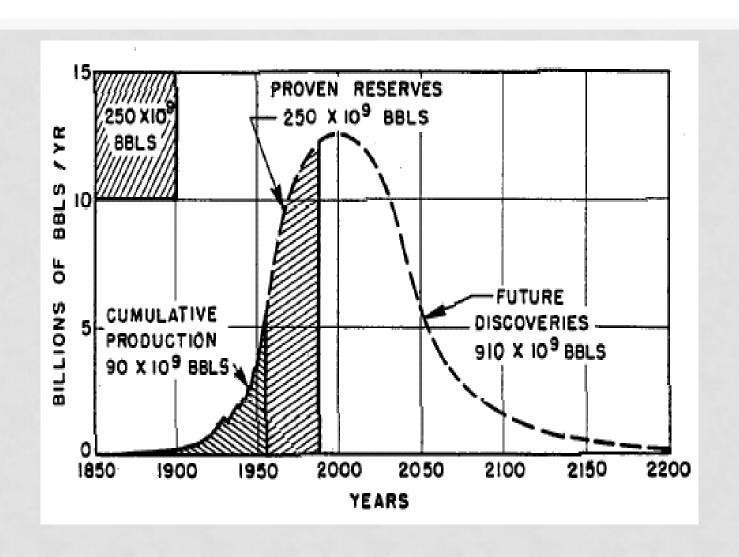
# OVER ALL VIEW OF ALTERNATIVE FUELS

IS THERE LIFE AFTER PETROLEUM?

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#### HUBBERT'S PEAK THEORY



For any given geographical area, from an individual oil field to the planet as a whole, the rate of oil production tends to follow a bell-shaped curve

In <u>1956</u>, Hubbert proposed that crude oil production in a given region over time would follow a bell-shaped curve without giving a precise formula. The idea is that after oil reserves are discovered, oil production increases exponencially at first as more efficient facilities are installed. At some unique point, a peak is reached and oil production begins an exponential decline.

US oil production peaked in 1971.

#### WHEN WILL THE PEAK TAKE PLACE?

- U.S oil production peak
- Peak of world oilified discoveries
- Hubbert's estimation for world peak
- Nowadays, the facts are
  - Oil production in decline in 33 out of 48
  - 2 out of 3 largest oil fields have peaked

## CLASSIFICATION OF ALTERNATIVE FUELS

- Gasoline type biofuels
- Diesel type biofuels
- Other types of internal combustion
- External combustion
- No combustion

#### GASOLINE TYPE BIOFUELS

- Ethanol
- Buthanol: Alcohol, can be fermented from Biomass,
- Methanol: Easy to facturate from methane but highly toxic and volatile
- P-Series: Mixture derived from natural gas, not widely used
- Hydrogen

#### DIESEL TYPE BIOFUELS

- Vegetable oils
  - Waste vegetable oils
  - Straight vegetable oils
- Biodiesel

### OTHER TYPES OF INTERNAL COMBUSTION

- Liquefied Natural Gas (LNG)
- Compressed Natural Gas (CNG)
- Propane
- Synfuel: Liquied obtained from natural gas or coal

#### **EXTERNAL COMBUSTION**

• Steam: Based upon the rankine cycle

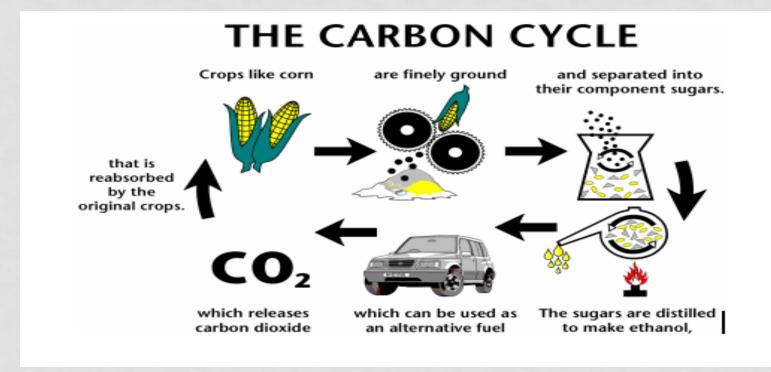
Organic waste

#### NO COMBUSTION

- Electric: Interesting alternative but dont forget electricity comes from combustion
- Solar cell cars: At first too large for consumer cars, but smaller with time

#### **BIOETHANOL**

Alcohol product produced from corn, sorghum, potatoes, wheat, sugar cane, even biomass such as cornstalks and vegetable waste.



