

## 工程热力学单元习题及答案

习题 1-4:

解: 两容器和大气存在以下关系:

$$\begin{cases} P_{\text{右}} - P_0 = P_A \\ P_{\text{右}} - P_{\text{左}} = P_B \\ P_{\text{左}} - P_0 = P_C \end{cases}$$

$$(1) \quad P_c = P_A - P_B = 2.5 \text{ bar} = 0.25 \text{ MPa}$$

$$(2) \quad P_{\text{左}} = P_c + P_0 = 2.5 \text{ bar}$$

习题 1-10:

解: 压力计使用前后右端有  $\frac{P_0 V_0}{T_0} = \frac{P_1 V_1}{T_1}$ 。

$$\text{则 } P_1 = \frac{P_0 V_0}{T_0} \frac{T_1}{V_1} = \frac{1 \times (400 + 150)}{288} \frac{303}{400} = 1.447 \text{ bar}$$

$$\text{又 } P_1 + \rho_{\text{水银}} g h_1 = P_{\text{测}} + \rho_{\text{水}} g h_2$$

$$\begin{aligned} P_{\text{测}} &= P_1 + \rho_{\text{水银}} g h_1 - \rho_{\text{水}} g h_2 \\ &= 1.447 \times 10^5 + 13600 \times 9.8 \times 300 \times 10^{-3} - 1000 \times 9.8 \times 1000 \times 10^{-3} \\ &= 174884 \text{ Pa} = 1.75 \text{ bar} \end{aligned}$$

习题 2-4:

答: 根据题意, 初始时

$$K a_1 = P_1 A$$

$$V_1 = a_1 A = 3 \times 10^{-5} \text{ m}^3$$

代入数据得

$$P_1 = 4.5 \times 10^5 \text{ Pa}$$

$$m_1 = \frac{P_1 V_1}{R_{CO_2} T_1} = 2.335 \times 10^{-3} kg$$

末端时

$$K a_2 = P_2 A$$

$$V_2 = a_2 A = 2.55 \times 10^{-5} m^3$$

代入数据得

$$P_2 = 3.825 \times 10^5 Pa$$

已知  $m_1 = m_2$

$$T_2 = \frac{P_2 V_2}{R_{CO_2} m_2} = 216.47 K$$

此系统为一闭口系统，有能量方程

$$Q = \Delta U + W$$

过程中气体放出的热量

$$Q = m c_v (T_2 - T_1) + \int_{a_1}^{a_2} \frac{k a}{A} d(A a)$$

$$m c_v (T_2 - T_1) = 2.335 \times 10^{-3} \times \frac{29.1}{44} \times (-83.68) J = -129.2 J$$

因为  $1 kg \cdot m = 9.81 \times 10^{-3} kJ$

$$\begin{aligned} \int_{a_1}^{a_2} \frac{k a}{A} d(A a) &= \int_{a_1}^{a_2} k a da = \frac{k(a_2^2 - a_1^2)}{2} = -187.31 kg \cdot cm \\ &= -18.37 J \end{aligned}$$

$$Q = \Delta U + W = -147.57 J$$

过程中气体放出的热量 147.57 J。

习题 2-5:

答:  $q = 5.69 kJ/kg$   $T_2 = -20^\circ C$   $T_1 = 15^\circ C$

$$q = c_n \Delta T \quad c_n = q / \Delta T = 2.857 kJ / (kg K)$$

已知气体为氧气

$$c_n = \frac{n-k}{n-1} c_v$$

可得  $n=1.53$

$$\left(\frac{P_2}{P_1}\right)^{\frac{n-1}{n}} = \frac{T_2}{T_1}$$

$$P_2 = 103 \text{ bar}$$

习题 2-9:

答:  $Q = W$

$$200 + 20 + 0 - 210 = 1400 + 0 + 2250 + W_4$$

$$W_4 = -3640 \text{ KJ}$$

习题 3-9:

解:

根据熵的计算

$$S_2 - S_1 = C_V \ln \frac{P_2}{P_1} + C_P \ln \frac{V_2}{V_1}$$

$$\frac{S_2 - S_1}{C_V} = \ln \frac{P_2}{P_1} + k \ln \frac{\rho_1}{\rho_2}$$

$$\frac{P_2}{\rho_2^k} = \frac{P_1}{\rho_1^k} e^{\frac{S_2 - S_1}{C_V}}$$

习题 3-16:

解: 由  $\frac{T_1}{T_2} = \left(\frac{V_2}{V_1}\right)^{n-1}$  得

$$\frac{273 + 300}{60 + 300} = 3^{n-1}$$

$$n = 1.42$$

$$q_n = m C_n (T_2 - T_1)$$

$$W_n = m \frac{R_g}{1-n} (T_1 - T_2)$$

由

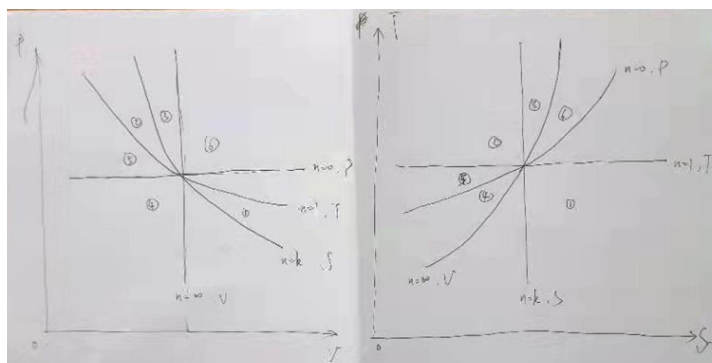
$$\begin{cases} 20 = 2 \times \frac{1.42 - k}{0.42} \times \frac{1}{k-1} R_g \times (60 - 300) \\ -100 = 2 \times \frac{R_g}{1-1.42} \times (300 - 60) \end{cases}$$

得

$$R_g = 0.0875 \text{ kJ}/(\text{kg} \cdot \text{K}), \quad k = 1.35$$

$$C_n = \frac{1.42 - 1.35}{0.42} \times \frac{1}{1.35 - 1} \times 0.0875 = 0.04167 \text{ kJ}/(\text{kg} \cdot \text{K})$$

习题 3-24:



习题 3-25:

