

# BIOLOGICAL HOMOCHIRALITY

Noise-Induced Symmetry Breaking  
Far from Equilibrium





# Homochirality: one of the universal features of life

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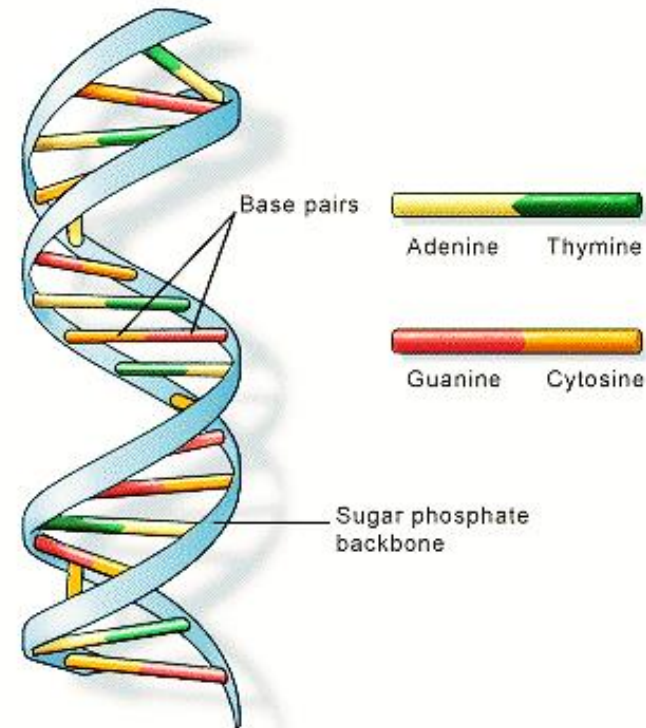
Together with **Canonical genetic code** , **homochirality** is one of the universal features of life on Earth.

- **Canonical genetic code:** refers to the set of rules by which the genetic information in DNA and RNA is translated into proteins.
- **Homochirality:** the single-handedness of all biological amino acids and sugars

The only universal process common to all life is, of course, **evolution**, and so it is natural to seek an explanation for biological homochirality in these terms.

# Homochirality: Where does it show itself?

The famous **double helix structure of DNA** is a **result of the chirality** of the sugar molecules in its backbone. Despite the diversity of proteins and their functions virtually all chiral biological amino acids are L-chiral, while all sugars are D-chiral.

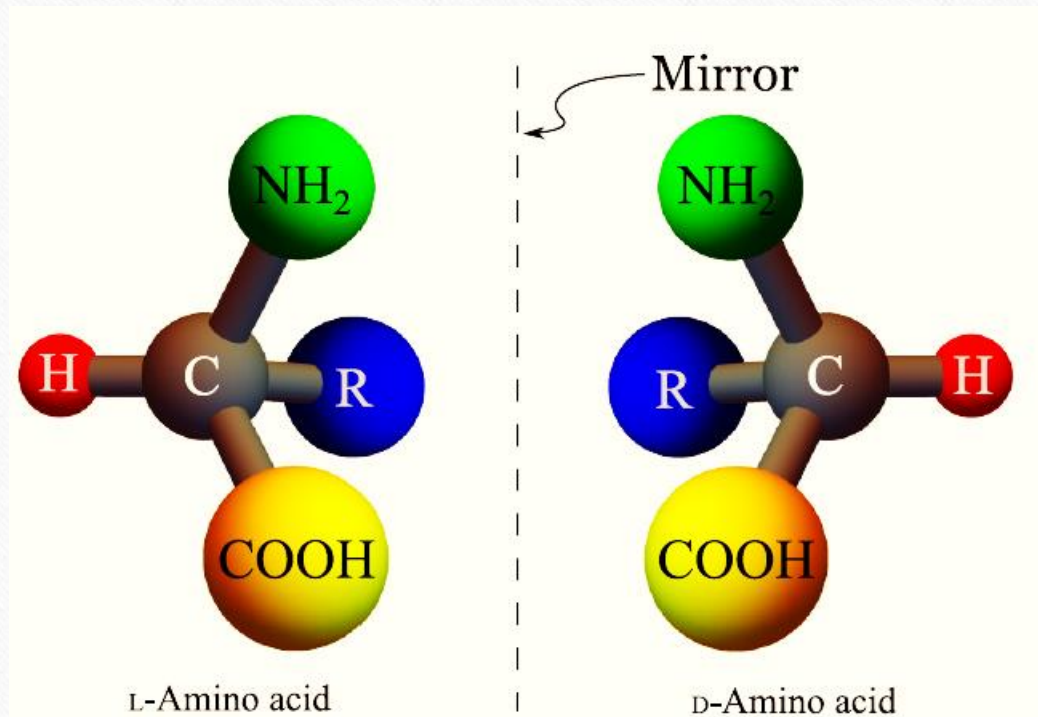


U.S. National Library of Medicine



# Homochirality: What does it mean?

Molecules that are not superimposable on their mirror image are called **chiral** (Greek for hand), and the atom surrounded by four different groups is known as the **chiral center** of the molecule.



# Some Homochirality Jargon

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- **enantiomer:** each category of chiral molecules.
- **racemic solution:** solutions of 50% right-handed and 50% left-handed molecules.
- **homochiral solution:** solution of all left-handed or all right-handed molecules.

