

Chapter 1st → Electric charges and Field

Electric charge → Electric charge is an intrinsic property of elementary particle which it automatically accompanies.

- * SI unit of electric charge → coulomb (c)
- * It is scalar quantity

Dimension of charge is

current = charge / time

$$q = \epsilon Q$$

$$q = q t$$

Dim. → [AT]

$$[M^0 L^0 T^1 A^1]$$

Production of charge :-

① Frictional electricity → Electricity produced by friction.

Ex When a glass rod is rubbed by silk cloth. In this rubbed substance became electrified. It is bcz of friction that the substance get charged on rubbing.

Electrostatic instruments → chargers of net

(i) Induction method \Rightarrow In this method neutral body is charged by bringing a charged body nearby.

Properties of Electric charge

(i) Additive nature of electric charge \Rightarrow Since the charge is scalar quantity it follows simple algebraism. It means that if a system contains charges q_1, q_2, \dots, q_n then its total charge will be

$$Q = q_1 + q_2 + \dots + q_n$$

(ii) Conservation of charge \Rightarrow Charge can neither be created nor be destroyed it can only be transformed from one system to another. Net charge will always remain same.

(iii) Quantisation of charge \Rightarrow This law states that quantity of a charge on a particle is always an integral multiple of least amount of charge. e is the basic unit of charge.
 $q = ne$.

When a glass rod is rubbed with silk cloth then glass rod acquires (+)ve charge and silk acquires equal (-)ve charge.

Coulomb's law

This law states that the force acting b/w two charges get separated by a distance r is directly proportional to the product of the charges and inversely proportional to the square of distance b/w them.



If two point charges q_1 and q_2 are separated by distance r then Force of attraction or repulsion b/w them is such that

$$\text{Law of Force } F \propto q_1 q_2 \quad (i)$$

$$\text{Resultant Force } F \propto \frac{1}{r^2} \quad (ii)$$

Combining (i) and (ii)

$$F \propto q_1 q_2$$

$$F = \frac{k q_1 q_2}{r^2}$$

$k \rightarrow$ electrostatic force constant

$$F = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2}$$

$\epsilon_0 \rightarrow$ permittivity of free space or vacuum,
 $k = 9 \times 10^9$

Comparing electrostatic and gravitational forces

Similhan'ips

- I Bath forces obey inverse square law.
 $F \propto \frac{1}{r^2}$
 - II Bath forces are proportional to product of masses or charges.
 $F \propto Mm$
 $F \propto q_1 q_2$
 - III Bath are central forces i.e. they act along the line joining the centres of the two bodies.
 - IV Bath are conservative forces
 - V Bath can operate in vacuum.

Dissimilarities: ~~size~~ (1) orientation?

- (i) Gravitational forces is attractive while electrostatic force may be attractive or repulsive.
 - (ii) Gravitational force does not depend on nature of substance while electrostatic force depends on nature of substance.
 - (iii) Electrostatic forces are much stronger than gravitational forces.

Numerically if two charges system is kept in vacuum then immersed as it is in liquid then what will be the effect on force.

Sol:

$$\text{In vacuum } F = \frac{k q_1 q_2}{r^2}$$

$$\text{By coulomb's law, } F = k q_1 q_2 \frac{1}{r^2}$$

$$F = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2} \quad (1)$$

In liquid

$$F = \frac{1}{4\pi\epsilon} \frac{q_1 q_2}{r^2} \quad (2)$$

(1) / (2)

$$\frac{F_0}{F} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2} \times \frac{4\pi\epsilon r^2}{q_1 q_2}$$