解: ① 已知  $T_{A1} = T_{B1} = 300\text{K}$ ,  $P_{A1} = P_{B1} = 1\text{bar}$ ,  $V_{A1} = V_{B1} = 0.5\text{m}^3$ ,  $P_{A2} = 202\text{bar}$ , 活塞是自由的, 故  $P_{B2} = P_{A2} = 202\text{bar}$ 。

$$R_{g_{N_2}} = \frac{R}{M_{N_2}} = 296.95 \text{ J}/(\text{kg} \cdot \text{K})$$

$$R_{g_{O_2}} = \frac{R}{M_{O_2}} = 259.83 \text{ J}/(\text{kg} \cdot \text{K})$$

$$m_A = \frac{P_{A1} V_{A1}}{R_{g_{N_2}} T_{A1}} = 0.56 \text{ kg}$$

$$m_B = \frac{P_{B1} V_{B1}}{R_{g_{O_2}} T_{B1}} = 0.64 \text{ kg}$$

由于 B 内进行可逆绝热过程,

$$T_{B2} = T_{B1} \left( \frac{T_{B2}}{T_{B1}} \right)^{\frac{k-1}{k}} = 1367.08K$$

$$V_{B2} = \frac{m_B R g_{O_2} T_{B2}}{P_{B2}} = 0.01m^3$$

②

$$V_{A2} = 1 - V_{B2} = 0.99m^3$$

$$T_{A2} = \frac{P_{A2} V_{A2}}{R g_{N_2} m_A} = 120258.34K$$

③

$$Cv_{N_2} = \frac{R g_{N_2}}{k - 1} = 742.375J/(kg \cdot K)$$

$$Cv_{O_2} = \frac{R g_{O_2}}{k - 1} = 649.575J/(kg \cdot K)$$

取 A+B 为热力系, 则

$$\begin{aligned} Q &= \Delta U_A + \Delta U_B = m_A Cv_{N_2} (T_{A2} - T_{A1}) + m_B Cv_{O_2} (T_{B2} - T_{B1}) \\ &= 50313.9kJ \end{aligned}$$

取 B 为热力系, 则

$$W_B = -\Delta U_B = -m_B Cv_{O_2} (T_{B2} - T_{B1}) = -443.62kJ$$

故  $W_A = -W_B = 443.62kJ$

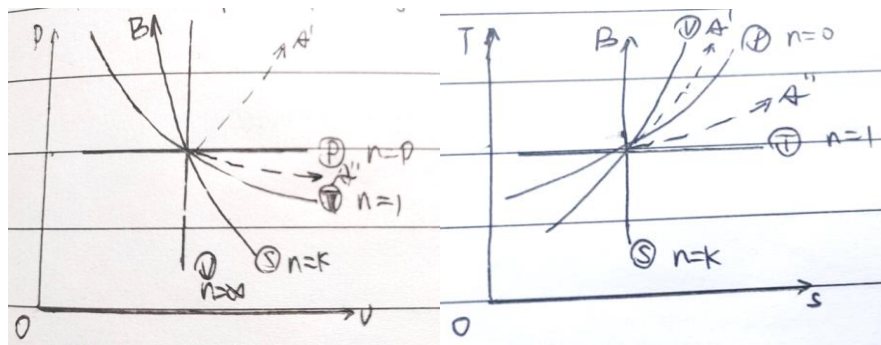
④

$$\Delta S_{O_2} = 0$$

$$Cp_{N_2} = Cv_{N_2} + R g_{N_2} = 1039.325J/(kg \cdot K)$$

$$\Delta S_{N_2} = m_A \left( Cp_{N_2} \ln \frac{T_{A2}}{T_{A1}} - R g_{N_2} \ln \frac{P_{A2}}{P_{A1}} \right) = 2.61kJ/K$$

⑤



习题 4-4:

解:  $\eta_1 = \left(1 - \frac{T_2}{T_1}\right)/2 = \left(1 - \frac{200+273}{1000+273}\right)/2 = 0.314$

$$\eta_2 = \left(1 - \frac{T_2}{T'_1}\right)/2 = \left(1 - \frac{200 + 273}{1100 + 273}\right)/2 = 0.328$$

$$Q_1 = \frac{w}{\eta_1} \quad Q'_1 = \frac{w'}{\eta_2} \quad \text{取} \quad w = 1\text{J} \quad w' = 1\text{J}$$

$$\frac{Q_1 - Q'_1}{Q_1} = \frac{1/\eta_1 - 1/\eta_2}{1/\eta_1} = 4.3\%$$

习题 4-6:

①两机输出功相等。②两机热效率相等。

解: (1)  $w_1 = w_2 \quad \eta_1 Q_1 = \eta_2 Q_A$

$$\eta_1/\eta_2 = Q_A/Q_1 \quad \eta_1 = 1 - Q_A/Q_1 \quad Q_A/Q_1 = 1 - \eta_1$$

$$\eta_1/\eta_2 = 1 - \eta_1$$

$$\frac{1 - T/T_1}{1 - T_2/T} = 1 - (1 - T/T_1) = T/T_1$$

$$T_1 = 627 + 273 = 900\text{K} \quad T_2 = 27 + 273 = 300\text{K}$$

$$T = \frac{1}{2}(T_1 + T_2) = 600\text{K}$$

(2)  $\eta_1 = \eta_2 \quad 1 - T/T_1 = 1 - T_2/T$

$$T = \sqrt{T_1 T_2} = 519.6\text{K}$$

习题 4-9:

解: (1)  $Q_{\text{排}} = 0.65 \times [20 - (-10)] = 19.5 \text{ kW}$

$$\varepsilon_{\max} = \frac{T_1}{T_1 - T_2} = \frac{273 + 20}{20 - (-10)} = 9.77$$

$$w_{\min} = \frac{Q_{\text{排}}}{\varepsilon_{\max}} = \frac{19.5}{9.77} = 2 \text{ kW}$$

$$(2) \frac{Q_{\text{吸}}}{w} = \frac{T_2}{T_1 - T_2} \quad \frac{Q_{\text{吸}}}{2} = \frac{273 + 20}{T_1 - (273 + 20)}$$

$$Q_{\text{排}} = 0.65 \times (T_1 - T_2) \quad \frac{w}{T_1 - T_2} = \frac{Q_{\text{吸}}}{T_2} = \frac{Q_{\text{排}}}{T_1}$$

$$\frac{293 \times 2}{(T_1 - 293) \times 293} = \frac{0.65 \times (T_1 - 293)}{T_1} \quad T_1 = 324 \text{ K}$$

