Generalised Gradient Approximation

Elious

Lets recap some LDA

Assumption: electron density of inhomogenoues system is locally homogenous. The exchange-correlation energy is thus given by:

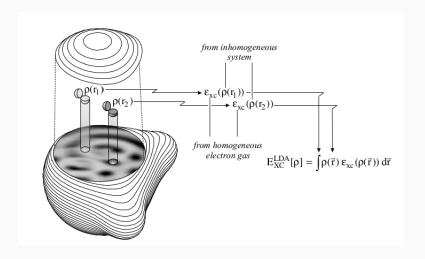
$$E_{xc}^{LDA} = \int \rho(\mathbf{r}) \epsilon_{xc}^{h}(\rho(\mathbf{r})) d\mathbf{r}$$
 (1)

and

$$E_{xc}^{LSDA} = \int \rho(\mathbf{r}) \epsilon_{xc}^{h}(\rho_{\uparrow}(\mathbf{r}), \rho_{\downarrow}(\mathbf{r})) d\mathbf{r}$$
 (2)

where $\epsilon_{xc}^h(\rho(\mathbf{r})) \to \text{exchange-correlation energy density at } \mathbf{r}$, evaluated by taking the E_{xc} of the uniform gas with density $\rho(\mathbf{r})$

Intuition for LDA



Taken from: W.Koch, M.C. Holthausen, Chemists Guide To DFT

Expectations



Figure 1: LDA should work fine

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Figure 2: LDA should be useless

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However, the energy details are not so good for the inhomohenous systems. Comparing with the experiments, the unsigned standard deviation in atomisation energies are:

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But, Why should the LDA even work at all for the inhomogenous system???

Lets find some holes in LDA

$$E_{xc}[\rho] = \frac{1}{2} \iint \frac{\rho(\mathbf{r}_1)\rho_{xc}(\mathbf{r}_1; \mathbf{r}_2)}{|\mathbf{r}_1 - \mathbf{r}_2|} d\mathbf{r}_1 d\mathbf{r}_2$$
 (5)

where

 $\rho(\mathbf{r}_1)d\mathbf{r}_1 \to \text{Probability density of finding an electron in } d\mathbf{r}_1 \text{ near } \mathbf{r}_1$ $\rho_{xc}(\mathbf{r}_1;\mathbf{r}_2)d\mathbf{r}_2 \to \text{Probability density of finding an electron in } d\mathbf{r}_2$ near \mathbf{r}_2 given there is an electron in $d\mathbf{r}_1$ near \mathbf{r}_1 . This is called the **exchange-correlation hole**. We can show

$$\int \rho_{xc}(\mathbf{r}_1; \mathbf{r}_2) d\mathbf{r}_2 = -1 \tag{6}$$

and

$$\rho_{xc}(\mathbf{r}_1; \mathbf{r}_2) \leqslant 0 \tag{7}$$

we now define:

$$\rho_{xc}(\textbf{r}_1;\textbf{r}_2) = \rho_x(\textbf{r}_1,\textbf{r}_2) + \rho_c(\textbf{r}_1,\textbf{r}_2) \tag{8} \label{eq:8}$$

The $\rho_{xc}(\mathbf{r}_1; \mathbf{r}_2)$ is factorised as the product:

$$\rho_{xc}(\mathbf{r}_1; \mathbf{r}_2) = \rho(\mathbf{r}_2) f_{xc}(\mathbf{r}_1; \mathbf{r}_2)$$
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here

$$-1 \leqslant f_{xc}(\mathbf{r}_1; \mathbf{r}_2) \leqslant 0 \tag{10}$$

Combining eqn(9 and 10) we can see that the factorisation is just another way of saying that there is **reduction of electron density** from \mathbf{r}_2 . The factor $f_{xc}(\mathbf{r}_1;\mathbf{r}_2)$ is called the pair-correlation factor.

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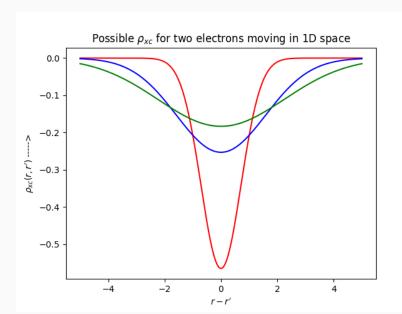
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- 1. The XC-hole is non-spherical as it depends on $\rho(\textbf{r}_2)$ which is usually non-uniform.
- 2. For LDA XC-hole is spherical as $\rho(\mathbf{r}_2)$ is uniform.

Intuition for XC-hole



Exchange hole

Exchange (Fermi) hole: We can show that for the Kohn-sham system $\rho_{xc}(\mathbf{r}_1;\mathbf{r}_2)$ arises due to *Pauli-repulsion* of same spin electrons and from there

$$\rho_{\mathsf{x}}(\mathbf{r}_1; \mathbf{r}_2) \leqslant 0 \tag{11}$$

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- 1. From eqn(11) the hole is negative everywhere \rightarrow an electron of a spin σ won't allow another electron of the same spin to occupy its orbital.
- 2. Also since its negative everywhere \rightarrow it must be responsible for the self-interaction correction.

