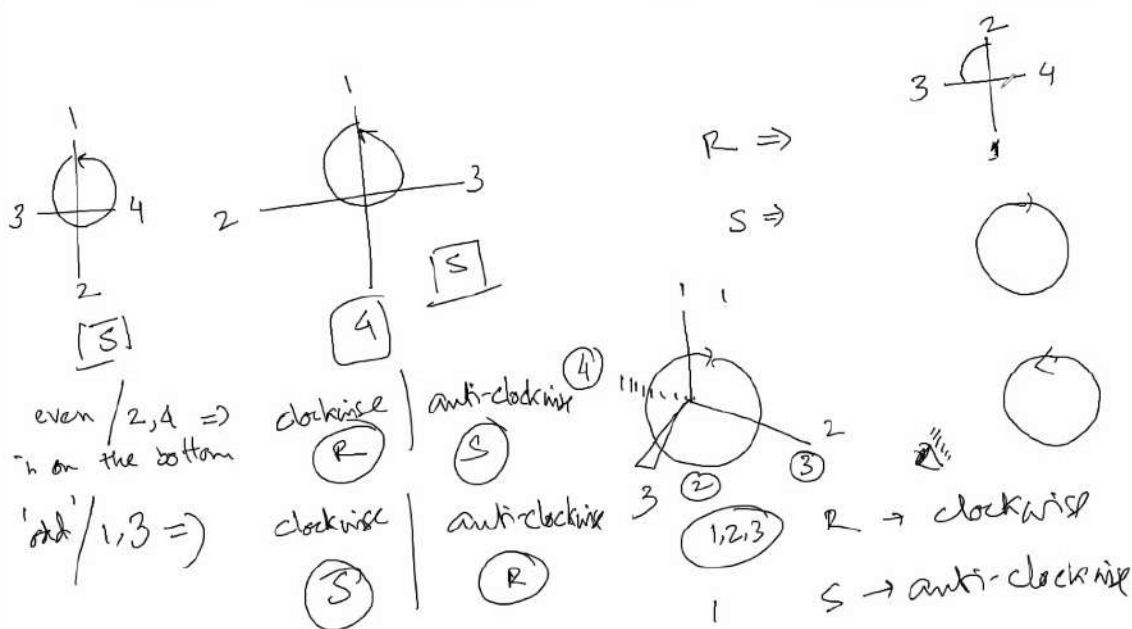
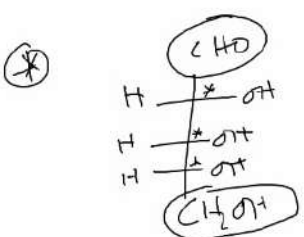


 Sahil Deshmukh



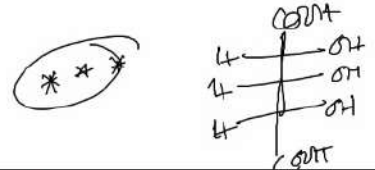
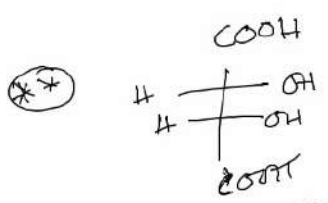
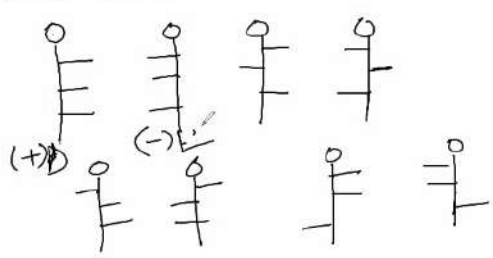


Ribose
(Aldo pentose)

Total no. of possible stereoisomers?

$$2^3 = 8$$

$$2^n$$



Sanchit Das



Adityachar Dwivedi



Dibyendu Sarkar



Annwoy Roy Ch...