习题 4-14:

解: (1)  $T_m$  可由计算熵增办法证明。将热源  $T_A$ 、冷源  $T_B$  和热机考虑为一个孤立系, 因整个过程是可逆, 因此  $\Delta S_{\text{孤}} = 0$

$$\begin{aligned} \Delta S_{\text{孤}} &= \Delta S_A + \Delta S_{\text{热机}} + \Delta S_B \\ &= m \int \frac{\delta Q_A}{T} + m \int \frac{\delta Q_B}{T} + 0 \\ &= m C_p \int_{T_A}^{T_m} \frac{dT}{T} + m C_p \int_{T_B}^{T_m} \frac{dT}{T} \\ &= m C_p \ln \frac{T_m}{T_A} + m C_p \ln \frac{T_m}{T_B} = 0 \end{aligned}$$

$$\ln \frac{T_m}{T_A} = \ln \frac{T_B}{T_m} \quad \frac{T_m}{T_A} = \frac{T_B}{T_m}$$

所以  $T_m = \sqrt{T_A \cdot T_B}$

(2) 可逆热机作出的总功  $W_o = Q_A - Q_B$

$$\begin{aligned} W_o &= m C_p (T_A - T_B) - m C_p (T_m - T_B) \\ &= m C_p (T_A - 2T_m + T_B) \\ &= m C_p (T_A + T_B - 2\sqrt{T_A \cdot T_B}) \end{aligned}$$

(3) 抽掉 A, B 之间的热机后, 则  $Q_A = Q_B$  即

$$mC_p(T_A - T_m) = mC_p(T_m - T_B)$$

所以

$$T_m = \frac{1}{2}(T_A + T_B)$$

热源熵增

$$\Delta S_A = m \int \frac{\delta Q_A}{T} = mC_p \int_{T_A}^{T_m} \frac{dT}{T} = mC_p \ln \frac{T_m}{T_A} = mC_p \frac{T_A - T_B}{2T_A}$$

冷源熵增

$$\Delta S_B = m \int \frac{\delta Q_B}{T} = mC_p \int_{T_B}^{T_m} \frac{dT}{T} = mC_p \ln \frac{T_m}{T_B} = mC_p \frac{T_A - T_B}{2T_B}$$

整个孤立系熵增：

$$\begin{aligned} \Delta S_{\text{孤}} &= \Delta S_A + \Delta S_B \\ &= mC_p \ln \left[ \frac{T_A + T_B}{2T_A} \cdot \frac{T_A + T_B}{2T_B} \right] \end{aligned}$$

$$\Delta S = mC_p \ln \frac{(T_A + T_B)^2}{4T_A T_B}$$

习题 4-16：

**解：**将两个物体和制冷机组成的整体视为一孤立系，则有：

$$\Delta S = \Delta S_{\text{热}} + \Delta S_{\text{冷}} + \Delta S_{\text{机}} \geq 0$$

$$\text{其中， } \Delta S_{\text{机}} = 0, \quad \Delta S_{\text{热}} = \int_{T_i}^{T_1} cm \frac{dT}{T}, \quad \Delta S_{\text{冷}} = \int_{T_i}^{T_f} cm \frac{dT}{T}$$

$$\text{则有：} \left( \int_{T_i}^{T_1} cm \frac{dT}{T} + \int_{T_i}^{T_f} cm \frac{dT}{T} \right) \geq 0$$

$$\text{可求得：} T_1 \geq \frac{T_i^2}{T_f}$$

可逆热机做功为：

$$W = Q_1 - Q_2 = cm(T_1 - T_i) + cm(T_f - T_i) \geq cm \left( \frac{T_i^2}{T_f} + T_f - 2T_i \right)$$



$$W_{min} = cm \left( \frac{T_i^2}{T_f} + T_f - 2T_i \right)$$

则最小做功

习题 5-12:

**解：** 从水蒸气表中查出，4bar 时的状态参数为：

$$T_1 = 143.642 \text{ } ^\circ\text{C}$$

$$v_1' = 0.0010835 \text{ m}^3/\text{kg}$$

$$v_1'' = 0.46246 \text{ m}^3/\text{kg}$$

$$h_1' = 604.87 \text{ kJ/kg}$$

$$h_1'' = 2738.49 \text{ kJ/kg}$$

$$s_1' = 1.7769 \text{ kJ/K/kg}$$

$$s_1'' = 6.8961 \text{ kJ/K/kg}$$

从水蒸气表中查出，0.6 bar 时的状态参数为：

$$T_2 = 85.9496 \text{ } ^\circ\text{C}$$

$$v_2' = 0.0010331 \text{ m}^3/\text{kg}$$

$$v_2'' = 2.7324 \text{ m}^3/\text{kg}$$

$$h_2' = 359.91 \text{ kJ/kg}$$

$$h_2'' = 2652.97 \text{ kJ/kg}$$

$$s_2' = 1.1454 \text{ kJ/K/kg}$$

$$s_2'' = 7.531 \text{ kJ/K/kg}$$

设膨胀过程为可逆过程，则熵保持不变，膨胀后干度为  $x_2$ ，则有：

$$x \cdot s_1'' + (1-x) \cdot s_1' = x_2 \cdot s_2'' + (1-x_2) \cdot s_2'$$

代入后解得：  $x_2 = 0.74$

膨胀后的容积：

$$m = \frac{V_1}{x \cdot v_1'' + (1-x_1) \cdot v_1'} = 3.2417 \text{ kg}$$

$$V_2 = m(x_2 \cdot v_2'' + (1-x_2) \cdot v_2') = 6.5556 \text{ m}^3$$

$$h_1 = x \cdot h_1'' + (1-x) \cdot h_1' = 2311.766 \text{ kJ/kg}$$

$$h_2 = x_2 \cdot h_2'' + (1-x_2) \cdot h_2' = 2056.7744 \text{ kJ/kg}$$

在此过程中，

$$\delta Q = \delta W + dU = \delta W + dh - \delta PV = 0$$

$$\text{因此， } W = -(H_2 - P_2 V_2) + (H_1 - P_1 V_1) = -85813.3924$$

习题 5-13：

**解：**（1）用完全气体状态方程

$$V = \frac{RT}{P} = \frac{\frac{8314}{18} \times (273 + 600)}{80 \times 10^5} = 0.05 \text{ m}^3 / \text{kg}$$

