced
- two molecules of NAD^+ are reduced to two molecules of NADH
- there is a NET gain of two molecules of ATP (produced by substrate level phosphorylation).



Glycolysis harvests chemical energy by oxidizing glucose to pyruvate

ATP is formed in glycolysis by **substrate-level phosphorylation** during which an enzyme transfers a phosphate group from a substrate molecule to ADP and ATP is formed.



Glycolysis harvests chemical energy by oxidizing glucose to pyruvate

- The steps of glycolysis have two main phases.
 - the energy investment phase and the energy payoff phase



1. Phosphorylation of glucose by ATP.

2-3. Rearrangement, followed by a second ATP phosphorylation.

4-5. The 6-carbon molecule is split into two 3-carbon molecules—one G3P, another that is converted into G3P in another reaction.

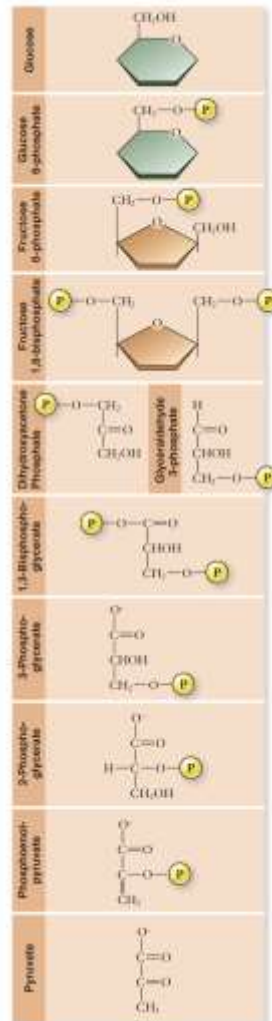
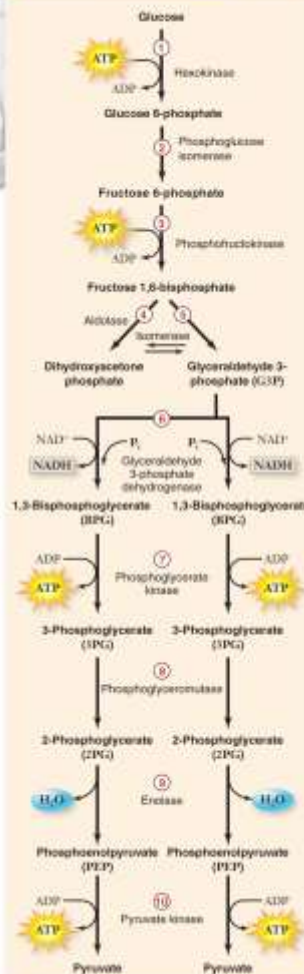
6. Oxidation followed by phosphorylation produces two NADH molecules and two molecules of BPG, each with one high-energy phosphate bond.

7. Removal of high-energy phosphate by two ADP molecules produces two ATP molecules and leaves two PPG molecules.

8-9. Removal of water yields two PEP molecules, each with a high-energy phosphate bond.

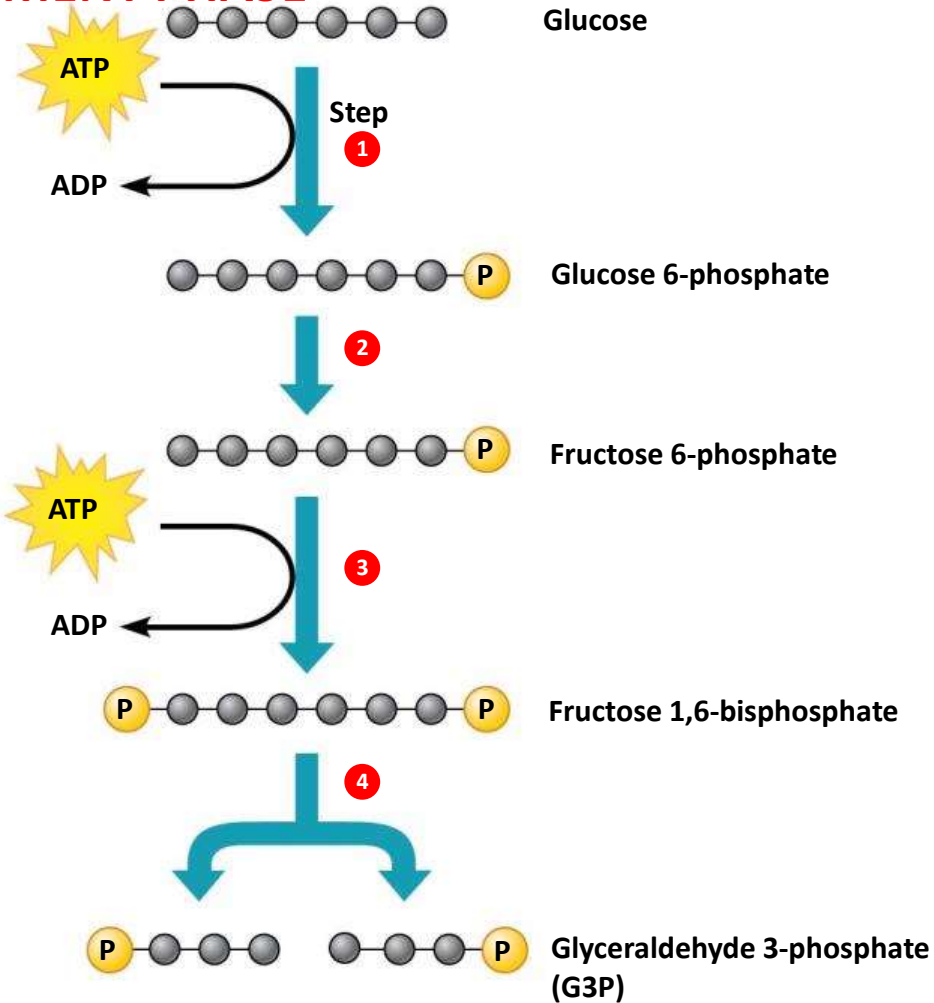
10. Removal of high-energy phosphate by two ADP molecules produces two ATP molecules and two pyruvate molecules.

Glycolysis: The Reactions



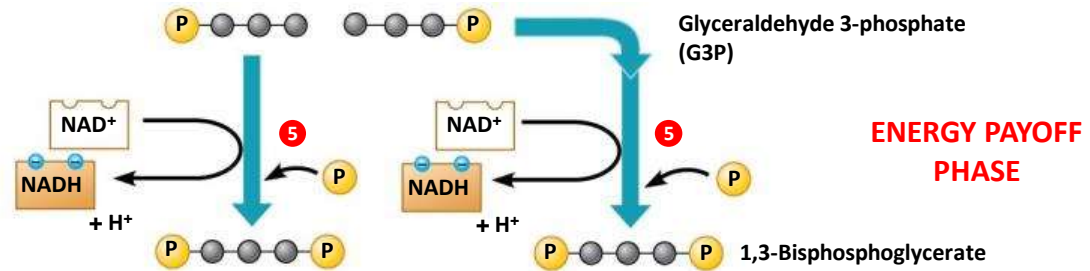
ENERGY INVESTMENT PHASE

Steps 1 – 3 Glucose is energized, using ATP.