Souradeep Dhar

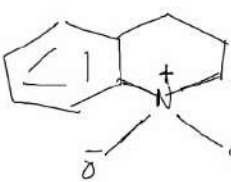


View

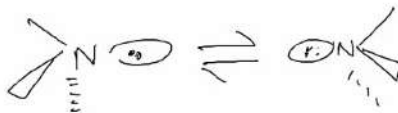
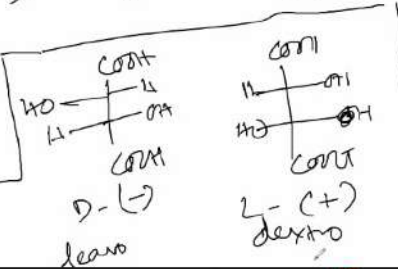
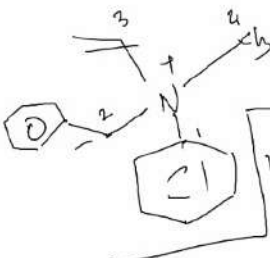
Compounds

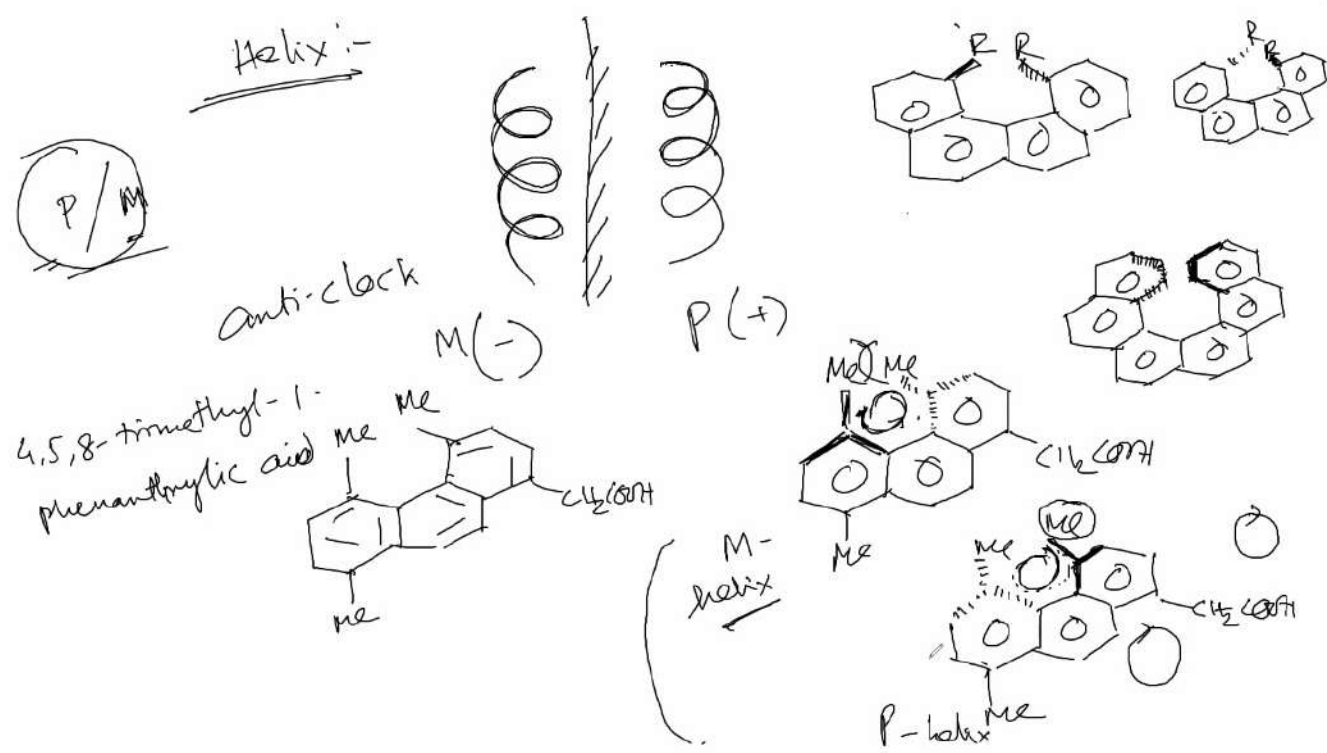
'C' chiral

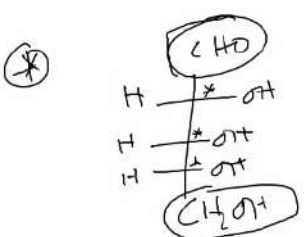
'Q' wastermany
ammonium salts



→ 'N' chiral center



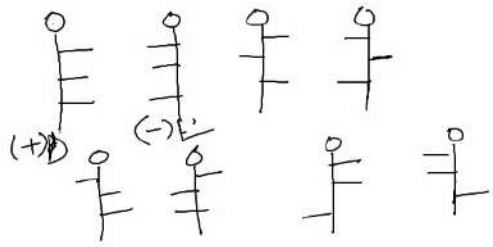




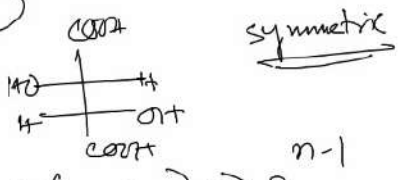
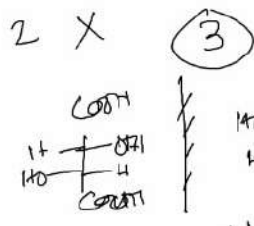
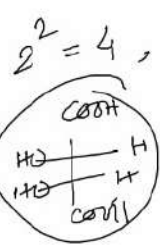
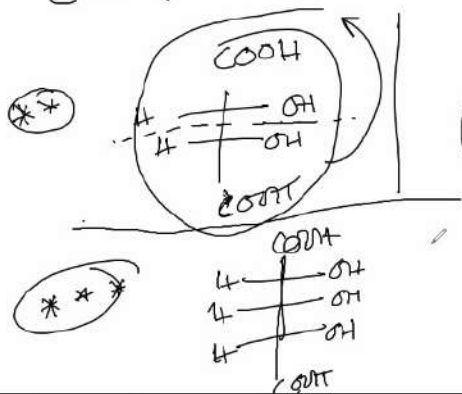
Total no. of possible stereoisomers?

$$2^3 = 8$$

$$2^n$$



