

Periodic Table of the Elements

Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1	<div>1.00794 1 H</div>												<div>4.00260 2 He</div>					
2	<div>6.941 3 Li</div>	<div>9.01218 4 Be</div>												<div>12.011 6 C</div>	<div>14.0067 7 N</div>	<div>15.9994 8 O</div>	<div>18.9984 9 F</div>	<div>20.180 10 Ne</div>
3	<div>22.98977 11 Na</div>	<div>24.305 12 Mg</div>											<div>26.98154 13 Al</div>	<div>28.0855 14 Si</div>	<div>30.97376 15 P</div>	<div>32.065 16 S</div>	<div>35.453 17 Cl</div>	<div>39.948 18 Ar</div>
4	<div>39.0983 19 K</div>	<div>40.08 20 Ca</div>	<div>44.9559 21 Sc</div>	<div>47.867 22 Ti</div>	<div>50.9415 23 V</div>	<div>51.996 24 Cr</div>	<div>54.9380 25 Mn</div>	<div>55.845 26 Fe</div>	<div>58.9332 27 Co</div>	<div>58.693 28 Ni</div>	<div>63.546 29 Cu</div>	<div>65.409 30 Zn</div>	<div>69.723 31 Ga</div>	<div>72.64 32 Ge</div>	<div>74.9216 33 As</div>	<div>78.96 34 Se</div>	<div>79.904 35 Br</div>	<div>83.798 36 Kr</div>
5	<div>85.4678 37 Rb</div>	<div>87.62 38 Sr</div>	<div>88.9059 39 Y</div>	<div>91.224 40 Zr</div>	<div>92.9064 41 Nb</div>	<div>95.94 42 Mo</div>	<div>98 43 Tc</div>	<div>101.07 44 Ru</div>	<div>102.906 45 Rh</div>	<div>106.42 46 Pd</div>	<div>107.868 47 Ag</div>	<div>112.41 48 Cd</div>	<div>114.818 49 In</div>	<div>118.71 50 Sn</div>	<div>121.760 51 Sb</div>	<div>127.60 52 Te</div>	<div>126.904 53 I</div>	<div>131.29 54 Xe</div>
6	<div>132.905 55 Cs</div>	<div>137.33 56 Ba</div>	<div>138.9055 57 La</div>	<div>178.49 72 Hf</div>	<div>180.948 73 Ta</div>	<div>183.84 74 W</div>	<div>186.207 75 Re</div>	<div>190.23 76 Os</div>	<div>192.217 77 Ir</div>	<div>195.08 78 Pt</div>	<div>196.967 79 Au</div>	<div>200.59 80 Hg</div>	<div>204.383 81 Tl</div>	<div>207.2 82 Pb</div>	<div>208.980 83 Bi</div>	<div>209 84 Po</div>	<div>210 85 At</div>	<div>222 86 Rn</div>
7	<div>(223) 87 Fr</div>	<div>(226) 88 Ra</div>	<div>(227) 89 Ac</div>	<div>(261) 104 Rf</div>	<div>(262) 105 Db</div>	<div>(266) 106 Sg</div>	<div>(272) 107 Bh</div>	<div>(277) 108 Hs</div>	<div>(276) 109 Mt</div>	<div>(281) 110 Ds</div>	<div>(280) 111 Rg</div>	<div>(285) 112 Cn</div>	<div>(284) 113** Uut</div>	<div>(289) 114 Uuq</div>	<div>(288) 115 Uup</div>	<div>(292) 116 Uuh</div>	<div>(?) 117 Uus</div>	<div>(294) 118 Uuo</div>

KEY

Atomic Mass →

Symbol →

Atomic Number →

Electron Configuration →

← Selected Oxidation States

Relative atomic masses are based on ¹²C = 12 (exact)

Note: Numbers in parentheses are mass numbers of the most stable or common isotope.

*denotes the presence of (2-8-) for elements 72 and above

**The systematic names and symbols for elements of atomic numbers 113 and above will be used until the approval of trivial names by IUPAC.