（2）用对比态方程

$$P_r = \frac{P}{P_{cr}} = \frac{80}{220.9} = 0.362$$

$$T_r = \frac{T}{T_{cr}} = \frac{873}{647.3} = 1.349$$

查图 5-2 通用压缩因子图得

$$Z \approx 0.96$$

$$\text{则 } v = \frac{ZRT}{P} = 0.96 \times 0.05 = 0.048 \text{ m}^3 / \text{kg}$$

(3) 查水蒸汽表,

$$v = 0.048394 \text{ m}^3 / \text{kg}$$

习题 5-14:

解: 4bar、200°C 为过热蒸汽, 查表得  $h_1 = 2860.4 \text{ kJ/kg}$ ,

$$v_1 = 0.53426 \text{ m}^3 / \text{kg}, \quad \delta = 7.1708 \text{ kJ} / \text{kg} \cdot \text{K}。 \text{ 而 } x = 0.2 \quad P_2 = P_1 = 4 \text{ bar}$$

查饱和蒸汽表,  $t_2 = 143.63 \text{ }^\circ\text{C}$

$$\text{而 } h' = 604.7, \quad h'' = 2737.6$$

$$\text{则 } h_2 = h''x + (1-x)h' = 0.2 \times 2737.6 + 0.8 \times 604.7 = 1031.28$$

$$|q_p| = h_1 - h_2 = 2860.4 - 1031.28 = 1829.12 \text{ kJ/kg}$$

$$\dot{M} = \frac{10^6}{1829.12} = 546.71 \text{ kg}$$

习题 6-5: (题中有两  $\text{H}_2$ , 将 5% 的  $\text{H}_2$  改为  $\text{N}_2$ )

解析: 设煤气摩尔数  $n_1$  为 1mol;

$$m_{\text{CO}} = 0.07 \times 28 = 1.96 \quad m_{\text{H}_2} = 0.48 \times 2 = 0.96 \quad m_{\text{CH}_4} = 0.40 \times 16 = 6.4 \\ m_{\text{N}_2} = 0.05 \times 28 = 1.4;$$

$$\text{煤气质量 } m_1 = 10.72 \text{ g} \quad \text{空气质量 } m_2 = 10.72 \times 8 = 85.76 \text{ g}$$

$$\text{混合气质量 } m = 96.48 \quad ;$$

$$\text{空气摩尔数 } n_2 = \frac{m_2}{M} = 2.9786 \text{ mol}$$

$$\text{混合气质量数 } M = \frac{m}{n} = \frac{10.72 \times 9}{1 + 2.9786} = 24.2479 \text{ g/mol};$$

各组分质量成分:

$$\omega_{\text{CO}} = \frac{1.96}{96.48} = 2.03\% \quad \omega_{\text{H}_2} = \frac{0.96}{96.48} = 1\% \quad \omega_{\text{CH}_4} = \frac{6.4}{96.48} = 6.63\%$$

$$\omega_{N_2} = \frac{1.4+85.76*78\%}{96.48} = 70.78\% \quad \omega_{O_2} = \frac{85.76*21\%}{96.48} = 19.56\%$$

当  $P = 1.2 \text{ bar}$ ,  $T = 373 \text{ K}$ ,  $V = 10 \text{ m}^3$  时;

$$R_g = \sum \omega R_{gi} = 2.03\% * 296.928 + 1\% * 4157 + 6.63\% * 519.635 + 70.78\% * 296.928 + 19.56\% * 259.8125 = 343.03;$$

$$\text{又 } PV = mR_g T$$

$$m = 9.378 \text{ kg}$$

习题 6-7:

$$\text{解: (1) } R_g = \omega_{O_2} R_{g_{O_2}} + \omega_{H_2O} R_{g_{H_2O}}$$

$$= \frac{1}{1+0.34} * R_{g_{O_2}} + \frac{0.34}{1+0.34} * R_{g_{H_2O}}$$

$$= 193.89 + 117.1957$$

$$= 311.0856 \text{ J/(Kg K)}$$

$$C_P = \omega_{O_2} C_{P_{O_2}} + \omega_{H_2O} C_{P_{H_2O}}$$

$$= \frac{1}{1+0.34} * 1.006 + \frac{0.34}{1+0.34} * 1.873$$

$$= 1226 \text{ J/(Kg K)}$$

$$C_P = R_g * \frac{\lambda}{\lambda - 1} \quad \lambda = 1.34$$

$$(2) \quad \frac{T_2}{T_1} = \left( \frac{P_2}{P_1} \right)^{\frac{k-1}{k}} \quad \text{代入数据} \quad T_2 = 468.57 \text{ K}$$

$$(3) \quad \Delta S_{\text{水}} = m_{\text{水}} \left( C_P \ln \frac{T_2}{T_1} - R_g \ln \frac{p_2}{p_1} \right) = 3.258 \text{ J/(Kg K)}$$

$$\Delta S_{\text{气}} = m_{\text{气}} \left( C_P \ln \frac{T_2}{T_1} - R_g \ln \frac{p_2}{p_1} \right) = -22.002 \text{ J/(Kg K)}$$

习题 6-18:

解析:

$t = 30$ ,  $t_{\infty} = 20$ , 由  $t_{30}$  可知  $P_{sg} = 0.0424 \text{ bar}$ ,

$$d = \frac{1.005 \times 10^3 (20 - 30) + d_3 L_3}{1.86(t_1 - t_{\infty}) + L_3} = \frac{-10050 + 36561.8}{18.6 + 2454.3} = 10.7 \text{ g}_a / \text{Kg}_a$$

$$\varphi = \frac{Bd}{(622 + d_1)P_{sg}} = \frac{1 \times 10.7}{(622 + 17) \times 0.0424} = 39.9\%$$

$$P_v = \varphi P_{sg} = 39.9\% \times 0.0424 = 0.01692 \text{ bar}$$

蒸汽饱和, 可知  $t_p = 10 + 4.176 = 14.176^\circ\text{C}$

习题 6-24:

解析: 由  $0.8$  及  $0^\circ\text{C}$  查图

$$h_1 = 7 \text{ kJ/kg}$$

其  $d$  加热至  $18^\circ\text{C}$  查图

$$h_2 = 27 \text{ kJ/kg}, \quad \varphi = 23\%$$

$$\text{得 } \Delta h = 27 - 7 = 20 \text{ kJ/kg}$$

习题 7-2:

解: 水的比热容是  $C_w = 4.187 \text{ kJ} / (\text{kg} \cdot ^\circ\text{C})$ , 温升  $14^\circ\text{C}$  时, 水套吸热。

$$Q = q_{m,w} \cdot C_w \cdot \Delta t = \frac{465 \text{ kg/h}}{3600 \text{ s/h}} \times 4.187 \text{ kJ} (\text{kg} \cdot ^\circ\text{C}) \times 14^\circ\text{C} = 7.5715 \text{ kJ/s}$$

$$\text{进气流量: } q_{m,a} = \frac{p_1 V_1}{R_g T_1} = \frac{0.1 \times 10^6 \times 250}{3600 \times 287 \times 293} = 0.08258 \text{ kg/s}$$

$$\text{单位质量空气放热: } q = \frac{Q}{q_{m,a}} = 91.6868 \text{ kJ/kg}, \text{ 因此加负号后换热量}$$
$$q = -91.6868 \text{ kJ/kg}$$

$$\text{根据 } q = \frac{n-K}{n-1} C_w (T_2 - T_1) \text{ 得到: } \frac{n-k}{n-1} = \frac{q}{C_w (T_2 - T_1)}$$

$$C_w = \frac{1}{k-1} R_g = \frac{1}{1.4-1} \times 287 \text{ J} / (\text{kg} \cdot \text{K}) = 717.5 \text{ J} / (\text{kg} \cdot \text{K})$$

$$\text{所以 } \frac{n-k}{n-1} = \frac{-91.6868 \times 10^3 \text{ J/kg}}{717.5 \text{ J} (\text{kg} \cdot \text{K}) \times (150 - 20)} = -0.983$$

$$n=1.2$$

$$\text{所以 } \frac{P_2}{P_1} = \left( \frac{T_2}{T_1} \right)^{\frac{n}{n-1}}, P_2 = \left( \frac{150+273}{20+273} \right)^{\frac{1.2}{1.2-1}}, P_1 = 0.905 \text{ MPa}$$

$$(2) \text{ 压气机的功率 } P = W_c \cdot q_{m,a}$$

$$\text{根据热力学第一定律: } q = \Delta h + W_c$$

$$\begin{aligned} W_c &= q - \Delta h = -91.6868 \text{ kJ/g} - \frac{k}{k-1} R(T_2 - T_1) \\ &= -91.6868 \text{ kJ/kg} - \frac{1.4}{1.4-1} \times 287 \times (150 - 20) \text{ J/kg} \\ &= -222,271.8 \text{ J/kg} \end{aligned}$$

习题 7-7: (删除原题中压比的已知条件)

**解:** 压气机增压比为  $\pi_c = 7.047$

压气机为绝热过程, 消耗的功:

$$W = \Delta h = C_p \times (T_2 - T_1) = 1.004 \text{ kJ/(kg} \cdot \text{K)} \times (534 \text{ K} - 288 \text{ K}) = 246.8 \text{ kJ/kg}$$

$$\text{功率: } P = q_m \cdot \omega = 13.28 \text{ kg/s} \times 246.98 \text{ kJ/kg} = 3279.95 \text{ kW}$$

$$(2) \text{ 若看作等熵过程, 压缩后温度为 } T_{2s} = \pi_c^{\frac{k-1}{k}} T_1 = 7.047^{\frac{1.4-1}{1.4}} \times 288 \text{ K} = 503.13 \text{ K}$$

$$\text{绝热效率: } \eta_c = \frac{T_{2s} - T_1}{T_2 - T_1} = \frac{503.13 - 288}{534 - 288} = 0.875$$

(3) 若将它看作一个多变过程, 则多变指数为:

$$n = \frac{\ln P_2 - \ln P_1}{\ln V_2 - \ln V_1} = 1.46, \text{ 其中 } V_1 = \frac{RgT_1}{P_1} = 0.9246 \text{ m}^3, V_2 = \frac{RgT_2}{P_2} = 0.2433 \text{ m}^3$$

习题 7-9:

$$\text{解: } q = \Delta h + w_s + \Delta \frac{1}{2} C_f^2, q = 0$$

$$\Delta h = C_p (T_2 - T_1) = 1.004 \text{ kJ/kg} \times (150^\circ \text{C} - 20^\circ \text{C}) = 130.52 \text{ kJ/kg}$$

$$\frac{1}{2} \Delta C_f^2 = \frac{1}{2} \times (50^2 - 10^2) = 1.2 \text{ kJ/kg}$$

$$W_s = -\Delta h - \frac{1}{2} \Delta C_f^2 = 131.72 \text{ kJ/kg} \text{ 每千克消耗 } 131.72 \text{ kJ 的功}$$

$$\textcircled{2} \quad T_{2s} = T_1 \cdot \left( \frac{p_2}{p_1} \right)^{\frac{k-1}{k}} = 293\text{K} \times \left( \frac{3}{1} \right)^{\frac{1.4-1}{1.4}} = 401.04\text{K}$$

$$\eta_c = \frac{T_{2s} - T_1}{T_2 - T_1} = \frac{401.04 - 293}{150 - 20} = 0.831$$

$$\textcircled{3} \quad \Delta S = C_p \ln \frac{T_2}{T_1} - Rg \cdot \frac{p_2}{p_1} = 1.004\text{kJ}(\text{kg} \cdot \text{k}) \times \ln \frac{423}{293} - 0.287\text{kJ}(\text{kg} \cdot \text{k}) \times \ln \frac{3}{1} = 0.05337\text{kJ} / (\text{kg} \cdot \text{k})$$

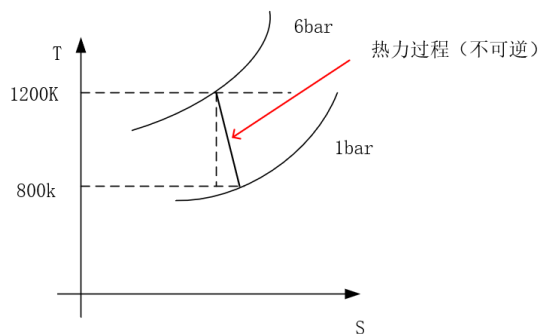
习题 7-12:

解:  $W_T = -\Delta h = C_p (T_1 - T_2)$

$$T_2 = T_1 - \frac{W_T}{C_p} = 1200 - \frac{404\text{kJ} / \text{kg}}{1.01\text{kJ} / (\text{kg} \cdot \text{k})} = 800\text{K}$$

$$S_g = \Delta S = C_p \ln \frac{T_2}{T_1} - Rg \ln \frac{p_2}{p_1} = 1.01\text{kJ}(\text{kg} \cdot \text{k}) \times \ln \frac{800}{1200} - 0.286\text{kJ}(\text{kg} \cdot \text{k}) \times \ln \frac{1}{6} = 102.92\text{J}(\text{kg} \cdot \text{k})$$

表示在 T-S 图上为:



习题 8-3:

答:

$$T^* = T + \frac{Cf^2}{2C_p}$$

$$C_p = 1.005$$

$$\Delta T_1 = T_1^* - T_1 = 1120.52\text{K}$$

$$\Delta T_2 = T_2^* - T_2 = 31.12\text{K}$$

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习题 8-6:

答:

$$C_p = \frac{k}{k-1} Rg$$

$$Ma = \frac{cf}{c}$$

$$T^* = T + \frac{C_f^2}{2C_p} = T + \frac{k-1}{2} Ma^2 T = T_{cr} + \frac{k-1}{2} T_{cr}$$

得证,

$$\frac{T}{T_{cr}} = \frac{(k+1)/2}{1 + \frac{k-1}{2} M^2}$$

习题 8-11:

答:  $T^* = T + \frac{C_f^2}{2C_p} = 850 + \frac{318^2}{2 \times 1004} = 900.34K$

$$P^* = P_1 \left( \frac{T^*}{T} \right)^{\frac{k}{k-1}} = 1.82bar$$

$$P_{cr} = \left( \frac{2}{k+1} \right)^{\frac{k}{k-1}} P^* = 0.962bar > 0.9bar$$

$$Cf_2 = 549m/s$$

习题 8-14:

答:

$$p_1 = 1 \times 10^5 Pa, \quad p_2 = 1.4 \times 10^5 Pa$$

$$T_1 = 273 + 30 = 303K$$

$$\begin{aligned} T_2 &= T_1 \left( \frac{p_2}{p_1} \right)^{k-1/k} = 303K \times \left( \frac{1.4}{1} \right)^{1.4-1/1.4} \\ &= 333.6K \end{aligned}$$

出口速度为零时, 所需的进口速度最小为:



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$$c_p = 1.004 \text{ kJ} / (\text{kg} \cdot \text{K})$$

$$\begin{aligned} c_{f1} &= \sqrt{2c_p(T_2 - T_1)} \\ &= \sqrt{2 \times 1004(333.6 - 303)} \\ &= 247.9 \text{ m/s} \end{aligned}$$

习题 8-15:

答:

$$p_1 = 3 \times 10^5 \text{ Pa}, \quad p_2 = 2 \times 10^5 \text{ Pa}$$

$$T_1 = 273 + 800 = 1073 \text{ K}$$

$$\begin{aligned} T_2 &= T_1 \left( \frac{p_2}{p_1} \right)^{k-1/k} = 1073 \text{ K} \times \left( \frac{2}{3} \right)^{1.4-1/1.4} \\ &= 955.6 \text{ K} \end{aligned}$$

忽略进口速度时，出口流速为:

$$\begin{aligned} c_p &= 1.004 \text{ kJ} / (\text{kg} \cdot \text{K}) \\ c_{f2} &= \sqrt{2c_p(T_1 - T_2)} \\ &= \sqrt{2 \times 1004(1073 - 955.6)} \\ &= 485.5 \text{ m/s} \\ v_2 &= \frac{R_g T_2}{p_2} = \frac{287 \times 955.6}{2 \times 10^5} \text{ m}^3 / \text{kg} = 1.37 \text{ m}^3 / \text{kg} \end{aligned}$$

出口截面积:

$$\begin{aligned} \dot{m} &= \frac{c_{f,2} A_2}{v_2} \\ A_2 &= \frac{\dot{m} v_2}{c_{f,2}} = \frac{1 \times 1.37}{485.5} = 0.0028 \text{ m}^2 \end{aligned}$$

习题 8-16:

答:

$$p_1 = 5 \times 10^5 \text{ Pa}, \quad p_2 = 3 \times 10^5 \text{ Pa}$$

$$T_1 = 1000 \text{ K}$$

若过程为可逆绝热过程:

$$T_{2s} = T_1 \left( \frac{p_2}{p_1} \right)^{k-1/k} = 1000K \times \left( \frac{3}{5} \right)^{1.4-1/1.4} = 864.2K$$

(1) 根据能量守恒方程：

$$T_2 = T_1 - \frac{c_{f2}^2}{2c_p} = 1000K - \frac{470^2}{2 \times 1004} = 890K$$

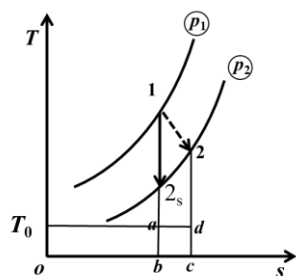
(2) 过程的熵增

$$\Delta s = c_p \ln \frac{T_2}{T_1} - R_g \ln \frac{p_2}{p_1} \quad \text{或}$$

$$\begin{aligned} \Delta s &= c_p \ln \frac{T_2}{T_{2s}} \\ &= 1.004 \times \ln \frac{890}{864.2} \\ &= 0.0295 kJ / (kg \cdot K) \end{aligned}$$

