



Appendix -II

Procedure for Online Payment of Fees and Helpline for Payment-Related Queries

After completing Step 2 of the **Online** Application Form, the candidate may remit the examination fee (**Step** 3) in the following manner:

- Check the validity of the Debit/ Credit Card and keep it ready with you while logging on to the
 website for submitting the Application Form. The candidates should enter the information asked
 for and make payment through a Debit/ Credit Card.
- Through Net Banking, check the balance in your account and keep all credentials ready with you while logging on to the website for making payments. The candidates should log in with his/her credentials of net banking and make payments accordingly.
- For payment through a UPI Service, check whether there is a requisite balance in the Bank Account / Wallet to which the concerned UPI is linked.

The prescribed Examination Fee (please see the Information at a Glance) can be paid through any Bank/Payment Gateway in any Payment Mode [service/processing charges per transaction and GST applicable thereon @ 18 % are to be borne by the candidate

rr												
S. N	Mode of Payment		ICICI BANK		HDFC BANK		CANARA BANK		STATE BANK OF INDIA			
		ICICI	NIL Charg	ge	HDFC	NIL Charge		Canara B	ank	NIL Charge	SBI	
1	Net Banking	Other Banks	4.00 + GS'	Т	Other Banks	4.00 + GST		Other Ba	nks	5.00 + GST	Other Banks	Rs 5.00+GST
	All Debit	ICICI or	Transact ion upto Rs 2000/-	0 %	HDFC	Transaction upto Rs 2000/-	0 %	Canara Bank	Nil	Nil		Transaction upto Rs 2000/-
2	Card	Other Banks	Transact ion above Rs 2000/-	0.5 % + GST	or Other Banks	Transaction above Rs 2000/-	0 %	or Other Banks	Charge			Transaction above Rs 2000/-
3	Credit Card	Domesti c	0.40% of value	Transaction	Domesti c	0.40% of Fee	+ GST	Domestic	3	0.80% of Fee + GST	Domestic	0.80% of Fee+ GST (Minimum Rs 11/-)
		Internati onal	2.35% of value	Transaction	Internati onal	Nil Charge		Internation	onal	2.35% of Fee+ GST	International	3.50% of Fee+ GST (Minimum Rs 11/-)
	Unified	ICICI or	Transact ion upto Rs 2000/-	0 %								Transaction upto Rs 2000/-
4	Payment Interface (UPI)	other banks	Transact ion above Rs 2000/-	5.0 % + GST		Nil Charge						Transaction above Rs 2000/-

Note: In case, the fee payment status is not 'OK', or the Confirmation Page of the Online Application is not generated after fee payment (completion of Step 3) of the candidate is advised to contact the helpline number/email of the concerned Bank/Payment Gateway Integrator, as mentioned below:-





(a) If Paying through the State Bank of India (SBI):

Level	Name	Email ID	Contact Number
1	Helpdesk	dgmcs.cc@sbi.co.in	18004253800
2	Helpdesk3	sbi.05222@sbi.co.in	08026599990
3	Customer Care	agmcustomer.Ihodel@sbi.co.in	1800112211
4	Through SMS	UNHAPPY (add text)	8008202020

(b) If Paying through Canara Bank:

Level	Name	Email ID	Contact Number
1	Helpdesk	pgsupport@billdesk.com	01202400850/1
			9971262371
2	Complaint	genadmnronoida@canarabank.com	01202400850/1
	Management Services		
3	Customer Care	genadmnronoida@canarabank.com	9971262371
4	Through SMS	genadmnronoida@canarabank.com	01202400850/1

(c) If Paying through HDFC Bank:

Level	Name	Email ID	Contact Number
1	Shri Vikram Singh	vikram.singh4@hdfcbank.com	9799810080
2.	Amit Singh	amit.singh26@hdfcbank.com	7428869770
3	Shri Ripon Bhattacharjee	ripon.bhattacharjee@hdfcbank.com	9625031697

(d) If Paying through ICICI Bank:

S. No.	Name	Contact Number
1.	Helpdesk	1800222884
2.	Deepak Kumar	8882892218
3.	Rakesh Bhatnagar	9953972004





Appendix-III

Computer Based Test (CBT)

The major examinations being conducted by NTA are Computer-Based Tests (CBT). A CBT requires candidates to sit in front of a computer terminal (node) allocated to them against their Roll number and Admit card. After logging the candidate will get detailed instructions for the examinations. At the designated time of the start of the examination, the candidates will be able to proceed and see the questions on the computer screen using the computer mouse. Candidates will have the option to change/modify/edit/answers already entered at any time during the examination.

Procedure for appearing in Computer Based Test (CBT):

(Sample/mock test will be available on the NTA website: www.nta.ac.in for hands-on practice)

- (a) A computer terminal (node) indicating the Roll Number will be allocated to each candidate. Candidates should find and sit on their allocated computers only. Any candidate found to have changed room/hall or the computer on their own other than the one allotted would lead to cancellation of candidature and no plea in this regard would be entertained.
- (b) For login, the candidate will have to enter a **login ID** and **password**. The computer terminal allotted to the candidate will display the WELCOME login screen, the Candidate's photograph and the subject opted by the candidate.

Candidate Login Page Student Login Candidate Welcome Screen



(c) After login, the candidate shall be able to see the detailed instructions for the examination. Candidates are advised to go through the instructions carefully regarding the type of questions and marking scheme. At the designated time of the start of the examination, the candidates will be able to proceed and see the questions on the computer screen.





Please read the instructions carefully

General Instructions:

- 1. The total duration of the examination is **180** minutes for **Paper 1** (**B.E./B.Tech**) and **210 minutes** for (B.Arch and B.Planning) both.
- 2. The clock will be set on the server. The countdown timer in the top right corner of the screen will display the remaining time available for you to complete the examination. When the timer reaches zero, the examination will end by itself. You will not be required to end or submit your examination.
- 3. The Question Palette displayed on the right side of the screen will show the status of each question using one of the following symbols:
- 1 You have not visited the question yet.
- You have not answered the question.
- You have answered the question.
- You have NOT answered the question, but have marked the question for review.
- The question(s) "Answered and Marked for Review" will be considered for evaluation.
- 4. The 'Marked for Review' status for a question simply indicates that you would like to look at that question again.
- 5. You can click on the ">" arrow which appears to the left of the question palette to collapse the question palette thereby maximizing the question window. To view the question palette again, you can click on "<" which appears on the right side of the question window.
- 6. You can click on your "Profile" image on the top right corner of your screen to change the language during the exam for the entire question paper. On clicking on the Profile image, you will get a drop-down to change the question content to the desired language.
- 7. You can click on to navigate to the bottom and to navigate to the top of the question area, without scrolling.

Navigating to a Question:

- 8. To answer a question, do the following:
 - a. Click on the question number in the Question Palette at the right of your screen to go to that question directly. Note that using this option does NOT save your answer to the current question.





- b. Click on **Save & Next** to save your answer for the current question and then go to the next question.
- c. Click on **Mark for Review & Next** to save your answer for the current question, mark it for review, and then go to the next question.

Answering a Question:

- 9. Procedure for answering a Multiple Choice Type question:
- a. To select your answer, click on the button of one of the options.
- b. To deselect your chosen answer, click on the button of the chosen option again or click on the **Clear Response** button.
- c. To change your chosen answer, click on the button of another option.
- d. To save your answer, you must click on the Save & Next button
- e. To mark the question for review, click on the 'Mark for Review & Next' button.
- 10. To change your answer to a question that has already been answered, first select that question for answering and then follow the procedure for answering that type of question.

Navigating through sections:

- 11. Sections in this question paper are displayed on the top bar of the screen. Questions in a section can be viewed by clicking on the Section name. The section you are currently viewing is highlighted.
- 12. After clicking the **Save and Next** button on the last question for a Section, you will automatically be taken to the first question of the next section.
- 13. You can shuffle between sections and questions anytime during the examination as per your convenience only during the time stipulated.
- 14. The candidate can view the corresponding section summary as part of the legend that appears in every section above the question palette.

Instruction for Question:

15. For the correctness of the translation for a particular language, only the **English version** will be considered as the final version for evaluation purposes.

The keyboard attached to the computer, if any, will be disabled during the entire duration of the examination. Depending on the type of question, the answers to questions can either be entered by clicking on the virtual on-screen keyboard (numeric or otherwise) using the computer mouse or by clicking the chosen option(s) using the computer mouse. Candidates will have the option to change/modify answers already entered at anytime during the entire duration of the examination.

In case the computer/mouse allotted to any candidate malfunctions during the test, he/she will be immediately allotted another computer system and the time lost due to this will be adjusted in the server so as to give the candidate the full allotted time.





The on-screen computer clock counter of every candidate will be set at the server. The countdown timer on the top right side of the computer screen will display the time remaining (in minutes) available for the candidate to complete the examination. When the timer reaches zero, the examination will end by itself. The candidates will not be required to end or submit the examination.

The Question Palette displayed on the screen will show the status of each question using one of the following symbols:



The question(s) "Answered and Marked for Review" status for a question indicates that the candidates would like to have a relook at that question again. A candidate has the option of answering a question and simultaneously "Marked for Review", these answers will be considered for evaluation. However, if a candidate has simply put "Marked for Review" for a question without answering it, the corresponding question marked for review without an answer will not be considered for evaluation. It may be noted that a candidate can return to any "Marked for Review" question at any time during the examination by clicking on the corresponding question number icon displayed on the Question Palette of the corresponding section.

- (d) The candidates can click on the ">" arrow which appears to the left of the question palette to collapse the question palette thereby maximizing the question viewing window. To view the question palette again, the candidate can click on "<" which appears on the right side of the question window.
- (e) Candidates can click on to navigate to the bottom and to navigate to the top of the question area, without scrolling. Using the computer mouse the candidate can scroll up and down the question viewing area for viewing the entire question.
- (f) The full paper can be viewed by clicking the "Question Paper" icon on the top right corner of the screen.
- (g) Blank Sheets for doing Rough Work/calculations shall be provided to the candidates. The Blanks Sheets would have a Header page for the candidates to write down his/her Name and Roll Number. All calculations/writing work is to be done only in the Blank Sheets provided at the Centre in the Examination Room/Hall and on completion of the test candidates must hand over the rough sheets to the invigilator on duty in the Room/Hall.
- (h) Navigating to a Question To navigate between questions within a Paper, the candidate needs to do the following:





- (a) Click on the question number in the Question Palette at the right of the screen to go to that numbered question directly. Note that using this option does NOT save the answer to the currently displayed question.
- (b) Click on "Save and Next" to save the answer to any question. Clicking on "Save and Next" will save the answer for the current question and the next question will be displayed on the candidate's computer screen.
- (c) Click on "Mark for Review and Next" to mark a question for review (without answering it) and proceed to the next question.



(i) Answering a Question

To navigate between questions within a Paper, the candidate needs to do the following:

Procedure for answering a Multiple-Choice type question:

- (a) To select the option(s), click on the corresponding button(s) of the option(s).
- (b) To deselect the chosen answer, click on the button of the chosen option again or click on the "Clear Response" button.
- (c) To save the answer, the candidate MUST click on the "Save and Next" button.
- (d) To mark the question for review (without answering it), click on the "Mark for Review and Next" button.

(j) Navigating through sections:

- (i) Sections in the question paper are displayed on the top bar of the screen. Questions in a section can be viewed by clicking on the section name. The section in which the candidate is currently viewing will be highlighted.
- (ii) After clicking the "Save and Next" button on the last question for a section, the candidate will automatically be taken to the first question of the next section.
- (iii) Candidates can shuffle between sections and questions within sections anytime during the examination as per convenience only during the time stipulated.
- (iv) The candidate can view the corresponding section summary as part of the legend that appears in every section above the question palette.
- (k) Procedure for answering questions that require inputs from the on-screen virtual keyboard (numeric or otherwise):





(a) The candidate will have to use the on-screen virtual keyboard (that would be displayed just below the question statement of these types of questions) and the attached computer mouse to enter his/her answer in the space provided for the answer.

On-Screen Virtual Keyboard



- (b) The answer can be changed, if required, anytime during the test. To save the answer, the candidate MUST click on the "Save and Next" button.
- (c) To mark the question for review (without answering it), click on the "Mark for Review and Next" button.

The candidates will have the option to change previously saved answer to any question, anytime during the entire duration of the test. To change the answer to a question that has already been answered, first select the corresponding question from the Question Palette, then click on "Clear Response" to clear the previously entered answer and subsequently follow the procedure for answering that type of question.

(l) **ROUGH WORK:**

All calculations/writing work is to be done only in the rough sheet provided at the Centre in the Examination Room/Hall and on completion of the test candidates must hand over the rough sheets to the invigilator on duty in the Room/Hall.





