## Diploma of Health Sciences Diploma of Science

#### **SLE155 Chemistry for the Professional Sciences**

#### Q9 The chemistry of carbon

[3+1+6+4+6=20 marks]

a) Write a condensed structural formula showing every atom but no bonds, and a molecular formula for the following alkene.

Hint: You do not need to name it!

[2 + 1 = 3 marks]

	Condensed formula
	Molecular formula
·	

b) Arrange the following alkanes in order of <u>increasing</u> boiling point. 2-methylbutane, 2,2-dimethylpropane and pentane

[1 mark]

c) i) Write structures for the following compounds:

[2+2=4 marks]

1-ethyl-3-methylcyclohexane	2-methylhex-3-yne

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### Q9 (continued) The chemistry of carbon

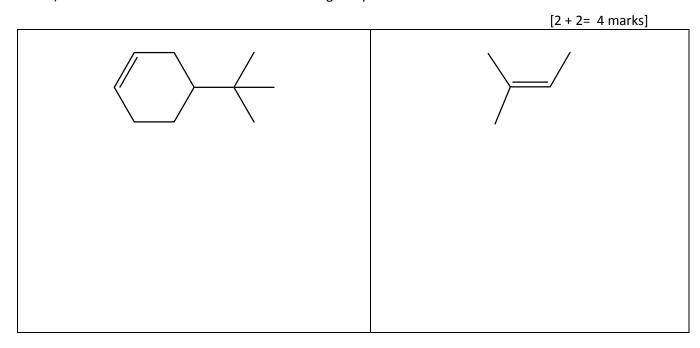
[3+1+6+4+6=20 marks]

c) ii) Write the structure for the following compound.

[2 marks]

	[Z marks]
1-bromo-2-chloro-4-ethylbenzene	

d) Write the names for each of the following compounds.



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#### Q9 (continued) The chemistry of carbon

[3+1+6+4+6=20 marks]

e) i) Complete the equations by predicting the major organic products formed in the following reactions.

Hint: You do not need to name the products.

[2 + 2 = 4 marks]

e) ii) Complete the equations below by predicting and naming the major organic product formed in the following reactions. Hint: you have to draw and name the major organic product.

[2 marks]