(3) 做功力损失：T-S 上面积  $A_{abcd}$ ,  $I = T_0 \Delta s$

摩擦产生的热量：T-S 图上面积  $A_{22sbc}$ ,  $Q = c_p (T_2 - T_{2s})$



#### 习题 9-4

解：

$$\eta_t = 1 - \frac{1}{\varepsilon^{k-1}} = 0.5$$

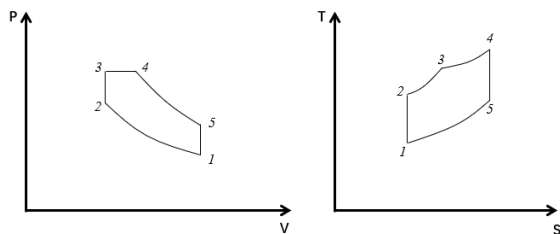
$$\therefore \varepsilon^{0.4} = 2$$

$$\frac{T_2}{T_1} = \varepsilon^{k-1} = 2$$

$$\therefore T_2 = 2T_1 = 2 \times 288 = 576K$$

# 习题 9-5

答：



循环中工质吸热  $q_1$ ，放热  $q_2$

$$T_1=363.15K, T_2=673.15K, T_3=863.15K, T_5=573.15K$$

$$p_2=p_1\left(\frac{T_2}{T_1}\right)^{\frac{k}{k-1}}=0.867Mpa$$

$$p_3=p_2\frac{T_3}{T_2}=1.11Mpa$$

$$p_4=p_3$$

$$p_5=p_1\left(\frac{T_5}{T_1}\right)=0.1578Mpa$$

$$T_4=T_5\left(\frac{p_4}{p_5}\right)^{\frac{k-1}{k}}=T_5\left(\frac{p_3}{p_5}\right)^{\frac{k-1}{k}}=1000.7K$$

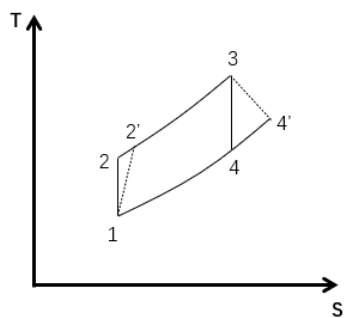
$$q_{in}=q_{2-3}+q_{3-4}=c_v(T_3-T_2)+c_p(T_4-T_3)=136.42+138.1=274.52K$$

$$q_{out}=c_v(T_5-T_1)=150.78K$$

$$\eta=\frac{q_{in}-q_{out}}{q_{in}}=45.07\%$$

# 习题 9-12

答：



$$\eta_c = 0.85, \quad \eta_t = 0.9, \quad T_1 = 288K$$

$$\pi = \frac{P_2}{P_1} = \frac{P_3}{P_4} = 9.5$$

$$T_3 = 1000K$$

$$T_2 = T_1 \left( \frac{P_2}{P_1} \right)^{\frac{k-1}{k}} = 288 \times 9.5^{\frac{1.4-1}{1.4}} = 548K$$

$$\eta_c = \frac{h_2 - h_1}{h_{2'} - h_1} = \frac{T_2 - T_1}{T_{2'} - T_1}$$

$$\therefore T_{2'} = T_1 + \frac{T_2 - T_1}{\eta_c} = 594K$$

$$\eta_c = \frac{h_3 - h_{4'}}{h_3 - h_4} = \frac{T_3 - T_{4'}}{T_3 - T_4}$$

$$T_4 = T_3 \left( \frac{P_4}{P_3} \right)^{\frac{k-1}{k}} = 526K$$

$$T_{4'} = T_3 + \eta_c (T_3 - T_4) = 573K$$

$$\omega_0 = \omega_{34'} - \omega_{12'} = (h_3 - h_{4'}) - (h_{2'} - h_1) = 121 \text{kJ/kg}$$

$$\eta_t = \frac{\omega_0}{q_{x0}} = \frac{\omega_0}{c_p (T_3 - T_{2'})} = 0.29$$

习题 9-14

答：

$$1) \text{ 当 } \pi = \frac{P_2}{P_1} = \left( \frac{T_2}{T_1} \right)^{\frac{k}{k-1}} = \frac{1273.15^{\left(\frac{1.4}{0.4}\right)}}{288.15} = 181.3 \text{ 时, } \eta_t \text{ 最大}$$

$$2) \text{ 当 } \pi \text{ 为 } 0 \text{ 时, } \eta_t \text{ 最小}$$

$$3) T_1 = 288.15K, T_3 = 1273.15K$$

$$\eta_t = 1 - \frac{1}{\pi^{\frac{k}{k-1}}}$$

$\eta_t$  随  $\pi$  的增大而提高

$$\omega_{net} = \omega_T - \omega_c = c_p (T_3 - T_4) - c_p (T_2 - T_1)$$

$$= c_p T_1 \left( \frac{T_3}{T_1} - \frac{T_4}{T_1} - \frac{T_2}{T_1} + 1 \right) = c_p T_1 \left( \tau - \tau \pi^{\frac{1-k}{k}} - \pi^{\frac{1-k}{k}} - 1 \right)$$

$$\text{令 } \frac{d\omega_{net}}{d\pi} = 0, \quad \pi_{opt} = \tau^{\frac{k}{2(k-1)}} = 13.46, w \text{ 最大}$$

习题 9-18

答：

$$T_1 = 223.13K, P_1 = 0.3bar, c_p = 1.004kJ / (kg \cdot K)$$

$$c_{f1} = 900m / s, c_{f2} = 100m / s$$

$$h_1 + \frac{1}{2}c_{f1}^2 = h_2 + \frac{1}{2}c_{f2}^2$$

$$T_2 = \frac{1}{2c_p}(c_{f1}^2 - c_{f2}^2) + T_1 = 622.3K$$

$$\pi = \frac{P_2}{P_1} = \left(\frac{T_2}{T_1}\right)^{\frac{k}{k-1}} = 36.23$$

$$q_1 = c_p(T_3 - T_2) = 400kJ / kg$$

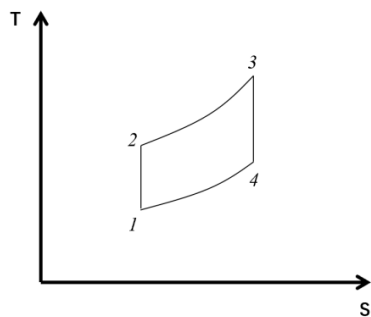
$$T_3 = 1020.7K$$

$$T_4 = T_3 \left(\frac{P_4}{P_3}\right)^{\frac{k-1}{k}} = T_3 \left(\frac{P_1}{P_2}\right)^{\frac{k-1}{k}} = 366K$$

