### THE TAMIL NADU DR. M.G.R. MEDICAL UNIVERSITY

# [BPHARM 0122] JANUARY 2022 Sub. Code: 2007 (MARCH 2021 EXAM SESSION)

# B.PHARMACY DEGREE COURSE (SEMESTER EXAMINATIONS) PCI Regulation 2017 – SEMESTER I PAPER IV – PHARMACEUTICAL INORGANIC CHEMISTRY Q.P. Code: 562007

Time: Three hours Maximum: 75 Marks

## I. Elaborate on: Answer any TWO questions.

 $(2 \times 10 = 20)$ 

- 1. (a) What is physiological Acid Base Balance?
  - (b) Give the composition and uses of Oral Rehydration Salt.
- 2. Write the method of preparation, assay and uses of
  - (a) Sodium Chloride.
  - (b) Ferrous sulphate.
- 3. Write a detailed note on mechanism of Antimicrobials and highlight on the use of Iodine and its preparation as antimicrobials.

### II. Write notes on: Answer any SEVEN questions.

 $(7 \times 5 = 35)$ 

- 1. Write the principle involved in the limit test of Sulphates.
- 2. Write a note on Indian pharmacopoeia.
- 3. Write about the preparation, assay and use of Calcium gluconate.
- 4. What are Expectorants? Give the preparation, properties, assay and use of Ammonium chloride.
- 5. Define Alum and give the formula, properties and uses of Potash alum.
- 6. Write the preparation, properties, assay and uses of Chlorinated lime.
- 7. Write short notes on Iodine and its preparations.
- 8. What are Cathartics? Give the preparation and properties of Bentonite.
- 9. What are Antidotes? Give the preparation, properties and uses of Sodium thiosulphate.

### III. Short answers on: Answer ALL questions.

 $(10 \times 2 = 20)$ 

- 1. Define Haematinics and give one example.
- 2. Give the storage conditions of Radio isotopes.
- 3. Define Astringent and give an example.
- 4. Uses of Citric acid in the limit test of Iron.
- 5. What is the use of activated Charcoal.
- 6. Give the formula and use of (a)Calcium carbonate (b)Zinc eugenol cement.
- 7. Define Dentifrice with example.
- 8. Enumerate the ideal properties of Antacids.
- 9. Give the principle involved in the limit test of Chloride.
- 10. Define Osmotic laxative.