Step 4 A six-carbon intermediate splits into two three-carbon intermediates.

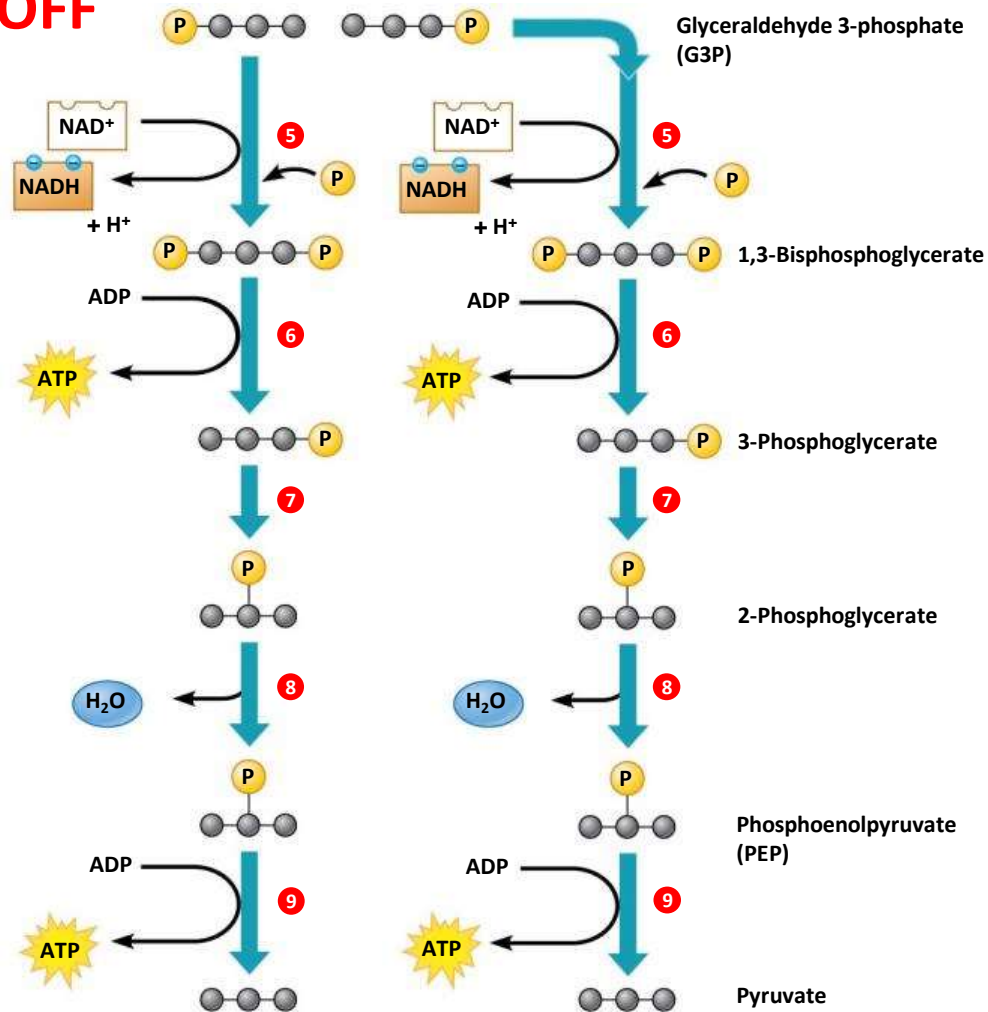
Step 5 A redox reaction generates NADH.



ENERGY PAYOFF PHASE

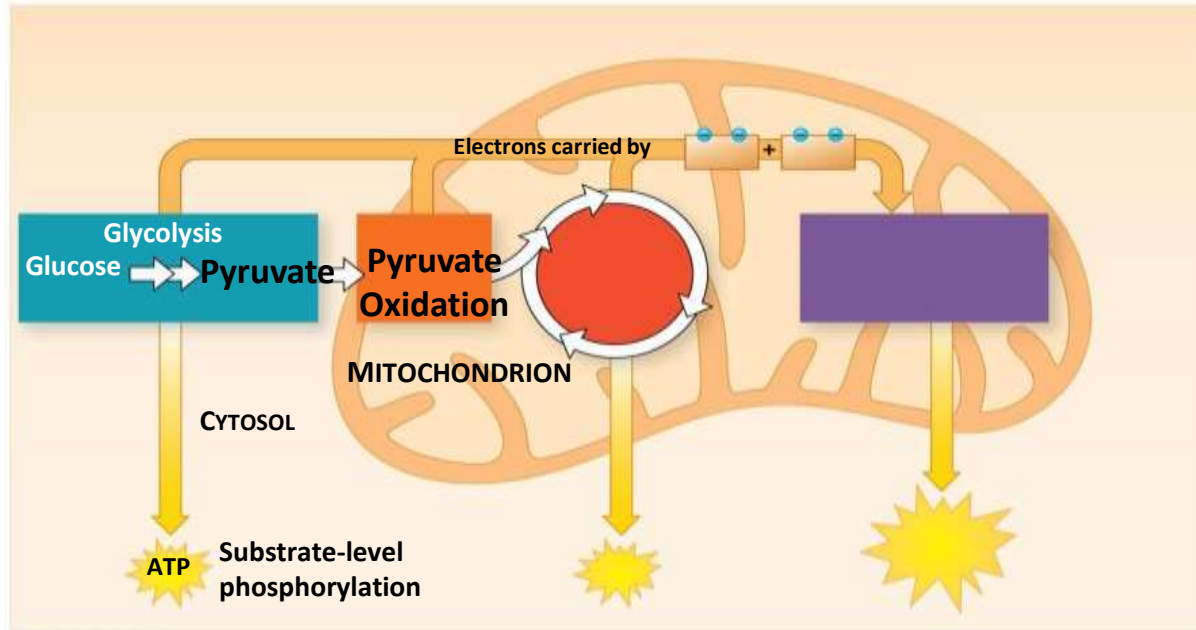
Step 5 A redox reaction generates NADH.

Steps 6 – 9 ATP and pyruvate are produced.

