

**KU LEUVEN**

**FACULTY OF SCIENCE**  
Department of Chemistry

# A very cool Title:

An even cooler subtitle

**DRAFT**

To remove, add ‘final’ to class options

**Mauro Gascón Navas**

Dissertation presented in partial fulfillment  
of the requirements for the degree of  
Erasmus Mundus Master of Science in  
Theoretical Chemistry and Computational  
Modelling

Supervisors:  
Prof. Dr. Thomas Jagau  
Prof. dr. ir. S. Leader

June 2025



# A VERY COOL TITLE:

AN EVEN COOLER SUBTITLE

Mauro GASCÓN NAVAS

Supervisors:

Prof. Dr. Thomas Jagau

Prof. dr. ir. S. Leader

Members of the

Examination Committee:

Prof. dr. ir. The Chairman, chair

Prof. dr. ir. The One

Prof. dr. ir. The Other

Prof. dr. External Jurymember

(Far Away)

Dissertation presented in partial  
fulfillment of the requirements for the  
degree of Erasmus Mundus Master  
of Science in Theoretical Chemistry  
and Computational Modelling

June 2025

© 2025 Mauro Gascón Navas  
Uitgegeven in eigen beheer, Mauro Gascón Navas, Leuven (Belgium)

Alle rechten voorbehouden. Niets uit deze uitgave mag worden vermenigvuldigd en/of openbaar gemaakt worden door middel van druk, fotokopie, microfilm, elektronisch of op welke andere wijze ook zonder voorafgaande schriftelijke toestemming van de uitgever.

All rights reserved. No part of the publication may be reproduced in any form by print, photoprint, microfilm, electronic or any other means without written permission from the publisher.

# Acknowledgements

...

Thank you thank you.

## **Instructions by the Arenberg Doctoral School:**

The scientific abstract should present the most important aims and conclusions of the dissertation in a brief text of ca. 2 pages.



# Abstract

...

## **Instructions by the Arenberg Doctoral School:**

The scientific abstract should present the most important aims and conclusions of the dissertation in a brief text of ca. 2 pages.





# Beknopte samenvatting

...aaa

## **Instructions by the Arenberg Doctoral School:**

The scientific abstract should present the most important aims and conclusions of the dissertation in a brief text of ca. 2 pages.



# List of Abbreviations

**MD** molecular dynamics. 3



# List of Symbols

$\Theta$       A nice symbol



# Contents

<b>Acknowledgements</b>	<b>i</b>
<b>Abstract</b>	<b>iii</b>
<b>Beknopte samenvatting</b>	<b>v</b>
<b>List of Abbreviations</b>	<b>vii</b>
<b>List of Symbols</b>	<b>ix</b>
<b>Contents</b>	<b>xi</b>
<b>List of Figures</b>	<b>xiii</b>
<b>List of Tables</b>	<b>xv</b>
<b>1 Introduction</b>	<b>3</b>
1.1 Biological Quinones . . . . .	3
1.2 Exotic Anions . . . . .	3
<b>2 Theoretical Background</b>	<b>5</b>
2.1 Self Consistent Field Methods . . . . .	5
2.1.1 Electron Correlation . . . . .	6
2.1.2 Coupled Cluster Theory . . . . .	8
2.1.3 Second Approximate Coupled Cluster . . . . .	9
2.2 Equation-of-Motion Methods . . . . .	10
2.2.1 Dyson orbitals . . . . .	11
<b>3 Computational Methods</b>	<b>13</b>
<b>4 Results and Discussion</b>	<b>15</b>

<b>5 This is conclusion</b>	<b>17</b>
<b>6 Manual</b>	<b>19</b>
6.1 Tips and Tricks . . . . .	19
6.1.1 Joint PhD Dissertation . . . . .	19
6.1.2 Image on the cover page . . . . .	19
6.1.3 Table of contents . . . . .	20
6.1.4 Small ebook size . . . . .	20
6.2 Settings for TeXstudio . . . . .	20
6.2.1 Support for TeXstudio structure panel . . . . .	20
6.2.2 Custom <i>makeindex</i> and <i>makeglossaries</i> commands . . .	21
6.2.3 Custom Build&View and Compile meta-commands . . .	21
6.3 Full cover page . . . . .	22
<b>A This is myappendix</b>	<b>23</b>
<b>Bibliography</b>	<b>25</b>