Ribose (Aldo pentose)



$n = \text{odd} (1, 3, 5, \dots) \Rightarrow 2^{n-1}$
 $n = \text{even} (2, 4, 6, \dots) \Rightarrow 2^{n-1} + 2^{(n-2)/2}$
 for $n=2$, $= 2^{2-1} + 2^{(2-2)/2} = 2^1 + 2^0 = 2 + 1 = 3$

even / 2, 4 \Rightarrow
"h" on the bottom
'odd' / 1, 3 \Rightarrow

[S]

clockwise
(R)

anti-clockwise
(S)

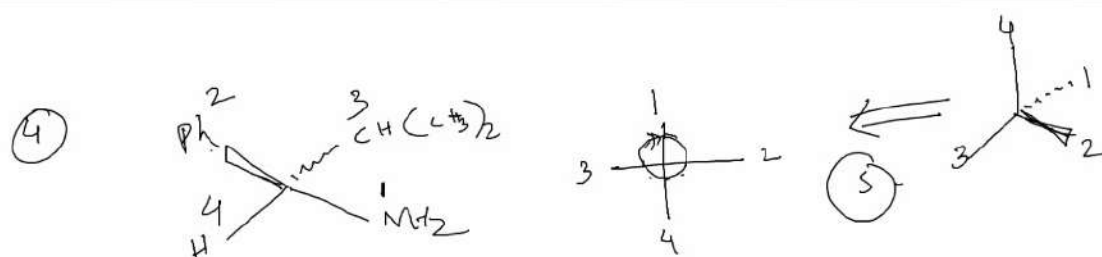
clockwise
S

anti-clockwise
(R)

R \Rightarrow
S \Rightarrow

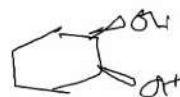
1, 2, 3

R \rightarrow clockwise
S \rightarrow anti-clockwise

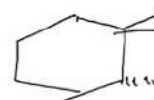


⑤ D-glyceraldehyde ? ⑥ D-tartaric acid !