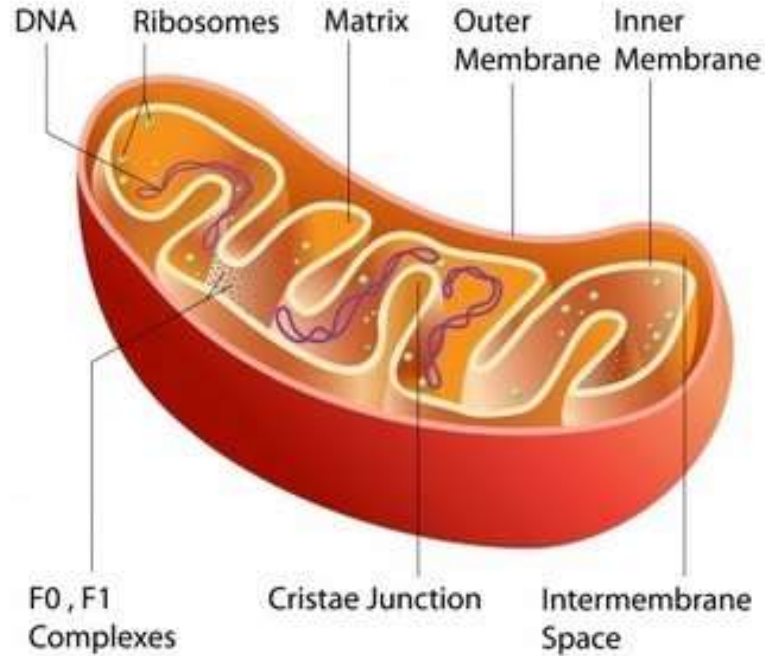


Pyruvate is oxidized in preparation for the citric acid cycle

- At the finish of glycolysis, 90% of the stored energy still remains in the pyruvate molecules
- Pyruvate molecules are then transported from the cytosol into a mitochondria where the citric acid cycle and oxidative phosphorylation will occur.



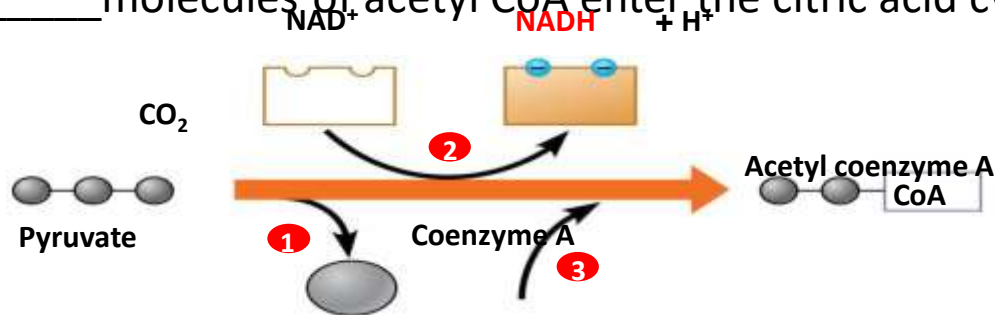
Anatomy of Mitochondria



Pyruvate is oxidized in preparation for the citric acid cycle

Pyruvate does not enter the citric acid cycle but undergoes some chemical grooming in which

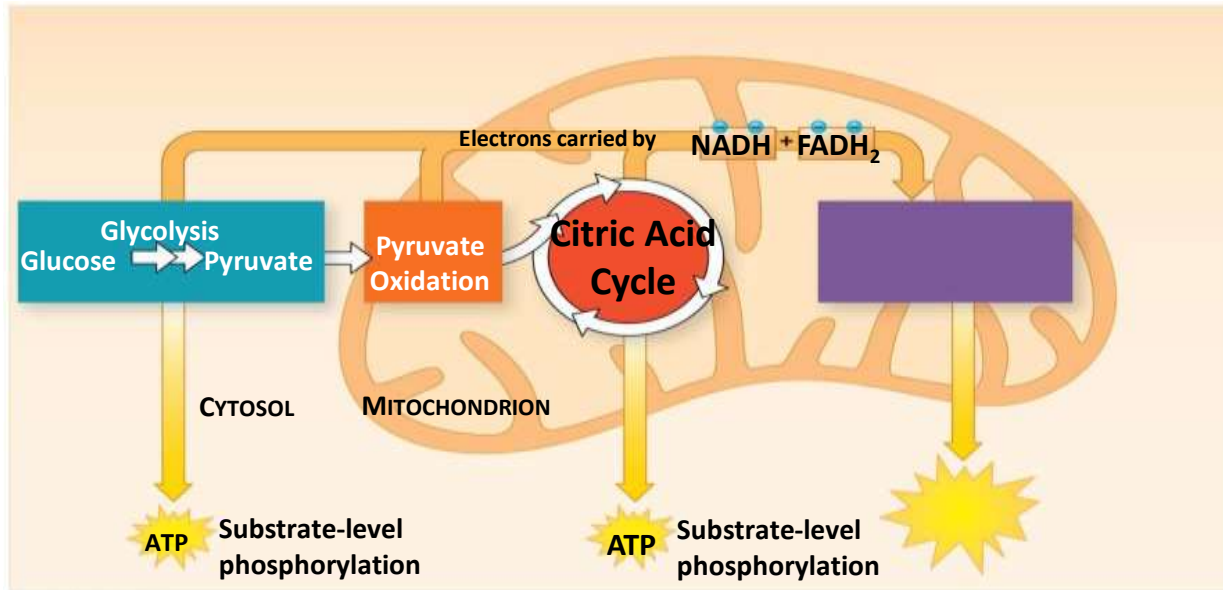
- a carboxyl group is removed and given off as CO_2
- the two-carbon compound remaining is oxidized while a molecule of NAD^+ is reduced to $\text{NADH} + \text{H}^+$
- coenzyme A joins with the two-carbon group to form acetyl coenzyme A, abbreviated as **acetyl CoA**.
- Then two molecules of acetyl CoA enter the citric acid cycle.

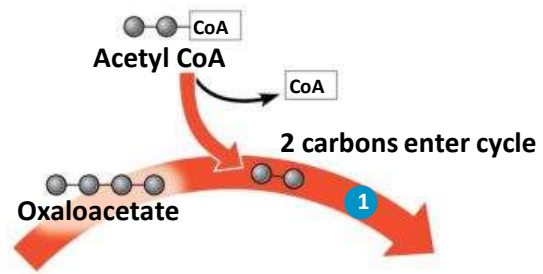


The citric acid cycle completes oxidation of organic molecules

The citric acid cycle (also called the Krebs cycle after the German-British researcher Hans Krebs) completes the oxidation of organic molecules, and generates many NADH and FADH₂ molecules.

