CMIP6 Model Documentation

Institute: IPSL

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Topic: Ocean Biogeochemistry

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Note: * indicates a required property

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1 Key Properties

Ocean Biogeochemistry key properties

1.1.1 Top level properties

Ocean Biogeochemistry key properties

1.1.1.1 Name *

Name of ocnbgchem model code

NEMO: Nucleus for European Modelling of the Ocean (TOP/PISCES)

1.1.1.2 Keywords *

Keywords associated with ocnbachem model code

Enter COMMA SEPARATED list:

1.1.1.3 Overview *

Overview of ocnbgchem model.

PISCES (Pelagic Interaction Scheme for Carbon and Ecosystem Studies) (Aumont and Bopp, 2006) simulates the cycling of carbon, oxygen, and of the major nutrients determining phytoplankton growth (phosphate, nitrate, ammonium, iron and silicic acid). The carbon chemistry of the model is based on the Ocean Carbon Model Intercomparison Project (OCMIP2) protocol (Najjar et al, 2007) and the parametrization proposed by Wanninkhof (1992) is used to compute air-sea gas exchange of CO2 and O2. PISCES includes a simple representation of the marine ecosystem with two phytoplankton size classes, representing nanophytoplankton and diatoms, as well as two zooplankton size classes, representing microzooplankton and mesozooplankton. Phytoplankton growth is limited by the availability of nutrients, temperature, and light. There are three nonliving components of organic carbon in the model: semi-labile dissolved organic carbon (DOC), with a lifetime of several weeks to years, as well as large and small detrital particles, which are fuelled by mortality, aggregation, fecal pellet production and grazing. Biogenic silica and calcite particles are also included._x000D_ _x000D_ Nutrients and/or carbon are supplied to the ocean from three different sources: atmospheric deposition, rivers and sediment mobilization. These sources are explicitly included but do not vary in time apart from a climatological seasonal cycle for the atmospheric input. Atmospheric deposition (Fe, N, P and Si) has been estimated from the INCA model (Aumont et al, 2008). River discharge of carbon and nutrients is taken from Ludwig et al (1996). Iron input from sediment mobilization has been parameterized as in Aumont and Bopp (2006). _x000D_ _x000D_ PISCES is used here not only to compute air-sea fluxes of carbon, but also to compute the effect of a biophysical coupling: the chlorophyll concentration produced by the biological component retroacts on the ocean heat budget by modulating the absorption of light as well as the oceanic heating rate (see Lengaigne et al (2007) for a detailed description).

1.1.1.4 Model Type *

Type of ocean biogeochemistry model

Selec	t SINGLE option:
	Geochemical - No living compartments
	NPZD - No plankton types

	PFT - Several plankton types
	Other - please specify:
1.1.1.5	Elemental Stoichiometry *
Describe	elemental stoichiometry (fixed, variable, mix of the two)
Selec	ct SINGLE option:
	Fixed - Fixed stoichiometry
	Variable - Variable stoichiometry
	Mix of both - Both fixed and mixed stoichiometry
l.1.1.6	Elemental Stoichiometry Details *
Describe	which elements have fixed/variable stoichiometry
Ente	er COMMA SEPARATED list:
1.1.1.7	Prognostic Variables *
	Il prognostic tracer variables in the ocean biogeochemistry component
•	olved inorganic carbon, Total alkalinity, dissolved oxygen, Calcite, Phosphate, Small par-
iculate organic n the b the diat	organic carbone, silicate, Nanophytoplankton, Microzooplankton, Semi-labile dissolved carbon, Diatoms, Mesozooplankton, Silicon content of the diatoms, Dissolved iron, Iron oig particles, Big particulate organic carbone, Iron in the small particles, Iron content of toms, Sinking biogenic silica, Iron content of the nanophytoplankton, Chlorophyll of the ytoplankton, Chlorophyll of the diatoms, Nitrate, Ammonium
1.1.1.8	Diagnostic Variables *
List of a	ll diagnotic tracer variables in the ocean biogeochemistry component (derived from prognostic variables
Ente	er COMMA SEPARATED list:
l.1.1.9	Damping
Describe	any tracer damping used (such as artificial correction or relaxation to climatology,)
Ente	er TEXT:
119	Passive Tracers Transport
	-
Time st	tepping method for passive tracers transport in ocean biogeochemistry
1.1.2.1	Method *
$Time \ ste$	pping framework for passive tracers
	Use ocean model transport time step