⑦ cis-cyclohexane-1,2-diol



⑧ trans-cyclohexane-1,2-diol






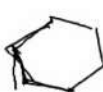


Bayer's Theory

1890

B.A. 109.5° Deviation

Angle strain

| | | |
|----------------------------------------------------------------------------------|--------------|-------------------------------------------------------------------------------------|
|  | cyclopropane |  |
|  | cyclobutane |  |
| | cyclopentane |  |
| | cyclohexane |  |

| | | |
|-------------|---------------|-----------------------------------------|
| 60° | 109.5° | $109.5^\circ - 60^\circ = 49.5^\circ$ |
| 90° | | $109.5^\circ - 90^\circ = 19.5^\circ$ |
| 108° | | $109.5^\circ - 108^\circ = 1.5^\circ$ |
| 120° | | $109.5^\circ - 120^\circ = -10.5^\circ$ |



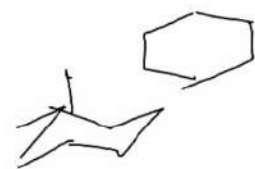
C-C cyclopentane 0.75°
C-C cyclohexane 5.28°

Drawbacks:

- Not able to explain the case for larger ring
- cyclopentane is more stable than cyclohexane
- larger ring system are not planar, rather twisted or puckered

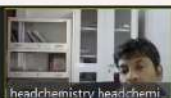
Bayer strain theory

$5 > 6 > 4 > 3$



Sanchit Das

 Sanchit Das



headchemistry headchemi

Sarthak

Sarthak

victus

 victus

Souradeep Dhar

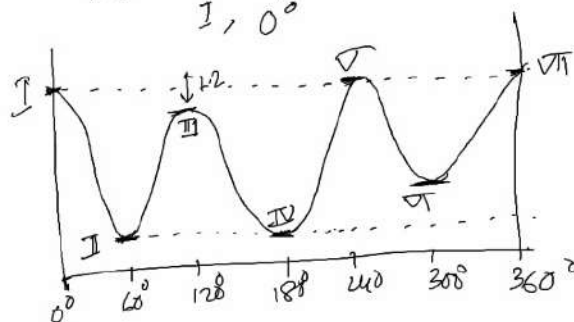
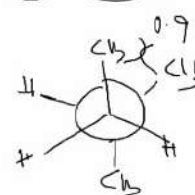
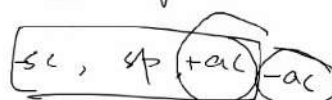
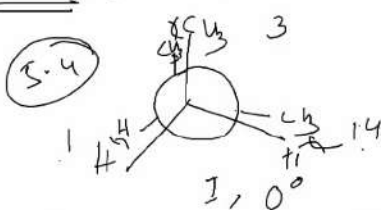
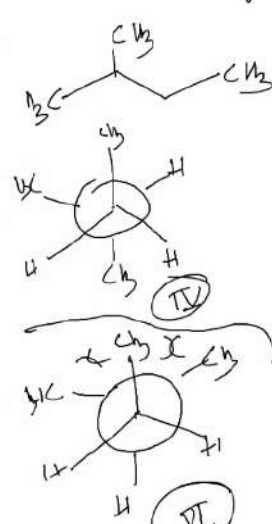
 Souradeep Dhar

Anshika Nangla

Anshika Nangla



2-methyl butane? conformational energy diagram



1.4 p 3

(4.2)