- Occurs in the Mitochondria

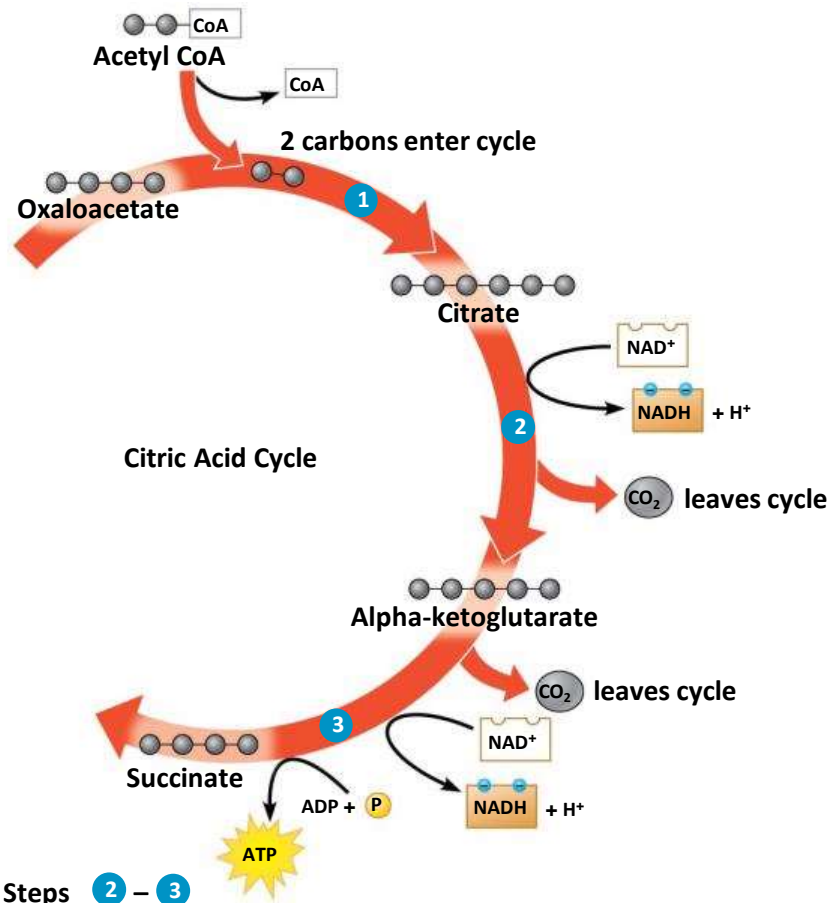




Citric Acid Cycle

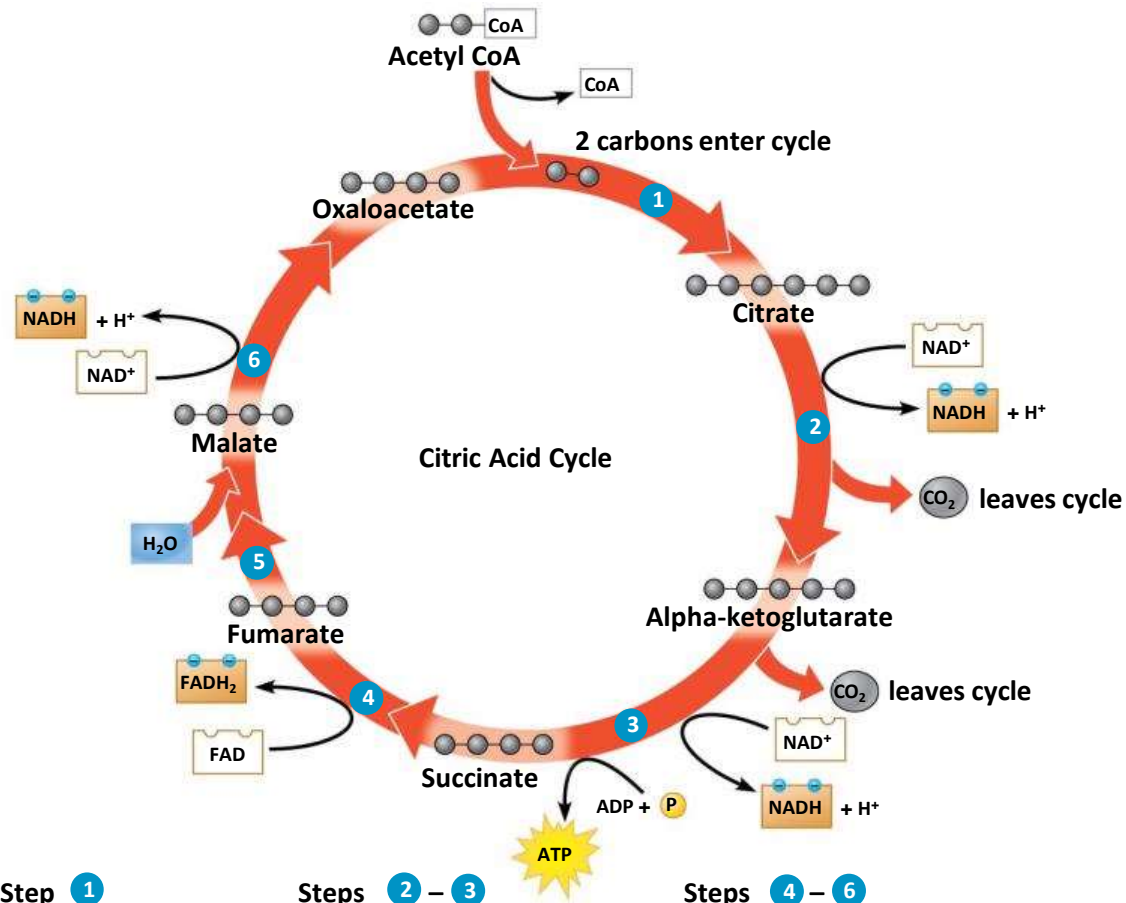
Step 1

Acetyl CoA stokes
the furnace.



Step 1
Acetyl CoA stokes
the furnace.

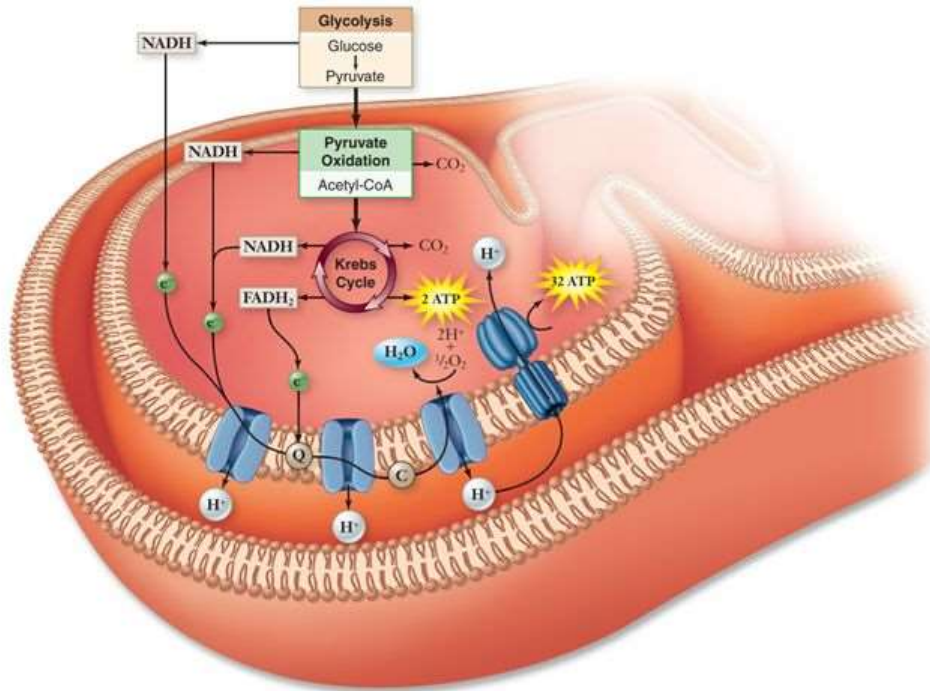
Steps 2 – 3
 NADH , ATP , and CO_2
are generated during
redox reactions.

