

Circuit Lab

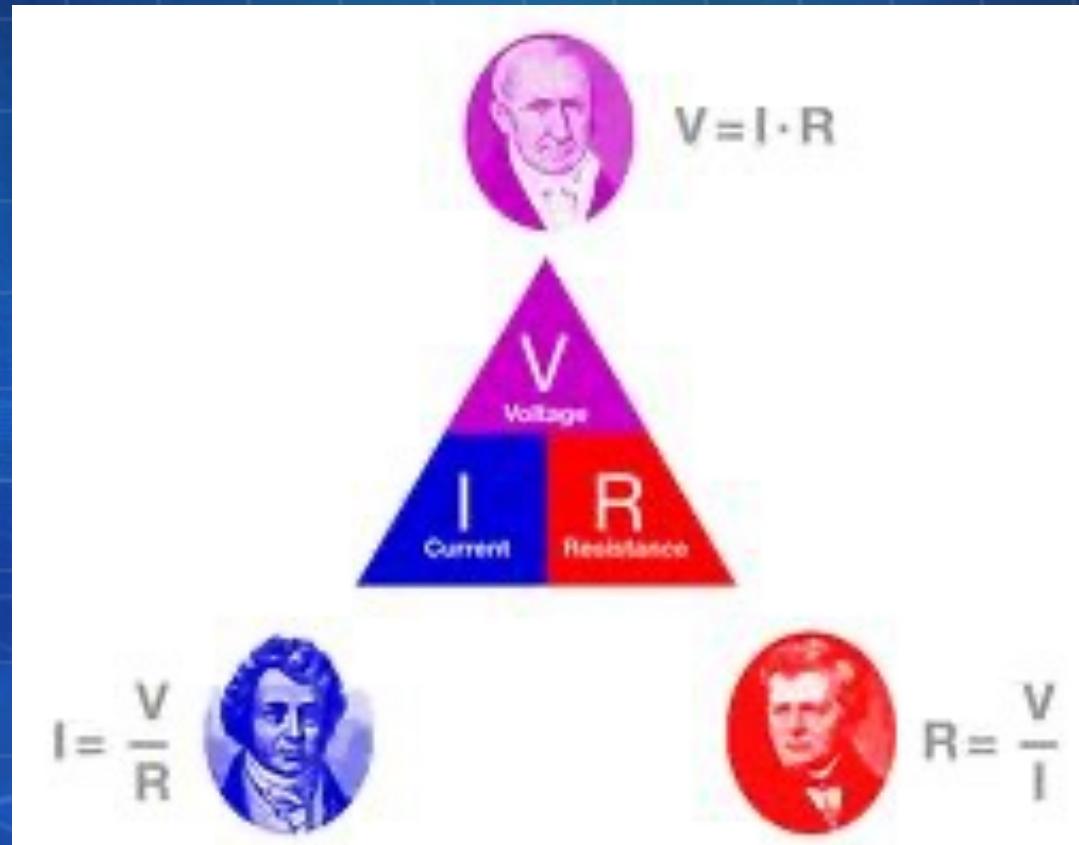
Practice #3—Kitchen Batteries, Series, and Parallel

Mr. Burleson
geaux15@hotmail.com

Agenda

- 15 minutes—Grading homework.
- 30 minutes—Learning Lesson of the Day
- 15 minutes—In Practice quick test on Lesson of the Day
- 25 minutes—Practical testing
- 5 minutes—Sending out homework

The Big Three—Volta, Ampere, and Ohm

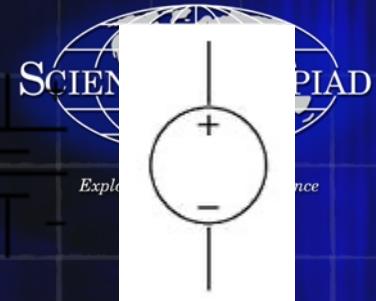


How to identify batteries?

- All batteries are Direct Current (DC)
- They are usually denoted by their Voltage
 - Normally 1.5V, 3V, 6V, 9V, 12V, and 24V
 - This is because most cells produce 1.5V and then are put in series.
- **POSITIVE (+) terminals** are either marked or painted **RED**, sometimes with a cover to reduce chance of a **SHORT CIRCUIT**
- **NEGATIVE (-) terminals** are normally marked or painted **BLACK** or **BLUE**.