$$\frac{F_0}{F} = \frac{\epsilon}{\epsilon_0}$$

$$\frac{F}{F_0} = \frac{\epsilon_0}{\epsilon}$$

$$(P) \text{ If } F_0 = \epsilon_0 \text{ and } F = \epsilon \epsilon_0$$

$$F = \epsilon \epsilon_0 \text{ (dielectric constant)}$$

$$\frac{F}{F_0} = \frac{1}{k} \quad F = F_0 \cdot \frac{1}{k}$$

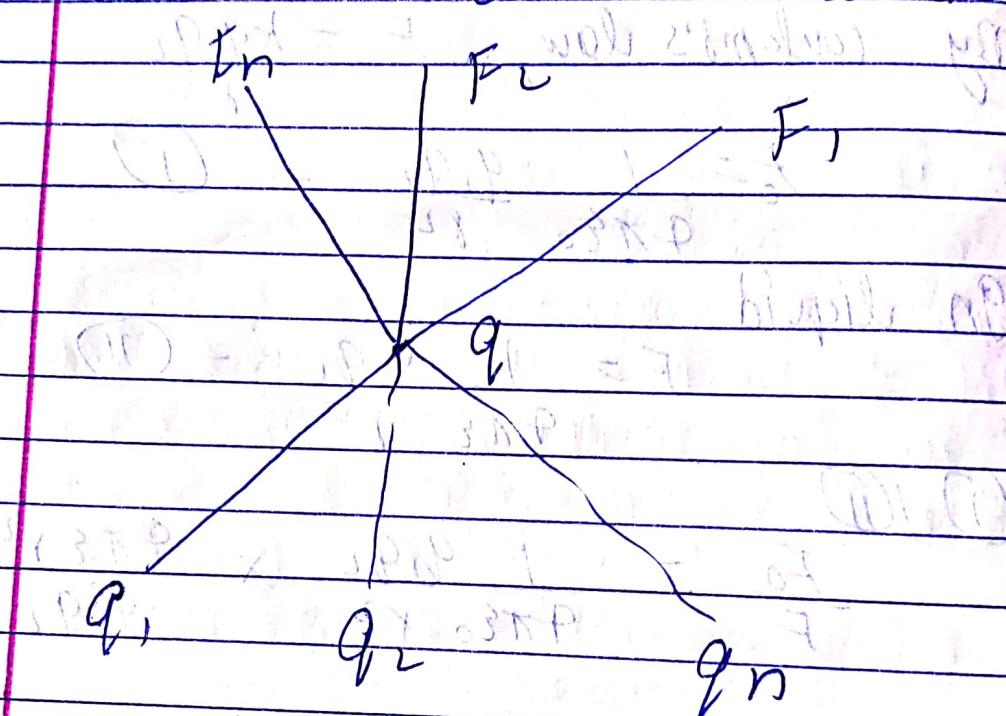
For liquid $k > 1$ and for vacuum $k = 1$.

$$F < F_0$$

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Forces b/w malleable charges or principle of superposition

→ It states that when a number of charges are interacting then the total force on a given charge is the vector sum of all forces exerted on it due to all other charges.



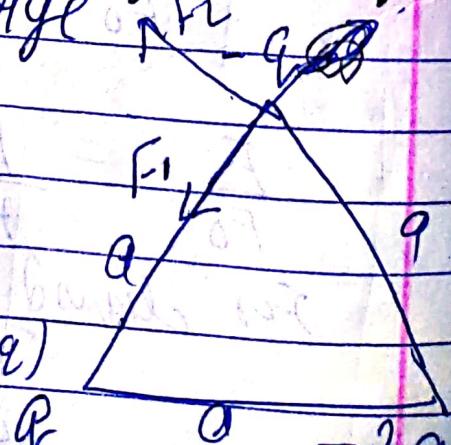
According to superposition principle Total force on charge q_1 is given by, $F_r = F_1 + F_2 + \dots + F_n$

* Find resultant force acting on $(+q)$.

⇒ Force acting on charge $(+q)$ due to $(-q)$ due to q

$$F_1 = \frac{kq_1}{r^2}$$

Force acting on charge $(-q)$ due to $(-2q)$



$$F_2 = 2 \underline{kq_2} \\ q_2$$

$$F_1 = 2F_1$$

Resultant force acting on (-q)