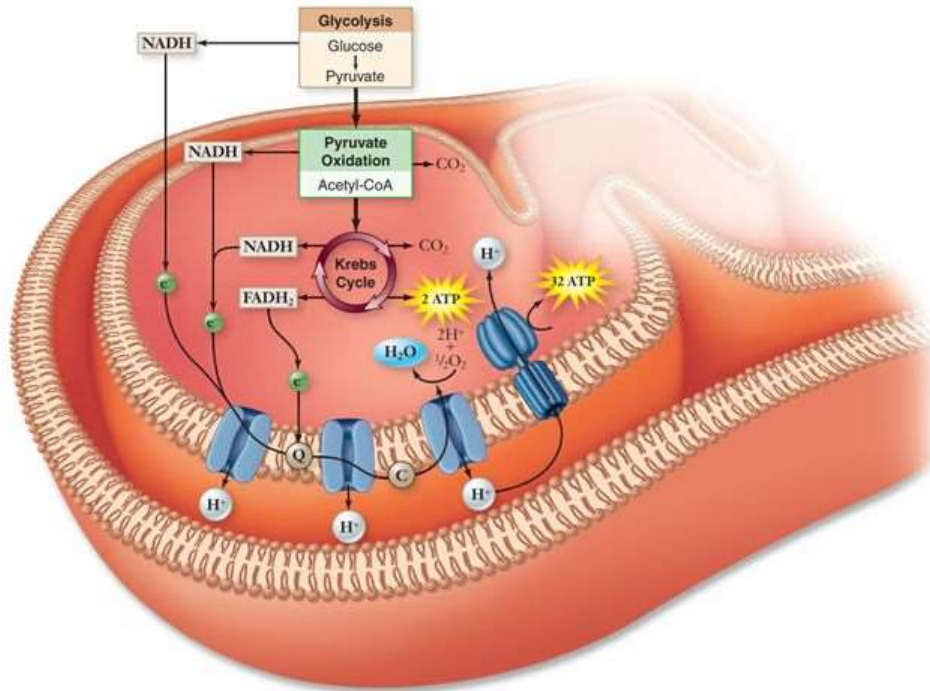


Step 1
Acetyl CoA stokes the furnace.

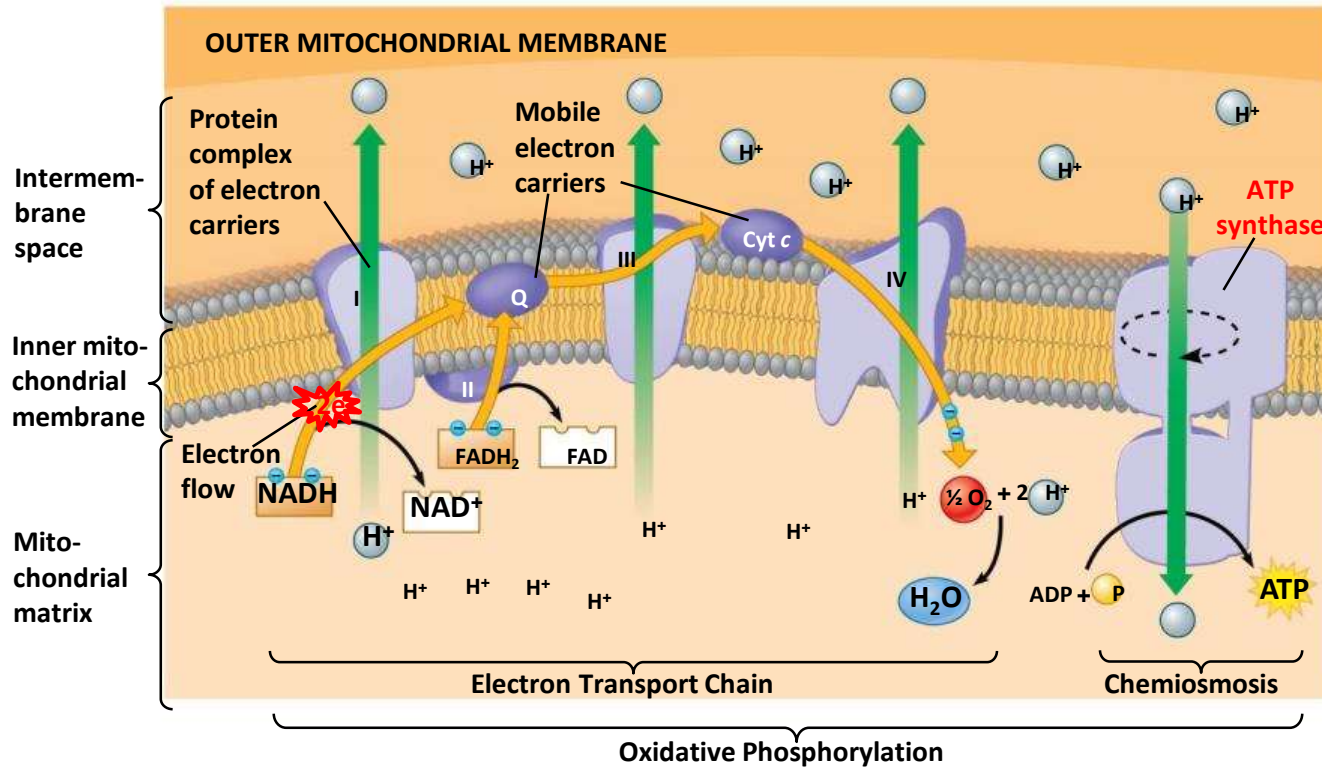
Steps 2 – 3
NADH, ATP, and CO₂ are generated during redox reactions.

Steps 4 – 6
Further redox reactions generate FADH₂ and more NADH.