Exploring the World of Science



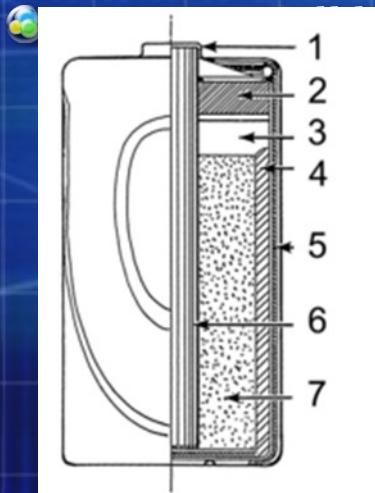
Batteries

Dry CELL

- Electrolyte is immobilized as a paste

- Zinc-carbon

- Alkaline batteries



batteries

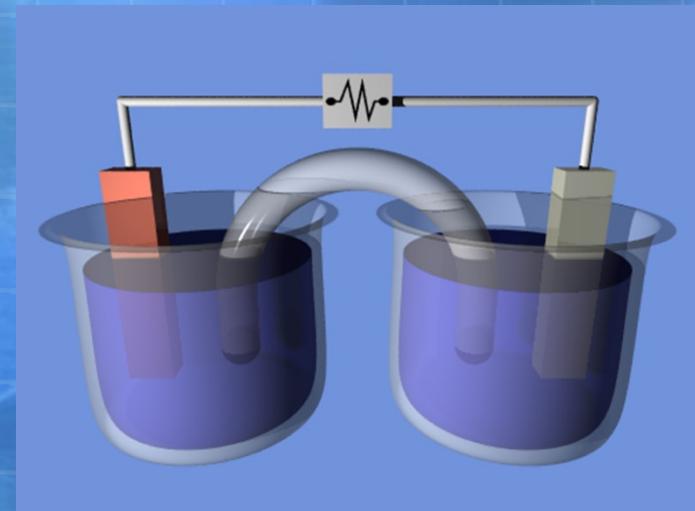
Line art drawing of a dry cell:
1. brass cap,
2. plastic seal,
3. expansion space,
4. porous cardboard,
5. zinc can,
6. carbon rod,
7. chemical mixture
(wikipedia.org)

WET CELL

- Has a liquid electrolyte

- Lead-acid (car batteries)

- Nickel-Cadmium



Kitchen Batteries

- You can create a simple battery with two different kinds of metal in a conducting solution
- To the right is an Aluminum and Copper strip in a glass of soda



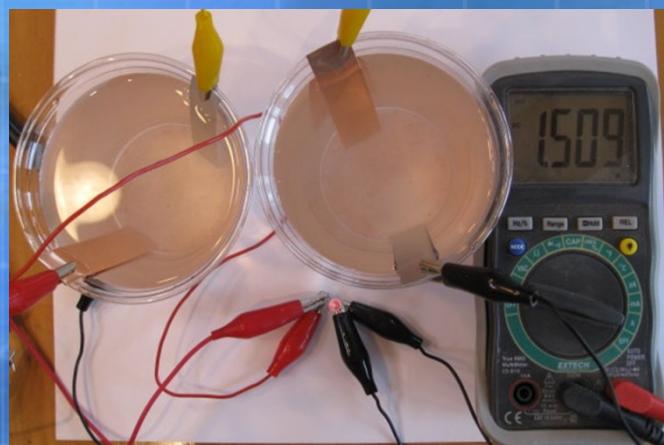
Copper-Aluminum Coke Battery

Kitchen Batteries

- You can create different types by changing the metals or changing the electrolyte solution
- You can add cells to increase the voltage



Zinc-Aluminum Coke Battery



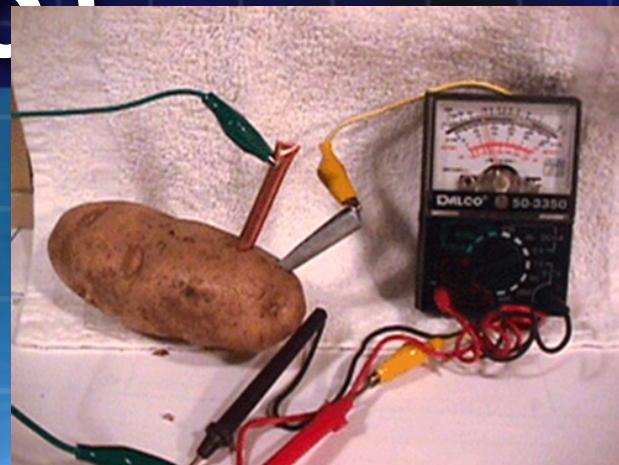
Two Cell Zinc-Aluminum Sea Water Battery

