SLE155 Chemistry for the Professional Sciences

Burwood and Geelong



Practice Questions Week 6

Hydrocarbons

Alkanes

Alkenes

Alkynes

Arenes



Pentane and cyclopentane are constitutional isomers.

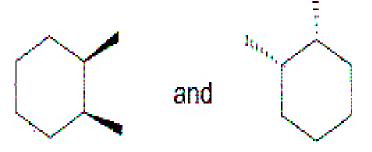
a. True

*b. False

The following structures represent the same molecule

*a. True

b. False



2,3-dimethyl-3-hexene occurs as E and Z isomers.

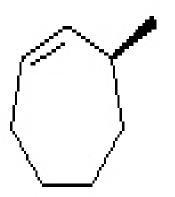
*a. True

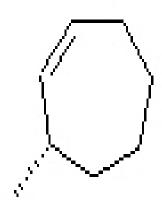
b. False

The following structures represent isomers.

a. True

*b. False





The following carbocations are listed in increasing order of stability (least first).

- a. True
- *b. False



The name of the following structure is 3,5-dibromobenzoic acid

*a. True

b. False

Which of the following molecules are constitutional isomers?

a.
$$CH_3CH_2OCH_3 \quad \text{and} \quad CH_3CH_2CH$$
 a.
$$CH_3CH_2OCH_3 \quad \text{and} \quad CH_3CCH_3$$
 b.
$$CH_3CH_2OCH \quad \text{and} \quad CH_3CCH_3$$

$$*_c. \quad CH_3CH_2OH \quad \text{and} \quad CH_3CCH_3$$
 d.
$$CH_3CH_2OH \quad \text{and} \quad CH_3CH$$
 d.

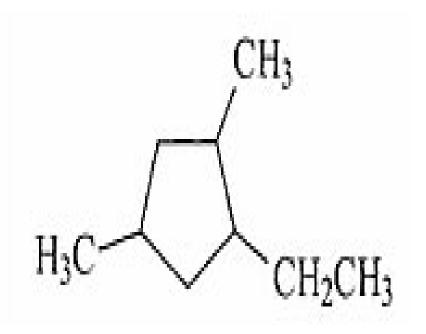
Which is the IUPAC name for the following cyclohexane?

*a. 1-ethyl-2,4-dimethylcyclopentane

b. 1,3-dimethyl-5-ethylcyclopentane

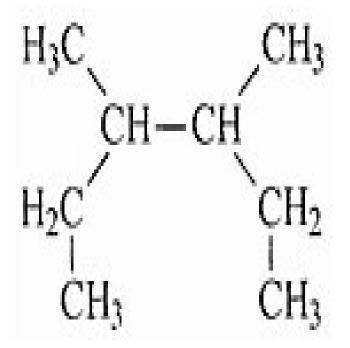
c. 2,4-dimethyl-1-ethylcyclopentane

d. 1-ethyl-3,5-dimethylcyclopentane



Which is the IUPAC name for the following compound?

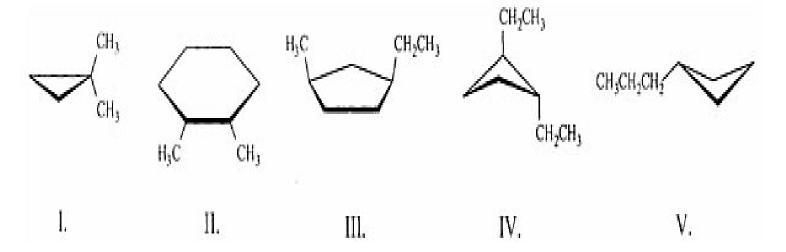
- a. 2-ethyl-3-methylpentane
- *b. 3,4-dimethylhexane
- c. 2,3-diethylbutane
- d. 3-methyl-4-ethylpentane



Which of the following cycloalkanes show cis-

trans isomerism?

a. I, Vb. I, II, Vc. II, III, V*d. II, III, IV



*a.

b

CH₃CH₂CH₂CH₂CH₃

c.



d.

Answer: a

Which compound has the lowest boiling point?