Appendix-IV

NTA: Test Practice Centres (TPCs)

What is a Test Practice Centre (TPCs)?

The Ministry of Human Resource Development mandated the NTA to set up, establish, and create a network of Test Practice Centres for candidates, especially in remote and rural areas to enable them to practice and be comfortable in taking a Computer Based Test (CBT). This facility is completely free of cost. Candidates can register online (on the NTA website) where they are provided a convenient TPC near to their location to practice on a given computer node. This facilitates and eases the process of being able to take a Computer Based Test (CBT). The entire experience of using a computer is close to the actual experience of taking a CBT. All efforts are made to provide practice tests and questions so that candidates can familiarize themselves with logging into the system, go through the detailed instructions regarding the test, use the mouse or numeric keyboard on-screen (virtual) to attempt each question, scroll down to the next question, navigate between questions, review and edit their options and submit questions.

The objective of TPCs is primarily to organize test practice for the upcoming NTA examinations.





Appendix - V

Procedure to be adopted for compilation of NTA scores for multi-session Papers (Normalization procedure based on PERCENTILE SCORE)

NTA may conduct examinations on multiple dates, generally in two sessions per day. The candidates will be given different sets of questions per session and it is quite possible that in spite of all efforts to maintain equivalence among various question papers, the difficulty level of these question papers administered in different sessions may not be exactly the same. Some of the candidates may end up attempting a relatively tougher set of questions when compared to other sets. The candidates who attempt the comparatively tougher examination are likely to get lower marks as compared to those who attempt the easier one. In order to overcome such a situation, a "Normalization procedure based on Percentile Score" will be used to ensure that candidates are neither benefitted nor disadvantaged due to the difficulty level of the examination. With the objective of ensuring that a candidate's true merit is identified and that a level playing field is created in the above context, the Normalization Procedure, set out below shall be adopted, for compiling the NTA scores for multi-session papers.

The process of Normalization is an established practice for comparing candidate scores across multi-session papers and is similar to those being adopted in other large educational selection tests conducted in India. For normalization across sections, NTA shall use the percentile equivalence.

Percentile Scores: Percentile scores are scores based on the relative performance of all those who appear for the examination. The marks obtained are transformed into a scale ranging from 100 to 0 for each session of examinees.

The Percentile Score indicates the percentage of candidates that have scored EQUAL TO OR BELOW (same or lower raw scores) that particular Percentile in that examination. Therefore the topper (highest score) of each session will get the same Percentile of 100 which is desirable. The marks obtained in between the highest and lowest scores are also converted to appropriate Percentiles.

The Percentile score will be the Normalized Score for the examination (instead of the raw marks of the candidate) and shall be used for the preparation of the merit lists.

The Percentile Scores will be calculated up to 7 decimal places to avoid the bunching effect and reduce ties.

The Percentile score of a Candidate is calculated as follows:

100 X Number of candidates appeared in the 'Session' with raw score EQUAL TO OR LESS than the candidate The total number of the candidates who appeared in the 'Session'

Note: The Percentile of the Total shall **NOT be** an aggregate or average of the Percentile of the individual subject. The percentile score is not the same as the percentage of marks obtained.

Example: Suppose a test was held in 4 sessions of examinees as per details given below:

(Allocation of Days and shifts was done randomly)

(a) The distribution of candidates was as follows:

Session-1: Day-1 Shift-1, **Session-2:** Day-1 Shift-2, **Session-3:** Day-2 Shift-1 and **Session-4:** Day-2 Shift-2





Session	Day/Shift	N	o of Candida	Marks	Marks	
		Absent	Appeared	Total	Highest	Lowest
Session-1	Day-1 Shift-1	3974	28012	31986	335	-39
Session-2	Day-1 Shift-2	6189	32541	38730	346	-38
Session-3	Day-2 Shift-1	6036	41326	47362	331	-49
Session-4	Day-2 Shift-2	9074	40603	49677	332	-44
Total	•	25273	142482	167755	346	-49
(Session-1 t	o Session-4)					

In this method of scoring the HIGHEST RAW SCORE in each paper (irrespective of the raw scores) will be the 100 Percentile indicating that 100% of candidates have scored equal to or lesser than the highest scorer/ topper for that session.

Highest Raw Score and Percentile Score: All the highest raw scores will have a normalized Percentile Score of 100 for their respective session.

Session	Total Candidate s Appeared	Highest Raw Score	Candidates who scored EQUAL OR LESS THAN Highest Raw Score	Percentile Score	Remarks
Session-1	28012	335	28012	100.0000000 [(28012/28012)*100]	i.e. all the highest raw scores would
Session -2	32541	346	32541	100.0000000 [(32541/32541)*100]	be normalized to
Session -3	41326	331	41326	100.0000000 [(41326/41326)*100]	100 Percentile Score for their respective
Session -4	40603	332	40603	100.0000000 [(40603/40603)*100]	session.

Lowest Raw Score and Percentile Score: The percentile Score of all the lowest raw scores will depend on the total number of candidates who have taken the examination for their respective sessions.





Session	Total Candidates Appeared	Lowest Raw Score	Candidates who scored EQUAL OR LESS THAN Lowest Raw Score	Percentile Score	Remarks
Session -1	28012	-39	1	0.0035699 [(1/28012)*10	i.e. Percentile Score of all the
G	20541	20	1	0]	lowest raw scores
Session -2	32541	-38	1	0.0030730 [(1/32541)*10 0]	is different i.e. Percentile Score depends on the
Session -3	41326	-49	1	0.0024198 [(1/41326)*10 0]	total number of candidates who have taken the
Session -4	40603	-44	1	0.0024629 [(1/40603)*10 0]	examination for their respective sessions.

The following is a further explanation of the interpretation of the raw scores and Percentile Score in Session-3 (Day-2 and Shift-1) with 41326 candidates who have taken the examination.

Candidate	Percentile Score	No of Candidates	Raw Score	Remark
A	100.0000000 [(41326/41326)*100]	1	331	This indicates that amongst those who appeared, 100% have scored either EQUAL TO OR LESS THAN candidate A (331 raw scores). It also indicates that no candidate has scored more than candidate A (331 raw scores).
В	90.1224411 [(37244/41326)*100]	77	121	This indicates that amongst those who appeared, 90.1224411% have scored either EQUAL TO OR LESS THAN candidate B (121 raw scores). It also indicates that the remaining candidates have scored





				more than candidate B (121 raw scores).
				5667 65).
С	50.4549194	381	41	This indicates that amongst those
	[(20851/41326)*100]			who appeared, 50.4549194%
	[(20031/11320) 100]			have scored either EQUAL TO
				OR LESS THAN candidate C (41
				raw scores).
				It also indicates that the
				remaining those who appeared
				have scored more than candidate
				C (41 raw scores).
D	31.7040120	789	25	This indicates that amongst those
	[(13102/41326)*100]			who appeared, 31.7040120%
	[(10102/11020) 100]			have scored either EQUAL TO
				OR LESS THAN candidate D (25
				raw scores)
				It also indicates that the
				remaining candidates have scored
				more than candidate D (25 raw
				scores).
E	1.1034216	100	-15	Indicates that amongst those who
	[(456/41326)*100]			appeared, 1.1034216% have
				scored either EQUAL TO OR
				LESS THAN candidate E (-15
				raw score)
				It also indicates that the
				remaining candidates have scored
				more than candidate E (-15 raw
				score)

STEP-BY-STEP PROCEDURE FOR NORMALIZATION AND PREPARATION OF RESULT:

Step-1: Distribution of Examinees in two shifts:

Candidates have to be distributed into two sessions randomly so that each session has an approximately equal number of candidates. These two sessions would be as follows:

Session-1: Day-1 Shift-1, Session-2: Day-1 Shift-2

In the event of a more number of days or less number of shifts, the candidates will be divided accordingly.





This will ensure that there is no bias in the distribution of candidates who shall take the examination. Further, with a large population of examinees spread over the entire country, the possibility of such bias becomes remote.

Step 2: Preparation of Results for each Session:

The examination results for each session will be prepared in the form of

- ➤ Raw Scores
- > Percentiles Scores of Total raw scores.

The Percentiles would be calculated for each candidate in the Session as follows: Let TP1 be the Percentile Score of the Total Raw Score of that candidate.

Total Danagatile		No. of candidates appeared from the session with raw score EQUAL TO			
Total Percentile	100 X	OR LESS than the score of the Candidate			
(TP1):		Total No. of candidates who appeared in the session			

Step-3: Compilation of NTA score and Preparation of Result:

The Percentile scores for the Total Raw Score for all the sessions (Session-1: Day-1 Shift-1, Session-2: Day-1 Shift-2) as calculated in Step-2 above would be merged and shall be called the NTA scores which will then be used for the compilation of results and further processing for deciding the allocation.

In the event of the percentiles for the multi-shifts being dissimilar/unequal, the lowest will be the eligibility cut-off for that category for all candidates (i.e. all shifts).

For Example: In the examination held in two shifts, if the 40% marks correspond to a Percentile score of 78 in Shift 1 and 79 in Shift 2, then all those equal to or above 78 percentiles (Percentile score of 100 to 78) in both shifts will become eligible in General Category. A similar method will be adopted for the other categories to determine eligibility cut-offs. In case the examination is held in more number of shifts the same principle shall apply.





Appendix - VI

Syllabus for JEE (Main) - 2024

Syllabus for JEE Main Paper 1 (B.E./B.Tech.)- Mathematics, Physics, and Chemistry

MATHEMATICS

UNIT 1: SETS, RELATIONS, AND FUNCTIONS:

Sets and their representation: Union, intersection, and complement of sets and their algebraic properties; <u>Power set</u>; Relation, Type of relations, equivalence relations, functions; one-one, into and onto functions, <u>the composition of functions</u>.

UNIT 2: COMPLEX NUMBERS AND QUADRATIC EQUATIONS:

Complex numbers as ordered pairs of reals, Representation of complex numbers in the form a + ib and their representation in a plane, Argand diagram, algebra of complex number, modulus, and argument (or amplitude) of a complex number, Quadratic equations in real and complex number system and their solutions Relations between roots and co-efficient, nature of roots, the formation of quadratic equations with given roots.

UNIT3: MATRICES AND DETERMINANTS:

Matrices, algebra of matrices, type of matrices, determinants, and matrices of order two and three, evaluation of determinants, area of triangles using determinants, Adjoint, and evaluation of inverse of a square matrix using determinants and, Test of consistency and solution of simultaneous linear equations in two or three variables using matrices.

UNIT 4: PERMUTATIONS AND COMBINATIONS:

The fundamental principle of counting, permutation as an arrangement and combination as section, Meaning of P(n,r) and C(n,r), simple applications.

UNIT 5: BINOMIAL THEOREM AND ITS SIMPLE APPLICATIONS:

Binomial theorem for a positive integral index, general term and middle term, and simple applications.

UNIT 6: SEQUENCE AND SERIES:

Arithmetic and Geometric progressions, insertion of arithmetic, geometric means between two given numbers, Relation between A.M and G.M.

UNIT 7: LIMIT, CONTINUITY, AND DIFFERENTIABILITY:

Real-valued functions, algebra of functions, polynomials, rational, trigonometric, logarithmic, and exponential functions, inverse function. Graphs of simple functions. Limits, continuity, and differentiability. Differentiation of the sum, difference, product, and quotient of two functions. Differentiation of trigonometric, inverse trigonometric, logarithmic, exponential, composite, and implicit functions; derivatives of order up to two, Applications of derivatives:





Rate of change of quantities, monotonic-Increasing and decreasing functions, Maxima and minima of functions of one variable,

UNIT 8: INTEGRAL CALCULAS:

Integral as an anti-derivative, Fundamental integral involving algebraic, trigonometric, exponential, and logarithmic functions. Integrations by substitution, by parts, and by partial functions. Integration using trigonometric identities.

Evaluation of simple integrals of the type

$$\int \frac{dx}{x^{2} + a^{2}} , \int \frac{dx}{\sqrt{x^{2} \pm a^{2}}} , \int \frac{dx}{a^{2} - x^{2}} , \int \frac{dx}{\sqrt{a^{2} - x^{2}}} , \int \frac{dx}{ax^{2} + bx + c} , \int \frac{dx}{\sqrt{ax^{2} + bx + c}} , \int \frac{(px+q)dx}{ax^{2} + bx + c}$$

$$\int \frac{(px+q)dx}{\sqrt{ax^{2} + bx + c}} \int \sqrt{a^{2} \pm x^{2}} dx , \int \sqrt{x^{2} - a^{2}} dx$$

. The fundamental theorem of calculus, properties of definite integrals. Evaluation of definite integrals, determining areas of the regions bounded by simple curves in standard form.

