CSIA Monitoring Scope

PAZ

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Purpose

Plot sampled cocentrations agains theoretical minimum mass vs. volume relations for multi-element CSIA. Data file:

- MonitoringScope_R.csv (R generated, Book 5 \approx Line 350)

Carbon

Conversion to ng

For a given sample of concentration C [$\mu g/L$], considering a 4 μL injection and a vial volume x_{vial} of 500 μL :

$$C \ [\mu g/L] \cdot \frac{10^{3} \ ng}{1 \ \mu g} \cdot \frac{1 \ mol_{S-met}}{283.796 \ g} \cdot \frac{12.0107 \ g}{1 \ mol_{C}} \cdot 15 \cdot \frac{4 \ [\mu L \ inj.]}{x_{vial} \ [\mu L \ vial]} \cdot X_{smp} \ [L] = M_{C}$$

The same equation re-arranged to compute it numerically so as to yield 10 ng of carbon, from a 500 μL vial:

$$C \; [\mu g/L] \cdot \frac{10^3 \; ng}{1 \; \mu g} \cdot \frac{1 \; mol_{S-met}}{283.796 \; g} \cdot \frac{12.0107 \; g}{1 \; mol_{C}} \cdot 15 \cdot \frac{4 \; [\mu L \; inj.]}{500 \; [\mu L \; vial]} \cdot \frac{1}{10 \; ng_{C}} = \frac{1}{X_{smp} \; [L]}$$

Chlorine

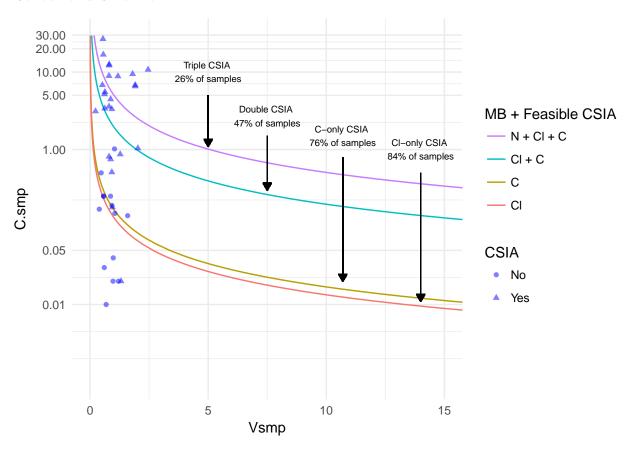
$$C \ [\mu g/L] \cdot \frac{10^3 \ ng}{1 \ \mu g} \cdot \frac{1 \ mol_{S-met}}{283.796 \ g} \cdot \frac{35.453 \ g}{1 \ mol_{Cl}} \cdot 15 \cdot \frac{4 \ [\mu L \ inj.]}{500 \ [\mu L \ vial]} \cdot \frac{1}{10 \ ng_{Cl}} = \frac{1}{X_{smp} \ [L]}$$

Nitrogen

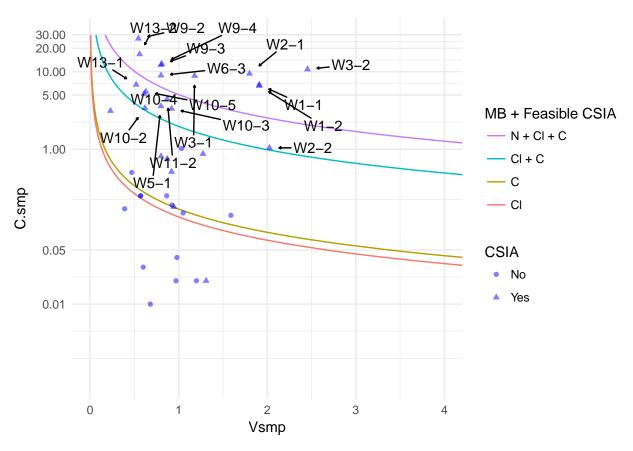
$$C \; [\mu g/L] \cdot \frac{10^3 \; ng}{1 \; \mu g} \cdot \frac{1 \; mol_{S-met}}{283.796 \; g} \cdot \frac{14.007 \; g}{1 \; mol_N} \cdot 15 \cdot \frac{4 \; [\mu L \; inj.]}{500 \; [\mu L \; vial]} \cdot \frac{1}{30 \; ng_N} = \frac{1}{X_{smp} \; [L]}$$

Plotting

Carbon and Chlorine



Plotting with labels



Percentages

- ## [1] "% Samples capable of triple CSIA"
- ## [1] 26.31579
- ## [1] "% Samples capable of double CSIA"
- ## [1] 47.36842
- ## [1] "% Samples capable of single C CSIA"
- ## [1] 76.31579
- ## [1] "% Samples capable of single Cl CSIA"
- ## [1] 84.21053