Use specific time step
1.1.2.2 Timestep If Not From Ocean Time step for passive tracers (if different from ocean) Enter INTEGER value:
1.1.3 Biology Sources Sinks Time stepping framework for biology sources and sinks in ocean biogeochemistry
1.1.3.1 Method *
Time stepping framework for biology sources and sinks
Select SINGLE option:
Use ocean model transport time step
Use specific time step
1.1.3.2 Timestep If Not From Ocean Time step for biology sources and sinks (if different from ocean) Enter INTEGER value:
1.2.1 Transport Scheme
Transport scheme in ocean biogeochemistry
1.2.1.1 Type *
Type of transport scheme
Select SINGLE option:
Offline
Online
1.2.1.2 Scheme *
Transport scheme used
Select SINGLE option:
Use that of ocean model
Other - please specify:

1.2.1.3	Use	Different	Scheme

 $Decribe\ transport\ scheme\ if\ different\ than\ that\ of\ ocean\ model$

Enter TEXT:

1.3.1 Boundary Forcing

Properties of biogeochemistry boundary forcing

1.3.1.1 Atmospheric Dep	osition *
Describe how atmospheric deposit	ion is modeled
Select SINGLE option:	
From file (climatology)	
From file (interannual	variations)
From Atmospheric Che	emistry model
1.3.1.2 River Input *	
Describe how river input is model	led
Select SINGLE option:	
From file (climatology)	
From file (interannual	variations)
From Land Surface mo	del
1.3.1.3 Sediments From I	Boundary Conditions
List which sediments are speficied	from boundary condition
Enter COMMA SEPARA	TED list:
1.3.1.4 Sediments From I	_
Enter COMMA SEPARA	
1.4.1 Gas Exchange Properties of gas exchange in	n ocean biogeochemistry
1.4.1.1 CO2 Exchange Pr	resent *
Is CO2 gas exchange modeled?	
☐ Fal	se

1.4.1.2 CO ₂ Exchange Type
Describe CO2 gas exchange
OMIP protocol
Other - please specify:
1.4.1.3 O2 Exchange Present *
Is O2 gas exchange modeled?
☐ True ☐ False
1.4.1.4 O2 Exchange Type
Describe O2 gas exchange
OMIP protocol
Other - please specify:
1.4.1.5 DMS Exchange Present * Is DMS gas exchange modeled ?
☐ False
1.4.1.6 DMS Exchange Type Specify DMS gas exchange scheme type
Enter TEXT:
1.4.1.7 N2 Exchange Present *
Is N2 gas exchange modeled ?
☐ False
1.4.1.8 N2 Exchange Type
Specify N2 gas exchange scheme type
Enter TEXT:
1.4.1.9 N2O Exchange Present *
Is N2O gas exchange modeled?
True

$Specify\ N2O\ gas\ exchange\ scheme\ type$
Enter TEXT:
1.4.1.11 CFC11 Exchange Present * Is CFC11 gas exchange modeled ? Select either TRUE or FALSE: True False
1.4.1.12 CFC11 Exchange Type Specify CFC11 gas exchange scheme type Enter TEXT:
1.4.1.13 CFC12 Exchange Present * Is CFC12 gas exchange modeled ? Select either TRUE or FALSE: True False
1.4.1.14 CFC12 Exchange Type Specify CFC12 gas exchange scheme type Enter TEXT:
1.4.1.15 SF6 Exchange Present * Is SF6 gas exchange modeled? Select either TRUE or FALSE: True False
1.4.1.16 SF6 Exchange Type Specify SF6 gas exchange scheme type Enter TEXT:
1.4.1.17 13CO2 Exchange Present * Is 13CO2 gas exchange modeled? Select either TRUE or FALSE: False