UNIT 9: DIFFRENTIAL EQUATIONS

Ordinary differential equations, their order, and degree, the solution of differential equation by the method of separation of variables, solution of a homogeneous and linear differential equation of the type

$$\frac{dy}{dx} + p(x)y = q(x)$$

UNIT 10: CO-ORDINATE GEOMETRY

Cartesian system of rectangular coordinates in a plane, distance formula, sections formula, locus, and its equation, the slope of a line, parallel and perpendicular lines, intercepts of a line on the co-ordinate axis.

Straight line

Various forms of equations of a line, intersection of lines, angles between two lines, conditions for concurrence of three lines, the distance of a point form a line, co-ordinate of the centroid, orthocentre, and circumcentre of a triangle,

Circle, conic sections

A standard form of equations of a circle, the general form of the equation of a circle, its radius and central, equation of a circle when the endpoints of a diameter are given, points of intersection of a line and a circle with the centre at the origin and sections of conics, equations of conic sections (parabola, ellipse, and hyperbola) in standard forms,

UNIT 11: THREE DIMENSIONAL GEOMETRY

Coordinates of a point in space, the distance between two points, section formula, directions ratios, and direction cosines, and the angle between two intersecting lines. Skew lines, the shortest distance between them, and its equation. Equations of a line





UNIT 12: VECTOR ALGEBRA

Vectors and scalars, the addition of vectors, components of a vector in two dimensions and three-dimensional space, scalar and vector products,

UNIT 13: STATISTICS AND PROBABILITY

Measures of discretion; calculation of mean, median, mode of grouped and ungrouped data calculation of standard deviation, variance, and mean deviation for grouped and ungrouped data.

Probability: Probability of an event, addition and multiplication theorems of probability, Baye's theorem, probability distribution of a random variate,

UNIT 14: TRIGONOMETRY

Trigonometrical identities and trigonometrical functions, inverse trigonometrical functions, and their properties,





PHYSICS

UNIT 1: PHYSICS AND MEASUREMENT

Units of measurements, System of Units, S I Units, fundamental and derived units, least count, significant figures, Errors in measurements, Dimensions of Physics quantities, dimensional analysis, and its applications.

UNIT 2: KINEMATICS

The frame of reference, motion in a straight line, Position- time graph, speed and velocity; Uniform and non-uniform motion, average speed and instantaneous velocity, uniformly accelerated motion, velocity-time, position-time graph, relations for uniformly accelerated motion, Scalars and Vectors, Vector. Addition and subtraction, scalar and vector products, Unit Vector, Resolution of a Vector. Relative Velocity, Motion in a plane, Projectile Motion, Uniform Circular Motion.

UNIT 3: LAWS OF MOTION

Force and inertia, Newton's First law of motion; Momentum, Newton's Second Law of motion, Impulses; Newton's Third Law of motion. Law of conservation of linear momentum and its applications. Equilibrium of concurrent forces.

Static and Kinetic friction, laws of friction, rolling friction.

Dynamics of uniform circular motion: centripetal force and its applications: vehicle on a level circular road, vehicle on a banked road.

UNIT 4: WORK, ENERGY, AND POWER

Work done by a constant force and a variable force; kinetic and potential energies, work-energy theorem, power.

The potential energy of spring conservation of mechanical energy, conservative and non-conservative forces; motion in a vertical circle: Elastic and inelastic collisions in one and two dimensions.

UNIT5: ROTATIONAL MOTION

Centre of the mass of a two-particle system, Centre of the mass of a rigid body; Basic concepts of rotational motion; moment of a force; torque, angular momentum, conservation of angular momentum and its applications;

The moment of inertia, the radius of gyration, values of moments of inertia for simple geometrical objects, parallel and perpendicular axes theorems, and their applications. Equilibrium of rigid bodies, rigid body rotation and equations of rotational motion, comparison of linear and rotational motions.





UNIT 6: GRAVITATION

The universal law of gravitation. Acceleration due to gravity and its variation with altitude and depth. Kepler's law of planetary motion. Gravitational potential energy; gravitational potential. Escape velocity, Motion of a satellite, orbital velocity, time period, and energy of satellite.

UNIT 7: PROPERTIES OF SOLIDS AND LIQUIDS

Elastic behaviour, Stress-strain relationship, Hooke's Law. Young's modulus, bulk modulus, and modulus of rigidity. Pressure due to a fluid column; Pascal's law and its applications. Effect of gravity on fluid pressure.

Viscosity. Stokes' law. terminal velocity, streamline, and turbulent flow.critical velocity. Bernoulli's principle and its applications.

Surface energy and surface tension, angle of contact, excess of pressure across a curved surface, application of surface tension - drops, bubbles, and capillary rise. Heat, temperature, thermal expansion; specific heat capacity, calorimetry; change of state, latent heat. Heat transfer-conduction, convection, and radiation.

UNIT 8: THERMODYNAMICS

Thermal equilibrium, zeroth law of thermodynamics, the concept of temperature. Heat, work, and internal energy. The first law of thermodynamics, isothermal and adiabatic processes.

The second law of thermodynamics: reversible and irreversible processes.

UNIT 9: KINETIC THEORY OF GASES

Equation of state of a perfect gas, work done on compressing a gas, Kinetic theory of gases - assumptions, the concept of pressure. Kinetic interpretation of temperature: RMS speed of gas molecules: Degrees of freedom. Law of equipartition of energy and applications to specific heat capacities of gases; Mean free path. Avogadro's number.

UNIT 10: OSCILLATIONS AND WAVES

Oscillations and periodic motion – time period, frequency, displacement as a function of time. Periodic functions. Simple harmonic motion (S.H.M.) and its equation; phase: oscillations of a spring -restoring force and force constant: energy in S.H.M. - Kinetic and potential energies; Simple pendulum - derivation of expression for its time period:

Wave motion. Longitudinal and transverse waves, speed of the travelling wave. Displacement relation for a progressive wave. Principle of superposition of waves, reflection of waves. Standing waves in strings and organ pipes, fundamental mode, and harmonics. Beats.

UNIT 11: ELECTROSTATICS

Electric charges: Conservation of charge. Coulomb's law forces between two point charges, forces between multiple charges: superposition principle and continuous charge distribution.

Electric field: Electric field due to a point charge, Electric field lines. Electric dipole, Electric field due to a dipole. Torque on a dipole in a uniform electric field.





Electric flux. Gauss's law and its applications to find field due to infinitely long uniformly charged straight wire uniformly charged infinite plane sheet, and uniformly charged thin spherical shell. Electric potential and its calculation for a point charge, electric dipole and system of charges; potential difference, Equipotential surfaces, Electrical potential energy of a system of two point charges and of electric dipole in an electrostatic field.

Conductors and insulators. Dielectrics and electric polarization, capacitors and capacitances, the combination of capacitors in series and parallel, and capacitance of a parallel plate capacitor with and without dielectric medium between the plates. Energy stored in a capacitor.

UNIT 12: CURRENT ELECTRICITY

Electric current. Drift velocity, mobility, and their relation with electric current. Ohm's law. Electrical resistance. V-l characteristics of Ohmic and non-ohmic conductors. Electrical energy and power. Electrical resistivity and conductivity. Series and parallel combinations of resistors; Temperature dependence of resistance.

Internal resistance, potential difference, and emf of a cell, a combination of cells in series and parallel. Kirchhoff's laws and their applications. Wheatstone bridge. Metre Bridge.

UNIT 13: MAGNETIC EFFECTS OF CURRENT AND MAGNETISM

Biot - Savart law and its application to the current carrying circular loop. Ampere's law and its applications to infinitely long current carrying straight wire and solenoid. Force on a moving charge in uniform magnetic and electric fields.

Force on a current-carrying conductor in a uniform magnetic field. The force between two parallel currents carrying conductors-definition of ampere. Torque experienced by a current loop in a uniform magnetic field: Moving coil galvanometer, its sensitivity, and conversion to ammeter and voltmeter.

Current loop as a magnetic dipole and its magnetic dipole moment. Bar magnet as an equivalent solenoid, magnetic field lines; Magnetic field due to a magnetic dipole (bar magnet) along its axis and perpendicular to its axis. Torque on a magnetic dipole in a uniform magnetic field. Para-, dia- and ferromagnetic substances with examples, the effect of temperature on magnetic properties.

UNIT 14: ELECTROMAGNETIC INDUCTION AND ALTERNATING CURRENTS

Electromagnetic induction: Faraday's law. Induced emf and current: Lenz's Law, Eddy currents. Self and mutual inductance. Alternating currents, peak and RMS value of alternating current/voltage: reactance and impedance: LCR series circuit, resonance: power in AC circuits, wattless current. AC generator and transformer.

UNIT 15: ELECTROMAGNETIC WAVES

Displacement current. Electromagnetic waves and their characteristics, Transverse nature of electromagnetic waves, Electromagnetic spectrum (radio waves, microwaves, infrared, visible, ultraviolet. X-rays. Gamma rays), Applications of e.m. waves.





UNIT 16: OPTICS

Reflection of light, spherical mirrors, mirror formula. Refraction of light at plane and spherical surfaces, thin lens formula, and lens maker formula. Total internal reflection and its applications. Magnification. Power of a Lens. Combination of thin lenses in contact. Refraction of light through a prism. Microscope and Astronomical Telescope (reflecting and refracting) and their magnifying powers.

Wave optics: wavefront and Huygens' principle. Laws of reflection and refraction using Huygens principle. Interference, Young's double-slit experiment, and expression for fringe width, coherent sources, and sustained interference of light. Diffraction due to a single slit, width of central maximum. Polarization, plane-polarized light: Brewster's law, uses of plane-polarized light and Polaroid.

UNIT 17: DUAL NATURE OF MATTER AND RADIATION

Dual nature of radiation. Photoelectric effect. Hertz and Lenard's observations; Einstein's photoelectric equation: particle nature of light. Matter waves-wave nature of particle, de Broglie relation.

UNIT 18: ATOMS AND NUCLEI

Alpha-particle scattering experiment; Rutherford's model of atom; Bohr model, energy levels, hydrogen spectrum. Composition and size of nucleus, atomic masses, Mass-energy relation, mass defect; binding energy per nucleon and its variation with mass number, nuclear fission, and fusion.

UNIT 19: ELECTRONIC DEVICES

Semiconductors; semiconductor diode: I-V characteristics in forward and reverse bias; diode as a rectifier; I-V characteristics of LED. the photodiode, solar cell, and Zener diode; Zener diode as a voltage regulator. Logic gates (OR. AND. NOT. NAND and NOR).

UNIT 20: EXPERIMENTAL SKILLS

Familiarity with the basic approach and observations of the experiments and activities:

- 1. Vernier calipers -its use to measure the internal and external diameter and depth of a vessel.
- 2. Screw gauge-its use to determine the thickness/ diameter of thin sheet/wire.
- 3. Simple Pendulum-dissipation of energy by plotting a graph between the square of amplitude and time.
- 4. Metre Scale the mass of a given object by the principle of moments.
- 5. Young's modulus of elasticity of the material of a metallic wire.
- 6. Surf ace tension of water by capillary rise and effect of detergents,
- 7. Co-efficient of Viscosity of a given viscous liquid by measuring the terminal velocity of a given spherical **body**,
- 8. Speed of sound in air at room temperature using a resonance tube,
- 9. Specific heat capacity of a given (i) solid and (ii) liquid by method of mixtures.
- 10. The resistivity of the material of a given wire using a metre bridge.
- 11. The resistance of a given wire using Ohm's law.





- 12. Resistance and figure of merit of a galvanometer by half deflection method.
- 13. The focal length of;
 - (i) Convex mirror
 - (ii) Concave mirror, and
 - (ii) Convex lens, using the parallax method.
- 14. The plot of the angle of deviation vs angle of incidence for a triangular prism.
- 15. The refractive index of a glass slab using a travelling microscope.
- 16. Characteristic curves of a p-n junction diode in forward and reverse bias.
- 17. Characteristic curves of a Zener diode and finding reverse break down voltage.
- 18. Identification of Diode. LED,. Resistor. A capacitor from a mixed collection of such items.





CHEMISTRY

PHYSICAL CHEMISTRY

UNIT I: SOME BASIC CONCEPTS IN CHEMISTRY

Matter and its nature, Dalton's atomic theory: Concept of atom, molecule, element, and compound:: Laws of chemical combination; Atomic and molecular masses, mole concept, molar mass, percentage composition, empirical and molecular formulae: Chemical equations and stoichiometry.