# The Mystery of Homochirality

Early  
Earth

The initial state was symmetric: solution of achiral molecules

The  
laws of physics are symmetric



We expect a symmetric final state: a biosphere made of a racemic solution of chiral molecules



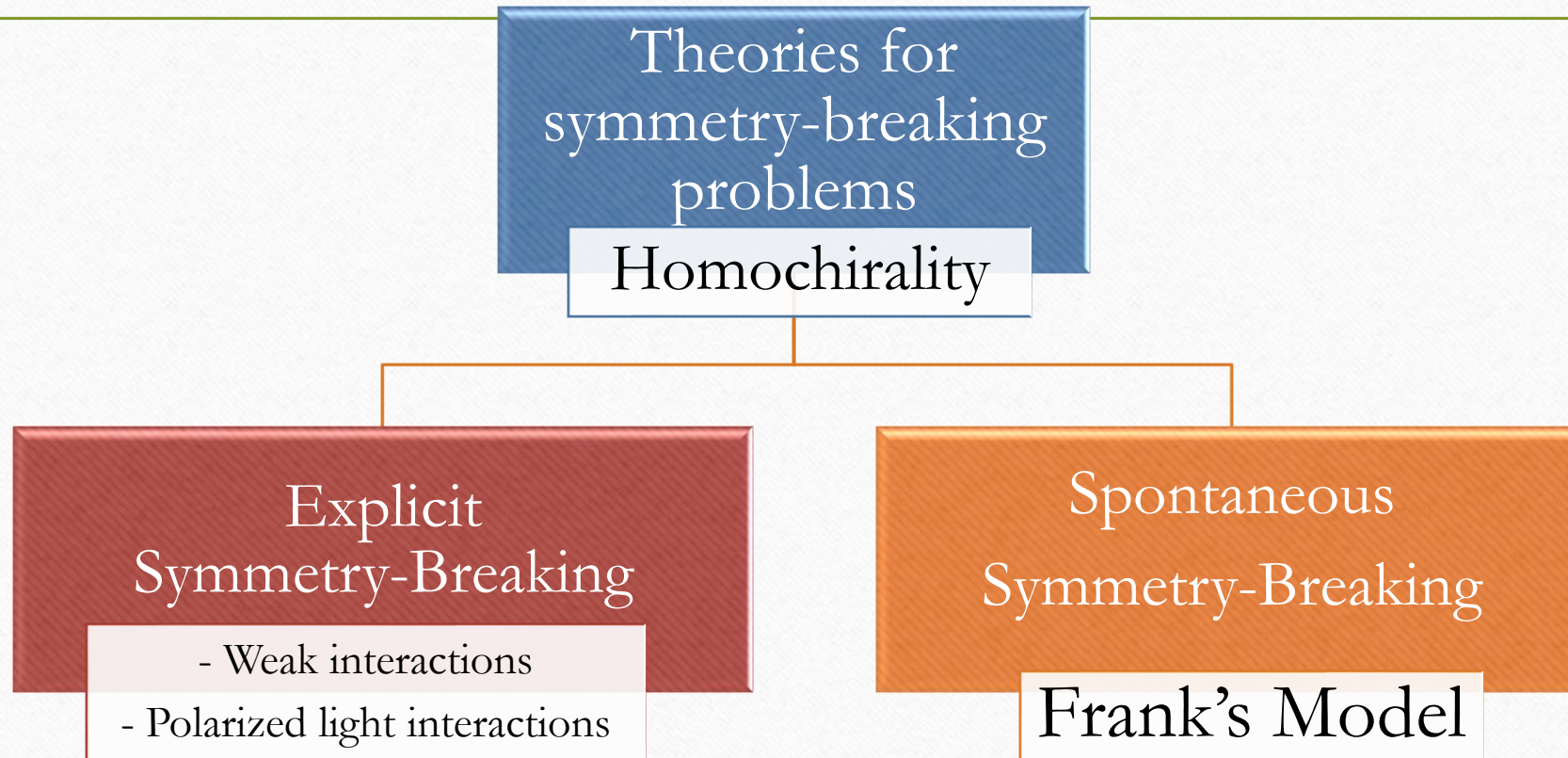


# Homochirality: A symmetry-breaking case

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A phenomenon in which the **initial state** and the **corresponding laws of physics** are symmetric with respect to a particular transformation, but the final state of the system violates that symmetry, is called a symmetry breaking

# Homochirality: Possible Explaining Theories





# Homochirality: Explicit Symmetry-Breaking Theories

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the laws of physics are only approximately symmetric, or there is an asymmetric perturbation to the system

- if life was formed from **chiral** organic molecules that were produced under a steady radiation of **circularly polarized** light, the **asymmetric interaction** of different enantiomers of chiral molecules with the light over hundreds of millions of years could lead to a significant enantiomeric excess
- Unlike electromagnetic interactions, the weak interaction violates mirror symmetry

A common weakness of explicit symmetry-breaking mechanisms is that the **homochirality achieved is only partial**

# Homochirality: Spontaneous Symmetry-Breaking Theories

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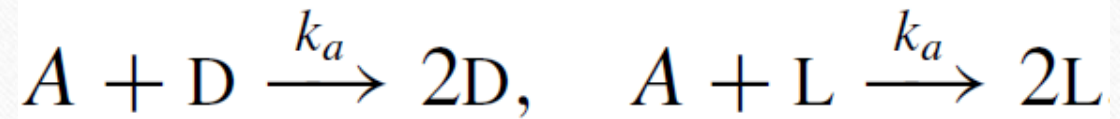
the governing laws are perfectly symmetric, and as a result, the symmetric state is a final solution, but it may be an **unstable solution**. In this case, even the slightest perturbation to the system moves the system away from the symmetric state.



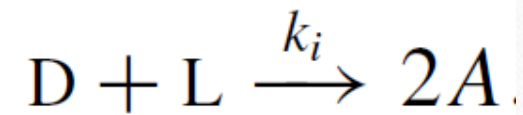
# Spontaneous Symmetry-Breaking Theories: Frank's Model