Kitchen Batteries (Potato)



Exploring the World of Science

- Use the potato as the electrolyte
- Use a Copper electrode and an Zinc electrode
- Put them close together, but not touching
- Produces about 1.2 V
- (from miniscience.com)



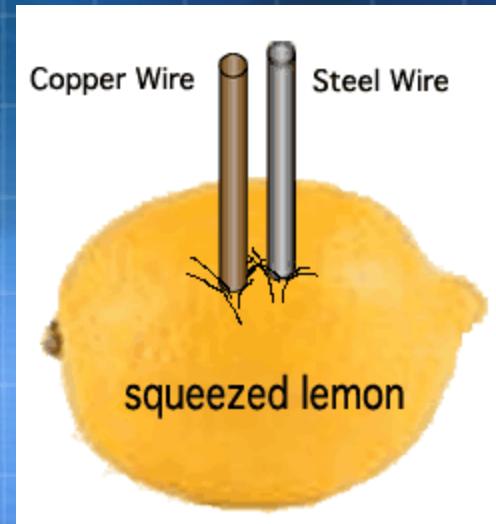
Three Cell Copper-Zinc Potato Battery

Kitchen Batteries (Lemon)



Exploring the World of Science

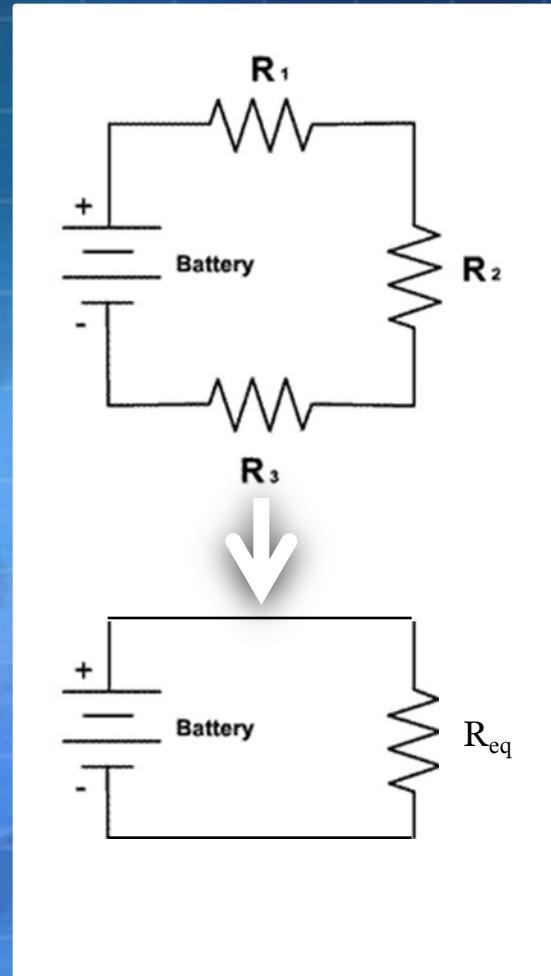
- Use the lemon as the electrolyte
- Use a Copper electrode and an Steel electrode
- Put them close together, but not touching
- Produces about 0.7 V
- (from energyquest.ca.gov)



One Cell Copper- Steel Lemon Battery, a Penny can be the Copper and a Paper Clip can be the Steel