UNIT 2: ATOMIC STRUCTURE

Nature of electromagnetic radiation, photoelectric effect; Spectrum of the hydrogen atom. Bohr model of a hydrogen atom - its postulates, derivation of the relations for the energy of the electron and radii of the different orbits, limitations of Bohr's model; Dual nature of matter, de Broglie's relationship. Heisenberg uncertainty principle. Elementary ideas of quantum mechanics, quantum mechanics, the quantum mechanical model of the atom, and its important features. Concept of atomic orbitals as one-electron wave functions: Variation of Ψ and Ψ^2 with r for 1s and 2s orbitals; various

quantum numbers (principal, angular momentum, and magnetic quantum numbers) and their significance; shapes of s, p, and d - orbitals, electron spin, and spin quantum number: Rules for filling electrons in orbitals – Aufbau principle. Pauli's exclusion principle and Hund's rule, electronic configuration of elements, and extra stability of half-filled and completely filled orbitals.

UNIT 3: CHEMICAL BONDING AND MOLECULAR STRUCTURE

Kossel-Lewis approach to chemical bond formation, the concept of ionic and covalent bonds.

Ionic Bonding: Formation of ionic bonds, factors affecting the formation of ionic bonds; calculation of lattice enthalpy.

Covalent Bonding: Concept of electronegativity. Fajan's rule, dipole moment: Valence Shell Electron Pair Repulsion (VSEPR) theory and shapes of simple molecules.

Quantum mechanical approach to covalent bonding: Valence bond theory - its important features, the concept of hybridization involving s, p, and d orbitals; Resonance.

Molecular Orbital Theory - Its important features. LCAOs, types of molecular orbitals (bonding, antibonding), sigma and pi-bonds, molecular orbital electronic configurations of homonuclear diatomic molecules, the concept of bond order, bond length, and bond energy.

Elementary idea of metallic bonding. Hydrogen bonding and its applications.

UNIT 4: CHEMICAL THERMODYNAMICS

Fundamentals of thermodynamics: System and surroundings, extensive and intensive properties, state functions, Entropy, types of processes.

The first law of thermodynamics - Concept of work, heat internal energy and enthalpy, heat capacity, molar heat capacity; Hess's law of constant heat summation; Enthalpies of bond





dissociation, combustion, formation, atomization, sublimation, phase transition, hydration, ionization, and solution.

The second law of thermodynamics - Spontaneity of processes; ΔS of the universe and ΔG of the system as criteria for spontaneity. ΔG° (Standard Gibbs energy change) and equilibrium constant.

UNIT 5: SOLUTIONS

Different methods for expressing the concentration of solution - molality, molarity, mole fraction, percentage (by volume and mass both), the vapour pressure of solutions and Raoult's Law - Ideal and non-ideal solutions, vapour pressure - composition, plots for ideal and non-ideal solutions; Colligative properties of dilute solutions - a relative lowering of vapour pressure, depression of freezing point, the elevation of boiling point and osmotic pressure; Determination of molecular mass using colligative properties; Abnormal value of molar mass, van't Hoff factor and its significance.

UNIT 6: EQUILIBRIUM

Meaning of equilibrium is the concept of dynamic equilibrium.

Equilibria involving physical processes: Solid-liquid, liquid-gas - gas and solid-gas equilibria, Henry's law. General characteristics of equilibrium involving physical processes.

Equilibrium involving chemical processes: Law of chemical equilibrium, equilibrium constants (K_p and K_c) and their significance, the significance of ΔG and ΔG° in chemical equilibrium, factors affecting equilibrium concentration, pressure, temperature, the effect of catalyst; Le Chatelier's principle.

Ionic equilibrium: Weak and strong electrolytes, ionization of electrolytes, various concepts of acids and bases (Arrhenius. Bronsted - Lowry and Lewis) and their ionization, acid-base equilibria (including multistage ionization) and ionization constants, ionization of water. pH scale, common ion effect, hydrolysis of salts and pH of their solutions, the solubility of sparingly soluble salts and solubility products, and buffer solutions.

UNIT 7: REDOX REACTIONS AND ELECTROCHEMISTRY

Electronic concepts of oxidation and reduction, redox reactions, oxidation number, rules for assigning oxidation number, and balancing of redox reactions.

Electrolytic and metallic conduction, conductance in electrolytic solutions, molar conductivities and their variation with concentration: Kohlrausch's law and its applications.

Electrochemical cells - Electrolytic and Galvanic cells, different types of electrodes, electrode potentials including standard electrode potential, half-cell and cell reactions, emf of a Galvanic cell and its measurement: Nernst equation and its applications; Relationship between cell potential and Gibbs' energy change: Dry cell and lead accumulator; Fuel cells.

UNIT 8: CHEMICAL KINETICS

Rate of a chemical reaction, factors affecting the rate of reactions: concentration, temperature, pressure, and catalyst; elementary and complex reactions, order and molecularity of reactions, rate law, rate constant and its units, differential and integral forms of zero and first-order





reactions, their characteristics and half-lives, the effect of temperature on the rate of reactions, Arrhenius theory, activation energy and its calculation, collision theory of bimolecular gaseous reactions (no derivation).

INORGANIC CHEMISTRY

UNIT 9: CLASSIFICATION OF ELEMENTS AND PERIODICITY IN PROPERTIES

Modem periodic law and present form of the periodic table, s, p. d and f block elements, periodic trends in properties of elements atomic and ionic radii, ionization enthalpy, electron gain enthalpy, valence, oxidation states, and chemical reactivity.

UNIT 10: P- BLOCK ELEMENTS

Group -13 to Group 18 Elements

General Introduction: Electronic configuration and general trends in physical and chemical properties of elements across the periods and down the groups; unique behaviour of the first element in each group.

UNIT 11: d - and f- BLOCK ELEMENTS

Transition Elements

General introduction, electronic configuration, occurrence and characteristics, general trends in properties of the first-row transition elements - physical properties, ionization enthalpy, oxidation states, atomic radii, colour, catalytic behaviour, magnetic properties, complex formation, interstitial compounds, alloy formation; Preparation, properties, and uses of $K_2Cr_2O_7$, and $KMnO_4$.

Inner Transition Elements

Lanthanoids - Electronic configuration, oxidation states, and lanthanoid contraction.

Actinoids - Electronic configuration and oxidation states.

UNIT 12: CO-ORDINATION COMPOUNDS

Introduction to coordination compounds. Werner's theory; ligands, coordination number, denticity. chelation; IUPAC nomenclature of mononuclear co-ordination compounds, isomerism; Bonding-Valence bond approach and basic ideas of Crystal field theory, colour and magnetic properties; Importance of co-ordination compounds (in qualitative analysis, extraction of metals and in biological systems).

ORGANIC CHEMISTRY

UNIT 13: PURIFICATION AND CHARACTERISATION OF ORGANIC COMPOUNDS

Purification - Crystallization, sublimation, distillation, differential extraction, and chromatography - principles and their applications.

Qualitative analysis - Detection of nitrogen, sulphur, phosphorus, and halogens.





Quantitative analysis (basic principles only) - Estimation of carbon, hydrogen, nitrogen, halogens, sulphur, and phosphorus.

Calculations of empirical formulae and molecular formulae: Numerical problems in organic quantitative analysis,

UNIT 14:SOME BASIC PRINCIPLES OF ORGANIC CHEMISTRY

Tetravalency of carbon: Shapes of simple molecules - hybridization (s and p): Classification of organic compounds based on functional groups: and those containing halogens, oxygen, nitrogen, and sulphur; Homologous series: Isomerism - structural and stereoisomerism.

Nomenclature (Trivial and IUPAC)

Covalent bond fission - Homolytic and heterolytic: free radicals, carbocations, and carbanions; stability of carbocations and free radicals, electrophiles, and nucleophiles.

Electronic displacement in a covalent bond

- Inductive effect, electromeric effect, resonance, and hyperconjugation.

Common types of organic reactions- Substitution, addition, elimination, and rearrangement.

UNITS 15: HYDROCARBONS

Classification, isomerism, IUPAC nomenclature, general methods of preparation, properties, and reactions.

Alkanes - Conformations: Sawhorse and Newman projections (of ethane): Mechanism of halogenation of alkanes.

Alkenes - Geometrical isomerism: Mechanism of electrophilic addition: addition of hydrogen, halogens, water, hydrogen halides (Markownikoffs and peroxide effect): Ozonolysis and polymerization.

Alkynes - Acidic character: Addition of hydrogen, halogens, water, and hydrogen halides: Polymerization.

Aromatic hydrocarbons - Nomenclature, benzene - structure and aromaticity: Mechanism of electrophilic substitution: halogenation, nitration.

Friedel-Craft's alkylation and acylation, directive influence of the functional group in monosubstituted benzene.

UNIT 16: ORGANIC COMPOUNDS CONTAINING HALOGENS

General methods of preparation, properties, and reactions; Nature of C-X bond; Mechanisms of substitution reactions.

Uses; Environmental effects of chloroform, iodoform freons, and DDT.

UNIT 17: ORGANIC COMPOUNDS CONTAINING OXYGEN

General methods of preparation, properties, reactions, and uses.

ALCOHOLS, PHENOLS, AND ETHERS





Alcohols: Identification of primary, secondary, and tertiary alcohols: mechanism of dehydration.

Phenols: Acidic nature, electrophilic substitution reactions: halogenation. nitration and sulphonation. Reimer - Tiemann reaction.

Ethers: Structure.

Aldehyde and Ketones: Nature of carbonyl group; Nucleophilic addition to >C=O group, relative reactivities of aldehydes and ketones; Important reactions such as - Nucleophilic addition reactions (addition of HCN. NH₃, and its derivatives), Grignard reagent; oxidation: reduction (Wolf Kishner and Clemmensen); the acidity of α -hydrogen. aldol condensation, Cannizzaro reaction. Haloform reaction, Chemical tests to distinguish between aldehydes and Ketones.

Carboxylic Acids

Acidic strength and factors affecting it,

UNIT 18: ORGANIC COMPOUNDS CONTAINING NITROGEN

General methods of preparation. Properties, reactions, and uses.

Amines: Nomenclature, classification structure, basic character, and identification of primary, secondary, and tertiary amines and their basic character.

Diazonium Salts: Importance in synthetic organic chemistry.

UNIT 19: BIOMOLECULES

General introduction and importance of biomolecules.

CARBOHYDRATES - Classification; aldoses and ketoses: monosaccharides (glucose and fructose) and constituent monosaccharides of oligosaccharides (sucrose, lactose, and maltose).

PROTEINS - Elementary Idea of α -amino acids, peptide bond, polypeptides. Proteins: primary, secondary, tertiary, and quaternary structure (qualitative idea only), denaturation of proteins, enzymes.

VITAMINS - Classification and functions.

NUCLEIC ACIDS - Chemical constitution of DNA and RNA.

Biological functions of nucleic acids.

Hormones (General introduction)

UNIT 20: PRINCIPLES RELATED TO PRACTICAL CHEMISTRY

Detection of extra elements (Nitrogen, Sulphur, halogens) in organic compounds; Detection of the following functional groups; hydroxyl (alcoholic and phenolic), carbonyl (aldehyde and ketones) carboxyl, and amino groups in organic compounds.

• The chemistry involved in the preparation of the following:

Inorganic compounds; Mohr's salt, potash alum.





Organic compounds: Acetanilide, p-nitro acetanilide, aniline yellow, iodoform.

- The chemistry involved in the titrimetric exercises Acids, bases, and the use of indicators, oxalic-acid vs KMnO₄, Mohr's salt vs KMnO₄
- Chemical principles involved in the qualitative salt analysis:

$$Cations-Pb^{2+,}\,Cu^{2+,}\,Al^{3+},\,Fe^{3+},\,Zn^{2+},\,Ni^{2+},\,Ca^{2+},\,Ba^{2+},\,Mg^{2+},\,NH_{4}^{+}$$

Anions- CO_3^{2-} , S^{2-} , SO_4^{2-} , $^{NO3-}$, NO^{2-} , Cl^- , Br^- , I^- (Insoluble salts excluded).

Chemical principles involved in the following experiments:

- 1. Enthalpy of solution of CuSO₄
- 2. Enthalpy of neutralization of strong acid and strong base.
- 3. Preparation of lyophilic and lyophobic sols.
- 4. Kinetic study of the reaction of iodide ions with hydrogen peroxide at room temperature.





Syllabus for JEE (Main) Paper 2A (B.Arch.) - Mathematics, Aptitude Test, and Drawing Test

Part - I MATHEMATICS

UNIT 1: SETS, RELATIONS, AND FUNCTIONS:

Sets and their representation: Union, intersection, and complement of sets and their algebraic properties; Power set; Relation, Type of relations, equivalence relations, functions; one-one, into and onto functions, the composition of functions.

UNIT 2: COMPLEX NUMBERS AND QUADRATIC EQUATIONS:

Complex numbers as ordered pairs of reals, Representation of complex numbers in the form a + ib and their representation in a plane, Argand diagram, algebra of complex number, modulus, and argument (or amplitude) of a complex number, Quadratic equations in real and complex number system and their solutions Relations between roots and co-efficient, nature of roots, the formation of quadratic equations with given roots.