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Auto-Catalysis Reaction



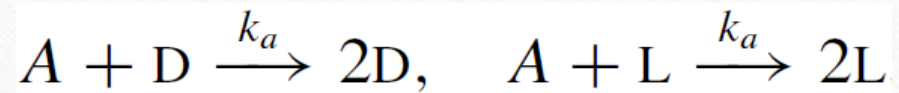
Chiral Inhibition Reaction



# Spontaneous Symmetry-Breaking Theories: Frank's Model

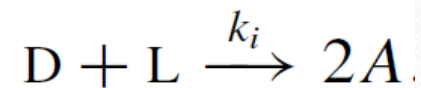
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Auto-Catalysis Reaction



self-replication

Chiral Inhibition Reaction



Law of Mass Action

$$\frac{d[A]}{dt} = 2k_i [D] [L] - k_a [A] ([D] + [L])$$

$$\frac{d[D]}{dt} = k_a [A] [D] - k_i [L] [D],$$

$$\frac{d[L]}{dt} = k_a [A] [L] - k_i [D] [L].$$



# Spontaneous Symmetry-Breaking Theories: Frank's Model

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The State of the System

$$\omega = \frac{[D] - [L]}{[D] + [L]}$$

$$\frac{d\omega}{dt} = \frac{1}{2}k_i([D] + [L])\omega(1 - \omega^2)$$

# Spontaneous Symmetry-Breaking Theories: Frank's Model

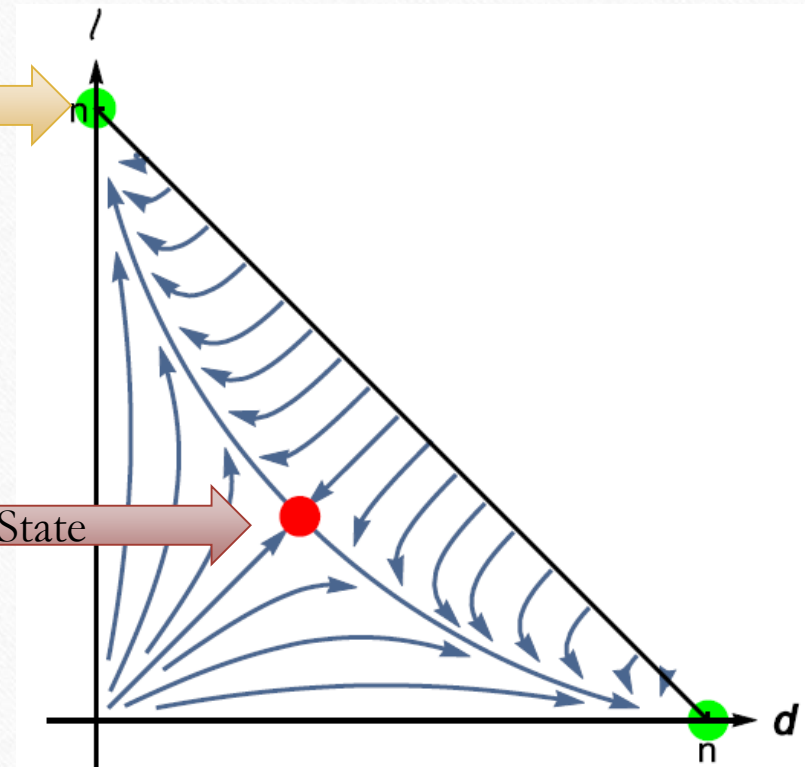
The State of the System

$$\omega = \frac{[D] - [L]}{[D] + [L]}$$

Homochiral State

$$\frac{d\omega}{dt} = \frac{1}{2}k_i([D] + [L])\omega(1 - \omega^2)$$

Racemic State



(a)  $A + D \rightarrow 2D$ ,  $A + L \rightarrow 2L$ ,  
 $D + L \rightarrow 2A$



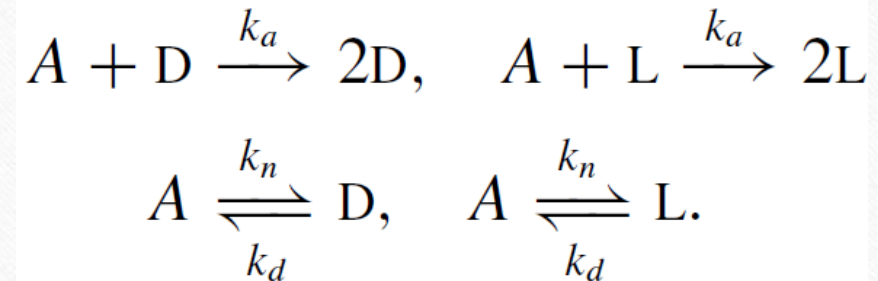
## The Problems with Frank's Model

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Although **autocatalysis** is an expected prerequisite for early life **self-replicators**, the mutually antagonistic relationship between the two chiral molecules (**chiral competition**) does not seem to be biologically necessary

## New Proposed Model

Let's remove the **chiral inhibition** reaction and replace it by **nonautocatalytic** and **decay reactions**

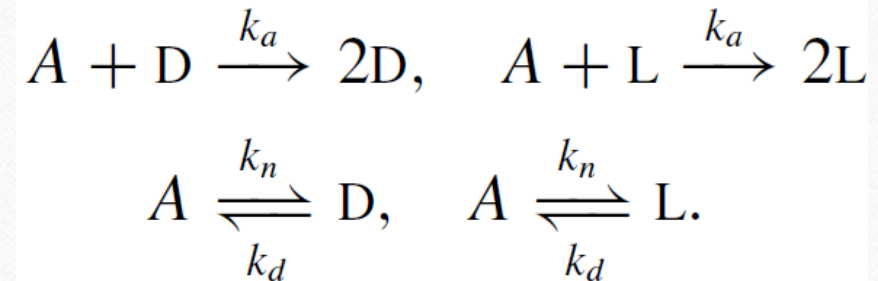


This model can be interpreted as a model for the evolution of early life where primitive **chiral self-replicators** can be produced randomly through **nonautocatalytic** processes at very low rates; the **self-replication** is modeled by **autocatalysis** while the **decay reaction** is a model for the **death process**.



