

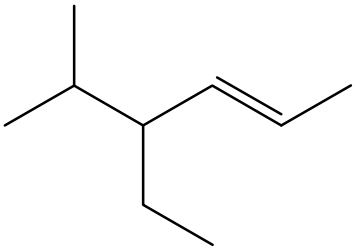
Diploma of Health Sciences
Diploma of Science
SLE155 Chemistry for the Professional Sciences

Q9 The chemistry of carbon

[3 + 1 + 6 + 4 + 6 = 20 marks]

- a) Write a condensed structural formula showing every atom but no bonds, and a molecular formula for the following alkene.
Hint: You do not need to name it!

[2 + 1 = 3 marks]

	Condensed formula
	<p>$(\text{CH}_3)_2\text{CHCH}(\text{CH}_2\text{CH}_3)\text{CH}=\text{CHCH}_3$ Or similar 2 marks ½ marks for minor error</p>
	Molecular formula
	<p>C_9H_{18} 1 mark, all or nothing</p>

- b) Arrange the following alkanes in order of **increasing** boiling point.
2-methylbutane, 2,2-dimethylpropane and pentane

All or nothing

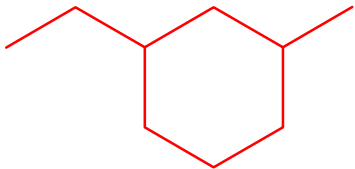
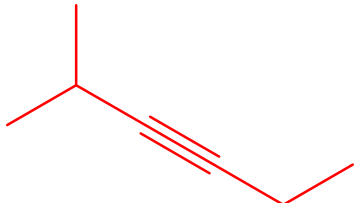
[1 mark]

2,2-dimethylpropane	<	2-methylbutane	<	pentane	1 mark
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- c) i) Write structures for the following compounds:

½ mark family, ½ mark chain length, ½ mark locant; ½ mark substituents

[2 + 2 = 4 marks]

<p>1-ethyl-3-methylcyclohexane</p>  <p>lowest numbering must be used, ethyl substituent is at C1</p>	<p>2-methylhex-3-yne</p>  <p>must be linear in the region of the triple bond</p>
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Q9 (continued) The chemistry of carbon

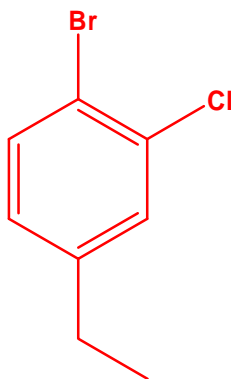
[3 + 1 + 6 + 4 + 6 = 20 marks]

c) ii) Write the structure for the following compound.

1 mark substituents , 1 mark positions

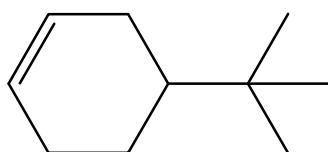
[2 marks]

1-bromo-2-chloro-4-ethylbenzene



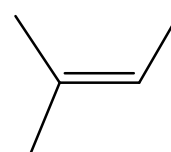
d) Write the names for each of the following compounds.

[2 + 2 = 4 marks]



4-tert-butylcyclohexene

1 mark cyclohexene, ½ mark substituent, ½ mark position



2-methylbut-2-ene

**1 mark butene, ½ mark position of double bond
½ mark substituent**

Lowest numbers must be used

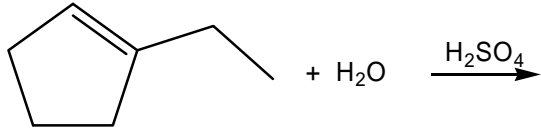
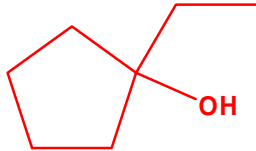
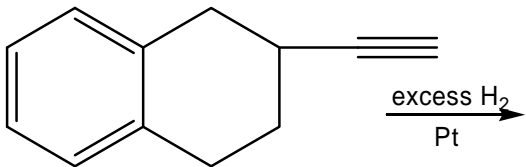
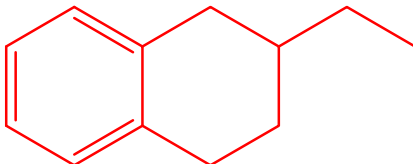
Q9 (continued) The chemistry of carbon

[3 + 1 + 6 + 4 + 6 = 20 marks]

- e) i) Complete the equations by predicting the major organic products formed in the following reactions.

Hint: You do not need to name the products.

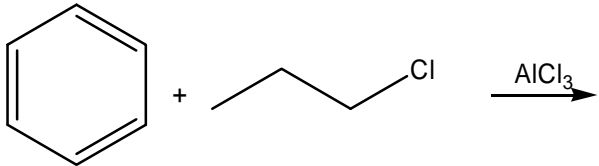
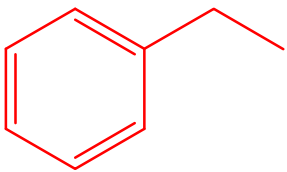
[2 + 2 = 4 marks]

 <p><chem>CC1=CCCC1.O>>CC1(O)CCCC1</chem></p>	 <p>2 marks if completely correct 1 mark for minor product</p>
 <p><chem>CC1=CC=CC2CCCCC1C#CC2>>CC1=CC=CC2CCCCC1CCC2</chem></p>	 <p>2 marks if completely correct Only 1 mark if aromatic ring is hydrogenated</p>

- e) ii) Complete the equations below by predicting and naming the major organic product formed in the following reactions. Hint: you have to draw and name the major organic product.

1 mark structure, 1 mark name

[2 marks]

 <p><chem>CC1=CC=CC=C1.CCCCl>>CC1=CC=CC=C1CCC</chem></p>	 <p>ethylbenzene + HCl</p>
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