#### BIOETHANOL

- Use in combustion engines
  - Hydrous ethanol

- Anhydrous (or dehydrated) ethanol
- ETBE

#### **BIOETHANOL - E100**

- Modifications in the engine to use the BioEthanol as pure fuel
  - To increase the relation of compression.
  - To change the mixture of fuel / air.
  - To place spark plugs resistant to major temperatures and pressures.
  - To place conduits resistant to the assault of tar after caulking.
  - To add a mechanism that should facilitate the take-off in cold.

#### BIOETHANOL - PROS

- Domestically produced
- Burns 10 % more efficiently than gasoline
- FFVs are available and becoming more affordable.

#### **BIOETHANOL - CONS**

- Less energy content than gasoline.
- Fueling stations yet difficult to find.
- Production is yet limited.
- Infraestructure for fueling and distribution is yet insufficient.

## EXAMPLE OF BIOETHANOL: FOCUS FFV

- 1.8-litre engine
- Produces 70 % less carbon dioxide than its petrol equivalent.
- It develops 123 bhp.
- Fuel E85 is a mixture of ethanol (85%) and petrol (15%).
- Price: 193 973 SEK

#### **HYDROGEN**

- Is the lightest element
- Is the most abundant element in the Universe
- Is not a direct energy source
- It can be obtained by means of solar energy, eolic or hidraulic electricity.
- Nowadays 95% is obtained from fossil fuels

#### **OBTAINING HYDROGEN**

Reformed with steam

• 
$$CH_4 + H_2O$$

• 
$$CO + H_2O$$

\_\_\_\_ 
$$CO + 3H_2$$

$$\longrightarrow$$
 CO<sub>2</sub> + H<sub>2</sub>

Water electrolysis

• 
$$H_2O + Power$$

$$\longrightarrow$$
 H<sub>2</sub> + 1/2O<sub>2</sub>

Photoelectrolysis

#### **OBTAINING HYDROGEN**

- Using the biomas
  - Biomass gasification
  - Pyrolysis
- Photobiologic production

#### **OBTAINING HYDROGEN**

- Hydrogen production on board the vehicle
  - Using methanol as fuel

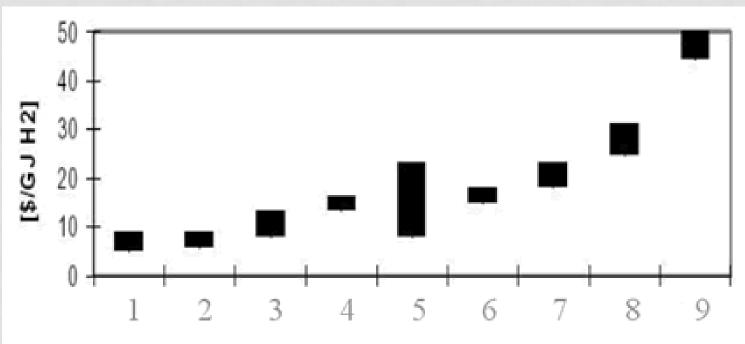
$$CH_3OH + \frac{1}{2}O_2$$
  $CO_2 + 2H_2$ 

• Using ethanol as fuel

$$CH_3CH_2OH + 3H_2O$$
  $CO + CO_2 + 6H_2$ 

**-----**

#### PRODUCTION COST OF HYDROGEN



- 1 Gas Natural-CO<sub>2</sub>
- 2 Carbón-CO<sub>2</sub>
- 3 Biomasa
- 4 Energía Nuclear
- 5 Rector de gas de alta temperatura (HTGR)

