

① Summary - Introduction to nuclear and particle physics

basic definitions:

${}^A_Z X_N$ A - atomic mass number ← based on ${}^{12}\text{C} = 12 \text{ u}$
Z - number of protons - atomic number
N - number of neutrons

Isotopes: a chemical element with different A and N but the same number of protons

- similar chemical behaviour
- but different physical behaviour

e.g. ${}^1\text{H}$, ${}^2\text{H}$, ${}^3\text{H}$, ${}^{12}\text{C}$, ${}^{13}\text{C}$, ${}^{14}\text{C}$, ${}^{35}\text{Cl}$, ${}^{37}\text{Cl}$, ${}^{235}\text{U}$, ${}^{238}\text{U}$

Isotone: same number of N but different Z ${}^2\text{H}$ & ${}^3\text{He}$

Isobar: same A : ${}^3\text{He}$, ${}^3\text{H}$, ${}^{40}\text{S}$, ${}^{40}\text{Cl}$, ${}^{40}\text{Ar}$

Isotopologues: molecules that differ only in the isotopes: ${}^{12}\text{CO}$, ${}^{18}\text{CO}$

Basic particles and forces:

Fermions → spin = $\frac{1}{2}$

quarks: up, charm, top
down, strange, bottom

interact through the strong force

leptons: electron, muon, tau
 ν_e , ν_μ , ν_τ

do not interact through the strong force

Bosons → spin = integer
force carriers

gluon
photon
Z boson
W boson

gauge or vector bosons
spin = 1

Higgs } scalar boson
spin = 0

elementary particles → no further sub-division

+ anti particles: same mass but opposite charge, except ν

Composite particles:

Hadrons: bound states of quarks : e.g. p^+ , n^0

two types - baryon: composed of 3 quarks : p^+ , n^0
anti baryon: 3 antiquarks

- meson: have the same number of quarks and anti quarks. e.g. π^0 , π^- , π^+

② Forces

- strong force : force carrier : gluons
↳ keeps p^+ and n^0 together
- electromagnetic force : force carrier : γ
- weak force : force carrier : W^+, W^-, Z^0 bosons
↳ responsible for β decay
- gravity : no known force carrier, not important on the scale of nuclear and particle physics

→ basic quantum mechanics

→ properties of nuclei