

## Assignment P2-1

1. In the molecular surface reaction of  $N_2$  interaction with iron (Fe) solid surface, the elemental iron is *bcc* in its ground state, which is metallic only due to 2 of its valence electrons. Within the free electron theory, calculate the Fermi energy ( $E_F$  in the units of eV) of elemental iron (Fe at  $T = 0$  K). Consider the atomic radius of Fe is  $1.86 \text{ \AA}$ .

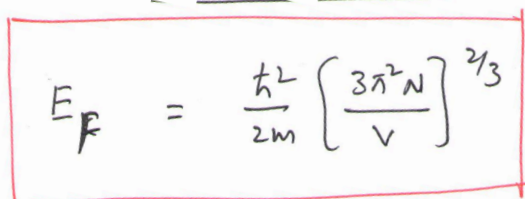
Plank constant ( $\hbar$  ; read it as h-bar) =  $1.05 \times 10^{-34} \text{ J.s}$

Mass of electron ( $m$ ) =  $9.1 \times 10^{-31} \text{ kg}$

1 eV =  $1.6 \times 10^{-19} \text{ J}$

### Solution:

From lecture notes, the Fermi energy can be written as:


$$E_F = \frac{\hbar^2}{2m} \left[ \frac{3\pi^2 N}{V} \right]^{2/3}$$

Since Fe crystal is *bcc*, the unit cell has 2 atoms per cell.

Volume of the *bcc* unit cell is:  $[4r/\sqrt{3}]^3$ ,

$r$  is the radius of Fe =  $1.86 \text{ \AA} = 1.86 \times 10^{-10} \text{ m}$

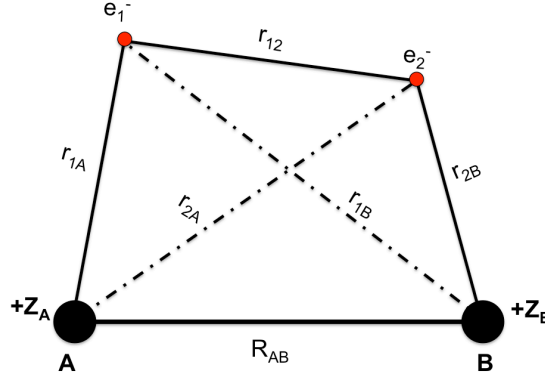
Therefore, volume of the unit cell =  $7.93 \times 10^{-29} \text{ m}^3$

Since Fe crystal has 2 atoms per unit cell, the crystal also will have 2 valence electrons per unit cell.

From this one can get the value of  $N/V$ .

Upon substitution of the values given for  $\hbar$  and mass of the electron, one can get the Fermi energy ( $E_F$ ) =  $4.98 \times 10^{-19} \text{ J} = 3.1 \text{ eV}$

2. Write the Hamiltonian that one would need to calculate the electronic wavefunction for a system of interacting particles. The system consisting of two atoms labeled as A and B each with an electron is pictorially depicted below.



Solution

$$\frac{p_1^2}{2m} - \frac{e^2}{4\pi\epsilon_0 r_{1A}} + \frac{p_2^2}{2m} - \frac{e^2}{4\pi\epsilon_0 r_{2B}} - \frac{e^2}{4\pi\epsilon_0 r_{1B}} - \frac{e^2}{4\pi\epsilon_0 r_{2A}} +$$

$$\frac{e^2}{4\pi\epsilon_0 r_{12}} + \frac{e^2}{4\pi\epsilon_0 R_{AB}} + \frac{p_A^2}{2M_A} + \frac{p_B^2}{2M_B}$$

For the electronic Hamiltonian, the last two terms (kinetic energy of the nucleus is ignored by invoking the Born-Oppenheimer approximation, see below). Note that the nucleus may be designated with the charge  $Ze$ , for example, the charge on the two atoms labeled as A and B would be  $Z_A e$  and  $Z_B e$ , respectively. The idea is: according to the given diagram, the notation must be strictly followed.

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For electrons ( $i$ ) and nuclei ( $\alpha$ ), the Schrödinger equation is approximated as (following the clamped nuclei model in other words, Born-Oppenheimer approximation)

$$\hat{H}_{el}\psi(r, R) = E_{el}\psi(r, R)$$

$$\hat{H}_{el} = \hat{T}_{el}(r) + \hat{V}_{eN}(r, R) + \hat{V}_{rr}$$

$$\hat{H}_{el} = -\frac{1}{2}\sum_i \nabla_i^2 - \frac{1}{2}\sum_{\alpha,i} \frac{Z_\alpha e^2}{r_i - R_\alpha} + \frac{1}{2}\sum_{i \neq j} \frac{e^2}{r_i - r_j}$$