



Year 12 Chemistry
Mastery Test: Ionic Equations – Acid/base reactions

Name: _____

Write balanced ionic equations for any reactions that occur when the following substances are mixed.

1. Solid aluminium oxide is added to a solution of hydrochloric acid

Equation _____

Observation _____

2. A solution of acetic acid (ethanoic acid) is added to solid sodium hydrogen carbonate

Equation _____

Observation _____

3. Carbonic acid is added to a solution of lithium hydroxide

Equation _____

Observation _____

4. Phosphoric acid is added to aqueous sodium hydroxide

Equation _____

Observation _____

5. Sulfuric acid is added to a solution of cobalt nitrate

Equation _____

Observation _____

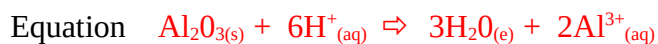


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ANSWERS

Write balanced ionic equations for any reactions that occur when the following substances are mixed.

1. Solid aluminium oxide is added to a solution of hydrochloric acid



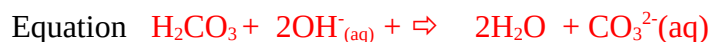
Observation **A white solid is added to a colourless solution and a colourless solution is formed.**

2. A solution of acetic acid (ethanoic acid) is added to solid sodium hydrogen carbonate



Observation **A white solid is added to a colourless solution and a colourless, odourless gas is formed in a colourless solution.**

3. Carbonic acid is added to a solution of lithium hydroxide



Observation **A white solid is added to a colourless solution and a colourless solution is formed.**

4. Phosphoric acid is added to aqueous sodium hydroxide



Observation **Two colourless solutions are mixed and a colourless solution is formed.**

5. Sulfuric acid is added to a solution of cobalt nitrate



Observation