UNIT3: MATRICES AND DETERMINANTS:

Matrices, algebra of matrices, type of matrices, determinants, and matrices of order two and three, evaluation of determinants, area of triangles using determinants, Adjoint, and evaluation of inverse of a square matrix using determinants and, Test of consistency and solution of simultaneous linear equations in two or three variables using matrices.

UNIT 4: PERMUTATIONS AND COMBINATIONS:

The fundamental principle of counting, permutation as an arrangement and combination as section, Meaning of P(n,r) and C(n,r), simple applications.

UNIT 5: BINOMIAL THEOREM AND ITS SIMPLE APPLICATIONS:

Binomial theorem for a positive integral index, general term and middle term, and simple applications.

UNIT 6: SEQUENCE AND SERIES:

Arithmetic and Geometric progressions, insertion of arithmetic, geometric means between two given numbers, Relation between A.M and G.M.

UNIT 7: LIMIT, CONTINUITY, AND DIFFERENTIABILITY:

Real-valued functions, algebra of functions, polynomials, rational, trigonometric, logarithmic, and exponential functions, inverse function. Graphs of simple functions. Limits, continuity, and differentiability. Differentiation of the sum, difference, product, and quotient of two functions. Differentiation of trigonometric, inverse trigonometric, logarithmic, exponential, composite, and implicit functions; derivatives of order up to two, Applications of derivatives: Rate of change of quantities, monotonic-increasing and decreasing functions, Maxima and minima of functions of one variable,





UNIT 8: INTEGRAL CALCULAS:

Integral as an anti-derivative, Fundamental integral involving algebraic, trigonometric, exponential, and logarithms functions. Integrations by substitution, by parts, and by partial functions. Integration using trigonometric identities.

Evaluation of simple integrals of the type

$$\int \frac{dx}{x^2 + a^2} , \int \frac{dx}{\sqrt{x^2 \pm a^2}} , \int \frac{dx}{a^2 - x^2} , \int \frac{dx}{\sqrt{a^2 - x^2}} , \int \frac{dx}{ax^2 + bx + c} , \int \frac{dx}{\sqrt{ax^2 + bx + c}} , \int \frac{(px + q)dx}{\sqrt{ax^2 + bx + c}} ,$$

$$\int \frac{(px + q)dx}{\sqrt{ax^2 + bx + c}} \int \sqrt{a^2 \pm x^2} dx , \int \sqrt{x^2 - a^2} dx$$

. The fundamental theorem of calculus, properties of definite integrals. Evaluation of definite integrals, determining areas of the regions bounded by simple curves in standard form.

UNIT 9: DIFFRENTIAL EQUATIONS

Ordinary differential equations, their order, and degree, the, solution of differential equation by the method of separation of variables, solution of a homogeneous and linear differential equation of the type

$$\frac{dy}{dx} + p(x)y = q(x)$$

UNIT 10: CO-ORDINATE GEOMETRY

Cartesian system of rectangular coordinates in a plane, distance formula, sections formula, locus, and its equation, the slope of a line, parallel and perpendicular lines, intercepts of a line on the coordinate axis.

Straight line

Various forms of equations of a line, intersection of lines, angles between two lines, conditions for concurrence of three lines, the distance of a point from a line, co-ordinate of the centroid, orthocentre, and circumcentre of a triangle,

Circle, conic sections

A standard form of equations of a circle, the general form of the equation of a circle, its radius and central, equation of a circle when the endpoints of a diameter are given, points of intersection of a line and a circle with the centre at the origin and sections of conics, equations of conic sections (parabola, ellipse, and hyperbola) in standard forms,

UNIT 11: THREE-DIMENSIONAL GEOMETRY

Coordinates of a point in space, the distance between two points, section formula, directions ratios, and direction cosines, and the angle between two intersecting lines. Skew lines, the shortest distance between them, and its equation. Equations of a line





UNIT 12: VECTOR ALGEBRA

Vectors and scalars, the addition of vectors, components of a vector in two dimensions and threedimensional space, scalar and vector products,

UNIT 13: STATISTICS AND PROBABILITY

Measures of discretion; calculation of mean, median, mode of grouped and ungrouped data calculation of standard deviation, variance, and mean deviation for grouped and ungrouped data.

Probability: Probability of an event, addition and multiplication theorems of probability, Baye's theorem, probability distribution of a random variate,

UNIT 14: TRIGONOMETRY

Trigonometrical identities and trigonometrical functions, inverse trigonometrical functions, and their properties,

Part -II APTITUDE TEST

UNIT - 1 Awareness of persons. Buildings, Materials.

Objects, Texture related to Architecture and Build-environment, Visualizing three-dimensional objects from two-dimensional drawings. Visualizing. Different sides of three-dimensional objects. Analytical Reasoning Mental Ability (Visual. Numerical and Verbal)

UNIT – **2** Three dimensional- perception: Understanding and appreciation of scale and proportions of objects, building forms and elements, colour texture harmony and contrast Design and drawing of geometrical or abstract shapes and patterns in pencil. Transformation of forms both 2D and 3D union, subtraction rotation, development of surfaces and volumes, Generation of plans, elevations, and 3D views of objects, creating two-dimensional and three-dimensional compositions using given shapes and forms.

Part - III DRAWING TEST

Sketching of scenes and activities from memory of urbanscape (public space, market, festivals, street scenes, monuments, recreational spaces, etc.). landscape (riverfronts. Jungle. Gardens, trees. Plants, etc.) and rural life.

To be conducted in a Drawing sheet.

Note: Candidates are advised to bring pencils. Own geometry box set, crasets and colour pencils, and crayons for the Drawing Test





Syllabus for JEE (Main) Paper 2B (B.Planning.) - Mathematics, Aptitude Test, and Planning

Part - I MATHEMATICS

UNIT 1: SETS, RELATIONS, AND FUNCTIONS:

Sets and their representation: Union, intersection and complement of sets and their algebraic properties; Power set; Relation, Type of relations, equivalence relations, functions; one-one, into and onto functions, the composition of functions.

UNIT 2: COMPLEX NUMBERS AND QUADRATIC EQUATIONS:

Complex numbers as ordered pairs of reals, Representation of complex numbers in the form a + ib and their representation in a plane, Argand diagram, algebra of complex number, modulus, and argument (or amplitude) of a complex number, Quadratic equations in real and complex number system and their solutions Relations between roots and co-efficient, nature of roots, the formation of quadratic equations with given roots.

UNIT3: MATRICES AND DETERMINANTS:

Matrices, algebra of matrices, type of matrices, determinants, and matrices of order two and three, evaluation of determinants, area of triangles using determinants, Adjoint, and evaluation of inverse of a square matrix using determinants and, Test of consistency and solution of simultaneous linear equations in two or three variables using matrices.

UNIT 4: PERMUTATIONS AND COMBINATIONS:

The fundamental principle of counting, permutation as an arrangement and combination as section, Meaning of P(n,r) and C(n,r), simple applications.

UNIT 5: BINOMIAL THEOREM AND ITS SIMPLE APPLICATIONS:

Binomial theorem for a positive integral index, general term and middle term, and simple applications.

UNIT 6: SEQUENCE AND SERIES:

Arithmetic and Geometric progressions, insertion of arithmetic, geometric means between two given numbers, Relation between A.M and G.M.

UNIT 7: LIMIT, CONTINUITY, AND DIFFERENTIABILITY:

Real-valued functions, algebra of functions, polynomials, rational, trigonometric, logarithmic, and exponential functions, inverse function. Graphs of simple functions. Limits, continuity, and differentiability. Differentiation of the sum, difference, product, and quotient of two functions. Differentiation of trigonometric, inverse trigonometric, logarithmic, exponential, composite, and implicit functions; derivatives of order up to two, Applications of derivatives: Rate of change of quantities, monotonic-increasing and decreasing functions, Maxima and minima of functions of one variable,





UNIT 8: INTEGRAL CALCULAS:

Integral as an anti-derivative, Fundamental integral involving algebraic, trigonometric, exponential, and logarithmic functions. Integrations by substitution, by parts, and by partial functions. Integration using trigonometric identities.

Evaluation of simple integrals of the type

$$\int \frac{dx}{x^{2} + a^{2}} , \int \frac{dx}{\sqrt{x^{2} \pm a^{2}}} , \int \frac{dx}{a^{2} - x^{2}} , \int \frac{dx}{\sqrt{a^{2} - x^{2}}} , \int \frac{dx}{ax^{2} + bx + c} , \int \frac{dx}{\sqrt{ax^{2} + bx + c}} , \int \frac{(px+q)dx}{\sqrt{ax^{2} + bx + c}} , \int \frac{(px+q)dx}{\sqrt{ax^{2} + bx + c}} , \int \frac{(px+q)dx}{\sqrt{ax^{2} + bx + c}} , \int \frac{dx}{ax^{2} + bx + c} , \int \frac{dx}{ax^{2} +$$

. The fundamental theorem of calculus, properties of definite integrals. Evaluation of definite integrals, determining areas of the regions bounded by simple curves in standard form.

UNIT 9: DIFFRENTIAL EQUATIONS

Ordinary differential equations, their order, and degree, the solution of differential equation by the method of separation of variables, solution of a homogeneous and linear differential equation of the type

$$\frac{dy}{dx} + p(x)y = q(x)$$

UNIT 10: CO-ORDINATE GEOMETRY

Cartesian system of rectangular coordinates in a plane, distance formula, sections formula, locus, and its equation, the slope of a line, parallel and perpendicular lines, intercepts of a line on the coordinate axis.

Straight line

Various forms of equations of a line, intersection of lines, angles between two lines, conditions for concurrence of three lines, the distance of a point form a line, co-ordinate of the centroid, orthocentre, and circumcentre of a triangle,

Circle, conic sections

A standard form of equations of a circle, the general form of the equation of a circle, its radius and central, equation of a circle when the endpoints of a diameter are given, points of intersection of a line and a circle with the centre at the origin and sections of conics, equations of conic sections (parabola, ellipse, and hyperbola) in standard forms,

UNIT 11: THREE-DIMENSIONAL GEOMETRY

Coordinates of a point in space, the distance between two points, section formula, directions ratios, and direction cosines, the angle between two intersecting lines. Skew lines, the shortest distance between them, and its equation. Equations of a line





UNIT 12: VECTOR ALGEBRA

Vectors and scalars, the addition of vectors, components of a vector in two dimensions and threedimensional space, scalar and vector products,

UNIT 13: STATISTICS AND PROBABILITY

Measures of discretion; calculation of mean, median, mode of grouped and ungrouped data calculation of standard deviation, variance, and mean deviation for grouped and ungrouped data.

Probability: Probability of an event, addition and multiplication theorems of probability, Baye's theorem, probability distribution of a random variate,

UNIT 14: TRIGONOMETRY

Trigonometrical identities and trigonometrical functions, inverse trigonometrical functions, and their properties,

Part -II APTITUDE TEST

UNIT - 1 Awareness of persons. Buildings, Materials, Objects, and Textures related to Architecture and Build-environment, Visualizing three-dimensional objects from two-dimensional drawings. Visualizing. Different sides of three-dimensional objects. Analytical Reasoning Mental Ability (Visual. Numerical and Verbal)

UNIT – **2** Three dimensional- perception: Understanding and appreciation of scale and proportions of objects, building forms and elements, colour texture harmony and contrast Design and drawing of geometrical or abstract shapes and patterns in pencil. Transformation of forms both 2D and 3D union, subtraction rotation, development of surfaces and volumes, Generation of plans, elevations, and 3D views of objects, creating two-dimensional and three-dimensional compositions using given shapes and forms.

Part - III PLANNING

UNIT-1 GENERAL AWARENESS

General knowledge questions and knowledge about prominent cities, development issues, government programs, etc.

UNIT-2 SOCIAL SCIENCES

The idea of nationalism, nationalism in India, pre-modern world, 19th-century global economy, colonialism, and colonial cities, industrialization, resources, and development, types of resources, agriculture, water, mineral resources, industries, national economy; Human Settlements

Power-sharing, federalism, political parties, democracy, the constitution of India





Economic development- economic sectors, globalization, the concept of development, poverty; Population structure, social exclusion, and inequality, urbanization, rural development, colonial cities,

UNIT-3 THINKING SKILLS

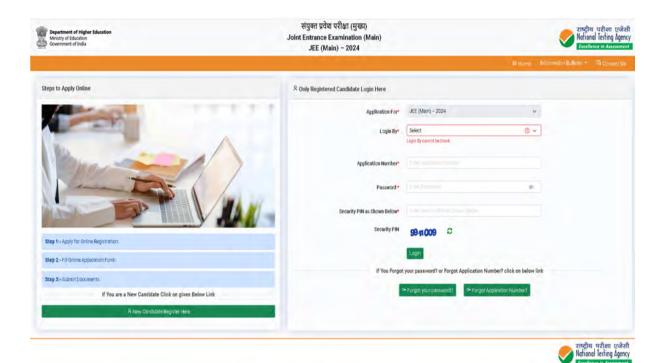
Comprehension (unseen passage); map reading skills, scale, distance, direction, area, etc.; critical reasoning; understanding of charts, graphs, and tables; basic concepts of statistics and quantitative reasoning.