$$q_2 = c_p(T_4 - T_1) = 143.44kJ / kg$$

$$w = q_1 - q_2 = 256.56kJ / kg$$

$$\eta = \frac{w}{q_1} = 64.14\%$$



习题 9-22

1) 不带回热

$$T_2 = T_1 \left( \pi \right)^{\frac{k-1}{k}} = 577.52 K$$

$$q_{in} = c_p (T_3 - T_2) = 738.72 \text{ KJ/ K g}$$

$$T_4 = T_3 \left( \frac{1}{\pi} \right)^{\frac{k-1}{k}} = 662 K$$

$$q_{out} = c_p (T_4 - T_1) = 372.35 \text{ KJ/ K g}$$

$$\eta = \frac{q_{in} - q_{out}}{q_{in}} = 49.6\%$$

2) 带回热

$$T_2 = T_1 \left( \pi \right)^{\frac{k-1}{k}} = 577.52 K$$

$$T_5 = T_4 \left( \frac{1}{\pi} \right)^{\frac{k-1}{k}} = 662 K$$

$$T_3 = T_5$$

$$T_6 = T_2$$

$$q_{in} = c_p (T_4 - T_3) = 651.3 \text{ KJ/ K g}$$

$$q_{out} = c_p (T_6 - T_1) = 286.42 \text{ KJ/ K g}$$

$$\eta = \frac{q_{in} - q_{out}}{q_{in}} = 56.02\%$$

### 习题 10-3

解:

$$\textcircled{1} p_2 = 0.004 \text{ MPa}$$

$$v_3 = 0.0010041 \text{ m}^3/\text{kg}, h_3 = 121.30 \text{ kJ/kg}, h_3'' = 2553.45 \text{ kJ/kg},$$

$$s_3 = 0.4221 \text{ kJ/(kg K)}, s_3'' = 8.4725 \text{ kJ/(kg K)}, x_2 = \frac{s_2 - s_3}{s_3'' - s_3} =$$

$$\frac{6.9676 - 0.4221}{8.4725 - 0.4221} = 0.813$$

$$h_2 = h_3 + x_2' (h_3'' - h_3) = 2098.64 \text{ kJ/kg}$$

$$w_p = v_3 (p_1 - p_2) = 0.0010041 \times (3.5 - 0.004) \times 10^3 = 3.51 \text{ kJ/kg}$$

$$h_4 = h_3 + w_p = 124.81 \text{ kJ/kg}$$

$$\text{循环热效率: } \eta_t = \frac{h_1 - h_2 - w_p}{h_1 - h_4} = 37.8\%$$

$$\textcircled{2} p_2 = 0.01 \text{ MPa}$$

$$v_3 = 0.0010103 \text{ m}^3/\text{kg}, h_3 = 191.76 \text{ kJ/kg}, h_3'' = 2583.72 \text{ kJ/kg},$$

$$s_3=0.649\text{kJ}/(\text{kg K}), \quad s_3''=8.1481, \quad x_2 = \frac{s_2-s_3}{s_3''-s_3} = \frac{6.9676-0.649}{8.1481-0.649} = 0.843$$

$$h_2 = h_3 + x_2'(h_3'' - h_3) = 2208.18 \text{ kJ/kg}$$

$$w_p=v_3(p_1-p_2)=0.0010103 \times (3.5-0.01) \times 10^3=3.53\text{kJ/kg}$$

$$h_4=h_3+w_p=195.29\text{kJ/kg}$$

$$\text{循环热效率: } \eta_t = \frac{h_1-h_2-w_p}{h_1-h_4} = 35.1\%$$

③  $p_2=0.1\text{MPa}$

$$v_3=0.0010432\text{m}^3/\text{kg}, \quad h_3=417.52\text{kJ/kg}, \quad h_3''=2675.14\text{kJ/kg},$$

$$s_3=1.3028\text{kJ}/(\text{kg K}), \quad s_3''=7.3589, \quad x_2 = \frac{s_2-s_3}{s_3''-s_3} = \frac{6.9676-1.3028}{7.3589-1.3028} = 0.935$$

$$h_2 = h_3 + x_2'(h_3'' - h_3) = 2528.39 \text{ kJ/kg}$$

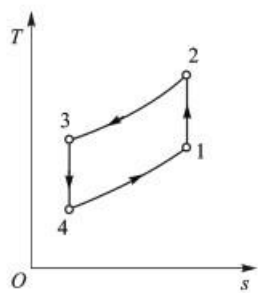
$$w_p=v_3(p_1-p_2)=0.001010432 \times (3.5-0.1) \times 10^3=3.55\text{kJ/kg}$$

$$h_4=h_3+w_p=421.07\text{kJ/kg}$$

$$\text{循环热效率: } \eta_t = \frac{h_1-h_2}{h_1-h_4} = 26.8\%$$

习题 11-1

答：空气制冷循环 T-S 图



由题目已知:  $T_1=293\text{K}$   $T_3=263\text{K}$   $P_1=P_4=1\text{bar}$   $P_2=P_3=6\text{bar}$

1) 求每千克空气制冷量

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$$T_4 = T_3 \left( \frac{P_4}{P_3} \right)^{\frac{K-1}{K}} = 263 \times \left( \frac{1}{6} \right)^{\frac{0.4}{1.4}} = 157.6K$$

$$T_2 = T_1 \left( \frac{P_2}{P_1} \right)^{\frac{K-1}{K}} = 293 \times 6^{\frac{0.4}{1.4}} = 488.9K$$

每千克空气制冷量：

$$q_c = h_1 - h_4 = 1.004 \times (293 - 157.6) = 136KJ / Kg$$

2) 每千克空气所需压缩功：

$$W_c = h_2 - h_1 = 1.004 \times (488.9 - 157.6) = 332.6KJ / Kg$$

3) 制冷系数：

$$\varepsilon = \frac{h_1 - h_4}{(h_2 - h_3) - (h_1 - h_4)} = \frac{1.004 \times (293 - 157.6)}{1.004 \times [(488.9 - 263) - (293 - 157.6)]} = 1.5$$

4) 在相同的温度范围内，逆向卡诺循环是制冷系数最高的循环

$$\varepsilon_{\max} = \varepsilon_c = \frac{T_3}{T_1 - T_3} = \frac{263}{293 - 263} = 8.8$$