$$F = \sqrt{F_1^2 + F_2^2 + 2F_1F_2 \cos(120^\circ)}$$

$$F = \int F^2 + (2F_1)^2 + 2F_1(2F_1)(-\frac{1}{c})$$

$$F = \sqrt{F_1^2 + qF_1^2 + qF_1^2} \quad (\text{Ansatz})$$

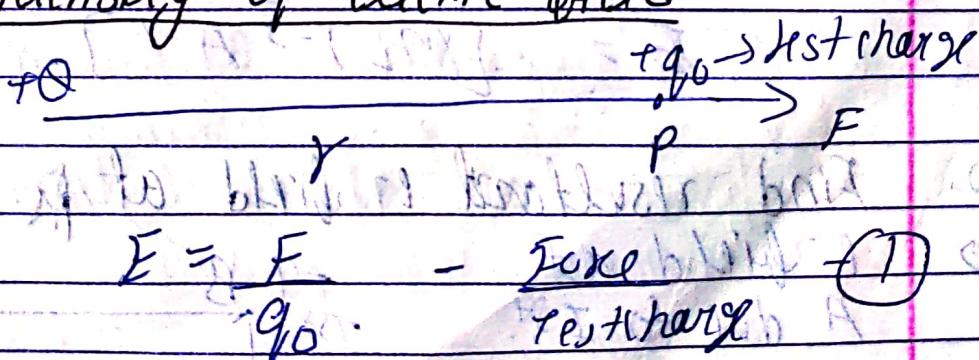
$$F = \sqrt{SF_1^2 - 2F_1^2}$$

$$F = \sqrt{3} F_{\text{wind}} + \omega_0 I_{\text{FF}}$$

$$F_{\text{grav}} = \frac{\sqrt{3} k q^2}{a^2}$$

- Electric field \Rightarrow The space surrounding a charge in which another charge experience an electric force is called electric field.

Electric field due to point charge or Intensity of electric field



By coulomb's law $F = \frac{q_1 q_2}{r^2}$

Substituting F in ①

$$E = \frac{1}{4\pi\epsilon_0} \frac{q}{r^2}$$

$$\boxed{E = \frac{1}{4\pi\epsilon_0} \frac{q}{r^2}} \quad \text{or} \quad E = k \frac{q}{r^2}$$

- * unit of e-field $\rightarrow N/C$
- * It is vector quantity
- * The test charge (q_0) must be very small so that it doesn't change value of E

Dimension of e-field

$$\text{Dim. } E = \text{Dim. } F$$

$$\text{Dim. } q$$

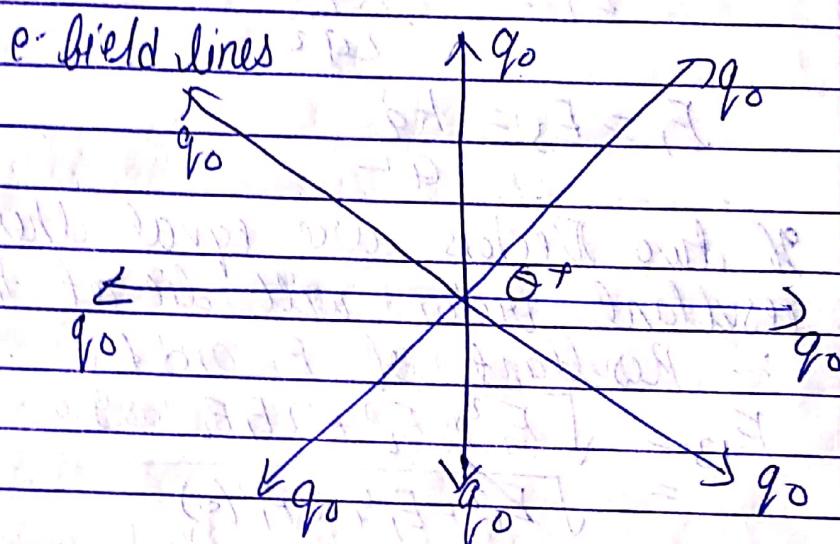
$$E = \text{of } [MLT^{-2}]$$

$$[AT]$$

$$E = [MLT^{-3}A^{-1}]$$

Electric field lines \Rightarrow Electric field lines are imaginary lines which are used to represent an e- field.

Dipole field \Rightarrow e- field produced by an e- dipole.



Electric Dipole \Rightarrow It is an arrangement in which two equal and opposite charges are kept at short distance.

- e- dipole moment \Rightarrow electric dipole movement is defined as a product of either of the charge and separation b/w them.

$$\text{+q} \quad -q$$

$$2a$$

$$p = q \times 2a$$

↑
distance of separation