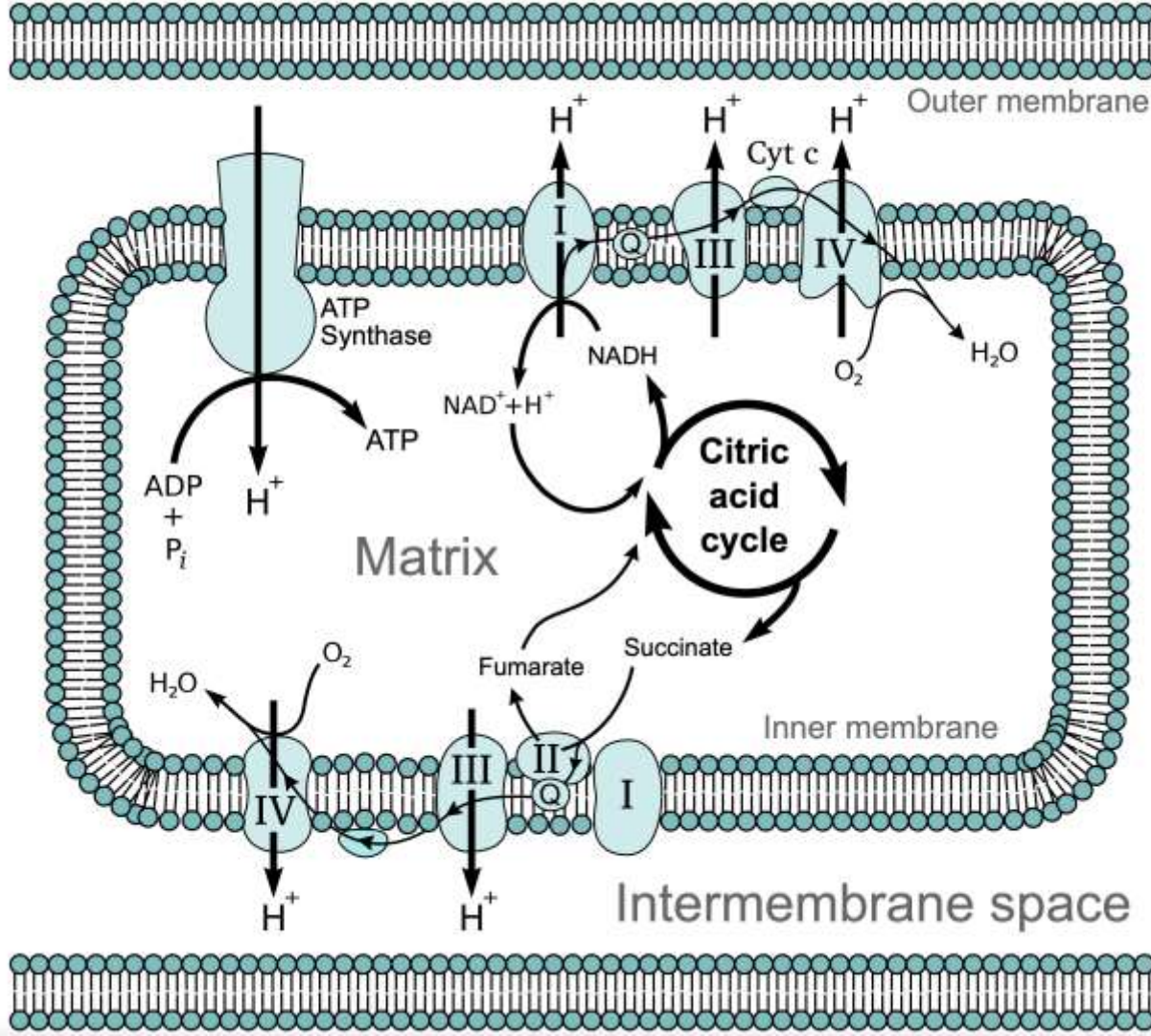




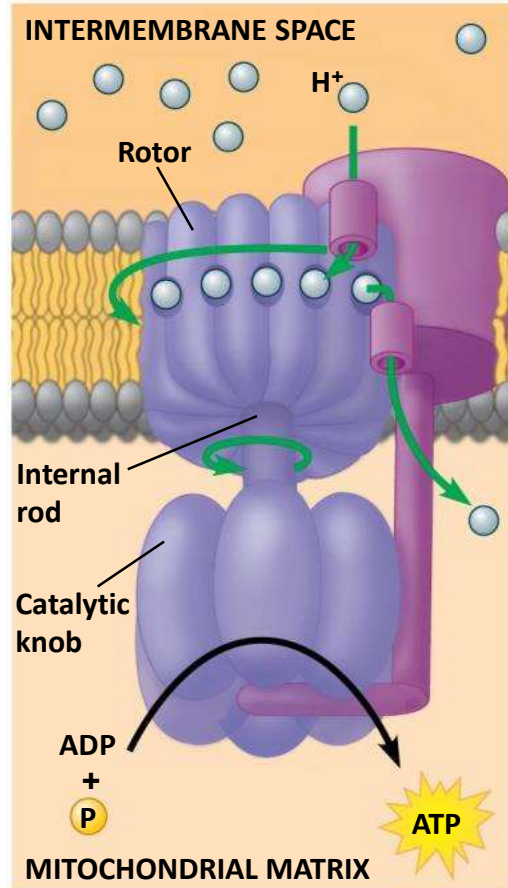
Most ATP production occurs by oxidative phosphorylation (The Electron Transport Chain)



The Electron Transport Chain



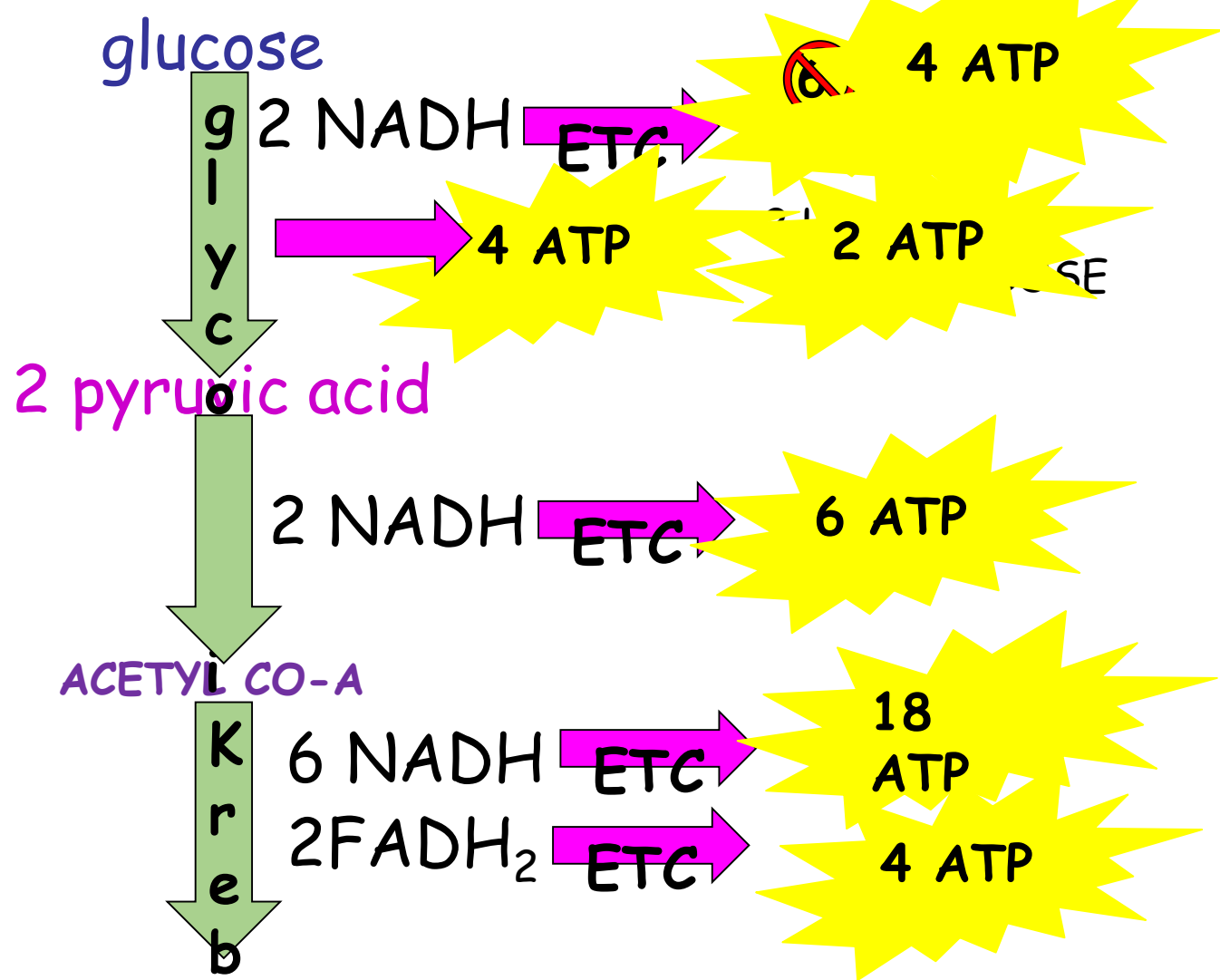
Most ATP production occurs by oxidative phosphorylation (The Electron Transport Chain)



