

**BAC-C102**  
**SEMESTER EXAMINATION DECEMBER 2023**  
**CLASS: B. Tech SEMESTER: I**  
**ENGINEERING CHEMISTRY**

Time: 3 hours

Max. Marks: 70

**Note:** Question Paper is divided into two sections: **A and B**. Attempt both the sections as per given instructions.

**SECTION-A (SHORT ANSWER TYPE QUESTIONS)**

**Instructions:** Answer any **five** questions in about 150 words each. Each question carries six marks.  
(5 X 6 = 30 Marks)

- Question-1: ✓ What is atomic radius? Why does atomic radius decrease across a period? Why are anions bigger than their parent atoms? ✓
- Question-2: ✓ What is hydrogen bond? Explain the different types of hydrogen bonds with examples. ✓
- Question-3: ✓ Define the rate of reaction. How does the rate of a reaction depend upon the concentration and surface area of the reactants? ✓
- Question-4: Define any two of the following with an example for each: (a) Enthalpy of a system (b) Standard electrode potential (c) Arrhenius concept of acids and bases
- Question-5: ✓ What are monomers and polymers? Explain with suitable examples. ✓
- Question-6: What is vulcanization of rubber? What are the advantages of vulcanized rubber?
- Question-7: ✓ What are nanomaterials? Explain the top down and bottom up approach of synthesis of nanomaterials. ✓
- Question-8: What are Fullerenes? Discuss the properties and applications of Fullerenes in brief.
- Question-9: Give the synthesis and uses of Paracetamol.
- Question-10: What is optical isomerism? How many optical isomers are possible in tartaric acid?

**SECTION-B (LONG ANSWER TYPE QUESTIONS)**

**Instructions:** Answer any **four** questions in detail. Each question carries 10 marks.

(4 X 10 = 40 Marks)



Question-1: Define the term electronegativity. How does it vary in the periodic table? Discuss the factors which influence the electronegativity of an atom.

Question-2: Define hybridization. Explain the structure of  $\text{BF}_3$  and  $\text{NH}_3$  molecules on the basis of hybridization.

Question-3: What is activation energy? How is activation energy of a chemical reaction determined?

Question-4: Write the Nernst equation. Discuss any two applications of the Nernst equation in detail.

Question-5: Discuss the preparation and uses of (a) Teflon (b) Nylon 6,6

Question-6: Write short notes on (i) Addition and condensation polymers (ii) Natural and synthetic rubbers

Question-7: What is nanotechnology? Discuss the applications of nanotechnology in various fields.

Question-8: What are substitution reactions? Describe  $\text{S}_{\text{N}}1$  and  $\text{S}_{\text{N}}2$  substitution reactions with examples.