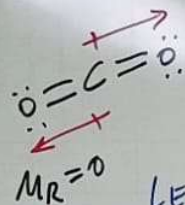


# Geometria molecular (VSEPR)

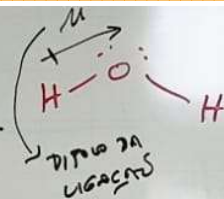
**Prof. Diego J. Raposo**

**UPE – Poli**

**2025.2**



## GEOMETRIA MOLECULAR



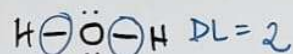
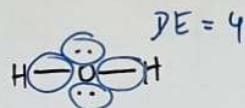
LEWIS: CONECTIVIDADE, LIGAÇÕES SIMPLES/MÚLTIPLAS, PARES DE ELÉTRONS LIVRES, POLARIDADE DAS LIGAÇÕES

MOLECULAS APOLARES

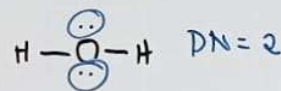
VSEPR (MODELO DE REPULSÃO DOS PARES DE ELÉTRONS DA CAMADA DE VALENCIA): FORMA E POLARIDADE DA MOLECULA



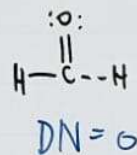
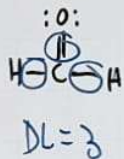
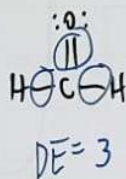
DOMÍNIO LIGANTE (DL)



DOMÍNIO NÃO LIGANTE (DN)



$$\text{DE} = \text{DL} + \text{DN}$$



DE

FORMA

2



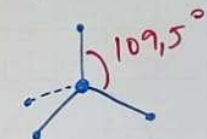
LINEAR

3



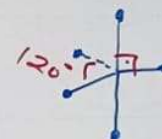
TRIGONAL PLANA

4



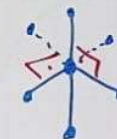
TETRAÉDRICA

5



BIPIRÂMIDE TRIGONAL

6



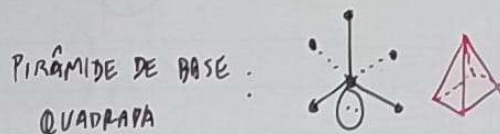
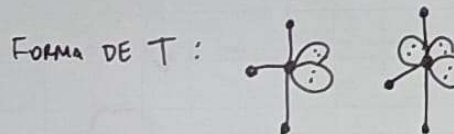
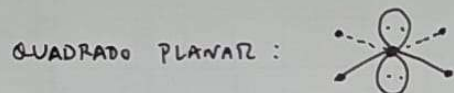
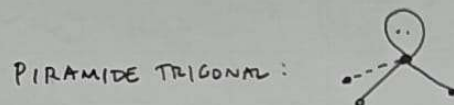
OCTAÉDRICA

# ALGUMAS GEOMETRIAS

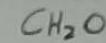
FORMA  $\leftrightarrow$  DE  
GEOMETRIA  $\leftrightarrow$  DL



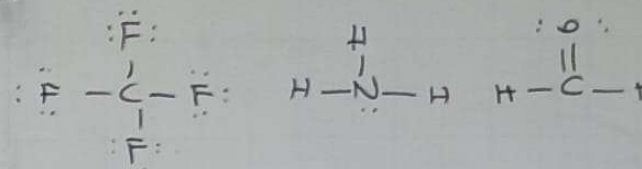
QUANDO  
DE = DL  
FORMA = GEOMETRIA



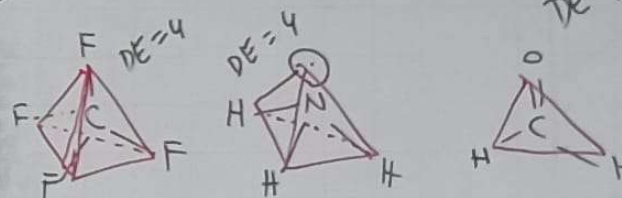
DETERMINE A GEOMETRIA MOLECULAR  
E AS PROPRIEDADES DAS MOLECULAS A  
SEGUIR:



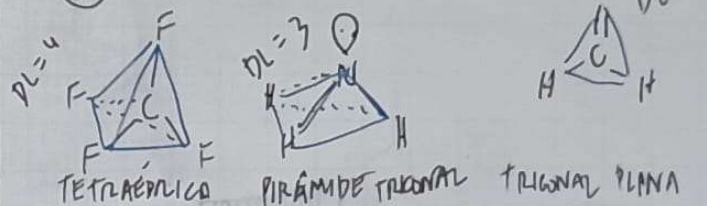
## 1) ESTRUTURAS DE LEWIS



## 2) CONTAR DE'S $\rightarrow$ FORMA

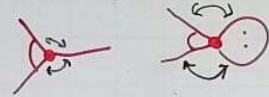
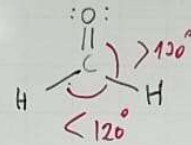
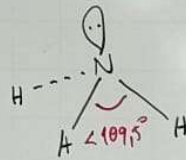
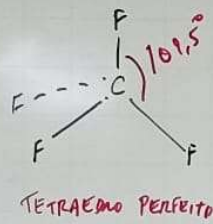


## 3) CONTAR DL'S $\rightarrow$ GEOMETRIA



④ DETERMINAR ÂNGULOS

PAIRES DE ELÉTRONS LIVRES  
LIGAÇÕES MÚLTIPLAS

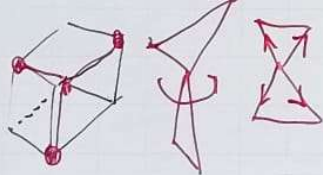
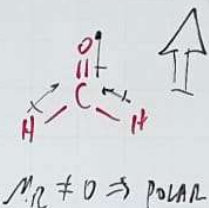
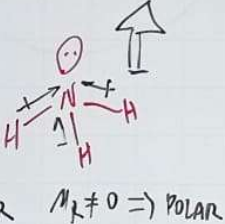
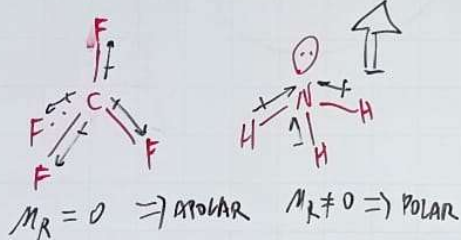


REPULSÃO ELÉTRICA

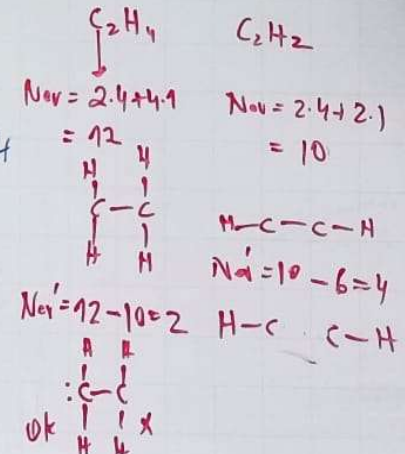
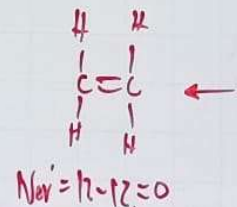
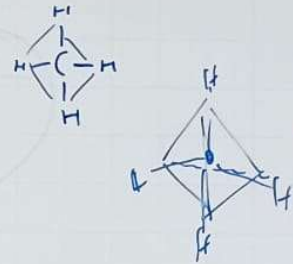
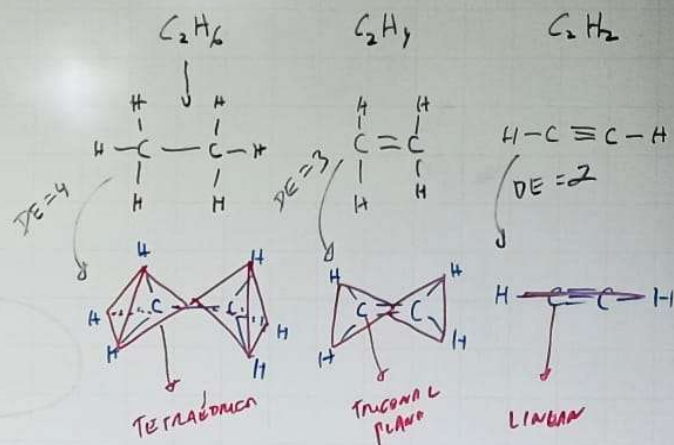


REPULSÃO ELÉTRICA

⑤ DETERMINAR POLARIDADE DA MOLÉCULA



⑥ MAIS DE UM ÁTOMO CENTRAL



**Obrigado e boa sorte!**