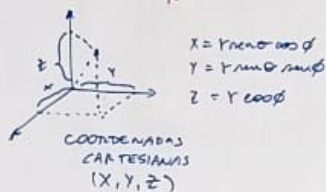
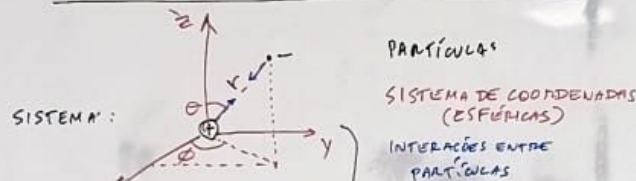


# EQ. DE SCHRODINGER PARA O ÁTOMO DE H



$$\Psi(r, \theta, \phi, n, l, m_l)$$

NÚMEROS INTEIROS CHAMADOS DE NÚMEROS QUÂNTICOS

$$\Psi_{n, l, m_l}(r, \theta, \phi)$$

ORBITAL

PARA ESCOLHER  $\Psi$  SELECIONAMOS  $n, l, m_l$  E RESPECIFICAMOS EM

## \* NÚMEROS QUÂNTICOS

$n$	NÚMERO QUÂNTICO FUNDAMENTAL
$l$	NÚMERO QUÂNTICO DO MOMENTO ANGULAR
$m_l$	NÚMERO QUÂNTICO MAGNÉTICO

CAMADA	$n$	$l$	SUBNÍVEL
K	1	0	s
L	2	0, 1	s, p
M	3	0, 1, 2	s, p, d
N	4	0, 1, 2, 3	s, p, d, f
O	5	0, 1, 2, 3, 4	s, p, d, f, g
...	...	...	...

$n$  SUBCAMADAS



ENERGIA DO ELÉTRON NO ORBITAL, TAMANHO DO ORBITAL

FORMA DO ORBITAL

ORIENTAÇÃO ESPACIAL DO ORBITAL



CAMADA (NÍVEL)

SUBCAMADA (SUBNÍVEL)

$n = 1, 2, 3, \dots$

ENERGIA, DISTÂNCIA DO NÚCLEO

$l = 0, 1, 2, 3, \dots, n-1$

CADA CAMADA POSSUI:

-  $n^2$  ORBITAIS

- ATÉ  $2n^1 e^-$

CADA SUBCAMADA POSSUI:

-  $2l + 1$  ORBITAIS

- ATÉ  $2(2l + 1) e^-$

$m_l = 0, \pm 1, \pm 2, \pm 3, \dots, \pm l$  (SÍMBOLOS VARIAM)

SUBNÍVEL	$l$	$m_l$	$n$	ORBITAIS	Nº DE ORBITAIS
s	0	0	1	1	1
p	1	0, $\pm 1$	2	3	4
d	2	0, $\pm 1, \pm 2$	3	5	9
f	3	0, $\pm 1, \pm 2, \pm 3$	4	7	16
...	...	...	...	...	...
...	...	...	n	$n^2$	$n^2$

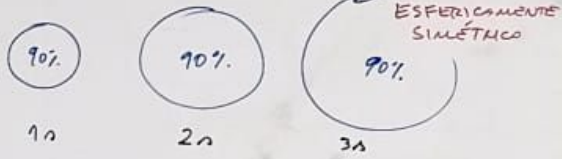
$2(2l + 1)$

# ORBITAIS

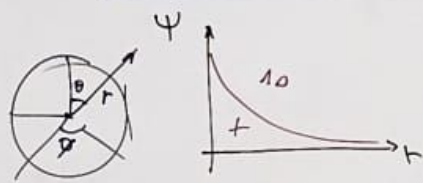
FORMA =  $l \Rightarrow 0, 1, 2, 3, \dots$   
s p d f

→ ORBITAL S

$$\psi(r, \theta, \phi) \xrightarrow[l=0]{ns} \psi(r)$$

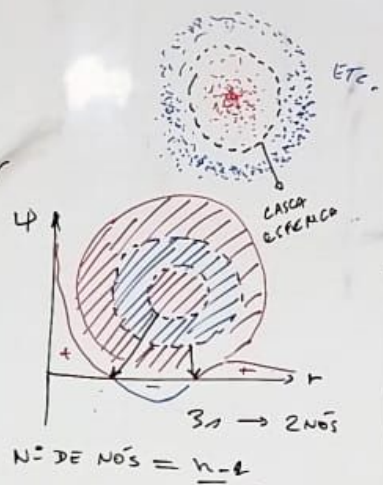
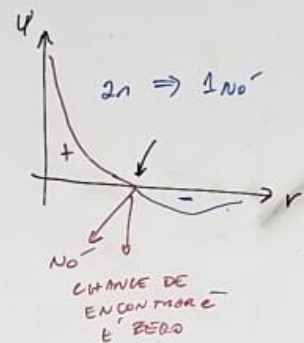
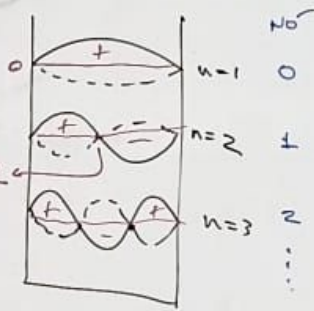