Most ATP production occurs by oxidative phosphorylation (The Electron Transport Chain)

Thus, after glycolysis and the citric acid cycle, the cell has gained

- 4 ATP
- 10 NADH
- 2 FADH₂
- To harvest the energy banked in NADH and FADH₂, these molecules must shuttle their high-energy _____ to an electron transport chain for oxidative phosphorylation
electrons
- This occurs in the _____ of mitochondria
membranes

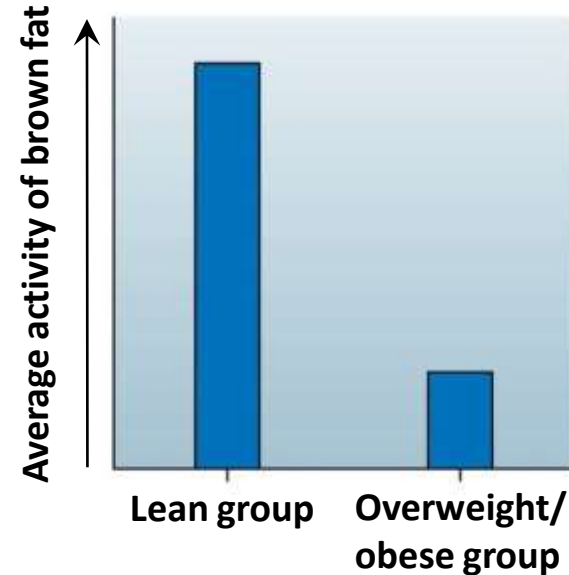


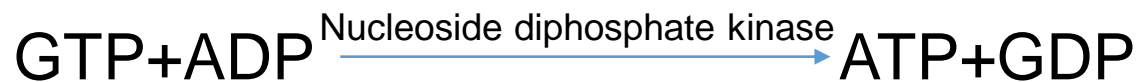
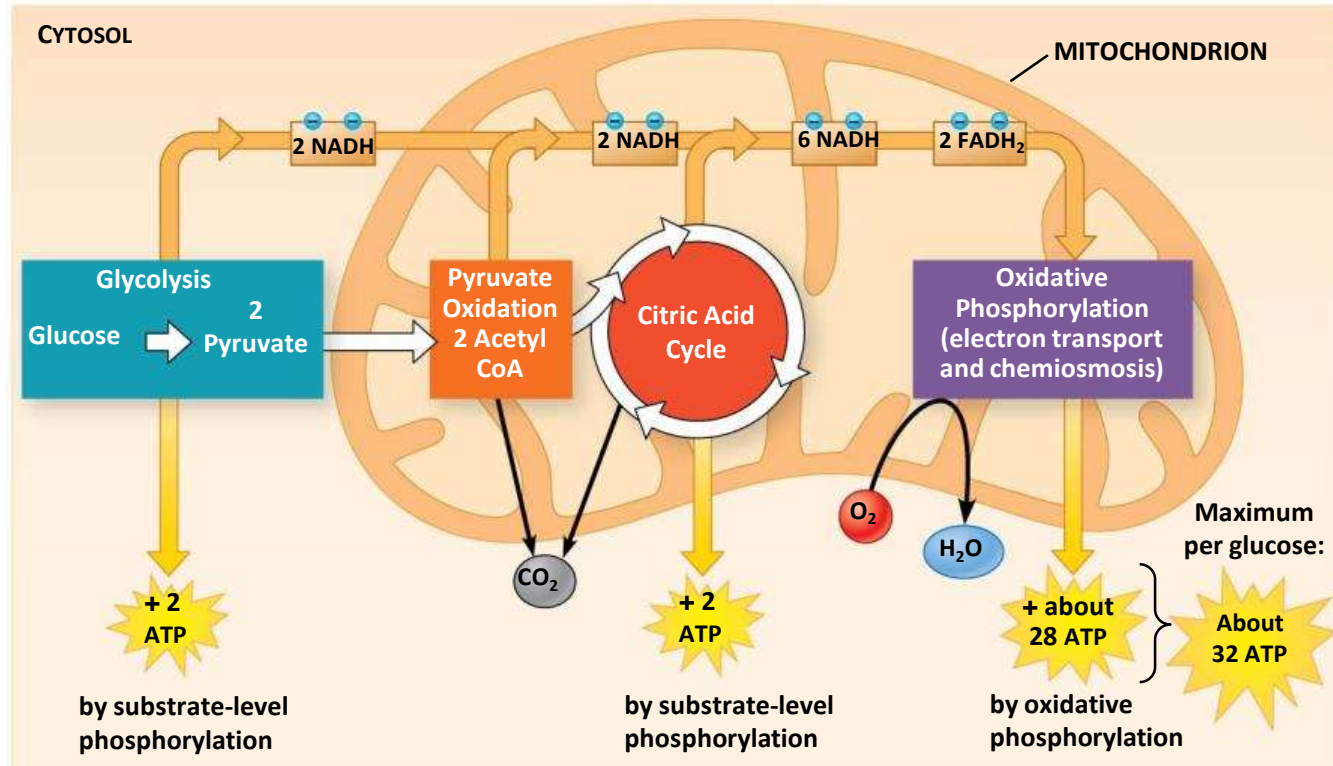
Overview: Oxidative phosphorylation

- The final stage of cellular respiration is oxidative phosphorylation (ETC), which involves electron transport and chemiosmosis and requires an adequate supply of oxygen.
- The electron transport chain is a series of proteins (electron carriers) built into a membrane that accept electrons from high energy NADH and FADH_2
- The proteins create an H^+ concentration gradient across the membrane and then use the energy of that gradient to drive ATP synthesis.
- When the electrons reach the bottom of the ETC they are picked up by the **final electron acceptor** O_2 to form water.
- In **chemiosmosis**, the H^+ diffuses back across the inner membrane, through **ATP synthase** complexes, driving the synthesis of ATP.

SCIENTIFIC THINKING: Scientists have discovered heat-producing, calorie-burning brown fat in adults

- Mitochondria in brown fat can burn fuel and produce heat without making ATP.
- Ion channels spanning the inner mitochondrial membrane
 - allow H^+ to flow freely across the membrane
 - dissipate the H^+ gradient that the electron transport chain produced, which does not allow ATP synthase to make ATP.
- Scientific studies of humans indicate that brown fat may be present in most people when activated by cold environments, the brown fat of lean individuals is more active.





