

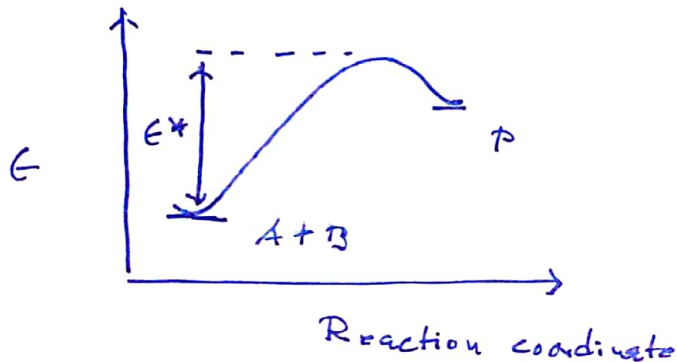
$$f(v) dv = 4\pi \left(\frac{m}{2\pi k_B T} \right)^{3/2} e^{-mv^2/2k_B T} v^2 dv$$

$$E = \frac{1}{2} m v^2 \Rightarrow v = \left(\frac{2E}{m} \right)^{1/2} dv = \left(\frac{2}{m} \right)^{1/2} \frac{1}{2} \frac{dE}{E^{1/2}}$$

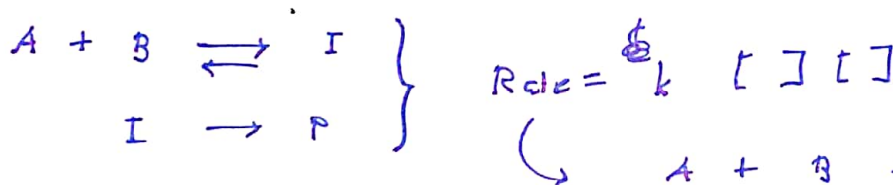
$$g(E) dE = 2\pi \left(\frac{1}{\pi k_B T} \right)^{3/2} e^{-E/k_B T} E^{1/2} dE$$

↑

fraction of no of molecules having energy between E to $E+dE$

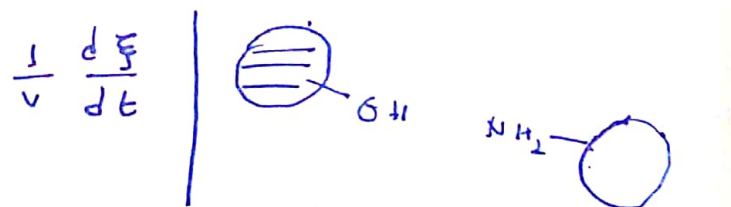


$$k(T) = A e^{-E_a/RT}$$

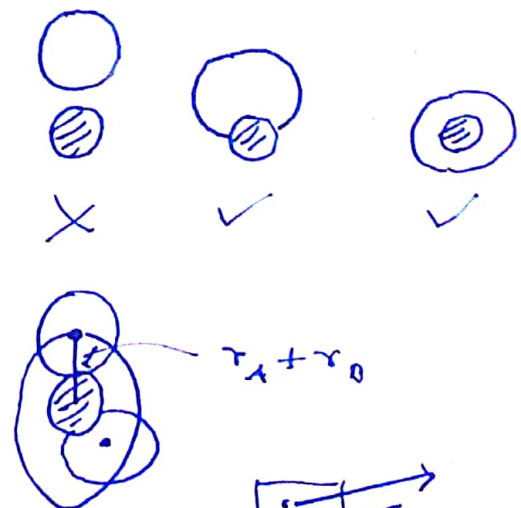
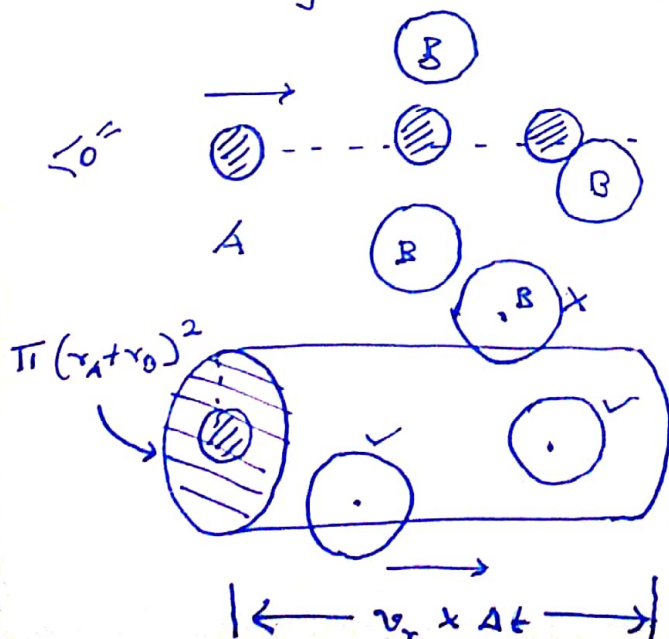