- ó Energía eólica onshore
- 7 Energía eólica offshore
- 8 Energía solar térmica
- 9 Energía solar fotovoltaica

## USE OF HYDROGEN IN AUTOMOTION

Hydrogen combustion in a MCIA

Fuel batteries

#### HYDROGEN COMBUSTION IN A MCIA

Mazda Rx8 Hydrogen



#### HYDROGEN COMBUSTION IN A MCIA

BMW 7 Series Hydrogen



#### FUEL BATTERIES

• Electrochemical systems where energy from a chemical reaction is directly turned into electricity.

• Hydrogen + Oxigen =>Electricity + Water

#### ADVANTAGES OF USING HYDROGEN AS A FUEL

- Abundant element in the Universe
- High efficiency
- Emission zero of pollutants
- Low working temperatures and pressures
- Silent functioning

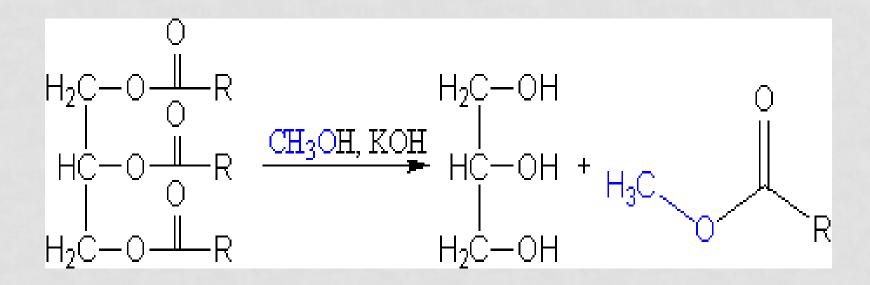
## DISADVANTAGES OF USING HYDROGEN AS A FUEL

- It is not a primary source.
- Obtaining pure hydrogen is really expensive.
- High storage and supply costs.
- High weight of fuel cells for the current prototypes
- High energetic expense to liquefy the hydrogen

#### BIODIESEL

- Diesel equivalent, processed fuel derived from biological sources.
- It is composed by mono-alkyl esters made from the transesterification of both vegetable oils and animal fats.
- Glicerine is produced as a second product.

#### **OBTAINING BIODIESEL**



#### SOURCES FOR BIODIESEL

- Vegetable oils
- Genetically modified vegetable oils
- Waste vegetable oils
- Bad quality vegetable oils

#### COMPARISON TO PETROLEUM

- Carbon monoxide: -50%
- Carbon dioxide: -78%
- Nitrogen oxide: +20%
  - BUT catalyc converters
- Biodegradable and non-toxic
- The U.S. Department of Energy confirms that biodiesel is less toxic than table salt and biodegrades as quickly as sugar.

#### NATURAL GAS

• Natural gas is a mixture of hydrocarbons, mainly methane (CH<sub>4</sub>).

• Other components: ethane, propane, nitrogen, helium, carbon dioxide, hydrogen sulfide, water vapour,...

#### PRODUCTION OF NATURAL GAS

- Water or sewage treatment.
- Gas wells.
- Crude oil production

#### USE OF NATURAL GAS AS A FUEL

- Light-duty applications
  - Natural gas vehicles
- Heavy-duty and medium-duty applications
  - Natural gas engines

#### ADVANTAGES OF NATURAL GAS

- Carbon monoxide 90 %
- Nitrogen oxide 60 %
- Carbon dioxide (greenhouse gas) 30-40%
- CO and particulate matter > 90 %
- NOx > 50 %

Advantages of natural gas vehicles compared to gasoline vehicles

#### CONCLUSIONS

Our ignorance is not so vast as our failure to use what we know

Marion King Hubbert (1903-1989)