Which is the IUPAC name for the following compound?

a. 1,1-dimethyl-4-ethyl-2,5-cyclohexadiene

b. 1-ethyl-4,4-dimethyl-2,5-cyclohexadiene

c. 3-ethyl-6,6-dimethyl-1,4-cyclohexadiene

*d. 6-ethyl-3,3-dimethyl-1,4-cyclohexadiene

$$H_3C$$
 H_3C
 CH_2CH_3

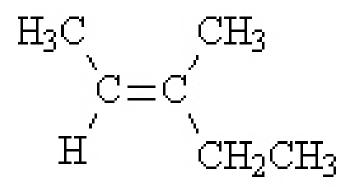
Which is the correct name for the following compound?

*a. E-3-methyl-2-pentene

b. Z-3-methyl-3-pentene

c. E-2-ethyl-2-butene

d. Z-3-ethyl-2-butene



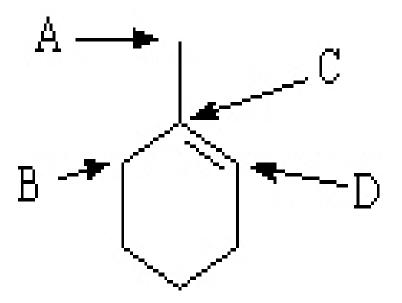
Using Markovnikov's rule, predict the position of the Cl atom in the major product from the reaction of 1-methylcyclohexene with HCl.

a. A

b. B

*c. C

d. D



Which is the major product from acid catalyzed hydration of 2-methyl-2-pentene?

- a. 2-methyl-3-pentanol
- b. 3-methyl-3-pentanol
- c. 4-methyl-2-pentanol
- *d. 2-methyl-2-pentanol

Which structures have the correct IUPAC names?

a. I, II

b. III, IV

c. II, III

*d. I, IV

I. 2-bromotoluene II. *para* methyl, *meta* chlorobromobenzene

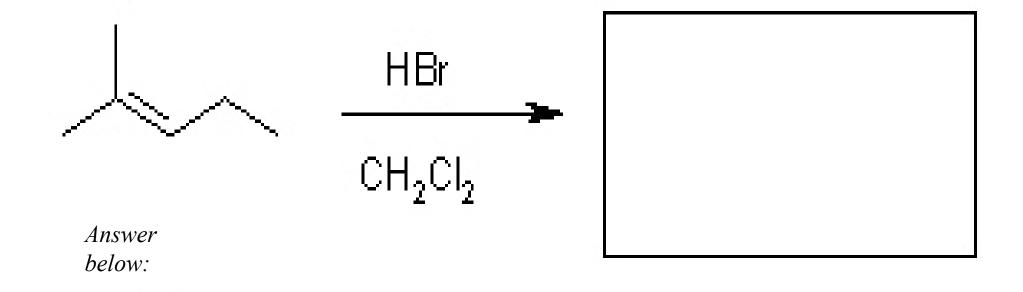
OCH₃
CH₂CH₃

III. meta chloroaniline

IV . para ethylanisole

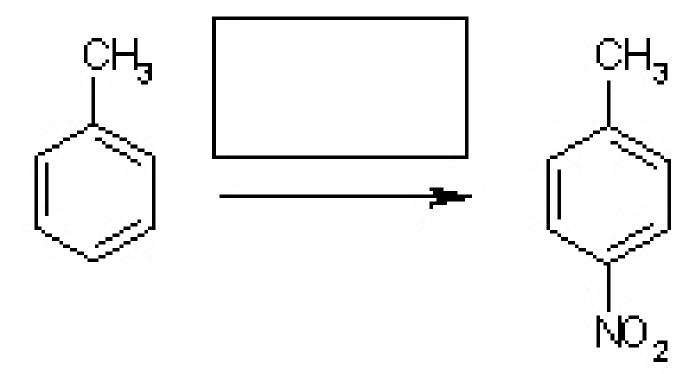


Complete the following reaction by providing the major product.





The reagent that completes the following reaction is?



Answer: HNO3, H2SO4