Quantum Chemistry

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(What do we mean by spin coordinate ω?) By multiplying a spatial orbital with any one of these spin function, we obtain spin orbital $\chi(\mathbf{x})$ where $\mathbf{x}$ specifies both spatial coordinate $\mathbf{r}$ and spin coordinate ω. X $-$ 5 So for every spatial orbital $\psi(\mathbf{r})$, we have 2 spin orbitals $\Psi(\mathbf{x_i})$ and X($\mathbf{x_i}$).

$\left\langle \psi(\vec{t}) \right| H \left| \psi(\vec{t}) \right\rangle$

$$H\psi = E\psi$$