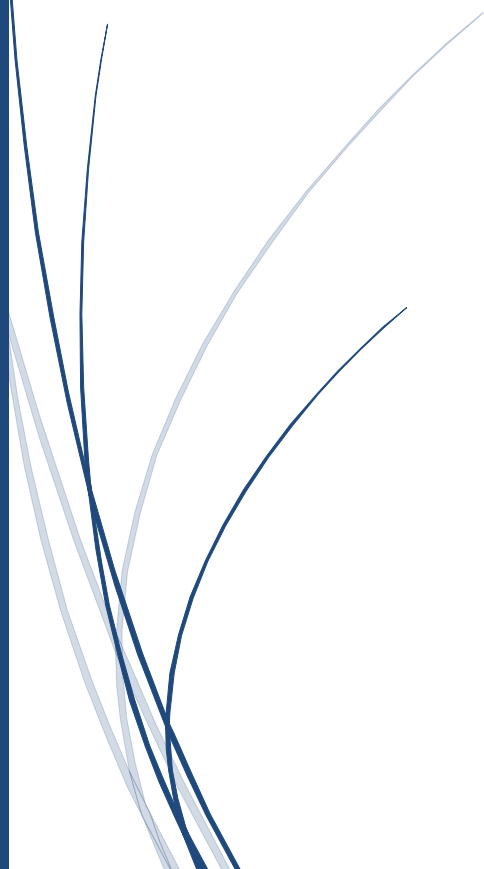


**Pharmaceutical Technology  
& Modern Pharmaceutics**

# **CHAPTERWISE NOTES**

## **Powders and Granules**



## PHARMACEUTICAL TECHNOLOGY & MODERN PHARMACEUTICS

### Powders and Granules

#### ➤ POWDER AND GRANULES

- \* Powders are mixtures of **finely divided drugs and/or chemicals** intended for **internal or external use**.
- \* Granules are prepared **agglomerates of powdered materials**. They may be used directly for their medicinal effects or in pharmaceutical processes, such as tablet manufacturing.

#### ◆ Particle Size and Analysis

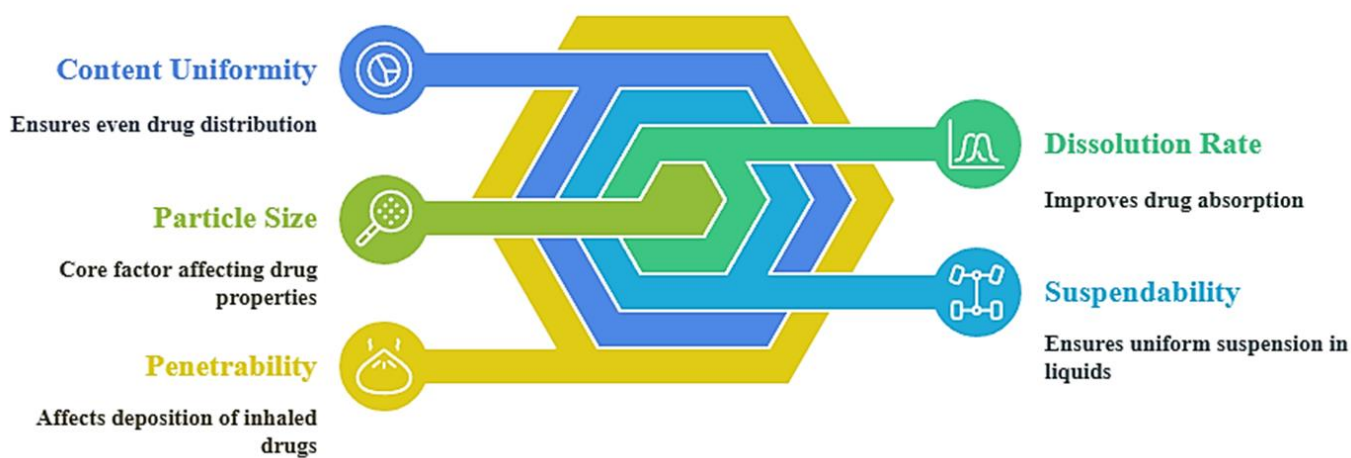
- \* Pharmaceutical powders and granules vary in size from extremely coarse (10 mm) to very fine (1  $\mu\text{m}$  or less).

#### ◆ Official Standard Pharmaceutical Powders (IP)

Grade of Powder	All particles must pass through Sieve No. (100%)	Sieve through which 40% of the particles pass (40%)
Coarse powder	10 / (1.70 mm)	44 / (355 $\mu\text{m}$ )
Moderately coarse powder	22 / (710 $\mu\text{m}$ )	60 / (250 $\mu\text{m}$ )
Moderately fine powder	44 / (355 $\mu\text{m}$ )	85 / (180 $\mu\text{m}$ )
Fine powder	85 / (180 $\mu\text{m}$ )	Not specified
Very fine powder	120 / (125 $\mu\text{m}$ )	Not specified

- \* Granules generally fall within the 4- to 12-sieve range but sometimes extend to the 12- to 20-sieve range in tablet making.