Appendix - VII

Replica of Online Application Form of JEE (Main)-2024 Examination





Information Bulletin: Joint Entrance Examination (Main) - 2024



संयुक्त प्रवेश परीक्षा (मुख्य) Joint Entrance Examination (Main) Department of Higher I Ministry of Education Government of India JEE (Main) - 2024 JEE (Main) - 2024 अतिलाइन अवेदन पत्र सर्वास्थल के लिए निर्देश और प्रक्रिया Cownload Information (full-timingno)). Please ensure your eligibility as per the criteria field down for JEE(Main), Ji 6. Examination Fee: The Fee for JEE(Main) - 2024 per Session (it as follows: sown for JEE(Main), JEE (Advanced), and periscipating State institutions (se applicable), न्याय मुनिवित करें कि आम JEE (Main), JEE (Advanced) और प्रोप्त गाम संकारों के लिए विमीरित सामार्थ के असुमार प्राप्त है For Centers In India (Fee in ₹) 1000/-800/-Outside India (Fee in ₹) General Paper 1: B.E./B.Tech OR Paper 2A B.Arch OR Paper-2B: B.Planning Gen-EWS/ OBC(NCL) SC/ST/PwD Paper 1: B.E.E.Tech and
Paper 2A: B.Arch
OR
Paper 1 B.E.E.Tech and
Paper 2B: B.F.E.Tech and
Paper 2B: B.F.E.Tech and
Paper 2B: B.F.E.Tech, Paper 2A: B.Arch and Paper 2B: B.Flanning
Paper 2A: B.Arch and Paper 2B: B.Flanning Third Gender Male 1000/pper 1: B. E. Fro. 18mm regression of the Paper 2B : B. Planning

pper 2A: B. Arch and Paper 2B : B. Planning

and Goods E. Service Taxes (GST) are to be paid by the candidate, as applicable. ferrole Third Gender Processing charges and Goods & Service Tasses (1971) or to one power year and an application Processing. Hope to be followed to apply confidenced to apply the first processing and the second processing and the second to apply the second to (Ref Uniform Amenda Media (Amenda Media (Amenda Media) (Amenda M Sing 2. Application Form :
The Candidates on log in with the system-generated Application Humber and pre-created Password for completing the Application Form including filling up of personal details, applying for the Paper colorating the Exemination Cities, providing the details of Equipment Qualifications, and upbracking the images and documents (if any).

Sufficient application (agriculture) (agricult l. The recent photograph should be either in colour or black & white with 80% face (without mask) visible including ears against a white background, जात की फोटो पर ले रंग या काले और सफेत में होनी फाड़िए, विवास 80% सेवृत (माका के किन) और बान सफेद पूर्व के बाथ | होता कर प्रधान के ता रंग के करता कर Free smould be in JPG/JPEG former (clearly legisle होजन की गई तहरीर और इसत दल सेवीजी अंग्रेडीईसी प्रारम्य (करहा रूपम से सुप्रवास) में होने चाड़िए। Bit. The size of the assumed photograph about be between 10 kb to 200 kb (clearly legibles) কৰিন জৰি প্ৰতিষ্ঠা জন কৰাৰ 16 design of the size of the s V. Size of the spanned copy of PWD pertificate between 10 io to 300 io (classly legible). PWD प्रमाणक की स्क्रेन की गई कॉकी का अरकार 10 के से से 200 के से क क्षेत्र (स्पष्ट रूप से सुचारण) Note. The Candidate has to uplosed only latal her own photograph, algoritum, and certificate(s) as monitored above (end not of anyoody ales) in a correct/proper manner, the facility for correction will not be given in the future, by case, it is found at any time in the future that the Candidate how used/uploaded the protograph, algoritum, and certificate(s) of someone clear in his/har application. From Application From Applicat बंदी के कि पूर्व है विकास भी दिन कारान Sep 1: Fee Payment, After completely Diep 1 and Step 2, the condistants have to pay the imposite examination fee. The fire can be examinated only order through like Benium, O not Card, Debt Card, UPL or Paymin Services. Proceeding charges and SST as applicable and charges are considered in a condistant in leak by the condistant and the complete of the condistant in the condistant in leak by the condistant in leak by the condistant and the condistant in leak by the B. For subsequent logins, the candidate will be able to login directly with their respective system-generated Application Number and the choisin Pleasw बाद के लागिन के लिए, उम्मीदबार अपने बंबविल किस्ताअनमें टेड एसिकोचल नंबर और गुने हुए प्रशावत के साथ गाँच लागिन करने में सड़म होंगे। BL The candidate is advised not to disclose or share their password with anyboop. Neither NTA for NIC will be responsible for the violation or misuse of the password of a candidate जम्मीदबार को सामा द जाती है कि ने अन्य भागवर्त किसी के साथ म बाहर न करे। किसी उम्मीदबार को समाव द के उल्लाम मा दुरुपांग के लिए एनटीए सिमीवर नहीं होगा। fV. Candidates can change his/her passwords after login (f desired. उम्मीदवार वाहें ले लागिन के बाद अपन प्रस्तव हैं बदल सकते हैं) V. Candidates should remember to log out at the end of their session so that the particulars of the candidate cannot be temperad with or modified by unauthorized persons. उम्मीदवारों को अपने कहा के आप में लोग अपन करना यह रखन चाहिए लांकि अनीयकृत कार्बिकों द्वारा उम्मीदवार के विकास के क्या के कुछाद य संयोधन न किया जा बाके। ন্ত্ৰমানুৱাৰ কা কৰে কয় ক কৰা বা লাগ নাব কৰে কৰে। The Password must be as per the following Pass I. Password must be **8 to 13 sharacters long.** যোগাৰ্ম্ভ ক के 13 Mar পাৰা হুণিদ পান্তিব। B. Password must have at least one Upper case lett पास्त्र में बम से क्षम एक अपरकेश अक्षर होना चाहिए। BL Passaword must have at least one Lower case let पासवर्ड में कमारे कम एक लोअर केस अक्षर होना चाहिए IV. Password must have at least one numeric value. भाषत हैं में कम से कम एक संस्थानक मान होना चाहिए. V. Plassword must have at least one special character eg.Mp#5%^&*-पासवर्ड में क्या से क्या एक विदेश अक्षर होना चाहिए विसे ! tgpaSs-&*-Here for ready over Passwood: The following options are available to meet Passwood.

I. Using Security Question and the American you have chosen outling Form Milling.

With Life to Select the American and the American Country of the Country Form Milling.

With Life to Select and the Country of the Country of the Country form a startiful read to Select and the Country of the Coun Mt. Using a reset link sem vis Email to your Registered Email address. अपके पंजीकृत देमेंल पते पर ईमेल के माध्यम से भेजे गए रीसेट तिक का उपयोग करन টি. The Absharkon Number printed on the computer-generated Confirmation Page must be membored in all such correspondences. It is threefore resemble to note cown the spokestion number printed on the Confirmation Page must be membored in all such correspondences. It is threefore resemble to note cown the spokestion number printed on the Confirmation Page.

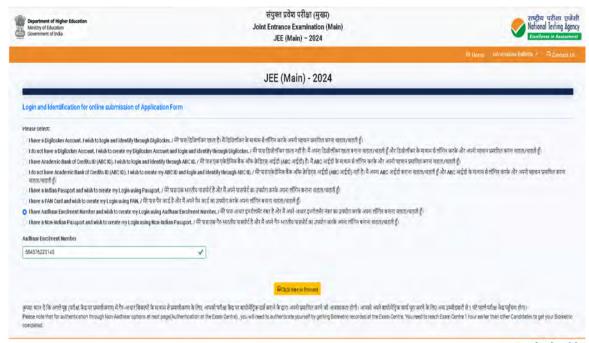
2 के प्रतिकृति प्





Information Bulletin: Joint Entrance Examination (Main) - 2024















Department of Higher Education Ministry of Education Oovermant of India Oovermant of India	Joint Entrance Examination (Main) JEE (Main) – 2024	National Testing Agenc
	one friends and	G (Limit - Moleculative Delivine * - Fi Constant (In
CF (UDDISTRUCT) Interestination Details		
Andhaar Enrolment Number:	58457622114b	
GP (20 Service Portion) Personnel Distante.		
्राच्या कार्यों वो सम्बाद्धा सकेरीत के अनुवारी As per class bible orthogone कार्यावर्त		Y
হিল্প / এবিনেজের বা নাম (Tather / Goardian Narre)* (তথ্য 10টি বী দলানা/মার্কবিট ঠ প্রসূপে Au per casa 1th certificables report		Y
দলে / এবিদনকজ জা নাম (Mother / Guardian Name)* (বৰু চাই বী চালাবে লাইবাঁট ই নপুলা। As per casa bits certificate reserved)	RENU	v
once of all over DEF (Conditioner) time of Nichola	16 🗸 🗸 Merch 🗸 🗸 2000	V +
(বৰ) চাই বী চন্দ্ৰভাগৰ বিষয়ে সনুসাৰ As on class title certification in succeed) বিশ্ব (Genden)	Male	√ ∝
यहात के प्रकार (identity Type)*		V v
ব্যাহার বিল্লা (Identification Number)*		*
ap and seeds (potentials arises reserves).	COMPLY (SECTION)	
(FCER UC) Present Address		
≒it (Pretrises No/ Name)*	PARTAPUR MEERUT	
हार-वृजनीयता (Sub-Locality)/Optional)		
स्थानियात (Locality)*	PHRIDAPUR	~
\$10 (Country)*	Initian	V ~
राज्य (State)	UTDARFRACEM	✓ ∀
The abouters		~ ~
पिन सीड (Pin Crafe)*	720161	Y
भी बाह्य समय (Mit bille Neurolius) *		V
shripn 447 til 195 að (Continn Mobile Hamber)*	R433481005	~
ट्रीत पतः (Ernall Address)*		
ईमेल पते की पुष्टि करें (Ocalism Email Address)*	het@yaprail.com	· ·
ক্টৰংখিক লথক বৃং (Alternate Contact No.) (Optional)		
C AND THE STATE OF		
© (PUE) (FIII) Permanent Address Dame As Present Address		
Will (Premises No./ Name)*	PARTAPURMEERUT	~
ुष्य-ग्रहानीयल (Sub-Locality)(Optional):	Processing and the second	
स्वर्गपत (Locality)	DATTIEVE	~
go (constr);		V + 1
	UTTAR PRACESH	V V
राज (state)*		
किल (District)*		✓ ~
ियन गरी.5 (Pin Code)(if you do not have PIN Code/ZIP Code, you can ful C00000)*	250103	~
Gr Chinne Paysward		
3files (Panaword)		+4
कुरिकास की पुष्टि करिकेट (continue Password)*		w.V
जुन्धि Fit (Decumity Question)*		
सृहक्। अवाब (Security Answer)*	163	V
Dir Security Pili		
Enter Security Pin(Case Sensitive)*	4179801	V
rance occount customs occurrent.	40 98-01 C 4179831	7.0
	III huber	