# List of Figures

1.1	Short caption for Table of Figures . . . . .	4
2.1	EOM-EA. . . . .	11



# List of Tables

2.1	Computational Scaling of quantum chemistry Methods. . . . .	9
-----	---	---



LIST OF TABLES ..... 1

**Instructions by the Arenberg Doctoral School:**

Report and discussion of the research in different chapters: These chapters, reporting and discussing research results, can be based on text that has already been published or accepted or submitted to journals or conferences. In that case, the bibliographic reference of the publication should be mentioned on the first page of the chapter. If there are multiple authors, the PhD researcher must clarify the own scientific contribution after that bibliographic reference on the first page of the chapter. There is no problem in editing or rewriting a text that has already been published or accepted for publication, for example to reach consistency and coherence in writing style and formatting, to add details that were left out from publication, to meet comments of the Examination Committee, etc; The doctoral candidate must determine whether the publisher allows public availability of the publications and in which form via the webpage Romeo/Sherpa.

Consistency in layout is required for the entire manuscript! References to figures, tables, appendices, and similar structures need to be consistent.



# Chapter 1

## Introduction

### Instructions by the Arenberg Doctoral School:

An in-depth introduction outlining the research in a broader context: Starting from a description of the state of the art in the domain, additionally, the research questions and objectives are formulated. Furthermore, this includes the global approach and research methods.

### 1.1 Biological Quinones

### 1.2 Exotic Anions

Illustration of how to include citations [2] and [3]. Lorem ipsum dolor sit amet, consetetur sadipscing elitr, sed diam nonumy eirmod tempor invidunt ut labore et dolore magna aliquyam erat, sed diam voluptua. At vero eos et accusam et justo duo dolores et ea rebum. Stet clita kasd gubergren, no sea takimata sanctus est Lorem ipsum dolor sit amet.

And yet another citation [1].

Introducing some symbol:  $\Theta$ .

Introducing an acronym: MD.

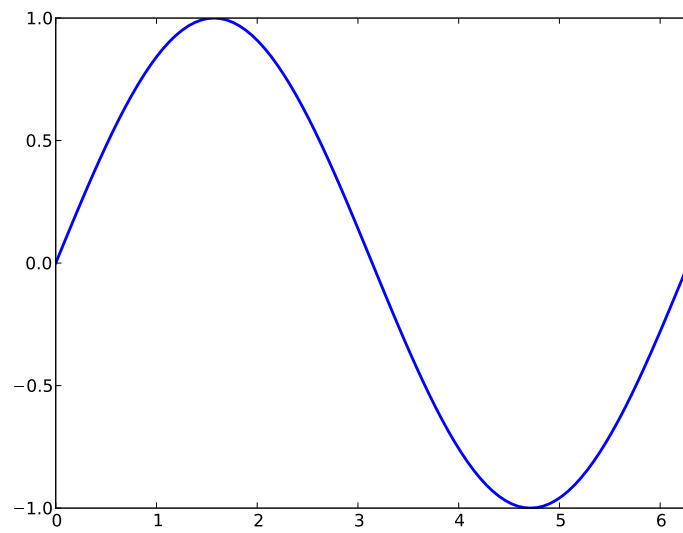


Figure 1.1: Illustration of how to include a figure (long text, should not go to Table of Figures).



## Chapter 2

# Theoretical Background

### 2.1 Self Consistent Field Methods

The Hartree-Fock (HF) method stands as a method in electronic structure theory. Its primary objective is to provide an approximate solution to the many-electron time-independent Schrödinger equation within the Born-Openheimer approximation, which governs the behavior of electrons within atoms and molecules:

$$\hat{H}_e \Psi_e = E_e \Psi_e \quad (2.1)$$

The HF method achieves this by assuming that each electron moves independently within an average electrostatic field generated by all the other electrons in the system. In the HF method the  $N$ -electron wavefunction is represented by a Slater determinant, which is formed by taking the antisymmetrized product of  $N$  individual one-electron wavefunctions, the spin-orbitals ( $\chi$ ):

$$\Psi(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N) = \frac{1}{\sqrt{N!}} \begin{vmatrix} \chi_1(\mathbf{r}_1) & \chi_2(\mathbf{r}_1) & \cdots & \chi_N(\mathbf{r}_1) \\ \chi_1(\mathbf{r}_2) & \chi_2(\mathbf{r}_2) & \cdots & \chi_N(\mathbf{r}_2) \\ \vdots & \vdots & \ddots & \vdots \\ \chi_1(\mathbf{r}_N) & \chi_2(\mathbf{r}_N) & \cdots & \chi_N(\mathbf{r}_N) \end{vmatrix} \quad (2.2)$$