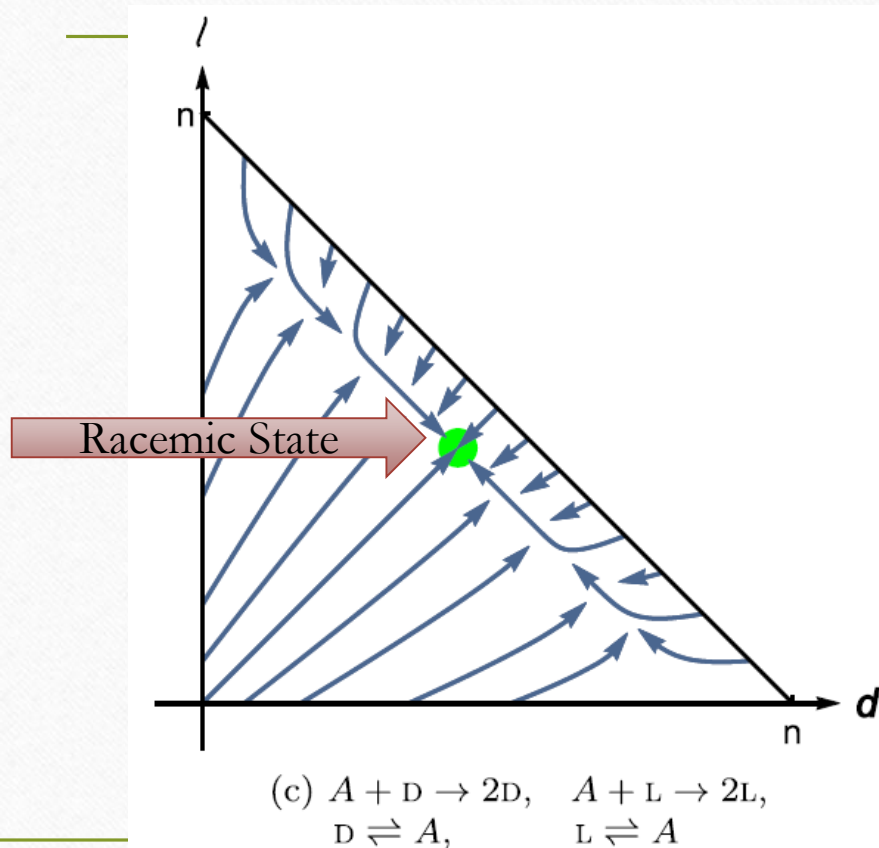
## New Proposed Model

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## New Proposed Model: Deterministic Approach



Spoiler!

when the effect of chemical number fluctuations from self-replication is taken into account, the system can transition to homochirality when the autocatalysis is the dominant mechanism for the production of the chiral molecules



## A Change of Approach: Revisiting our Assumptions

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**Law of mass action:** the **expected value** of the number of collisions per unit time is proportional to the product of the concentrations of the reactants.

**Near equilibrium,** a system of a large number of interacting chemicals follows Boltzmann statistics and can be approximated by its **expected value**.

the **expected value** of number of collisions is used instead of the actual **probability distribution of the number of collisions per unit time**.

# A Change of Approach: Far from Equilibrium Conditions

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interpret the **law of mass action** as the **probability** per unit time of occurrence of a chemical reaction.

write the **master equation** for the rate of change of the **probability** of the system having given concentrations of reactants and products



# The Stochastic Model: The Master Equation

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$$\frac{\partial P(\vec{x}, t)}{\partial t} = V \sum_{\vec{y}} [T(\vec{x}|\vec{y})P(\vec{y}, t) - T(\vec{y}|\vec{x})P(\vec{x}, t)]$$

The probability to  
find the system at  
state  $x$  at time  $t$

The transition rate  
from the state  $y$  to  
state  $x$

## The Stochastic Model: The Langevin Equation

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$$\frac{d\omega}{dt} = -\frac{2k_n k_d V}{N k_a} \omega + \sqrt{\frac{2k_d}{N}} (1 - \omega^2) \eta(t)$$

Deterministic Part

White noise



## The Stochastic Model: The Fokker – Planck Equation

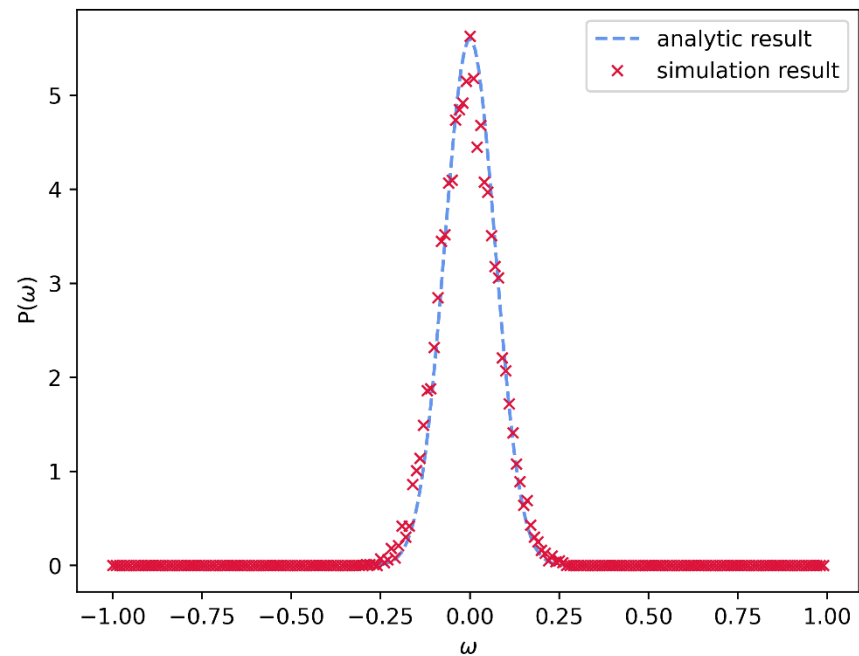
$$\begin{aligned}\frac{\partial P(\omega, t)}{\partial t} = & \frac{\partial}{\partial \omega} \left[ \frac{2k_n k_d V}{N k_a} \omega P(\omega, t) \right] \\ & + \frac{1}{2} \frac{\partial^2}{\partial \omega^2} \left[ \frac{2k_d}{N} (1 - \omega^2) P(\omega, t) \right]\end{aligned}$$