ESFERICAMENTE  
SIMÉTRICO

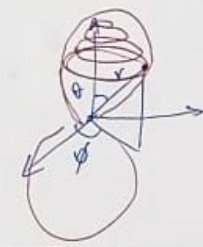


TAMANHO  
ENERGIA

NODO  
NO = 0

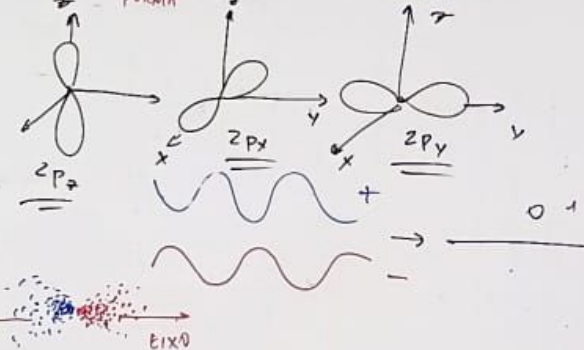
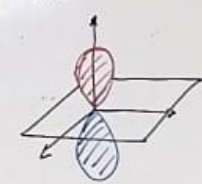
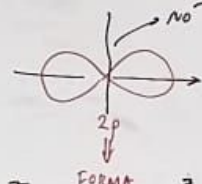


$$N^{\circ} \text{ DE NOS} = n - l$$



→ ORBITAIS p :

$$\psi(r, \theta, \phi) \xrightarrow{np} \psi(r, \phi)$$

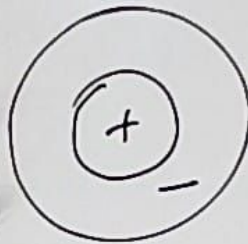


E

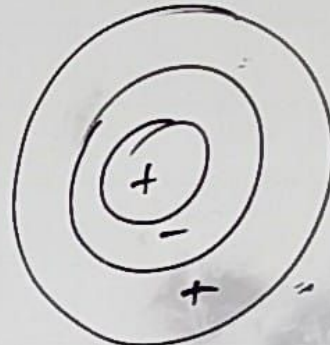
# ORBITAIS



1s

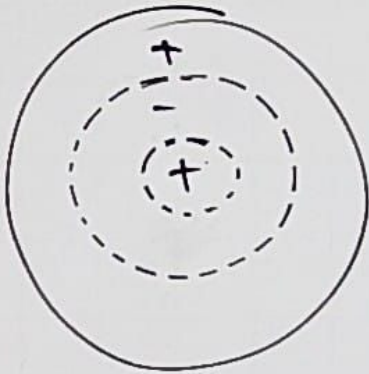


2s

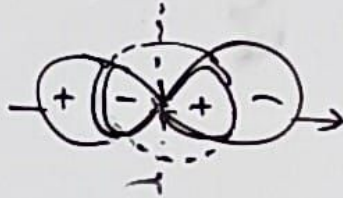


3s

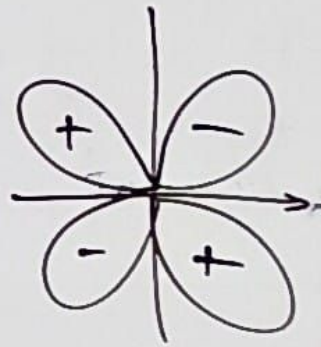
SE ELÉTRON NÃO ESTÁ LAÍ, COMO VAI INTERAGIR  
AÍ NÃO



3s



3p



3d

ENERGIA AUMENTA

NÚMERO  
QUANTICO

- QUANTOS VALORES POSSÍVEIS EXISTEM PARA  $l$  E  $m_l$  QUANDO (A)  $n=3$  E (B)  $n=5$

$n$ , DE ORBITAS	CAMADA	$1, 2, 3, \dots, n$	$0, 1, 2, \dots, n-1$ $l$	$0, \pm 1, \pm 2, \dots, \pm l$ $m_l$
$1^2$	K	1	0	0
4	L	2	0, 1	0, $\pm 1$
9	M	3	0, 1, 2	0, $\pm 1, \pm 2$
16	N	4	0, 1, 2, 3	!
<u>25</u>	O	5	0, 1, 2, 3, 4	!