The determinantal inherently satisfies both the Pauli exclusion principle, and the antisymmetry requirement of fermions. The energy expectation for a Slater

determinant according to HF is variational and can be computed as:

$$\begin{aligned}
 E_{HF} &= \sum_{i=1}^N \hat{F}_i \Psi \\
 &= \sum_{i=1}^N \hat{h}(i) + \sum_{i,j=1}^N (\hat{J}_j(i) - \hat{K}_j(i)) \\
 &= \sum_{i=1}^N \langle \chi_i | \hat{h} | \chi_i \rangle + \frac{1}{2} \sum_{i,j=1}^N \langle \chi_i \chi_j | | \chi_i \chi_j \rangle
 \end{aligned} \tag{2.3}$$

Where,  $\hat{F}$  is the Fock operator.  $\hat{F}$  is made up from  $\hat{h}$ , the one-electron core Hamiltonian operator (kinetic energy and electron-nucleus attraction);  $\hat{J}_j(i)$ , the Coulomb operator, describing the electrostatic repulsion between electron  $i$  and the average charge distribution of electron  $j$ , and  $\hat{K}_j(i)$  is the exchange operator, a purely quantum mechanical term arising from the antisymmetry principle.

The Hartree-Fock equations are inherently non-linear because the Fock operator depends on the wavefunctions of all the other electrons, their interactions are coupled. Consequently, these equations cannot be solved analytically and necessitate an iterative procedure known as the Self-Consistent Field (SCF) method.

### 2.1.1 Electron Correlation

The Hartree-Fock (HF) method is inherently limited by its neglect of the instantaneous correlation between the motions of electrons. In the HF approximation, each electron is treated as moving independently within a static, average field created by all other electrons. This mean-field approach fails to account for the fact that electrons, being negatively charged, will instantaneously repel each other, leading to a dynamic correlation in their movements as they try to avoid each other in space.

This neglect leads to an overestimation of the electron-electron repulsion energy, and is responsible for its inability to accurately predict certain phenomena, such as London dispersion forces. The difference between the exact non-relativistic energy of the system and the energy obtained in the Hartree-Fock limit is defined as the correlation energy and is always negative due to the variational principle. Correlated methods aim to include the effects of the instantaneous interactions between electrons that are neglected in the mean-field approximation of HF theory. Several correlated methods used during this work are:

## Møller-Plesset Perturbation Theory

Møller-Plesset (MP) perturbation theory offers a systematic way to improve upon the HF energy by treating the electron correlation as a perturbation to the Hartree-Fock Hamiltonian. The Fock operator is taken as the zeroth-order Hamiltonian, and the difference between the exact electron-electron repulsion and the Fock operator is considered the perturbation. The energy and wavefunction are then expanded as a series in terms of the perturbation strength. The first-order energy correction in MP theory is zero, so the first non-trivial correction to the HF energy appears at the second order, giving rise to the MP2 method. The MP2 energy correction for a closed-shell molecule is given by:

$$E_{MP2} = -\frac{1}{4} \sum_{ij}^{occ} \sum_{ab}^{virt} \frac{|\langle ij || ab \rangle|^2}{\epsilon_a + \epsilon_b - \epsilon_i - \epsilon_j} \quad (2.4)$$

Where  $i, j$  denote occupied molecular orbitals,  $a, b$  denote virtual molecular orbitals, and  $\epsilon$  are the corresponding orbital energies from the HF calculation. MP theory can be extended to higher orders (MP3, MP4, etc.) to achieve greater accuracy, although the computational cost increases significantly with each order.