### Influence of Particle Size in Pharmacy



## ➤ COMMINATION OF DRUGS

- \* Comminution refers to the **process of reducing the particle size** of a substance.
- \* It is an essential step in pharmaceutical formulation to enhance drug dissolution, absorption, and uniformity in dosage forms.

### ◆ Methods of Comminution

#### \* Small-Scale Comminution

- **Trituration: Grinding a drug in a mortar and pestle** to reduce its particle size.
  - A **porcelain mortar** (rough surface) provides finer grinding.
  - A **glass mortar** (smooth surface) is preferred when simple mixing is required.
- **Levigation:** Used in the preparation of ointments and suspensions.
  - A **paste is formed** by mixing the powder with a levigating agent (e.g., mineral oil, glycerin) in which the powder is insoluble.
  - The paste is then triturated to reduce the particle size.
  - The levigated paste is incorporated into the ointment base for a smooth texture.

#### \* Large-Scale Comminution

- **Milling and Pulverization:** Various types of mills and pulverizers are used to reduce particle size efficiently.
  - **FitzMill Comminuting Machine:** Uses rapidly moving blades to grind particles and pass them through a screen of desired size.

### ◆ Blending of Powders

#### \* Spatulation:

- Mixing small amounts of powders using a spatula on an ointment tile or paper.
- Suitable for substances prone to forming **eutectic mixtures** (e.g., phenol, camphor, menthol).

#### \* Trituration:

- Used for both mixing and comminution.
- Preferred when mixing potent substances with diluents using **geometric dilution** to ensure uniform distribution.
- Geometric dilution involves stepwise addition and mixing of diluent in equal proportions to the potent drug.

#### \* Sifting:

- Powders are passed through sifters to achieve a **light, fluffy texture**.
- Not suitable for potent drugs requiring precise blending.

#### \* Tumbling:

- Powders are mixed in a rotating chamber.
- Used in both **small-scale** and **large-scale** blending.
- Though effective, it is a time-consuming process.

### ◆ Segregation of Powders

- \* **Sifting or percolation:** Fine particles settle at the bottom, pushing larger particles upward.
- \* **Air entrapment (fluidization):** Fine particles remain suspended in air, creating uneven distribution.
- \* **Dusting:** Fine particles stay airborne longer than heavier particles.

### Geometric Dilution – Powder Mixing

- Used when **potent drug** is mixed with a **large quantity of diluent**.
- Ensures **uniform distribution** of drug in the mixture.

#### Example:

- **100 mg potent drug + 900 mg lactose**

#### Steps:

1. Mix **100 mg drug + 100 mg lactose** → **200 mg mixture**
2. Add **200 mg lactose** to 200 mg mixture → **400 mg mixture**
3. Add **400 mg lactose** to 400 mg mixture → **800 mg mixture**
4. Add **remaining lactose (200 mg)** → **1000 mg final mixture**

#### Key Point:

Always **double the mixture weight** by adding an equal weight of diluent at each step.

### ➤ CLASSIFICATION OF POWDERS

#### 1. Bulk Powders for Internal Use




- \* Dispensed when **accuracy of dosage is not critical**.
- \* Contains multiple doses and is supplied in **wide-mouthed containers** for easy access.
- \* Suitable for **non-potent substances** like antacids and laxatives.

#### Examples:

- \* **Compound Rhubarb Powder B.P.C.** (Gregory's Powder): Contains rhubarb, magnesium carbonate, and ginger.
- \* **Compound Bismuth Powder:** Contains calcium carbonate, magnesium carbonate, sodium bicarbonate, and bismuth carbonate.

#### 2. Bulk Powders for External Use

- \* These are non-potent substances used for external applications, supplied in containers designed for specific usage.

Types of Bulk Powders for External Use:	
	<ul style="list-style-type: none"> <li>* <b>Dusting Powders</b> <ul style="list-style-type: none"> <li>○ <b>Medical</b> → For superficial skin conditions, pathogen-free                             <ul style="list-style-type: none"> <li>■ Example: Clotrimazole dusting powder</li> </ul> </li> <li>○ <b>Surgical</b> → For wounds/body cavities; sterile                             <ul style="list-style-type: none"> <li>■ Example: Sterile talc for pleurodesis</li> </ul> </li> </ul> </li> </ul>
	<ul style="list-style-type: none"> <li>* <b>Insufflations</b> <ul style="list-style-type: none"> <li>○ Medicated powders introduced into body cavities (nose, ear, throat, vagina) via insufflator                             <ul style="list-style-type: none"> <li>■ Example: Boric acid insufflation powder</li> </ul> </li> </ul> </li> </ul>
	<ul style="list-style-type: none"> <li>* <b>Snuffs</b> <ul style="list-style-type: none"> <li>○ Finely divided powders inhaled into nostrils for antiseptic/decongestant action</li> </ul> </li> </ul>