Hard spheres

Gas phase



How many collisions happen per unit time per unit volume?



①

of collisions per unit time per unit volume

$$= \frac{\pi (r_A + r_B)^2 \times v_r \times \Delta t \times n_B \times n_A}{\Delta t}$$

$$= \sigma_{\text{total}} \times \frac{v_r}{m} \times n_B \times n_A$$

collision cross-section

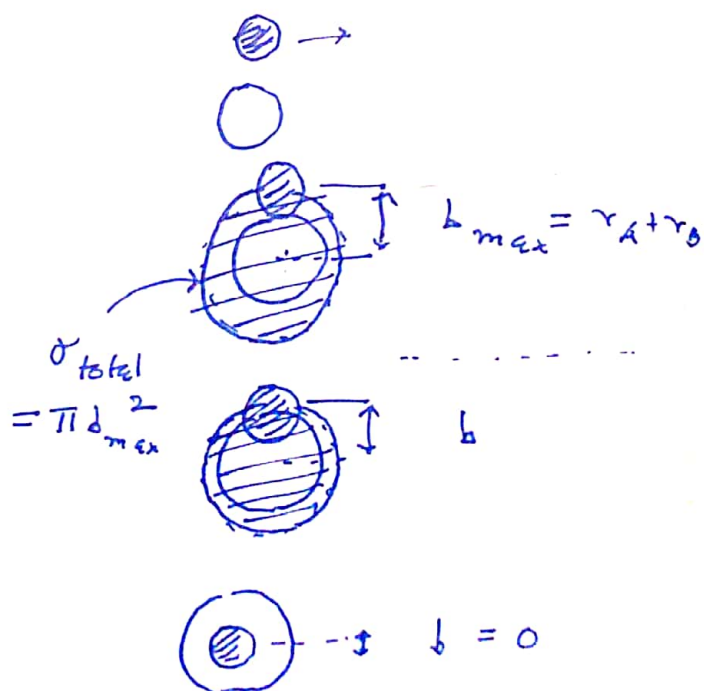
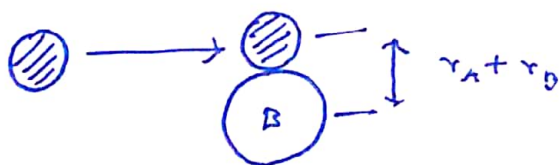
$$\text{Rate} = \sigma_{\text{total}} \times \int f(v) v_r dv_r \times n_A \times n_B$$