## Density Functional Theory

Density Functional Theory (DFT) provides an alternative approach to incorporating electron correlation by focusing on the electron density of the system rather than the wavefunction, reducing the degrees of freedom of the system from  $3N - 3$  to just 3. The fundamental principle of DFT is that the ground state energy of a system is a unique functional of its electron density:

$$\left( -\frac{1}{2} \nabla^2 + V_{ext}(\mathbf{r}) + V_H(\mathbf{r}) + V_{XC}[\rho(\mathbf{r})] \right) \psi_i(\mathbf{r}) = \epsilon_i \psi_i(\mathbf{r}) \quad (2.5)$$

Where  $V_{ext}$  represents the external potential,  $V_H(\mathbf{r}) = \int \frac{\rho(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}'$  is the hartree potential,  $V_{XC}$  is the Exchange-Correlation potential and  $\rho(\mathbf{r})$  is the electron density. The exchange-correlation functional is the most challenging part of DFT, as it is not known exactly and must be approximated. The accuracy of DFT calculations depends heavily on the choice of exchange-correlation functional.

## Configuration Interaction

Configuration Interaction (CI) methods improve upon HF by expressing the electronic wavefunction as a linear combination of the HF ground state determinant and excited state determinants:

$$|\Psi_{CI}\rangle = c_0|\Phi_0\rangle + \sum_{ia} c_{ia}|\Phi_{ia}\rangle + \sum_{ijab} c_{ijab}|\Phi_{ijab}\rangle + \dots \quad (2.6)$$

Where  $|\Phi_0\rangle$  is the HF ground state determinant,  $|\Phi_{ia}\rangle$  represents a wavefunction with a hole in spin-orbital  $i$  and a particle in the spin-orbital  $a$ , and  $c$  are the CI coefficients. Full CI, which includes all possible excited determinants, is exact within the basis set but computationally prohibitive for all but the simplest systems. Truncated CI methods, such as CIS (singles) and CISD (singles and doubles), are more practical but lack size extensivity.

### 2.1.2 Coupled Cluster Theory

The coupled cluster (CC) theory is considered the gold-standard method in quantum chemistry. Similarly to CI, the CC method expands the wavefunction as a linear combination of Slater determinants. However, the CC method results into a size-extensive and size-consistent wavefunction by using an exponential ansatz.

$$|\Psi_{CC}\rangle = e^{\hat{T}}|\Psi_0\rangle \quad (2.7)$$

Where  $\hat{T}$  is the cluster operator, which is the central component of CC theory and is defined as a sum of excitation operators:

$$\hat{T} = \hat{T}_1 + \hat{T}_2 + \hat{T}_3 + \dots + \hat{T}_N \quad (2.8)$$

where  $N$  is the total number of electrons in the system. Each term in this sum corresponds to a specific level of excitation:

- $\hat{T}_1 = \sum_i^{\text{occ}} \sum_a^{\text{virt}} t_i^a a_a^\dagger a_i$  represents single excitations.
- $\hat{T}_2 = \frac{1}{4} \sum_{i,j}^{\text{occ}} \sum_{a,b}^{\text{virt}} t_{ij}^{ab} a_a^\dagger a_b^\dagger a_j a_i$  represents double excitations.
- Higher-order excitation operators  $\hat{T}_3, \hat{T}_4, \dots$  describe simultaneous excitation of three, four, and more electrons.

The coefficients  $t_i^a, t_{ij}^{ab}$ , etc., are cluster amplitudes to be determined by solving the coupled cluster Schrödinger equation. The energy of the system is

obtained by projecting the Schrödinger equation onto the Hartree-Fock reference determinant:

$$E_{CC} = \langle \Psi_0 | e^{-\hat{T}} \hat{H} e^{\hat{T}} | \Psi_0 \rangle \quad (2.9)$$

Using the Baker-Campbell-Hausdorff (BCH) expansion, the exponential operators in Eq. 2.9 can be simplified to a series of commutators which ends at the fourth order. The cluster operator  $\hat{T}$  can be truncated at different levels of excitation:

- **CCD** (Coupled Cluster Doubles): This is the simplest approximation in the CC family, where the cluster operator is truncated to include only single excitations:  $\hat{T} \approx \hat{T}_2$ . Due to the Brilluin’s theorem, the amplitudes of single excitations are 0.
- **CCSD** (Coupled Cluster Singles and Doubles): This is one of the most widely used and generally accurate coupled cluster methods, where the cluster operator includes both single and double excitations:  $\hat{T} \approx \hat{T}_1 + \hat{T}_2$ .
- **CCSDT** (Coupled Cluster Singles, Doubles, and Triples):  $\hat{T} \approx \hat{T}_1 + \hat{T}_2 + \hat{T}_3$ . The hierarchy can be extended to include even higher levels of excitation, converging to the Full Configuration Interaction (Full CI) limit. Full CI includes all possible excitations within a given one-electron basis set and represents the exact solution to the non-relativistic Schrödinger equation in that basis.