✱ **Dentifrices (Tooth Powders)**

- Used with toothbrush for teeth cleaning
- Contains abrasives ( $\text{CaCO}_3$ ,  $\text{NaCl}$ ) and flavors
  - Example: Tooth powder with  $\text{CaSO}_4$ ,  $\text{NaCl}$ , peppermint oil

### 3. Simple and Compound Powders for Internal Use

- ✱ **Simple Powders:** Contain **only one ingredient** (either crystalline or amorphous) divided into doses.
- ✱ **Compound Powders:** Contain **two or more ingredients**, mixed and divided into single-dose portions.

### 4. Powders Enclosed in Cachets

- ✱ Solid unit dosage forms made from rice paper (Enclose measured doses of powders within two rice paper halves).
- ✱ Used to enclose **nauseous or unpleasant-tasting powders**.
- ✱ Available in different sizes (0.2 to 1.5 g capacity).
- ✱ Softened in water before swallowing.

### 5. Compressed Powders (Tablets)

- ✱ **Powders are compressed** into solid dosage forms for easier administration and dosage control.
- ✱ Common in pharmaceuticals due to **longer shelf life and precise dosing**.

## DISPENSING OF POWDERS INVOLVING SPECIAL PROBLEMS

### 1. Volatile Substances

- ✱ Some vegetable powders contain volatile oils that may be lost during grinding.
- ✱ Substances like **menthol, camphor, and essential oils** may evaporate when incorporated into powders.
- ✱ **Solution:** Use **double wrapping** (inner wax paper, outer thick paper) to minimize volatilization.

### 2. Hygroscopic and Deliquescent Powders

- ✱ **Hygroscopic powders** absorb moisture from the air.
- ✱ **Deliquescent powders** absorb so much moisture that they dissolve.
- ✱ **Examples:** Ammonium chloride, sodium bromide, zinc chloride.
- ✱ **Solution:**
  - Supplied in **granular form** to reduce surface area exposure.
  - **Double wrapping** is recommended.
  - **Aluminum foil or plastic covers** are advisable in humid weather.

### 3. Efflorescent Powders

- ✱ Some crystalline substances release **water of crystallization** when exposed to humidity, leading to dampness.
- ✱ **Examples:** Caffeine, citric acid, ferrous sulfate.
- ✱ **Solution:**
  - Use the **anhydrous form** of the substance.
  - Mix with an **inert absorbent** before blending with other ingredients.

#### 4. Eutectic Mixtures

- \* When two or more substances mix, they liquefy due to forming a compound with a lower melting point.
- \* **Examples:** Menthol, camphor, phenol, aspirin, chloral hydrate.
- \* **Solutions:**
  - Dispense as **separate sets of powders**.
  - Mix with an **inert absorbent** (e.g., magnesium carbonate, kaolin, starch).
  - If other ingredients are present, first **triturate eutectic substances**, then incorporate the rest.

#### AEROSOL POWDERS

- \* Used in dry-powder inhalers (DPI) for conditions like asthma.
- \* Micronized particles (**1–6 µm**) allow deep lung penetration.
- \* Often contain **pharmaceutical diluents** like crystalline alpha-lactose monohydrate to improve flow and stability.

#### GRANULES

- \* Granules are **agglomerates** of smaller powder particles. They are irregularly shaped but can also be prepared as **spherical** granules.
- \* They usually fall within the **4- to 12-mesh** sieve size range, but their size can vary depending on their intended application.



#### METHODS OF GRANULE PREPARATION

Granules can be prepared using **wet** and **dry** methods.

##### Wet Granulation

##### 1. Moistening Method:

- The powder or powder mixture is moistened.
- The resulting paste is passed through a screen of desired mesh size.
- The granules are placed on drying trays and dried by **air** or **heat**.
- Granules are moved periodically to prevent sticking.