Unmute Start Video

96
Participants



Share Screen


Record

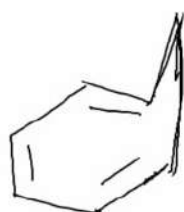
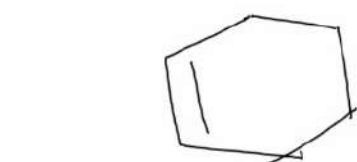
 Reactions

Apps

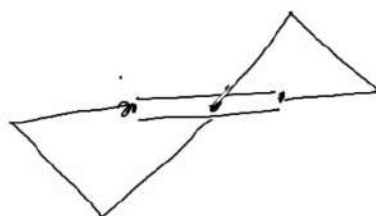
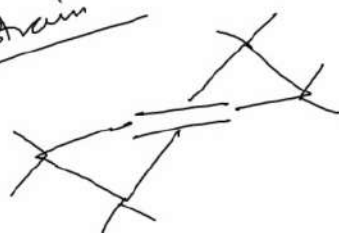
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View Options

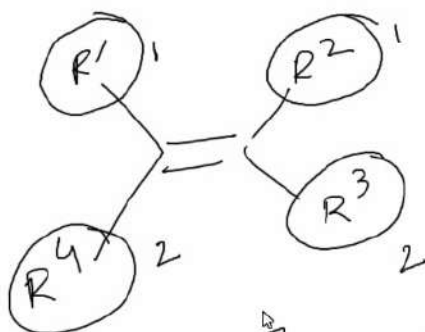
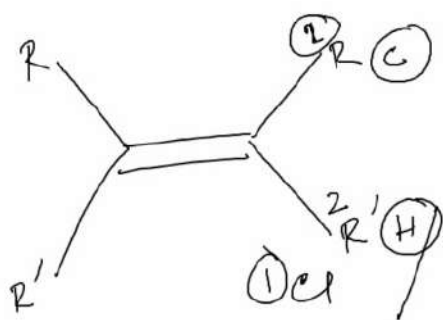


"transannular strain"



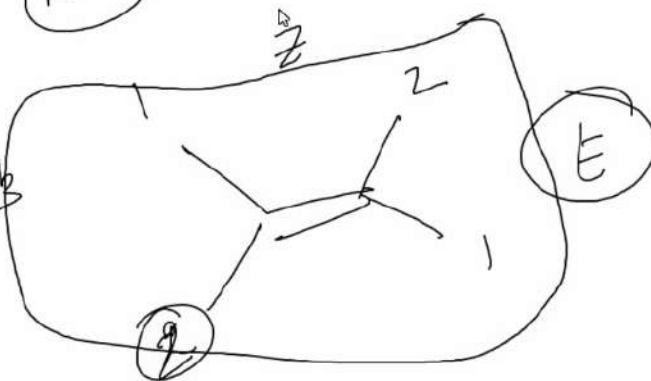
'transannular strain'





E - ?
Z - ?

Atomic no.
Br > Cl > COOH > Cl



Sanchit Das

Sanchit Das

headchemistry headchemistry

Diiraj pakad

Diya jasmine M

Diya jasmine M

Prince Raj

Bayer's Theory

1890

B.A. 109°

Deviation

Angle strain

cyclopropane



60°

$49^\circ 5'$

24.75°

cyclobutane



90°

$19^\circ 5'$

9.75°

cyclopentane



108°

$1^\circ 5'$

0.75°

cyclohexane



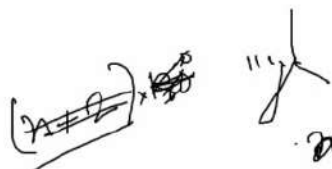
120°

-11°

-5°

Bayer Strain theory

$5 > 6 > 4 > 3$



$2 \times 3 =$

180°

360°

540°

720°

$2 \times 3 = ?$

2-methyl butane? conformational energy diagram

