Solving the phase boundary between two solid phases via the common tangent procedure

David C. de Busturia

Department of Chemistry. Imperial College London

Saturday 9th June, 2018

Outline

E(V), $E(V) + E_{ZP}$, F(V; T) and pressure of transition. Thermal evolution

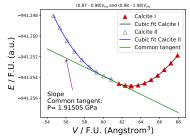
Phase Boundary

Outline I

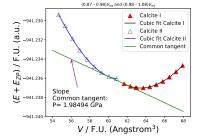
$$E(V)$$
, $E(V) + E_{ZP}$, $F(V; T)$ and pressure of transition. Thermal evolution

Phase Boundary

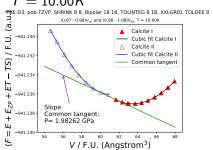
PBE-D3, pob-TZVP, SHRINK 8 8, Bipolar 18 18, TOLINTEG 8 18, XXLGRID, TOLDEE 8



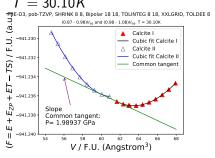
PBE-D3, pob-TZVP, SHRINK 8 8, Bipolar 18 18, TOLINTEG 8 18, XXLGRID, TOLDEE 8



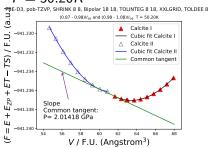
T = 10.00 K



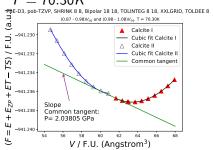
T = 30.10K



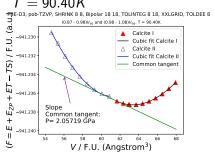
T = 50.20 K



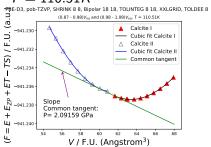
T = 70.30K



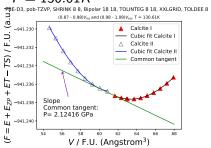
T = 90.40K



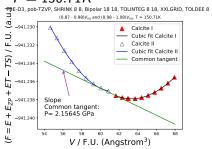
T = 110.51K



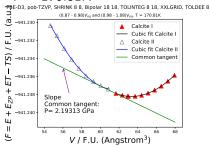
T = 130.61K



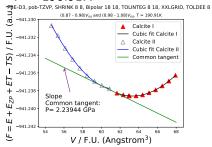
T = 150.71K



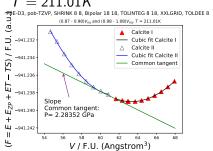
T = 170.81K



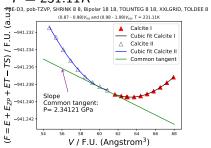
T = 190.91K



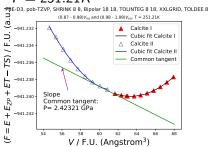
T = 211.01K



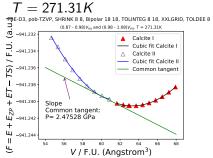
T = 231.11K



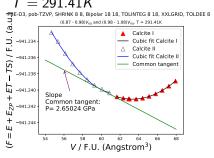
T = 251.21K



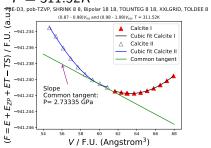
T = 271.31K



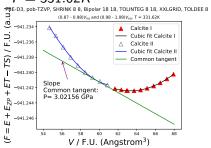
T = 291.41K



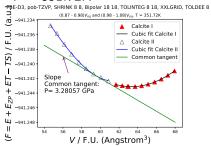
T = 311.52K



T = 331.62K



T = 351.72K



Outline I

$$E(V)$$
, $E(V) + E_{ZP}$, $F(V;T)$ and pressure of transition. Thermal evolution

Phase Boundary

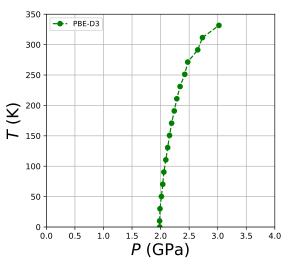


Figure 1: Pressure-temperature phase boundary