##### 2. Fluid Bed Processing:

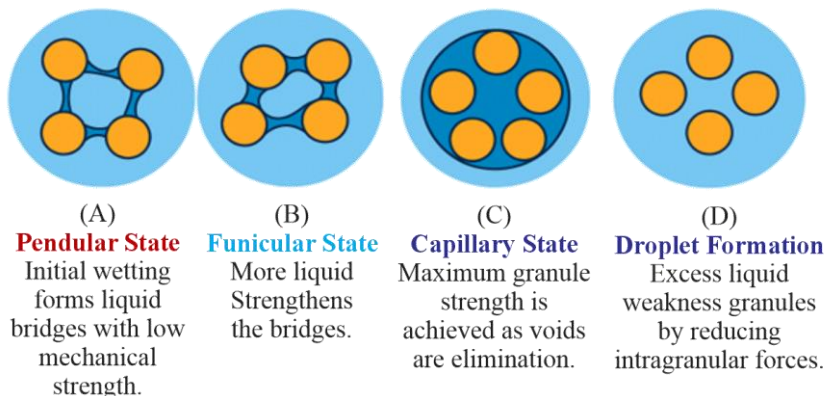
- Particles are dispersed in a conical chamber.
- A liquid excipient is sprayed onto them while they remain suspended.
- The mixture dries, forming **granules or pellets** of defined particle size.

##### Granule Formation Stages

**Pendular State** → **Funicular State** → **Capillary State** → **Droplet Formation**



### Stages in development of moist granules as proportion of liquid



#### ◆ **Dry Granulation**

##### **1. Roll Compaction Method:**

- Powder is passed through a **roll compactor**.
- Two mechanically rotating metal rolls press the powder into **dense sheets**.
- The compacted material is granulated into uniform particle size using a **granulating machine**.
- This method is often integrated into a **compactor-granulation system**.

##### **2. Slugging Method:**

- The powder is compressed into **large tablets or slugs** using high pressure (8,000–12,000 lb).
- Slugs are then granulated into **desired particle sizes** for tablet production.
- Some fine particles (fines) that do not form granules are collected and reprocessed.

#### ◆ **Properties and Advantages of Granules**

- \* **Better Flowability:** Compared to powders, granules flow freely from hoppers into tablet presses, improving the efficiency of tablet production.
- \* **Improved Stability:** Due to their smaller surface area, granules are **less affected by humidity** and are less likely to cake or harden over time.
- \* **Enhanced Wettability:** Granules dissolve or suspend more easily in liquids than fine powders, which may float on the surface.

#### ◆ **Granulated Pharmaceutical Products**

##### \* **Antibiotic Granules for Oral Suspension:**

- Contain the active drug along with **colorants, flavorants, and excipients**.
- Mixed with purified water just before dispensing to create a stable liquid.
- **Examples:**
  - **Biaxin** (Clarithromycin)
  - **Omnicef** (Cefdinir)
  - **Augmentin ES-600** (Amoxicillin/Clavulanate Potassium)
  - **Ceftin** (Cefuroxime Axetil)

##### \* **Lactinex Granules:**

- Contain **Lactobacillus acidophilus** and **Lactobacillus bulgaricus**.

- Used for **diarrhea treatment**, including antibiotic-associated diarrhea.
- Can be mixed with water, sprinkled on food, or eaten directly.

★ **Effervescent Granules:**

- Dissolve in water before use.
- Can also be compressed into effervescent tablets.
- Example: **Zantac EFFERdose Tablets** (Ranitidine).

**EFFERVESCENT GRANULATED SALTS**

- ★ Effervescent salts are **granules or coarse powders** that contain a medicinal agent in a **dry mixture** of sodium bicarbonate, citric acid, and tartaric acid.
- ★ When added to water, these components react to release **carbon dioxide**, creating effervescence.



◆ **Role of Citric and Tartaric Acids**

- ★ **Tartaric Acid Alone:** Results in granules that lose firmness and crumble.
- ★ **Citric Acid Alone:** Creates a sticky mixture that is difficult to granulate.
- ★ **Combination of Both:** Prevents these issues and ensures a stable granule formulation.

◆ **Methods of Preparation**

**1. Dry or Fusion Method**

- ★ **Binding Agent:** The water present in citric acid acts as the binding agent.
- ★ **Procedure:**
  1. **Citric acid crystals** are powdered and mixed with other powders of the same sieve size for uniformity.
  2. **Mixing is done quickly** in a low-humidity environment to prevent premature reaction.
  3. The mixture is **heated at 34°C to 40°C** in an oven.
  4. The released water of crystallization dissolves some powder, triggering a slight reaction and softening the mass.
  5. Once a spongy consistency is achieved, the mass is passed through sieves to obtain the desired granule size:
    - **No. 4 sieve** → Large granules
    - **No. 8 sieve** → Medium granules
    - **No. 10 sieve** → Small granules
  6. The granules are **dried at ≤54°C** and immediately stored in tightly sealed containers.

**2. Wet Method**

- ★ **Binding Agent:** Water added to alcohol acts as the moistening agent.
- ★ **Procedure:**
  1. All powders used can be **anhydrous**, unlike the fusion method.
  2. Water is added gradually in portions to form a **pliable mass**.
  3. Granules are then prepared and dried using the same method as in the fusion process.