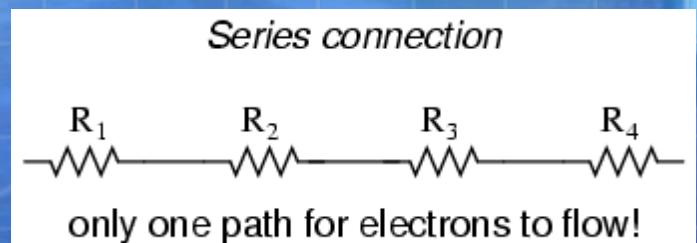
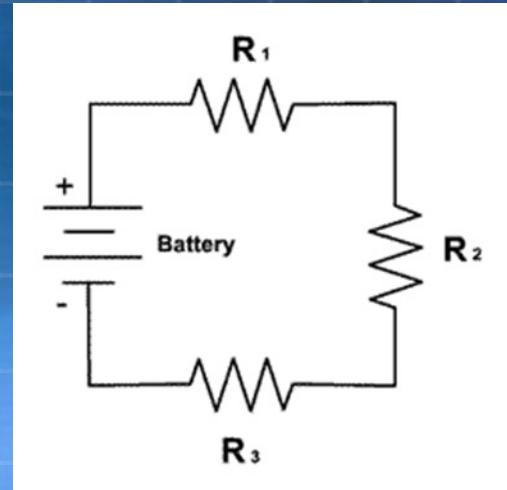
Equivalent Resistance or Load

- Multiple resistors can be combined into a single resistance to represent all of them called an Equivalent Resistance or R_{eq}
- The current flowing through both of the circuits on the right is exactly the same



Series Circuit

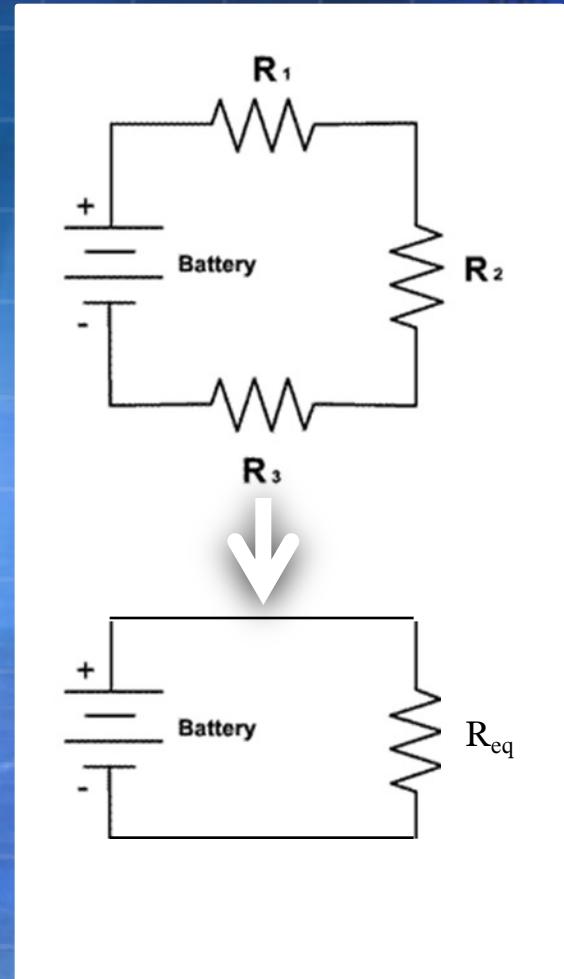
- ➊ Two or more resistors in series
- ➋ Current is the same through all resistors (electrons have no other path)
- ➌ Voltage is split between resistors
- ➍ $R_{eq} = R_1 + R_2 + R_3$
- ➎ R_{eq} is always greater than the largest single resistance
- ➏ Current = Voltage / R_{eq}



Series Circuit (example)

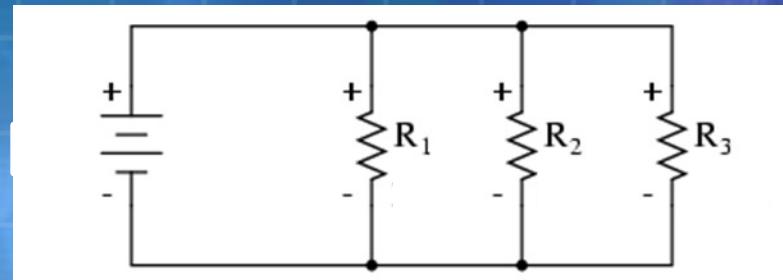
- $R_{eq} = R_1 + R_2 + R_3$
- $R_1 = 300 \Omega$
- $R_2 = 300 \Omega$
- $R_3 = 400 \Omega$
- $R_{eq} = 300 \Omega + 300 \Omega + 400 \Omega$
- $R_{eq} = 1,000 \Omega$ or $1 \text{ k}\Omega$
- If $V = 1,000 \text{ V}$ or 1 kV
- $I = V/R = 1\text{kV}/1 \text{ k}\Omega = 1 \text{ A}$

