





	y is supercritical point drying used after HF etching of a SOI wafer?
0	To provide a smooth wafer surface for proceeding to next fabrication steps
0	To completely remove HF molecules on the surface in order to prevent any hazardous consequences during wafer handling
<u></u>	To prevent collapse of free-standing Si structures on the wafer surface by capillary forces
0	To remove any organic residues remaining on the wafer surface after etching
<b>~</b>	
After iquid Often furthe 'Supe	nation the wet etching of SiO <sub>2</sub> , during the drying or the etching of the rinsing solution, the limited amount of underneath the functional structure can pull down the latter to the substrate by capillary forces. after such collapse, the functional structure cannot be released again and hence, cannot be used in an application. The solution to avoid such a problem is to use supercritical point drying. See rcritical drying for realization of suspended structures; test microstructures for quantifying stress in types" video from 2:30 to 3:10 for detailed explanations.
2. Wh	ich of the following steps is essential in a supercritical point drying cycle?
	CO <sub>2</sub> in the closed chamber is condensed into a liquid
0	HF is replaced by cold H <sub>2</sub> SO <sub>4</sub>
0	HF is filled back again in the chamber for one last removal
0	Water is replaced by high-pressure acetone
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	nation ercritical point drying, after the etching of the sacrificial layer in HF bath, HF is replaced by deionized
In supwater surface GO of the GO of	Later, water is replaced by ethyl alcohol which is a liquid with higher vapor pressure and lowell Rights Rescription. This is followed by the placement of the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with a closed cha
n supwatersurface  GPolyg  GPO  GPO  GPO  GPO  GPO  GPO  TOOM  TO	Later, water is replaced by ethyl alcohol which is a liquid with higher vapor pressure and lowe All Rights Rescribed. Later, water is replaced by the placement of the wafer in a closed chamber that can be filled with as the interpretation. This is followed by the placement of the wafer in a closed chamber that can be filled with as the water in a closed chamber that can be filled with as the water in a closed chamber that can be filled with as the water in a closed chamber that can be filled with as the water in a closed chamber that can be filled with as the water is no distinction between liquid and hase, hence no surface tension. Finally, the pressure is dropped, CO <sub>2</sub> is forced to evaporate and then temperature is reached. See "Supercritical drying for realization of suspended structures; test
n supwater surface e Polyg see tip gas p room micro	Later, water is replaced by ethyl alcohol which is a liquid with higher vapor pressure and lowe All Rights Rescribe tension. This is followed by the placement of the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with with the substrate was included by the pressure of the wafer in a closed chamber that can be filled with was included by the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer in a closed chamber that can be filled with as the wafer water wate
n supwater surface of the surface o	Later, water is replaced by ethyl alcohol which is a liquid with higher vapor pressure and loweAll Rights Rescrete tension. This is followed by the placement of the wafer in a closed chamber that can be filled with ashtowelficteroneccan apply thigh pressure so that the gascis-do robensed and liquide @ given is displayed with each towelficteroneccan apply this breasure so that the gascis-do robensed and liquide @ given is displayed with each towelficteroneccan and the substrate. Hereafter, by increasing the street the substrate is no distinction between liquid and hase, hence no surface tension. Finally, the pressure is dropped, CO <sub>2</sub> is forced to evaporate and then temperature is reached. See "Supercritical drying for realization of suspended structures; test structures for quantifying stress in thin layers" video from 3:05 to 6:30 for detailed explanations. Such of the following is true for surface-machined microstructures?  Stress may develop due to the different thermal dilation between the film and the substrate, while



## Explanation

An issue in surface micromachining is the presence of stress in the surface-micromachined microstructures. This results from the fact that the polySi layer is deposited at a high temperature and, when cooling down to room temperature, the thermal expansion coefficient of the thick substrate and the thin deposited layer is not the same. The stress becomes the most apparent after the removal of the sacrificial layer. The stress may be compressive or tensile. The tensile stress might not be visible because it will result in a straight beam or straight membrane just like a microstructure without stress. Ring-crossbar test structures are very useful for detecting and quantifying the stress in the thin film material. Both tensile and compressive stresses can be quantified with this test structures. See "Supercritical drying for realization of suspended structures; test microstructures for quantifying stress in thin layers" video from 6:30 to 11:50 for detailed explanations.

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