The Steady State  
Solution

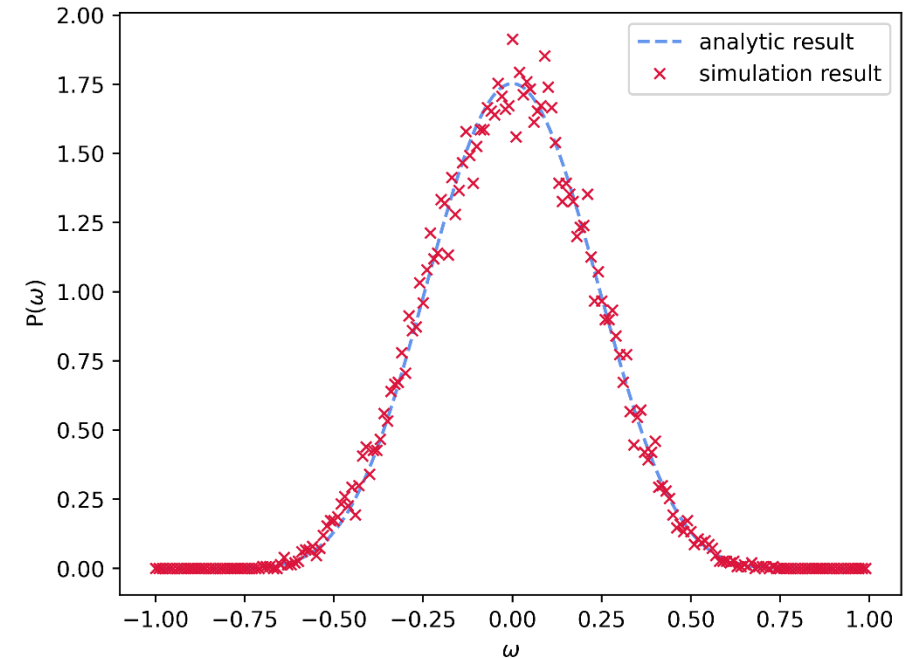
$$P_s(\omega) = \mathcal{N}(1 - \omega^2)^{\alpha-1}, \quad \text{with} \quad \alpha = \frac{k_n V}{k_a}$$

# The Stochastic Model: Gillespie simulations for $\alpha > 1$

probability density from simulation v.s. analytic model  
 $\alpha = 100$

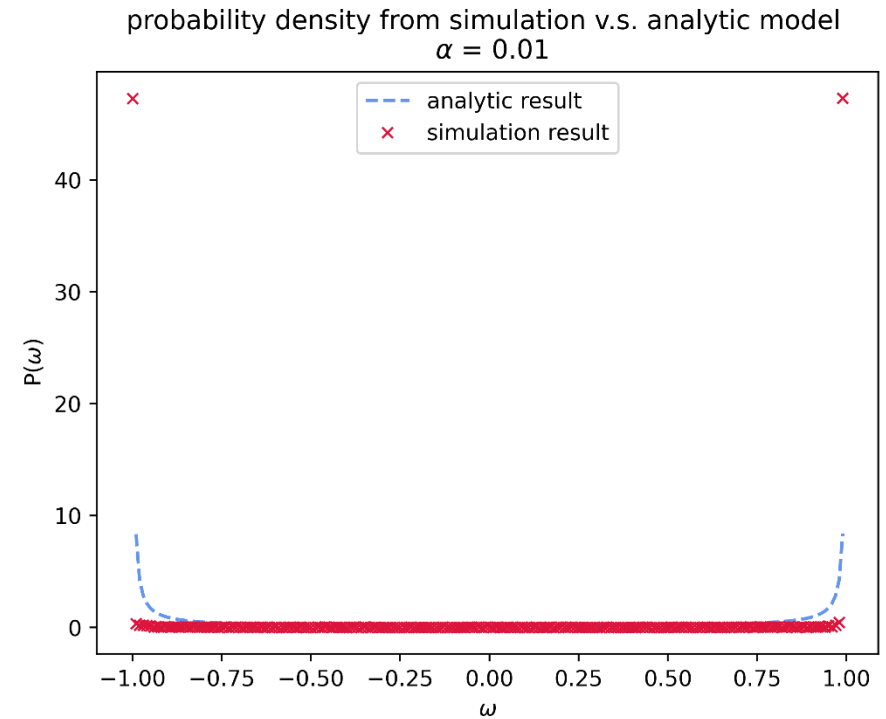
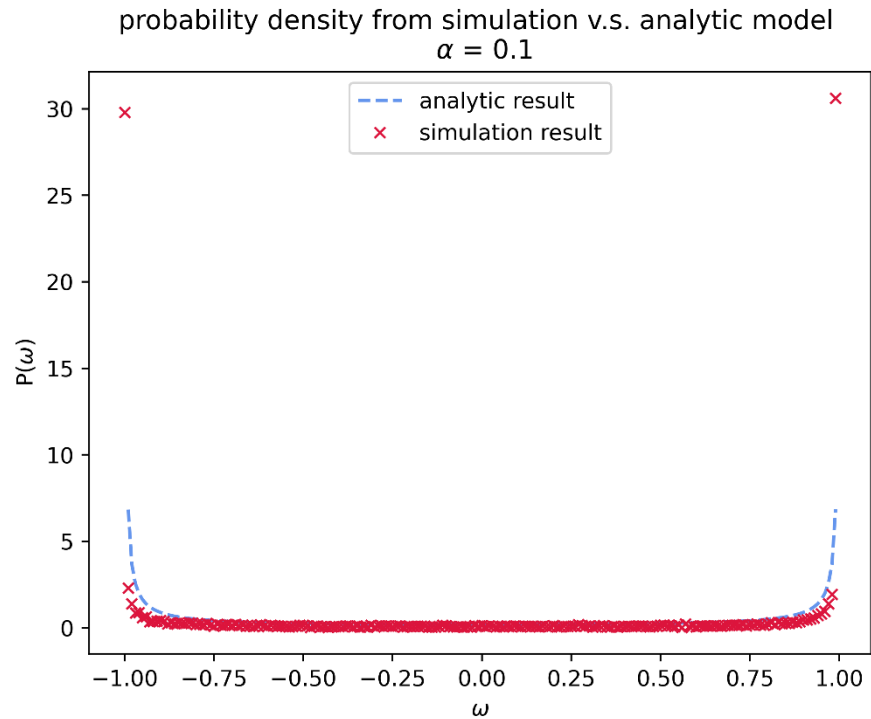


