

## Assignment 10

Indian Institute of Science Education and Research

CHM202: Energetics and dynamics of chemical reactions

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**Ques. 1** The enzyme-catalysed conversion of a substrate at 25°C has a Michaelis constant of  $0.024 \text{ mol dm}^{-3}$ . The rate of the reaction is  $1.15 \times 10^{-4} \text{ mol dm}^{-3} \text{ s}^{-1}$  when the substrate concentration is  $0.890 \text{ mol dm}^{-3}$ . What is the maximum velocity of this enzymolysis?

**Ques.2** The following results were obtained for the action of an ATPase on ATP at 20°C, when the concentration of the ATPase was  $20 \text{ nmol dm}^{-3}$ :

[ATP]/( $\mu\text{mol dm}^{-3}$ )	0.60	0.80	1.4	2.0	3.0
$v/(\mu\text{mol dm}^{-3} \text{ s}^{-1})$	0.81	0.97	1.30	1.47	1.69

Determine the Michaelis constant, the maximum velocity of the reaction, the turnover number, and the catalytic efficiency of the enzyme.

**Ques.3** The slope and intercept obtained from (1/Rate) against (1/substrate concentration) of an enzyme catalyzed reaction are 300 and  $2 \times 10^5$ , respectively. Calculate the Michaelis-Menten constant of the enzyme in this reaction.

**Ques.4** Consider an enzyme-catalysed reaction that follows Michaelis-Menten kinetics with  $K_M = 0.75 \text{ mmol dm}^{-3}$ . What concentration of a competitive inhibitor characterized by  $K_I = 0.56 \text{ mmol dm}^{-3}$  will reduce the rate of formation of product by 75 percent when the substrate concentration is held at  $0.10 \text{ mmol dm}^{-3}$ ?

**Ques.5** Some enzymes are inhibited by high concentration of their own substrates. (a) Show that when substrate inhibition is important the reaction rate  $v$  is given by

$$v = \frac{v_{max}}{1 + \frac{[S]_0}{K_1} + \frac{K_M}{[S]_0}}$$

Where  $K_1$  is the equilibrium constant for dissociation of the inhibited enzyme-substrate complex. (b) What effect does substrate inhibition have on a plot of  $1/v$  against  $1/[S]_0$ ?