$$= \sigma_{\text{total}} \times \langle v_r \rangle \times n_A \times n_B$$

$$= k(T) \times [A] \times [B]$$

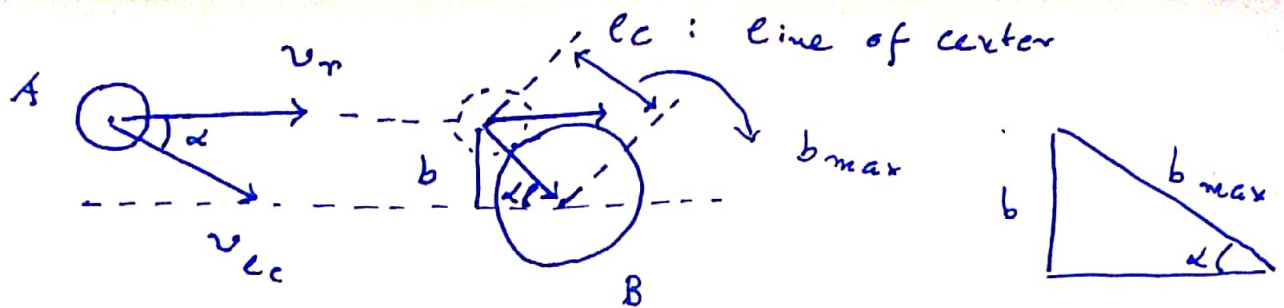
$n_A \times n_B$

$$k(T) = \sigma_{\text{total}} \times \langle v_r \rangle$$



$\geq \epsilon^*$

$b = \text{impact parameter}$

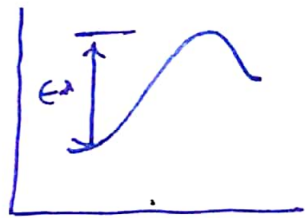


$$\epsilon_{l_c} = \frac{1}{2} \mu v_{l_c}^2 = \frac{1}{2} \mu v_r^2 \cos^2 \alpha = \frac{1}{2} \mu v_r^2 (1 - \sin^2 \alpha)$$

$$= \frac{1}{2} \mu v_r^2 \left(1 - \frac{b^2}{b_{max}^2} \right) = \epsilon_r \left(1 - \frac{b^2}{b_{max}^2} \right) \geq \epsilon^*$$



$$\sigma(\epsilon_r) = \int_0^{b^*} 2\pi b db \quad \left\{ \begin{array}{l} b^2 \leq b_{max}^2 \left(1 - \frac{\epsilon^*}{\epsilon_r} \right) \\ \pi b^2 \leq \pi b_{max}^2 \left(1 - \frac{\epsilon^*}{\epsilon_r} \right) \\ \sigma(\epsilon_r) \leq \sigma_{total} \left(1 - \frac{\epsilon^*}{\epsilon_r} \right) \end{array} \right.$$



reactive cross-section

$$k(T) = \int_{\epsilon^*}^{\infty} \sigma(\epsilon_r) \times v_r \times \frac{f(\epsilon_r) d\epsilon_r}{\frac{1}{2} \mu v_r^2 = \epsilon_r}$$

$$= \sigma_{total} \times \frac{2\pi \left(\frac{1}{\pi k_B T} \right)^{3/2} \times \int_{\epsilon^*}^{\infty} \left(1 - \frac{\epsilon^*}{\epsilon_r} \right) \times \left(\frac{2\epsilon_r}{\mu} \right)^{1/2} \times \epsilon_r^{1/2} e^{-\epsilon_r/k_B T} d\epsilon_r}{\epsilon^*}$$

$$= () \times \int_{\epsilon^*}^{\infty} \left(\frac{\epsilon_r - \epsilon^*}{\epsilon_r} \right) \times \left(\frac{2}{\mu} \right)^{1/2} \times \cancel{\epsilon_r^{1/2}} \times \cancel{\epsilon_r^{1/2}} e^{-\epsilon_r/k_B T} d\epsilon_r$$

$$= () \times e^{-\frac{\epsilon^*}{k_B T}} \times \left(\frac{2}{\mu} \right)^{1/2} \int_{\epsilon^*}^{\infty} (\epsilon_r - \epsilon^*) e^{-\frac{(\epsilon_r - \epsilon^*)}{k_B T}} d\epsilon_r$$

$$= () \times (k_B T)^2 \int_{\epsilon^*}^{\infty} \left(\frac{\epsilon_r - \epsilon^*}{k_B T} \right) e^{-\frac{(\epsilon_r - \epsilon^*)}{k_B T}} d\left(\frac{\epsilon_r - \epsilon^*}{k_B T} \right)$$

