LATEX Chemistry Package CheatSheet

Eureka

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→ Basic usage

```
% \usepackage[version=4]{mhchem}
\ce{CO2 + C -> 2CO}
\ce{Hg^2+ ->[I-] HgI ->[I-] [Hg^{II}]^2-}
\ce{[Pt(\eta^2-C2H4)C13]-}
\ce{Fe(CN)_{$\frac{6}{2}$}}$
\ce{RNO2 <=>[+e] RNO2^{-.}}
```

$$CO_2 + C \longrightarrow 2CO$$
 (1)

$$Hg^2 \xrightarrow{I^-} HgI \xrightarrow{I^-} [Hg^{II}I_4]^{2-}$$
 (2)

$$[Pt(\eta^2-C_2H_4)Cl_3]^-$$
 and $Fe(CN)_{\frac{6}{2}}$ (3)

$$RNO_2 \stackrel{+e}{\rightleftharpoons} RNO_2^{-\bullet}$$
 (4)

→ Parenthesis, Brackets, Braces

 $1/ \ce{x Na(NH4)HP04 -> [\Delta] (NaP03)_x + x NH3^ + x H20}$

$$x \operatorname{Na}(\operatorname{NH}_4)\operatorname{HPO}_4 \xrightarrow{\Delta} (\operatorname{NaPO}_3)_x + x \operatorname{NH}_3 \uparrow + x \operatorname{H}_2 O$$
 (5)

→ Bonds

$$C_6H_5-CHO$$
 (6)

$$A - B = C \equiv D \tag{7}$$

$$A \equiv B \equiv C \equiv D \tag{8}$$

$$A \cdots B \cdots C$$
 (9)

→ Further Examples

$$Zn^{2+} \xrightarrow[+2 \text{ H}^{+}]{\text{Hydroxid}} Zn(OH)_{2} \downarrow \xrightarrow[+2 \text{ H}^{+}]{\text{Hydroxozikat}} [Zn(OH)_{4}]^{2-}$$
(10)

$$K = \frac{[\mathrm{Hg}^{2+}][\mathrm{Hg}]}{[\mathrm{Hg}_2^{2+}]} \tag{11}$$