

2.8

(a) $Q = C_V \Delta T = \rho V c_V \Delta T = 4.92 \times 10^5 J,$

(b) $Q = C_p \Delta T = \rho V c_p \Delta T = 6.94 \times 10^5 J,$

(c) 某时间容器中有质量为 m , 温度为 T 的气体, 由理想气体状态方程知

$$\frac{m_0}{M} RT_0 = n_0 RT_0 = nRT = \frac{m}{M} RT$$

即 $m_0 T_0 = mT, m = m_0 \frac{T_0}{T}$

$$dQ = m C_p dT \Rightarrow Q = \int_{T_0}^{T_f} C_p m_0 \frac{T_0}{T} dT = m_0 C_p T_0 \ln \frac{T_f}{T_0} = 6.70 \times 10^5 J.$$

2.10

$$c_w m_w \Delta T_w = c_{Cu} m_{Cu} \Delta T_{Cu}$$

得到 $c_{Cu} = 619.5 J/(kg \cdot K).$

2.11

$$\Delta Q = \Delta U - W' = u - u_0 - p_0 V_0 = 0$$

即 $u - u_0 = p_0 V_0.$

2.21

$$\begin{aligned} V dp + 2p dV &= 0 \\ d(pV) &= nR dT = p dV + V dp \end{aligned}$$

得到 $p dV = -nR dT$

$$dQ = dU + p dV = n(C_V - R) dT$$

即热容 $C = C_V - R$

比热容为 $c = \frac{C}{\mu} = \frac{C_V - R}{\mu}.$

2.22

绝热过程 $dQ = dU + p dV = \left(\frac{\partial U}{\partial T}\right)_V dT + \left[\left(\frac{\partial U}{\partial V}\right)_T + p\right] dV = C_{V,m} dT + \frac{RT}{V-b} dV = 0$

解微分方程得到 $C_{V,m} \ln T = -R \ln(V-b) + \text{const}.$

两边取 e 指数得到 $T(V-b)^{R/C_{V,m}} = \text{const}.$

2.23

与 2.22 相同, 绝热过程满足 $T(V-b)^{R/C_V} = \text{const}.$

2.24

$$dQ = C_V dT + p dV = C dT$$

两式才能消dT.

$$\text{即 } p dV = (C - C_V) dT = d(pV) - V dp = R dT - V dp$$

$$\text{得到 } V dp = (C_V + R - C) dT = (C_p - C) dT$$

$$\text{二式相除得 } \frac{p dV}{V dp} = - \frac{C_V - C}{C_p - C}$$

$$\text{得到 } p V^{\frac{C_V - C}{C_p - C}} = p V^n = \text{const.}$$

目标 \star p-V.

C_V, C_p 需要 R.