

## Photo Description



This picture shows golden-brown fried food pieces sitting on top of fresh green lettuce and colorful vegetables. The fried pieces have a bumpy, crispy coating that looks crunchy. You can see different colored vegetables like orange carrots and purple cabbage mixed with the green lettuce leaves.

## Scientific Phenomena

The Anchoring Phenomenon is the transformation of food through cooking and heating. When foods are heated in oil, the water inside turns to steam and escapes, while the outside becomes crispy and changes color. This happens because heat energy breaks apart and rearranges the molecules in the food, creating new textures, colors, and even new substances through chemical reactions.

## Core Science Concepts

1. Heat Transfer - Heat energy moves from the hot oil into the cooler food, cooking it from the outside in
2. States of Matter - Water in the food changes from liquid to gas (steam) when heated
3. Physical vs. Chemical Changes - The food undergoes both physical changes (getting hot, losing water) and chemical changes (browning, new flavors forming)
4. Energy Transformation - Electrical or gas energy becomes heat energy, which then changes the food's properties

### Pedagogical Tip:

Use familiar foods like popcorn or toast to help students connect to the concept of heat changing matter. Students can observe and describe changes they see, hear, and smell during cooking demonstrations.

### UDL Suggestions:

Provide multiple ways for students to express their observations through drawing, acting out the process, or using graphic organizers to compare before/after states of the food.

## Zoom In / Zoom Out

1. Zoom In: At the molecular level, heat energy makes tiny particles (molecules) move faster and faster until chemical bonds break and reform, creating new substances with different properties than the original raw food.
2. Zoom Out: This connects to the larger food system where energy from the sun helps plants grow, humans harvest and process the food, and we use various energy sources to transform raw ingredients into meals that provide energy for our bodies.

### Discussion Questions

1. What changes do you notice between raw vegetables and cooked food? (Bloom's: Analyze | DOK: 2)
2. How do you think heat energy traveled from the stove to change this food? (Bloom's: Apply | DOK: 2)
3. What evidence tells you that the food has been heated? (Bloom's: Evaluate | DOK: 3)
4. If you could watch the tiny particles in this food while it cooked, what do you think you would see happening? (Bloom's: Create | DOK: 3)

### Potential Student Misconceptions

1. Misconception: "The food just gets hot, nothing else changes."  
Clarification: Heating food creates both physical and chemical changes - new colors, textures, and even new substances form.
2. Misconception: "All the water disappears when you cook food."  
Clarification: Some water turns to steam and escapes, but some water stays inside the food in different forms.
3. Misconception: "Heat only makes things melt."  
Clarification: Heat can cause many different changes - melting, evaporating, chemical reactions, and changes in texture and color.

### NGSS Connections

- Performance Expectation: 2-PS1-1 (can be adapted up) - Plan and conduct an investigation to describe and classify different kinds of materials by their observable properties
- Disciplinary Core Ideas: 2-PS1.A - Different kinds of matter exist and many of them can be either solid or liquid, depending on temperature
- Crosscutting Concepts: Cause and Effect - Simple tests can be designed to gather evidence to support or refute student ideas about causes

### Science Vocabulary

- \* Heat energy: The energy that makes things warmer and can change how they look and feel
- \* Chemical change: When heat or other forces create completely new substances with different properties
- \* Physical change: When something looks different but is still made of the same materials
- \* Molecule: Tiny particles too small to see that make up all matter
- \* Steam: Water that has turned into an invisible gas when heated
- \* Temperature: How hot or cold something is

### External Resources

Children's Books:

- From Milk to Ice Cream by Bridget Heos
- Cooking by Sally M. Walker
- Heat by David Dreier

YouTube Videos:

- "States of Matter for Kids" - Simple explanation of how heating changes solids, liquids, and gases: <https://www.youtube.com/watch?v=ZnbKmPdpVFs>

- "Physical and Chemical Changes" - Kid-friendly examples of different types of changes in matter: [https://www.youtube.com/watch?v=37pir0ej\\_SE](https://www.youtube.com/watch?v=37pir0ej_SE)