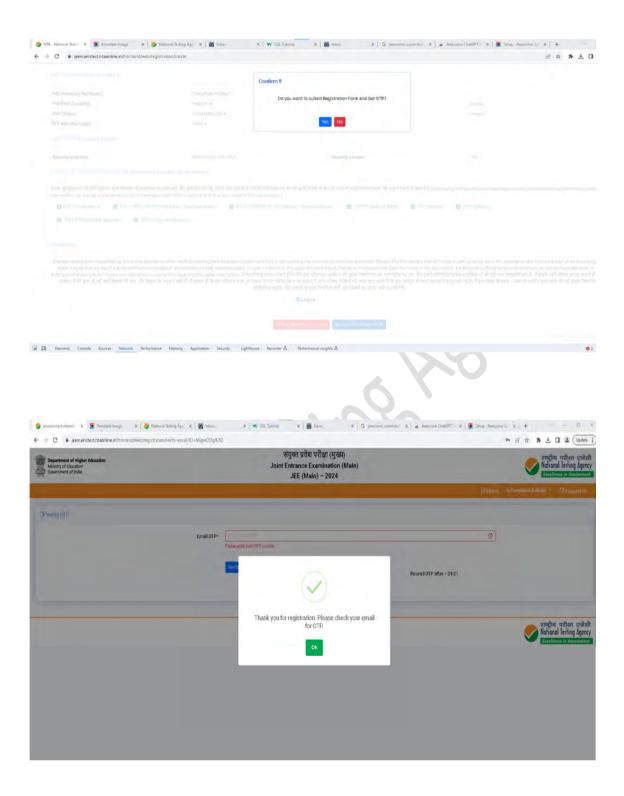




	Servey the following particulars or	erefully: If you would like to change any particulars entered.	
or (Price) (Personal Dotale		KATEON / GRAY button or press TrapART and rever liver button	
अभ्यर्शी का नाम (Candidate's Name)	HARSH CHICARA	अभवनी की जन्म शिनि (Candidateh Data of Birth)	19-03-2001
येता / अभिभाषक का नाम (Father / Quardian Name)	TOGERS	माता / अधिभावक का नाम (Mother / Guardian Name) पहचान रोख्या (Identification Number)	964576223145
हापान के प्रकार (identity Type) तेग (Gender)	Marie - and Goy's Identity with photograph	पहचान तस्या (identification Number)	594578223145
	Mare		
Name (City) Present Address			
SIII (Premises No/ Name)	PARIAPUB MERBUT	्राप-स्थानीपता (Sub-Locality)(Optional)	
मानीपता (Locality)	FARTAPUR	čiri (Country)	Mi04A
T-FI (State)	UTTAR PRACESH	Fires (District)	Meirut
चित्र कोट (Pin Code)	250103	इंसल पर्छ (Email Address)	ht/gychel.com
रोबाट्स नेबर (Mobile Number)	91 6433461006	वैक्रीनेफ लंगके तुम (Alternate Contact No.) (Optional)	
(1 30) Perminent Address			
eभ्यं वी का नाम (Cundidate's Name)	HARSH CHIKARA	अभ्यनी की जन्म तिथि (Candidate's Date of Birth)	19 03-2000
रेश / अधिभाषक का नाम (Father / Quardian Name)	YORKS4	Rick / officerow do HH (Mother / Guardian Name)	RMA)
spare is user (Identity Type)	any Other Valid Gold Identity with photograph	Signat delli (identification Number)	5845/67/1145
Hrt (Gender)	Male		
HE IVE Propert Address			
RII (Premises No/ Name)	FARTAFUR MECRUT	्राय-रणानीपता (\$ub-Locality)(Optional)	
मानीयरा (Locotty)	PARTAPUR	©TI (Country)	MEM
(State)	UTTAR PRACEICH	Press (Distance)	Marrie
िन कोंड (Pin Code) विवादत नेकर (Mobile Number)	7351CA 91-9433481008	ਹੁੰਦੇਰ ਪਹਾਂ (Email Address) ਰੈਲੀ-ਪੋਧਾ ਵੱਧਾਂਕੇ ਦੂਸ (Alternate Contact No.) (Optional)	hc1@yesevalindes
	Anadesistand	describe that dis (viceures courses us?) (obtions)	
nt/11(1) Permanend Address.			
प्रभागी का नाग (Candidate's Name)	HARSH CHRARA	अभागी की जन्म विभि (Candidate's Date of Birth)	16-00-2000
पेशा / अधिभाषक का नाम (Father / Guardian Name)	1996514	साराष्ट्र / अभिकारण का नाम (Mother / Guardian Name)	NON/
aputer (is towarz (identity Type)	Any Other Velid Govt Identity with photograph	पहुन्तन शेलार (identification Number)	56A876222145
तेग (deader)	Many		
(1-479 Series Present Address			
Ed (Premines No/ Name)	IMBITATUR MERILIF	(PE-PRI-BUILI (Sub-Locality)(Optional)	
मा-विपता (Locality) तन्त्र (State)	PORTOPUN LITTAR PRADEZHI	CVI (Country) Plets (District)	Meets Meets
मेन करिट (Min Code)	250103	ÇRONIII (Email Address)	holgyaneloon
रीबाट्स नंबर (Mobile Number)	91-8423461005	वैकल्पिक अंचके सुन्न (Alternate Contact No.) (Optional)	35.97000000
	31 33331100	3,77.12 3.12 4.12 3.12 3.12 3.12 3.12 3.12 3.12 3.12 3	
Tell Von Personnell Address			
and का नाम (Candidate's Name)	HARSH CHIKARA	अध्यर्थी की जन्म तिथि (Candidates Date of Birth)	15-03-2000
रेता / अधिभाषक का नाम (Father / Quardian Name)	YORKSH	माता / अभिरक्षाचक का नाम (Mother / Buardian Name)	property.
ig पान के प्रकार (identity Type) तेम (Gender)	Any Dines Valid Gord Edentry with photograph	पश्चान संस्था (identification Number)	M45/5Z23145
	Mile		
(vivi) of the contract of the			
Fill (Premises No/ Name)	PARTAPUR MICIPAT	उप-रण्ड-विपता (Swb-Locality)(Options()	
न्या-विषया (Locality)	PARTAPUR	ेच-रच-तपता (Sue-Locality)(Optional) देश (Country)	Incia
FVF (State)	LITTAR FRACESH	Firm (District)	Mescul
पेन को स (Pin Code)	200,03	(ilia vin (Email Address)	NOT ENDOWNER CORE
रीवाद्या नेवर (Mobile Number)	91-8455451006	वैक्टियक लंधके सुप्र (Alternate Contact No.) (Optional)	
W/TTLIT! Perconnent Andrens			
assid as any reaching a second	Maleria Piero APA	averall all using EMA consequences	16/05/0000
स्थानी का शाम (Candidate's Name) वेटा / अधिकायक का शाम (Father / Guardian Name)	HARPICENCARA YORGSIA	anarell की जन्म शिर्त (Candidate's Date of Birth) मामा / अभिनामच्या का नाम (Mother / Guardian Name)	16-03-9000 REPAJ
प्रत / अधिभागविक का नाम (Father / Guardian Name) प्रभाग के अकार (Identity Type)	Any Other Yeld Gord Identity with photograph	सास / अधिकारक का नाम (Mother / Guardian Name) पर पान राज्या (identification Number)	98.57.6222145
Bri (Gender)	Male	- A	THE PROPERTY CONT.
SI A SI CONT PROGRAM Address			
EII (Premines No/ Name)	BARIAPUR MERRUT	्राप-रणा-विपता (Sub-Locality)(Optional)	
णा-विपत्त (Locality)	FARTAPLE	Str (Country)	MUDAN
(State)	LITTAR FRADEIN	Silvers (Disasters)	Menut
bit sits (Pin Oode)	289100	Çibe पता (Email Address)	holgyapmakom
रीकट्टा नेवर (Mobile Number)	VI-8403461006	distribute of a (Alternate Contact No.) (Optional)	
(CVVI) Perminent Athlines			
The state of the s			
and a monday was	August and	month of train 10th income	44 00 0000
श्यर्थी का नाम (Candidate's Name) रेश / अभिभाषक का नाम (Father / Operation Name)	HARSH CHICARA YOGSOH	ক্রাকর্ণী ক্রী ক্রাক্টার্কি (Candidate's Date of Birth) মান্ত্রা / ক্রীক্ষাক্রক কা নাম (Mother / Guardian Name)	16-03-2000 Minu.
रता / जाम भाषक का नाम (Father / Guardian Marie) व्हपान के प्रकार (Identity Type)	Any Other Valid Govt Identity with photograph	गात / जा पंचाय का नाम (Mother / Guardian Name)	M45/6723145
dri (Gender)	Mare:		_,
SHE III Punant Adapan			
RII (Premises No/ Name)	PARTAPUR MECRUT	्रय-स्थानीयमा (Sub-Locality)(Optional)	
मानीयता (Locality)	FARTAPUR	ČTI (Country)	MDEA
FPI (State)	UTTARFRADEDH	Flore (District)	Meesil
it-s with (Pin Code)	260163	(film tull (Email Address)	PC1@ystatrielicom
रोक्ट्स नेवर (Mobile Number)	91-8435431006	वैक्शियक लेगके तुप (Alternate Contact No.) (Optional)	
(ATT (AT) Perconnect Address			
	attention finishment and	and the second s	1.2
		देश (Country)	INCIA
भा-तिपता (Locality)	FARTAFUR		
Perfect (Localty)	LITTARFRADESH	Flore (District)	Meenit
भा-तिपता (Locality)			