Fermentation: Anaerobic Harvesting of Energy

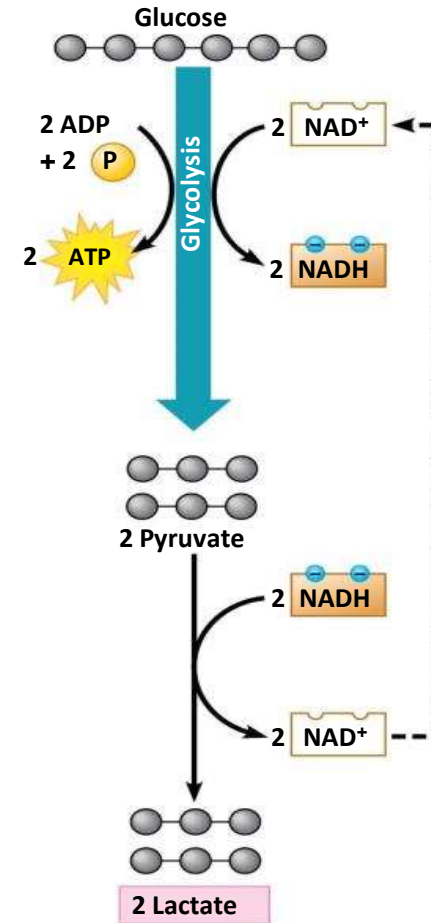
Fermentation enables cells to produce ATP without oxygen

- Fermentation is a way of harvesting chemical energy that does not require oxygen.
- Fermentation uses glycolysis, produces two ATP molecules per glucose, and reduces NAD^+ to NADH.
- Fermentation also provides an anaerobic path for recycling NADH back to NAD^+ .

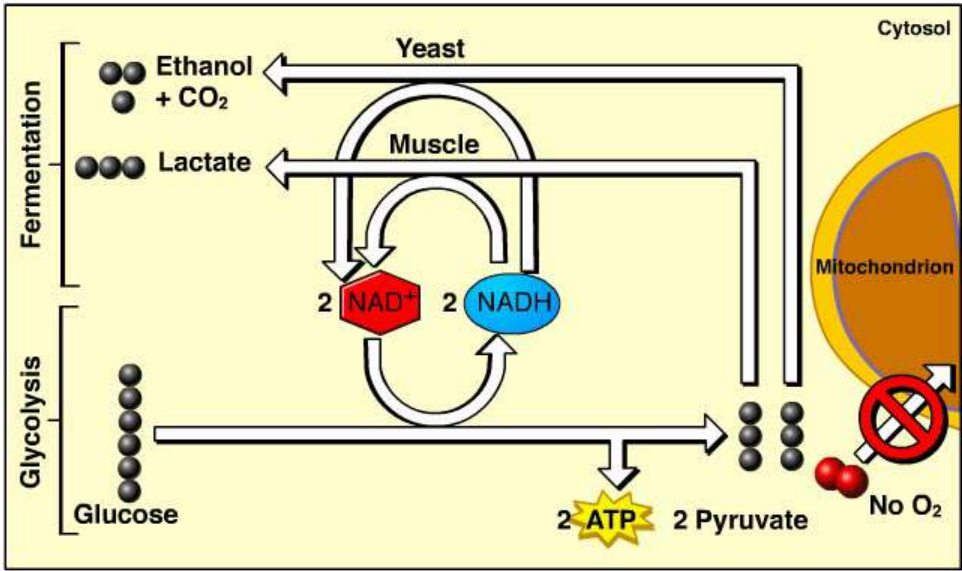


Lactic Acid Fermentation

- Your muscle cells and certain bacteria can regenerate NAD^+ through **lactic acid fermentation**, in which NADH is oxidized back to NAD^+ and pyruvate is reduced to lactate.
- Lactate is carried by the blood to the liver, where it is converted back to pyruvate and oxidized in the mitochondria of liver cells.
- The dairy industry uses lactic acid fermentation by bacteria to make cheese and yogurt.
- Other types of microbial fermentation turn soybeans into soy sauce and cabbage into sauerkraut.

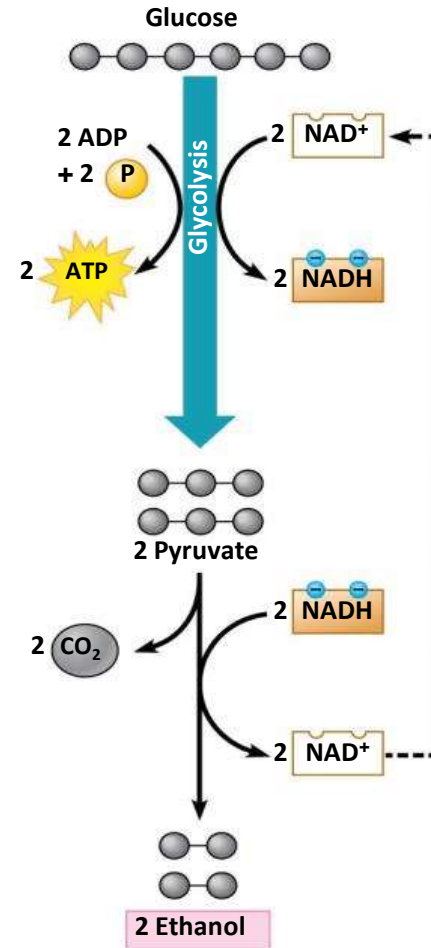


Fermentation Overview



Alcohol Fermentation

- The baking and winemaking industries have used **alcohol fermentation** for thousands of years.
- In this process, yeast (single-celled fungi) oxidize NADH back to NAD^+ and convert pyruvate to CO_2 and ethanol.



Fermentation enables cells to produce ATP without oxygen

- Obligate anaerobes (bacterial cells) require anaerobic conditions, are poisoned by oxygen, and live in stagnant ponds and deep soils. Ex. *Clostridium botulinum*, causes botulism.

Botox



- Facultative anaerobes can make ATP by fermentation or oxidative phosphorylation and include yeasts and many bacteria.

EVOLUTION CONNECTION: Glycolysis evolved early in the history of life on Earth

- Glycolysis is the universal energy-harvesting process of life. The role of glycolysis in fermentation and respiration dates back to life long before oxygen was present, when only prokaryotes inhabited the Earth, about 3.5 billion years ago.
- The ancient history of glycolysis is supported by its occurrence in all the domains of life and location within the cell, using pathways that do not involve any membrane-enclosed organelles of the eukaryotic cell.

Connections Between Metabolic Pathways

Cells use many kinds of organic molecules as fuel for cellular respiration

- Although glucose is considered to be the primary source of sugar for respiration and fermentation, ATP is generated using carbohydrates, fats, and proteins.
- Fats make excellent cellular fuel because they contain many hydrogen atoms and thus many energy-rich electrons and yield more than twice as much ATP per gram as compared to a gram of carbohydrate.
- Proteins can also be used for fuel, although your body preferentially burns sugars and fats first.

