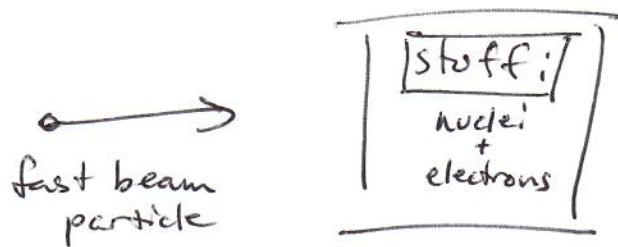


Collisions, Energy loss, & Scattering

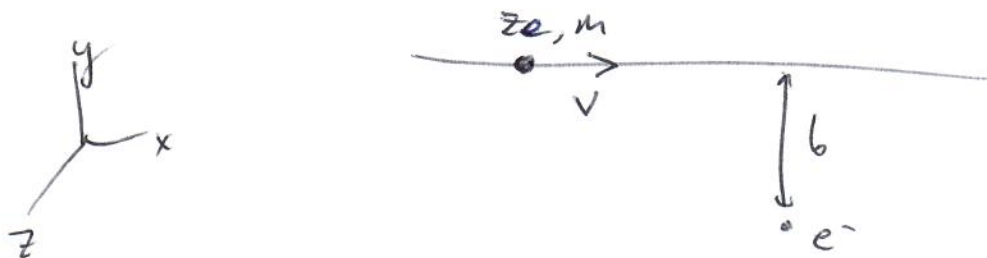
Last unit on the properties & dynamics of single charges.



Nuclei are heavy & cause large-angle quasi-elastic scattering. Electrons are light and can absorb energy without significant deflection. Both are important effects.

energy transfer in a coulomb collision

start w/ heavy beam particle & atomic e^- .



Atomic e^- have typical speeds $\sim 1\% c$.

Assume beam is much faster.

Then the transverse momentum kick is (for small deflections)

$$\Delta p \approx \int_{-\infty}^{\infty} dt (E_y) (ze)$$

$$= \int_{-\infty}^{\infty} \frac{ze^2}{4\pi\epsilon_0} \frac{\gamma b dt}{(b^2 + \gamma^2 v^2 t^2)^{3/2}}$$

see last lecture

can absorb γ into t - result indep of γ !

$$= \frac{2Ze^2}{4\pi\epsilon_0 b v}$$

If the e^- is nonrelativistic, it will gain energy

$$\Delta E_e \approx \frac{(\Delta p)^2}{2m_e} = \frac{2Z^2 e^4}{(4\pi\epsilon_0)^2 m_e v^2 b^2}$$

The beam deflection angle is



$$\frac{\Delta p}{p} \approx \theta = \frac{2Ze^2}{4\pi\epsilon_0 p b v}$$

We need $\Delta p/p \ll 1$ for the approximation
used to compute Δp to hold (small deflection)
and $p = \gamma m v$, so

$$\frac{Z Z e^2}{4\pi\epsilon_0 \gamma m v^2} \ll 1$$

$$\text{or } b_{\min} = \frac{Z e^2}{4\pi\epsilon_0 \gamma m v^2} \quad (\text{ignore } Z's)$$

When $b \rightarrow b_{\min}$, our calculations are breaking.

Another way to see it: we want the
electron to move a small distance cf. b .

The duration of the collision is $\Delta t \approx \frac{b}{\gamma v}$

(start at E_y - the γ is due to the
"packing up" of the transverse fields w/ γ .)

During the collision, the e^- moves a distance

$$d \sim \left(\frac{\Delta p}{2m} \right) \times \Delta t = \frac{Z e^2}{\gamma m v^2} = b_{\min}.$$

\uparrow avg velocity

There's also a b_{\max} . If the collision lasts too long, the fact that the electron is bound to a nucleus will matter, and it becomes a 3-body problem.

The nucleus moves too and its mass enters the ΔE formula.

require $\Delta t \lesssim \frac{1}{\omega}$ $\omega = \text{orbital frequency of the } e^-$

$\sim \frac{b}{\gamma v}$ (cf. $E_y \sim \frac{\gamma b}{(b^2 + \gamma^2 v^2 t^2)^{3/2}}$ - field is important over times $\lesssim \frac{b}{\gamma v}$)

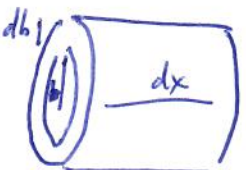
$\Rightarrow b_{\max} = \frac{\gamma v}{\omega}$

(numbers: $\omega \sim \frac{1\% c}{1 \text{ \AA}}$ \leftarrow atomic e^- speed $\sim 3 \times 10^{16} \text{ Hz}$
 \leftarrow atomic radius

for $\gamma v \sim c$, would get $b_{\max} = 10^{-8} \text{ m}$
 $= 100 \text{ \AA}$)

If a block of matter has N atoms/unit vol
and Z e^- /atom,

$$dn = 2\pi b db dx N Z \text{ electrons}$$

in  \leftarrow this material.

So the energy loss is

$$\frac{dE}{dx} = 2\pi N Z \int_{b_{min}}^{b_{max}} \Delta E(b) b db$$

$$\approx 2\pi N Z \int_{b_{min}}^{b_{max}} b db \frac{2Z^2 e^4}{mv^2} \frac{1}{b^2}$$

$$= \frac{4\pi N Z^2 e^4}{mv^2} \ln\left(\frac{b_{max}}{b_{min}}\right)$$

$$\text{with } b_{max}/b_{min} \approx \frac{\gamma^2 mv^3}{Ze^2w}$$

The log forgives all our \approx in the precise
form of $b_{min, max}$.