- ** R_{eq} = the sum of the resistors or if they are all the same value then multiply them by that number
 - Two 300Ω resistors in series = $2 \times 300\Omega = 600\Omega$
 - Three 300Ω resistors in series = $3 \times 300\Omega = 900\Omega$
 - Two 200Ω resistors in series = $2 \times 200\Omega = 400\Omega$



Parallel Circuit

- ➊ Two or more resistors in parallel
- ➋ Voltage is the same through all resistors
- ➌ Current is split between resistors (electrons are split between all branches)
- ➍ $\frac{1}{R_{eq}} = \frac{1}{R_1} + \frac{1}{R_2} + \frac{1}{R_3}$
- ➎ R_{eq} is always less than the smallest single resistance
- ➏ Current (from battery) = Voltage / R_{eq} , which is then split among the three branches



Parallel Circuit (Example)



Exploring the World of Science

- $1/R_{eq} = 1/R_1 + 1/R_2$

- $R_1 = 2 \Omega$

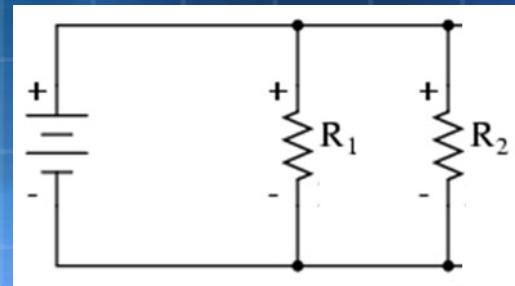
- $R_2 = 2 \Omega$

- $1/R_{eq} = 1/2\Omega + 1/2\Omega = 1/\Omega$

- $R_{eq} = 1 \Omega$

- If $V = 6 V$

- $I = V/R = 6 V / 1 \Omega = 6A$, which is split among the two resistors



Parallel Circuit (Example)



Exploring the World of Science

- 1/R_{eq} = 1/R₁ + 1/R₂

- R₁ = 4 Ω

- R₂ = 4 Ω

- 1/R_{eq} = 1/4 Ω + 1/4 Ω = 1/2 Ω

- R_{eq} = 2 Ω

- If V = 6 V

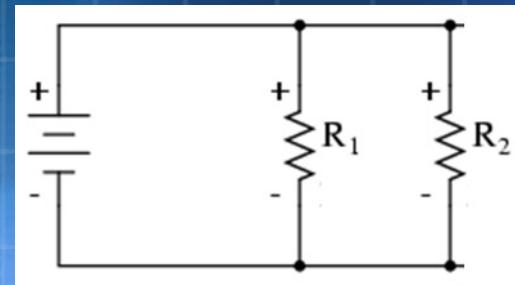
- I = V/R = 6 V / 2 Ω = 3 A, which is then split among the two resistors

- **If the resistors have the same value, R_{eq} = the resistance is divided by the number in parallel

- Two 4Ω resistors in parallel are 4Ω/2 = 2Ω

- Two 6Ω resistors in parallel are 6Ω/2 = 3Ω

- Three 6Ω resistors in parallel are 6Ω/3 = 2Ω





In Practice Quiz

- Please do the quiz using your binder!!!
- Ask Questions!!!!

Practical

- ➊ Create kitchen battery using the lemons and potatoes provided
 - ➌ Measure the voltage and determine the **POSITIVE (+)** and **NEGATIVE (-)** terminals
 - ➌ Identify the electrolyte, cathode, and anode.
- ➋ Create a series circuit and measure the voltage of all parts
 - ➌ What is the current through each one?
- ➌ Create a parallel circuit and measure the voltage of all parts
 - ➌ What is the current through each one?

Homework

- Describe the difference between wet and dry cell batteries? Give two examples.
- Give two examples of “kitchen batteries”, including terminal notation
 - Which is the positive terminal and which is the negative terminal?
 - Which is the cathode and which is the anode?
 - What materials are used for the terminals?
 - What is the electrolyte?
- If I need to have the same VOLTAGE for all elements, would parallel or series circuit be better?
- If I need to have the same CURRENT for all elements, would parallel or series circuit be better?
- Homework Generator Level 1 VIR (for series circuits)
- Homework Generator Level 2 Parallel (for parallel circuits)