probability density from simulation v.s. analytic model  
 $\alpha = 10$

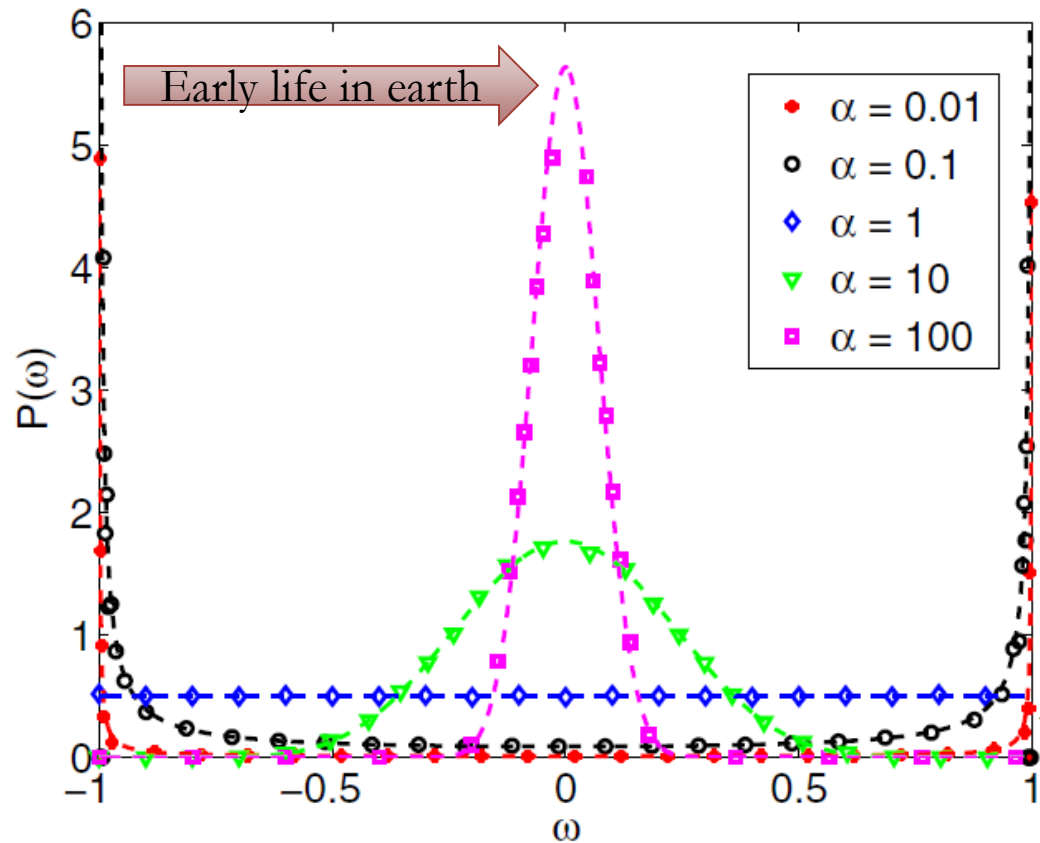




# The Stochastic Model: Gillespie simulations for $\alpha < 1$



# The Stochastic Model



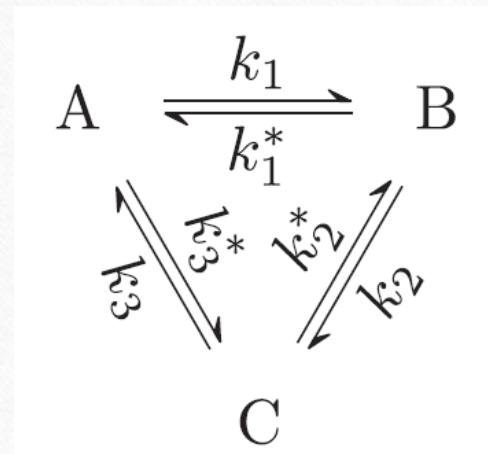
$$\alpha_c = 1$$

The parameter  $\alpha$  is proportional to the ratio of the nonautocatalytic production rate,  $k_n$ , to the self replication rate,  $k_a$ .



## Thermodynamics of the System: principle of microscopic reversibility

the **principle of microscopic reversibility** states that **at equilibrium**, the rate of the forward reaction and the reverse reaction are equal for all reactions.



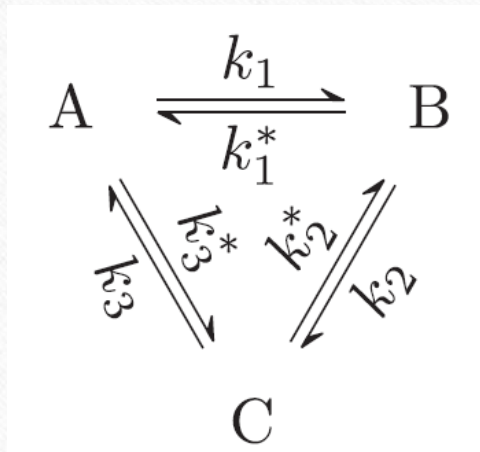
$$k_1[A] = k_1^*[B], \quad k_2[B] = k_2^*[C], \quad k_3[C] = k_3^*[A].$$

Wegscheider's  
Condition

$$\frac{k_1}{k_1^*} \frac{k_2}{k_2^*} \frac{k_3}{k_3^*} = 1.$$

## Thermodynamics of the System: Wegscheider's Condition

The constants are constants! Therefore even away from equilibrium the condition should hold. In other words, Wegscheider's condition is **the condition for the existence of static equilibrium solution**.



