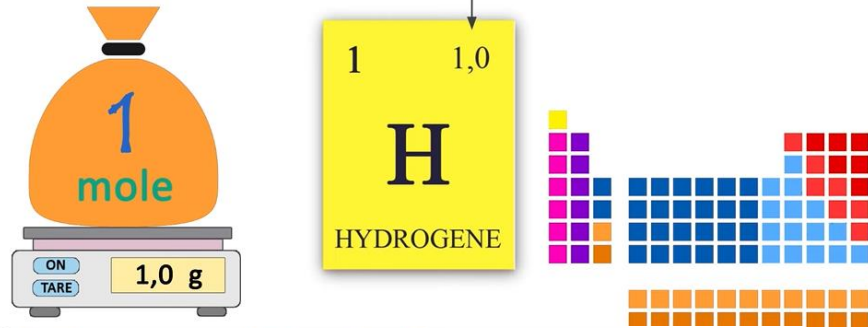


# La MOLE 2

## Masse Molaire atomique



$$M(H) = 1,0 \text{ g.mol}^{-1}$$

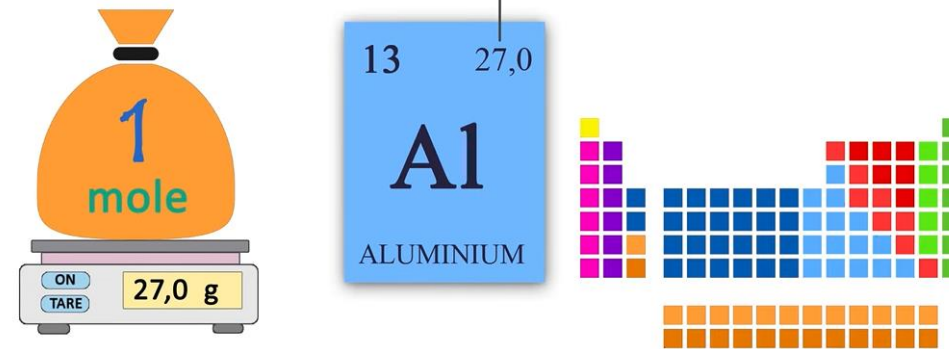


## Masse Molaire atomique

## Masse Molaire atomique



$$M(Al) = 27,0 \text{ g.mol}^{-1}$$



$$n = \frac{m(Al)}{M(Al)} = \frac{80,0}{27,0} = 2,96 \text{ mol}$$