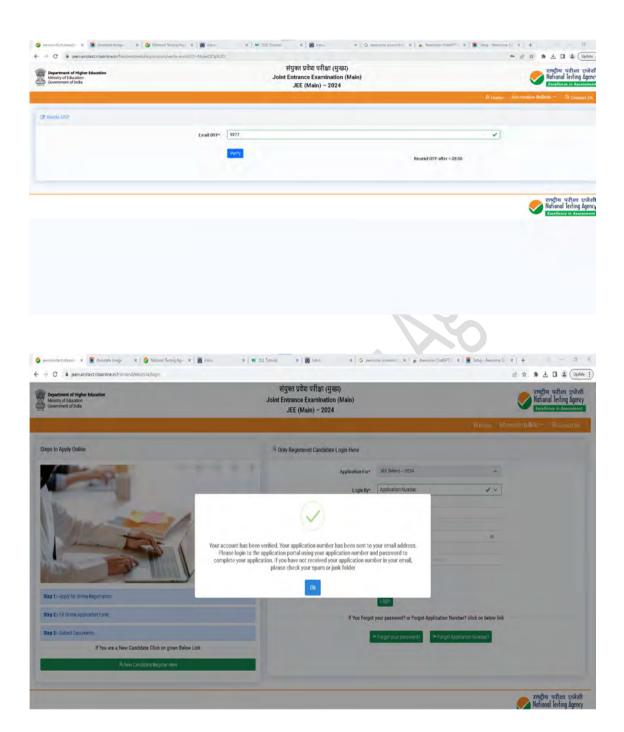






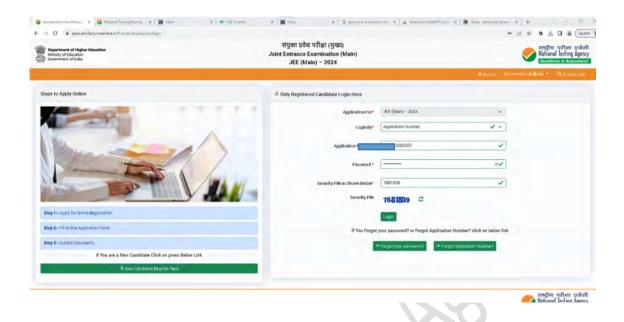


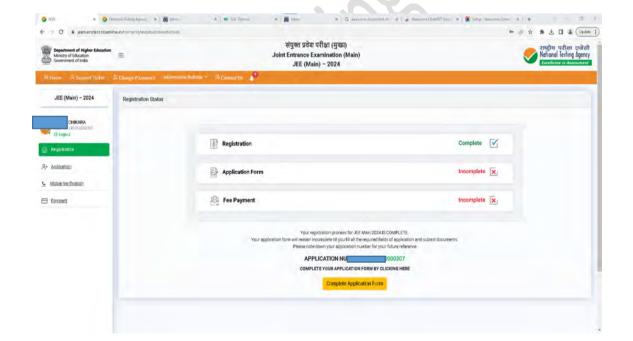






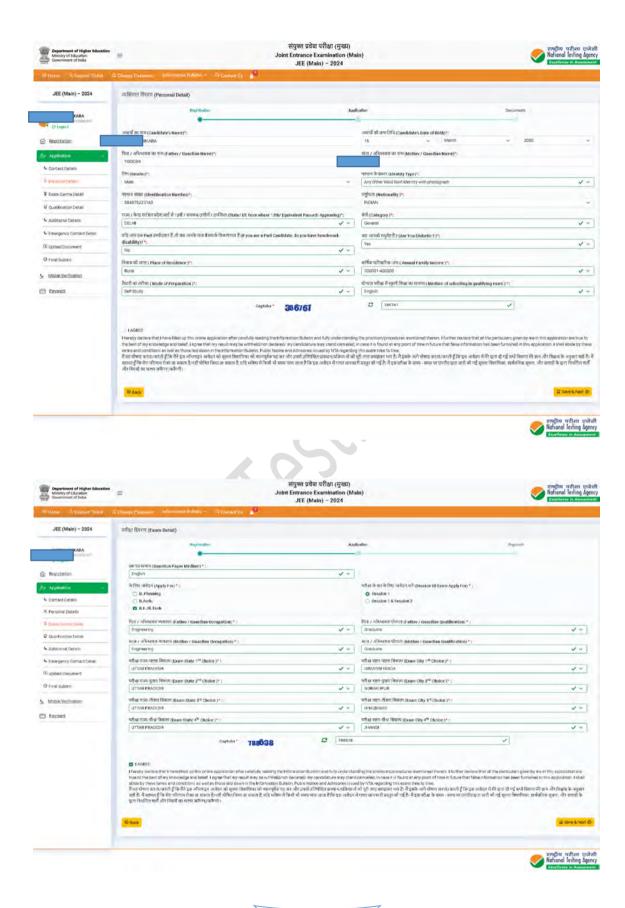






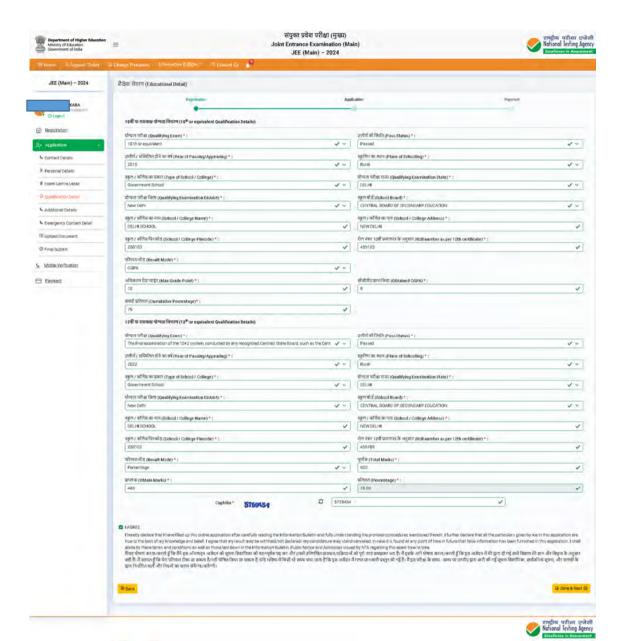






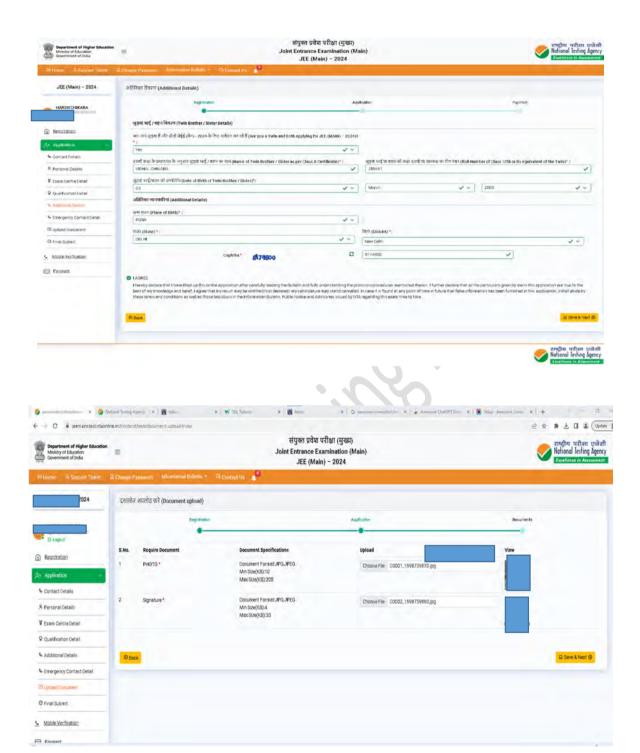






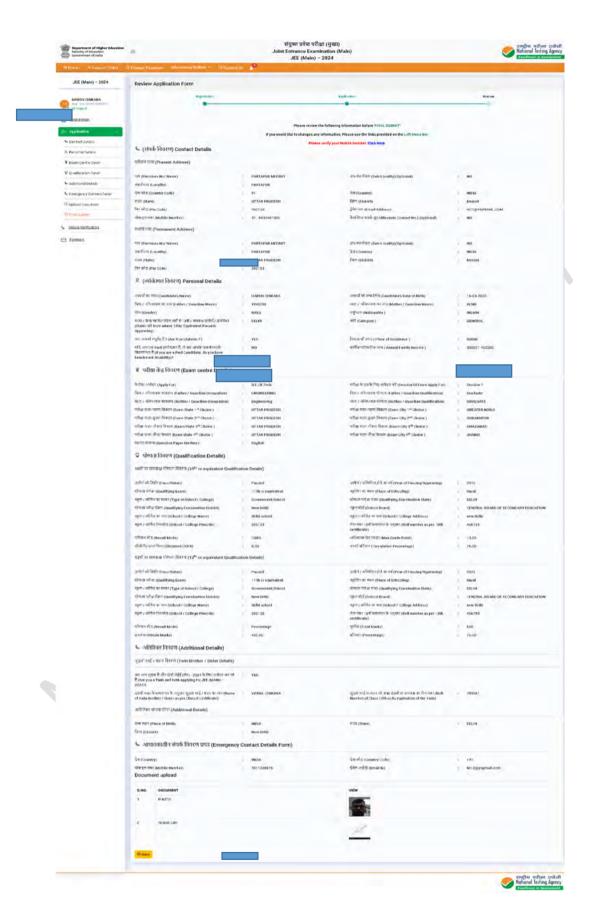








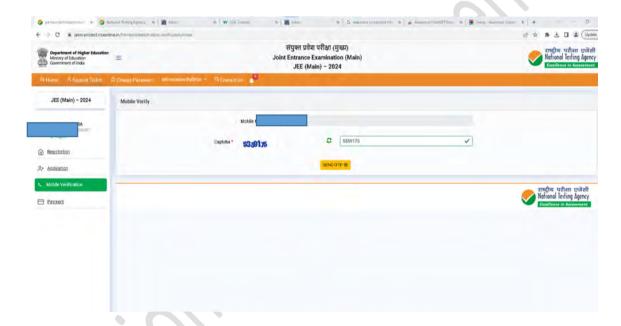






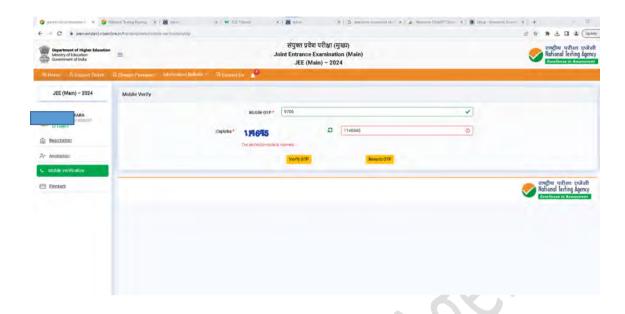


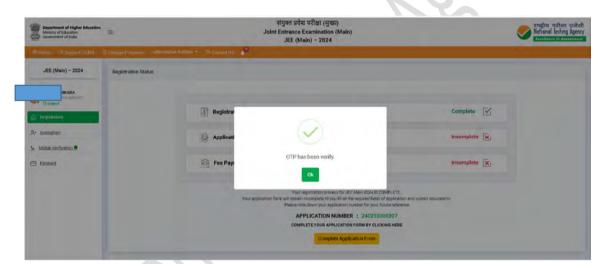






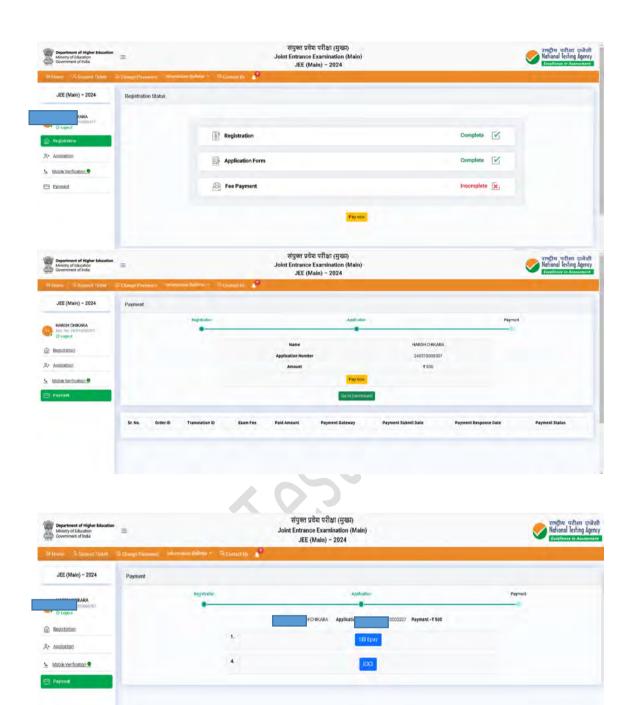






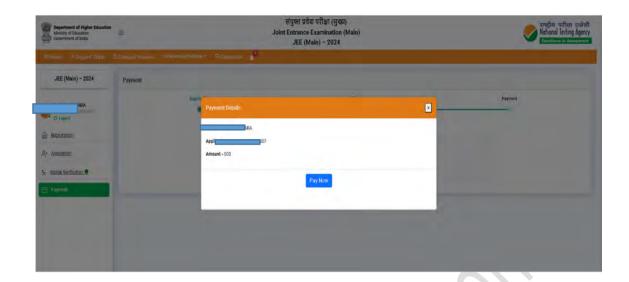


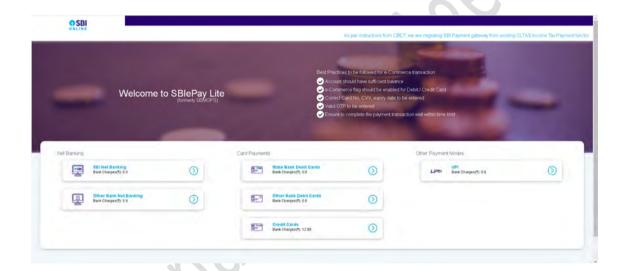






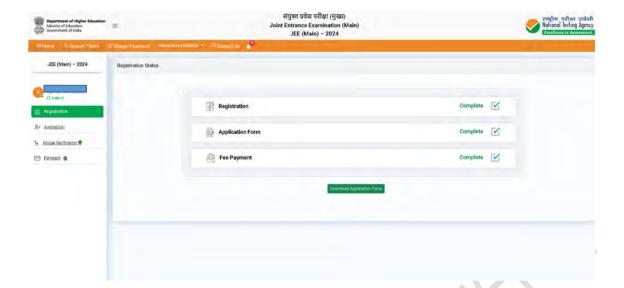






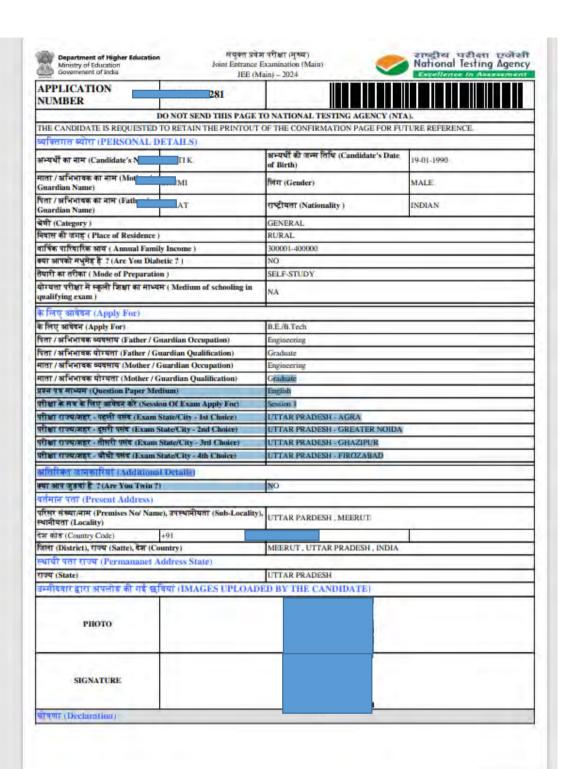












I declare that information furnished by me in the application form is true in all respect and nothing is concealed. In case any entry or information is found to be false, this shall entail automatic cancellation of my admission besides rendering me liable to such actions as deemed fit by the University. I hereby undertake that I have carefully gone through the eligibility conditions prescribed in the prospectus for the programme. I am applying after satisfying that I meet the eligibility conditions. If at any stage it is found that I do not fullfill the minimum prescribed criteria, my admission, if granted, shall stand cancelled and I shall have no right to admission whatsoever.

IP Address : 125.63.102.166

Date of Downloading: 31-10-2023 10:58:33 AM









011-40759000

INFORMATION BULLETIN



National Testing Agency

(An autonomous organization under the Department of Higher Education, Ministry of Education, Government of India)

Address for Correspondence First Floor, NSIC-MDBP Building, Okhla Industrial Estate, New Delhi - 110020

Help Line: For Technical support, contact following during working days between 10.00 a.m. to 5.00 p.m.

www.nta.ac.in http:// jeemain.nta.ac.in

e-mail: jeemain@nta.ac.in