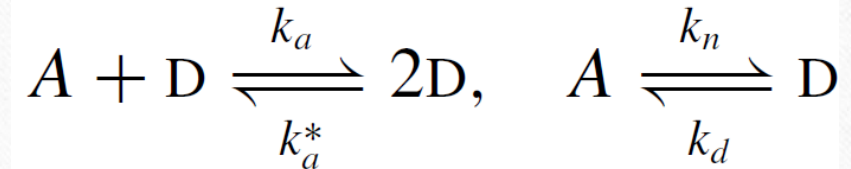
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Wegscheider's  
Condition

$$\frac{k_1}{k_1^*} \frac{k_2}{k_2^*} \frac{k_3}{k_3^*} = 1.$$

Thermodynamics of the System: Does our system have an equilibrium solution?

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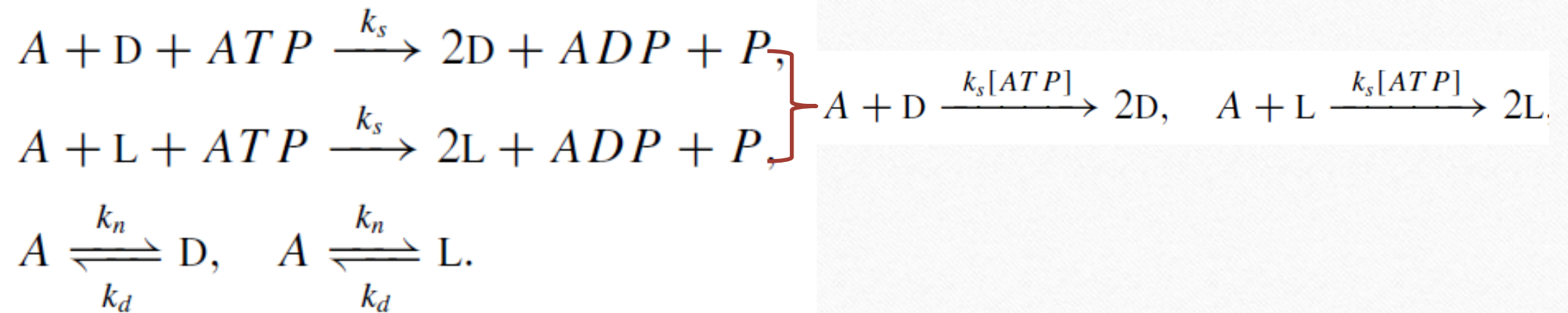
$$k_a[A][D] = k_a^*[D]^2, \quad k_n[A] = k_d[D]$$

$$\frac{k_a}{k_a^*} = \frac{k_n}{k_d}$$



## Thermodynamics of the System: closed-system set of reactions

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## Thermodynamics of the System: what are the evidence?

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It is a fact that all biological systems are driven out of equilibrium.

*But*

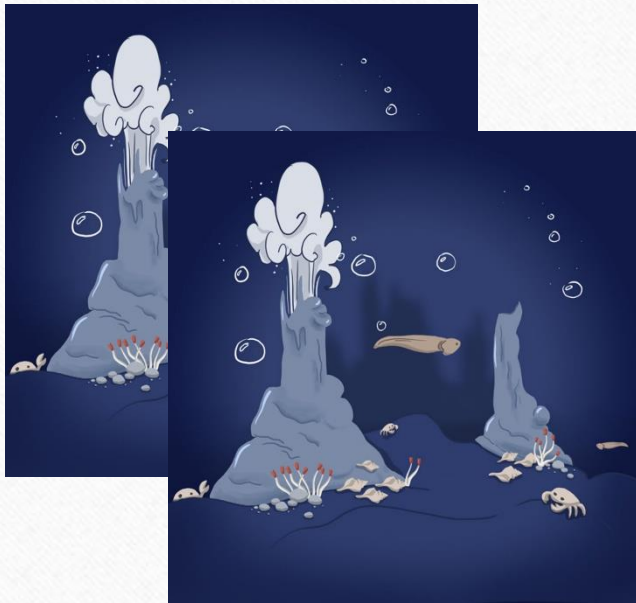
There is more reasoning behind the assumption that the homochiral system is driven out of equilibrium:

there is no closed dilute system with a completely homochiral equilibrium.

$$G = U + pV - TS$$

## Homochirality: The Spatially Extended System

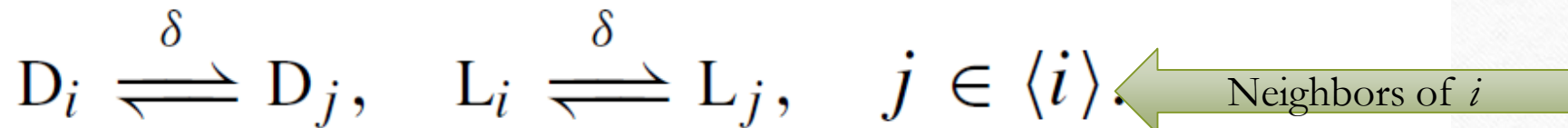
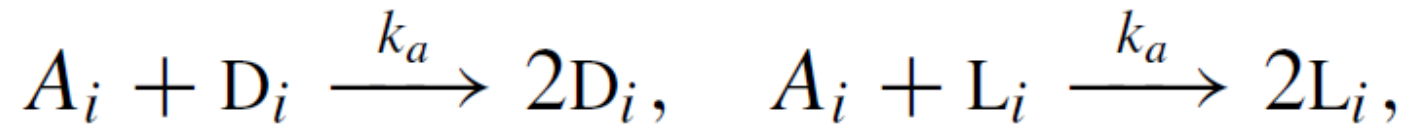
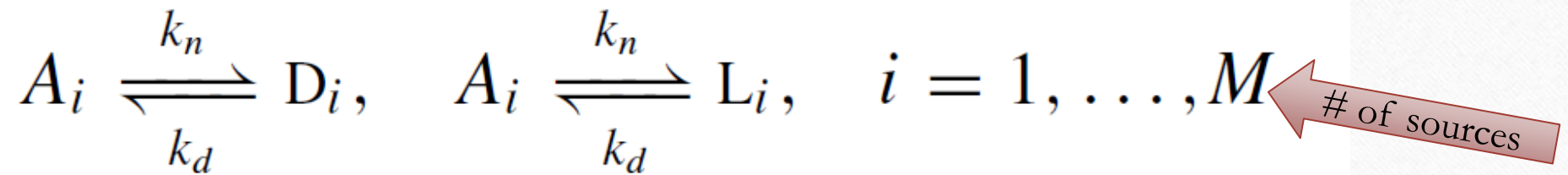
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- $\delta$ : **rate of diffusion** between different sources  
(for example hydrothermal vents)
- For low values of  $\delta$ , the chirality of each source is **independent** of the others
  - For high values, the sources **synchronize** their chirality.