We are now left with the **Correlation (Coulomb) hole** which mainly arises due to $1/r_{ij}$ nature of the coulomb repulsion

$$\int \rho_c(\mathbf{r}_1; \mathbf{r}_2) d\mathbf{r}_2 = 0 \tag{13}$$

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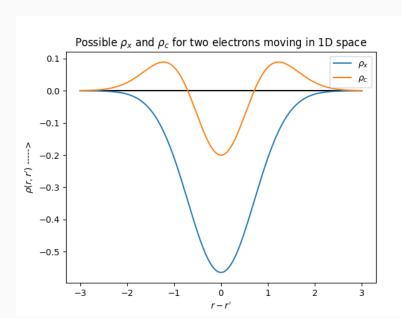
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- 2. Since this is due to $1/r_{ij}$, ρ_c will also be responsible for the electron-electron cusp in the many-electron wavefunction.

Intuition for X and C-hole



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Answer: LDA is for homogenous electron gas→tends to homogenise the properties for inhomogenous systems.Eg: It leads to overbinding in molecules.

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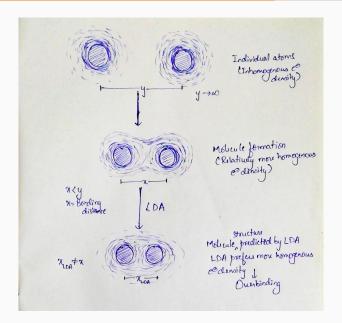
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- 6. Exchange energy of the molecule is too negative ightarrow Overbinding

Overbinding in LDA - the process



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So we can try:

$$\rho^{inh}(\mathbf{r}) \to \rho^{h}(\mathbf{r}) \left[1 + \nabla \rho^{h}(\mathbf{r}')|_{\mathbf{r}'=\mathbf{r}} + \mathcal{O}(\nabla^{2} \rho^{h}(\mathbf{r})) \right]$$
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and what's done in GE is:

$$E_{xc}^{GE}(\rho) \to \int \rho \epsilon_{xc}(\rho) d\mathbf{r} + \int C_{xc} \rho \frac{|\nabla \rho|}{\rho^{\frac{4}{3}}} + \dots$$
 (15)

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Due to the breaking of the above universal conditions of exact holes \rightarrow The relationship between on-top hole and its extension is lost \rightarrow the E^{GE}_{xc} (which represents the attraction between an electron density and its hole) will now have inconsistent behaviour.

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So we have

$$E_{xc}^{GE}[\rho] + XC_{properties} \rightarrow E_{xc}^{GGA}[\rho]$$
 (16)

where E_{xc}^{GGA} is the Generalised Gradient Approximation

$$E_{XC}^{GGA}[\rho] = \int \rho(\mathbf{r}) \varepsilon_{xc}^{h}(\rho(\mathbf{r})) (1 + \mu s^{2} + \mathcal{O}(s^{4})) d\mathbf{r}$$
 (17)

where μ is a parameter and s is the dimensionless quantity:

$$s = \frac{|\nabla \rho(\mathbf{r})|}{\rho^{\frac{4}{3}}(\mathbf{r})} \tag{18}_{18/2}$$

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LDA v/s GGA →Atomisation energies(in eV)

System	$\Delta E^{ m UHF}$	$\Delta E^{ m LSD}$	ΔE^{PW91}	$\Delta E^{ m PBE}$	$\Delta E^{ m expt}$
H_2	84	113	105	105	109
LiH	33	60	53	52	58
CH_4	328	462	421	420	419
NH_3	201	337	303	302	297
OH	68	124	110	110	107
H_2O	155	267	235	234	232
HF	97	162	143	142	141
Li ₂	3	23	20	19	24
LiF	89	153	137	136	139
Be_2	-7	13	10	10	3
C_2H_2	294	460	415	415	405
C_2H_4	428	633	573	571	563
HCN	199	361	326	326	312
CO	174	299	269	269	259
N_2	115	267	242	243	229
NO	53	199	171	172	153
O_2	33	175	143	144	121
F_2	-37	78	54	53	39
P_2	36	142	120	120	117
Cl ₂	17	81	64	63	58
Mean abs. error	71.2	31.4	8.0	7.9	

Taken from: *GGA made simple*, John P. Perdew, Kieron Burke, Matthias Ernzerhof, Phys. Rev. Lett. 78, 1396 (1997)

References

- CECAM summer school(2017) video lectures by Levy, Perdew, Kieron Burke https://www.youtube.com/channel/UCfLssAro7SMxgaeKTNFFeeA
- 2. ABC of DFT, Kieron Burke
- 3. A Chemist's Guide to Density Functional Theory, Wolfram Koch, Max C. Holthausen
- 4. Electronic Structure Calculations for Solids and Molecules-Theory and Computational Methods, Jorge Kohanoff
- GGA made simple John P. Perdew, Kieron Burke, Matthias Ernzerhof, Phys. Rev. Lett. 78, 1396 (1997)