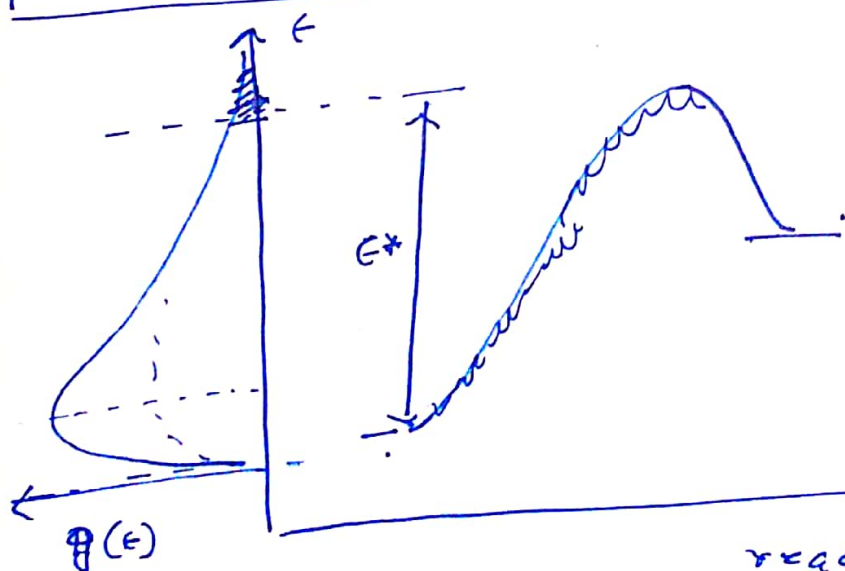
$$= \sigma_{total} \times \underbrace{2\pi \left(\frac{1}{\pi k_B T} \right)^{3/2} \times \left(\frac{2}{\mu} \right)^{1/2}}_c \times \underbrace{e^{-E^*/k_B T} \times (k_B T)^2}_d \times \underbrace{\int_0^\infty x e^{-x} dx}_e$$

$\frac{E_r - E^*}{k_B T} = x$
 $E_r \rightarrow E^*, x \rightarrow 0$
 $E_r \rightarrow \infty, x \rightarrow \infty$

$$= \sigma_{total} \times \frac{2^{3/2} \times (k_B T)^{1/2}}{\pi^{1/2} \times \mu^{1/2}} \times e^{-E^*/k_B T}$$

$$= \sigma_{total} \times \left(\frac{8 k_B T}{\pi \mu} \right)^{1/2} \times e^{-E^*/k_B T}$$

$$k(T) = \sigma_{total} \times \langle v_r \rangle \times e^{-E^*/k_B T}$$



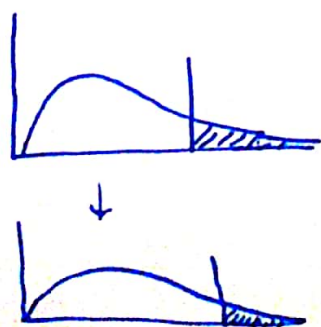
$$k(T) = A e^{-E_a/RT}$$

$$k(T) = \sigma_{total} \times \langle v_r \rangle \times e^{-E^*/k_B T}$$

$$E_a \leftrightarrow E^*$$

$$\frac{d \ln k}{dT} = ?$$

reaction coordinate



$$E_a = \langle E_r \rangle_{E_r \geq E^*} - \langle E_r \rangle$$



Activation energy

$$E^* \leftarrow (\text{energy of activation})$$

threshold energy