## Homochirality: The Spatially Extended Model




## Homochirality: The Spatially Extended Model Results

$$\begin{aligned} \frac{d\omega_i}{dt} = & -\frac{2k_n k_d V}{N k_a} \omega_i + \delta \sum_{j \in \langle i \rangle} (\omega_j - \omega_i) \\ & + \sqrt{\frac{2k_d}{N} (1 - \omega_i^2)} \eta_i(t) + \sqrt{\frac{\delta}{N}} \xi_i(\vec{\omega}, t), \end{aligned}$$

correlated  
Gaussian noise

$M = 2$  Model: different  $\delta$  regimes

$$\frac{d\omega_i}{dt} = -\frac{2k_n k_d V}{N k_a} \omega_i + \delta \sum_{j \in \langle i \rangle} (\omega_j - \omega_i) + \sqrt{\frac{2k_d}{N}} (1 - \omega_i^2) \eta_i(t) + \sqrt{\frac{\delta}{N}} \xi_i(\vec{\omega}, t),$$


$\delta$  should be compared with  $\frac{2k_d \alpha}{N}$



$M = 2$  Model: critical  $\alpha$  for low  $\delta$  regime


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for  $\delta \approx 0$

$$Q_s(\omega) = \mathcal{Z}(1 - \omega^2)^{\alpha + \frac{\delta N}{2k_d} - 1}$$

$$\alpha_c \approx 1 - \delta \frac{N}{2k_d}$$

$M = 2$  Model: critical  $\alpha$  for high  $\delta$  regime

$$\frac{d\omega_i}{dt} = -\frac{2k_n k_d V}{N k_a} \omega_i + \delta \sum_{j \in \langle i \rangle} (\omega_j - \omega_i) + \sqrt{\frac{2k_d}{N}} (1 - \omega_i^2) \eta_i(t) + \sqrt{\frac{\delta}{N}} \xi_i(\vec{\omega}, t),$$


$\delta$  should be compared with  $\frac{2k_d \alpha}{N}$

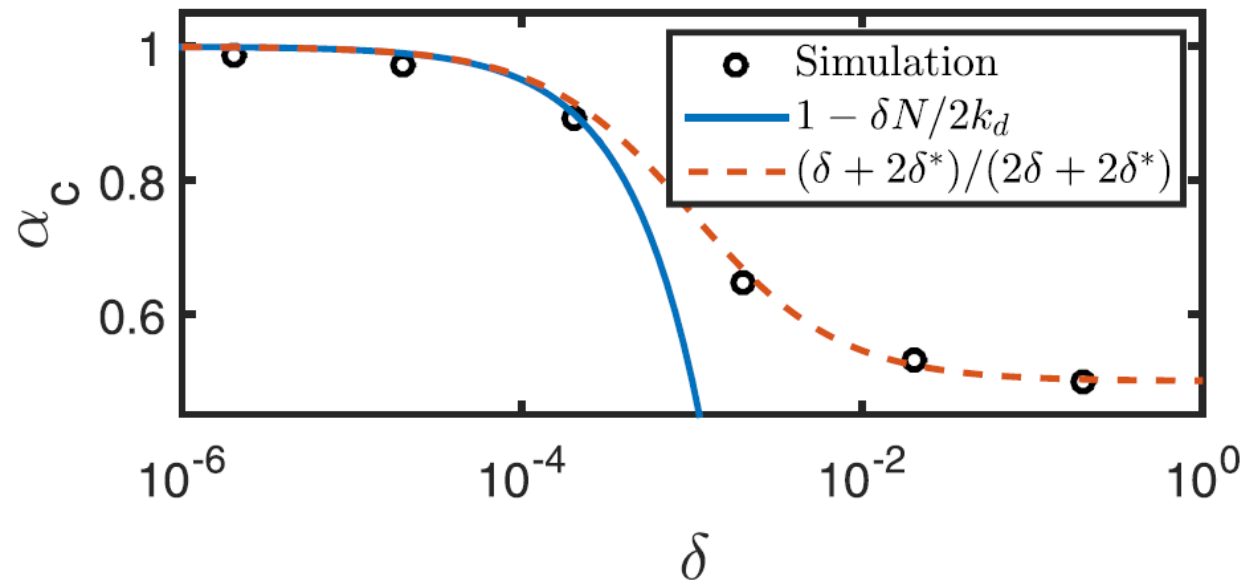
$$\alpha = \frac{k_n V}{k_a}$$

For  $\delta \gg k_d/N$

the whole system can be considered well mixed and has the critical value of  $\alpha$ ,  $\alpha_c^{\text{system}} = 1$ , from the well-mixed results

$$\text{for } \delta \gg 0, \quad \alpha_c \approx \frac{1}{2}$$

## $M = 2$ Model: Robustness Analysis



$$\alpha_c = \frac{\delta + 2\delta^*}{2\delta + 2\delta^*}, \quad \delta^* = \frac{k_d}{N}.$$

$$\alpha = \frac{k_n V}{k_a}$$



## The Spatially Extended Model: Global Homochirality

$$\phi(t, \vec{x}_1, \vec{x}_2) = \langle \omega(t, \vec{x}_1) \omega(t, \vec{x}_2) \rangle$$

$$\zeta = \sqrt{\frac{n \mathcal{D} k_a}{2 k_n k_d}}.$$

