

Your Results for: "Multiple choice questions"

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Summary of Results

56% Correct of 18 Scored items:10 Correct:  56%8 Incorrect:  44%[More information about scoring](#)

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- 1.** **CORRECT** In which region of the electromagnetic spectrum does an absorption at 600 nm come?
- Your Answer:** Visible.
-
- 2.** **INCORRECT** A solution of $0.001 \text{ mol dm}^{-3}$ NiSO_4 is placed in an optical cell of pathlength 1 cm, and the absorption spectrum is recorded. The absorptions have characteristic λ_{max} and ϵ_{max} values. What are the correct units of ϵ_{max} ?
- Your Answer:** $\text{cm dm}^3 \text{ mol}^{-1}$
- Correct Answer:** $\text{dm}^3 \text{ mol}^{-1} \text{ cm}^{-1}$
-
- 3.** **CORRECT** In which region of the electromagnetic spectrum does an absorption at 177 nm come?
- Your Answer:** Vacuum-UV.
-
- 4.** **CORRECT** What does the notation $\sigma^* \leftarrow n$ mean?
- Your Answer:** Absorption; transition from a non-bonding MO to σ^* MO.
-
- 5.** **INCORRECT** How do values of λ_{max} for the $n^* \leftarrow n$ transitions vary among a series of conjugated polyenes?

Your Answer: (blank)

6. **INCORRECT** What is a *red shift*?

Your Answer: The shifting of an absorption to higher energy.

Correct Answer: The shifting of an absorption to lower energy.

7. **CORRECT** What is a chromophore?

Your Answer: A group of atoms in a compound responsible for the absorption of electromagnetic radiation.

8. **CORRECT** Which of the following is the principal chromophore in an azo-dye?

Your Answer: N=N

9. **INCORRECT** Lycopene ($\lambda_{\text{max}} = 469 \text{ nm}$) is present in tomatoes. What colour of light does lycopene absorb?

Your Answer: Red.

Correct Answer: Blue.

10. **CORRECT** In which of the following are the π -electrons *not* delocalized?

Your Answer: Hepta-1,6-diene.

11. **INCORRECT** All but one of the following groups can donate a lone pair of electrons into the carbon π -system of a conjugated polyene. Which is the odd one out?

Your Answer: SMe

Correct Answer: Me

12. **INCORRECT** Which statement is *incorrect* about the allyl anion and buta-1,3-diene?

Your Answer: Both have four π -electrons.

Correct Answer: Both have four π MOs.

13. **CORRECT** Which of the following arrangements of carbon-carbon double bonds along the backbone of a polyene does *not* correspond to a conjugated system?

Your Answer: $-\text{C}=\text{C}-\text{C}-\text{C}=\text{C}-$

14. **CORRECT** An aqueous solution of a dye has a strong absorption with $\lambda_{\text{max}} = 464 \text{ nm}$. The colour of the solution is:

Your Answer: orange.

15. **CORRECT** Which of the following absorption maxima is *not* in the visible range of the electronic spectrum?

Your Answer: $\lambda_{\text{max}} = 250 \text{ nm}$

16. **CORRECT** Which of the following statements is consistent with an electronic absorption being broad?

Your Answer: An electronic absorption includes vibrational and rotational structure.

17. **INCORRECT** Two solutions of the same compound were made up. Solution A was of concentration $0.98 \times 10^{-4} \text{ mol dm}^{-3}$, and solution B was $1.66 \times 10^{-2} \text{ mol dm}^{-3}$. The electronic spectrum of solution A contained one absorption at $\lambda_{\text{max}} = 230 \text{ nm}$, while the

spectrum of solution B contained absorptions at $\lambda_{\text{max}} = 230$ and 365 nm. The difference in the spectra was because:

Your Answer: the value of λ_{max} depends on concentration.

Correct Answer: ϵ_{max} for the absorption at 365 nm is much smaller than ϵ_{max} of the band at 230 nm.

18. **INCORRECT**

A compound **X** is characterized in its electronic spectrum by an absorption with $\lambda_{\text{max}} = 217$ nm ($\epsilon_{\text{max}} = 21\,000 \text{ dm}^3 \text{ mol}^{-1} \text{ cm}^{-1}$). Of the compounds given below, **X** is most likely to be:

Your Answer: water.

Correct Answer: buta-1,3-diene.

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