1.4.1.10 N2O Exchange Type

1.4.1.18 13CO2 Exchange Type Specify 13CO2 gas exchange scheme type
Enter TEXT:
1.4.1.19 14CO2 Exchange Present * Is 14CO2 gas exchange modeled? Select either TRUE or FALSE: True False
1.4.1.20 14CO2 Exchange Type
Specify 14CO2 gas exchange scheme type Enter TEXT:
1.4.1.21 Other Gases
Specify any other gas exchange
Enter TEXT:
1.5.1 Carbon Chemistry
Properties of carbon chemistry biogeochemistry
1.5.1.1 Type *
1.5.1.1 Type * Describe how carbon chemistry is modeled

Describe how carbon chemistry is modeled
Describe how carbon chemistry is modeled Select SINGLE option:
Describe how carbon chemistry is modeled Select SINGLE option: OMIP protocol
Describe how carbon chemistry is modeled Select SINGLE option: OMIP protocol Other protocol
Describe how carbon chemistry is modeled Select SINGLE option: OMIP protocol Other protocol 1.5.1.2 Ph Scale
Describe how carbon chemistry is modeled Select SINGLE option: OMIP protocol Other protocol 1.5.1.2 Ph Scale If NOT OMIP protocol, describe pH scale.
Describe how carbon chemistry is modeled Select SINGLE option: OMIP protocol Other protocol 1.5.1.2 Ph Scale If NOT OMIP protocol, describe pH scale. Sea water

1.5.1.3 Constants If Not OMIF

 ${\it If NOT\ OMIP\ protocol,\ list\ carbon\ chemistry\ constants.}$

Enter COMMA SEPARATED list:

1.6.1 Tuning Applied

 $Tuning\ methodology\ for\ ocean\ biogeochemistry\ component$

1.6.1.1 Description *

General overview description of tuning: explain and motivate the main targets and metrics retained. and Document the relative weight given to climate performance metrics versus process oriented metrics, and on the possible conflicts with parameterization level tuning. In particular describe any struggle and with a parameter value that required pushing it to its limits to solve a particular model deficiency.

Enter TEXT:

1.6.1.2 Global Mean Metrics Used

List set of metrics of the global mean state used in tuning model/component

Enter COMMA SEPARATED list:

1.6.1.3 Regional Metrics Used

List of regional metrics of mean state used in tuning model/component

Enter COMMA SEPARATED list:

1.6.1.4 Trend Metrics Used

List observed trend metrics used in tuning model/component

Enter COMMA SEPARATED list:

2 Tracers

Ocean biogeochemistry tracers

2.1.1 Top level properties

 $Ocean\ biogeochemistry\ tracers$

•	-	•	-	76 T			
2.	Ι.	Ι.		IN	a	m	e

 $Commonly\ used\ name\ for\ the\ tracers\ in\ ocnbgchem\ model.$

Enter TEXT:

2.1.1.2 Overview

Overview of ocean biogeochemistry tracers in ocnbgchem model.

Enter TEXT:

2.1.1.3	Sulfur	Cycle	Present	*
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2.1.1.4 Nutrients Present *

 $List\ nutrient\ species\ present\ in\ ocean\ biogeochemistry\ model$

Select MULTIPLE options:		
	Nitrogen (N)	
	Phosphorous (P)	
	Silicon (S)	
	Iron (Fe)	
	Other - please specify:	

2.1.1.5 Nitrous Species If N

 ${\it If\ nitrogen\ present,\ list\ nitrous\ species.}$

Select MULTIPLE options:

Nitrates (NO3)
Amonium (NH4)
Other - please specify:

2.1.1.6	Nitrous Processes If N
$If\ nitroge$	n present, list nitrous processes.
Selec	et MULTIPLE options:
	Dentrification
	N fixation
	Other - please specify:
2.2.1]	Ecosystem
Ecosyste	em properties in ocean biogeochemistry
2.2.1.1	Upper Trophic Levels Definition *
Describe	how upper trophic levels are defined in model (e.g. based on size)
Ente	r TEXT:
	Upper Trophic Levels Treatment *
	how upper trophic levels are treated in model
Ente	r TEXT:
2.2.2	Phytoplankton
Phytople	ankton properties in ocean biogeochemistry
2.2.2.1	Type *
Type of p	hytoplankton
Selec	et SINGLE option:
	None
	Generic
	PFT including size based (specify both below) - Plankton functional type including size based
	Size based only (specify below)
	PFT only (specify below)
2.2.2.2	Pft
Phytoplan	nkton functional types (PFT) (if applicable)
Selec	et MULTIPLE options:
	Diatoms

	Nfixers			
	Calcifiers			
	Other - please specify:			
2.2.2.3 Size Classes				
Phytoplankton size classes (if applicable)				
Select MULTIPLE options:				
	Microphytoplankton			
	Nanophytoplankton			
	Picophytoplankton			
	Other - please specify:			
$2.2.3 \mathrm{Z}$	Zooplankton			
Zooplank	cton properties in ocean biogeochemistry			
2.2.3.1	Type *			
Type of zooplankton				
	None			
	Generic			
	Size based (specify below)			
	Other - please specify:			
2.2.3.2 Size Classes				
Zooplankte	on size classes (if applicable)			
Select	t MULTIPLE options:			
	Microzooplankton			
	Mesozooplankton			
	Other - please specify:			
0 0 1 T	Nachard Ormania Mattan			

2.3.1 Disolved Organic Matter

 $Disolved\ organic\ matter\ properties\ in\ ocean\ biogeochemistry$

2.3.1.1	Bacteria Present *
Is there b	bacteria representation ?
Selec	ct either TRUE or FALSE:
	True
2.3.1.2	Lability *
Describe	treatment of lability in dissolved organic matter
	None
	Labile - Less than a few days
	Semi-labile - Few days to a few years
	Refractory - Over a few years
	Other - please specify:
2.4.1	Particules
Particu	late carbon properties in ocean biogeochemistry
2.4.1.1	Method *
	articulate carbon represented in ocean biogeochemistry?
	Diagnostic
	Diagnostic (Martin profile)
	Diagnostic (Balast)
\boxtimes	Prognostic
	Other - please specify:
2.4.1.2	Types If Prognostic
If progno	$stic,\ type(s)\ of\ particulate\ matter\ taken\ into\ account$
Selec	ct MULTIPLE options:
	POC
	PIC (calcite)
	PIC (aragonite
	BSi
	Other - please specify:

2.4.1.3	Size If Prognostic			
If prognostic, describe if a particule size spectrum is used to represent distribution of particules in water volume				
Selec	t SINGLE option:			
	No size spectrum used			
	Full size spectrum			
	Discrete size classes (specify which below)			
2.4.1.4	Size If Discrete			
If prognos	tic and discrete size, describe which size classes are used			
Enter	TEXT:			
2415	Cipling Speed If Dreamastic			
	Sinking Speed If Prognostic tic, method for calculation of sinking speed of particules			
	t SINGLE option:			
	Constant			
	Function of particule size			
	Function of particule size Function of particule type (balast)			
	Other - please specify:			
	Other - please specify.			
2.5.1 I	Dic Alkalinity			
DIC and alkalinity properties in ocean biogeochemistry				
2.5.1.1	Carbon Isotopes *			
Which can	rbon isotopes are modelled (C13, C14)?			
Selec	t MULTIPLE options:			
	C13			
	C14)			
2.5.1.2	Abiotic Carbon *			
$Is\ abiotic$	carbon modelled ?			
Select either TRUE or FALSE:				

True

☐ False

2.5.1.3 Alkalinity *				
$How \ is \ alkalinity \ modelled \ ?$				
Select SINGLE option:				
	Prognostic			
	Diagnostic)			