Method	Operation count	Memory
HF	$O(N^4)$	$O(N^4)$
MP2	$O(N^5)$	$O(N^4)$
CCD/CCSD	$O(N^6)$	$O(N^4)$
CCSDT	$O(N^8)$	$O(N^6)$
CC2	$O(N^5)$	$O(N^4)$

Table 2.1: Computational Scaling of quantum chemistry Methods.

### 2.1.3 Second Approximate Coupled Cluster

Second Approximate Coupled Cluster (CC2) belongs to the broader family of CCn approximate coupled cluster methods, where the ‘n’ in CCn indicates the level of approximation within a perturbative hierarchy. These methods aim to reduce the computational cost associated with standard CC truncations while still retaining a reasonable level of accuracy.

In CC2 the equations for the single amplitudes ( $t_i^a$ ) are the same as those in CCSD, while the doubles amplitudes ( $t_{ij}^{ab}$ ) are calculated using the non-iterative expression for MP2 (Eq 2.4). The resulting expression for the CC2 correlation energy is:

$$E_{CC2} = \sum_{ij} \sum_{ab} \frac{1}{4} \frac{|\langle ij || ab \rangle|^2}{\epsilon_a + \epsilon_b - \epsilon_i - \epsilon_j} + \sum_i \sum_a \hat{F}_{ai} t_i^a \quad (2.10)$$

The perturbative treatment of the doubles amplitudes in CC2, reduces the computational cost compared to CCSD, Table 2.1. While this approximation can lead to a less accurate description of electron correlation, the inclusion of zeroth-order singles amplitudes allows for an approximate description of orbital relaxation, which often leads to higher quality results compared to MP2.

## 2.2 Equation-of-Motion Methods

Equation-of-Motion Coupled Cluster (EOM-CC) methods are an extension of ground-state coupled cluster theory which provide a framework for calculating a variety of excited (EE), ionized (IP) and electron-attached (EA) states. In the EOM-CC, the target electronic state is described by applying a linear excitation operator  $\hat{R}$  to a reference state, which typically is the coupled cluster wavefunction of the ground state. The target state wavefunction can then be expressed as  $|\Psi_{\text{EOM}}\rangle = \hat{R}|\Psi_{\text{CC}}\rangle = \hat{R}e^{\hat{T}}|\Phi_{\text{HF}}\rangle$ .

The form of the operator  $\hat{R}$  is similar to the cluster operator and chosen to access the desired target state. In the case of EOM-EA, the electron attachment operator  $R_{EA}$  includes terms that describe the addition of one electron to an unoccupied orbital (a one-particle creation operator), terms that describe the addition of one electron accompanied by the excitation of another electron from an occupied to an unoccupied orbital (a two-particle and one-hole creation operator), and so on:

$$\hat{R}_{EA} = \hat{R}_{1EA} + \hat{R}_{2EA} + \dots = \sum_a r^a a_a^\dagger + \frac{1}{2} \sum_{ab} \sum_i r_i^{ba} a_b^\dagger a_a^\dagger a_i + \dots \quad (2.11)$$

Where  $a$  and  $b$  denote virtual orbitals,  $i$  denotes an occupied orbital, and  $r^a$  and  $r_i^{ba}$  are the coefficients (amplitudes) to be determined. The electron affinities (EAs) of the system are then obtained as the eigenvalues of the similarity-transformed Hamiltonian:

$$\bar{H}_N \bar{R}|\Psi_0\rangle = \Delta E_{\text{EOM}} \bar{R}|\Psi_0\rangle \quad (2.12)$$

$$\bar{H}_N = e^{-T} H e^T - \langle \Psi_0 | e^{-T} H e^T | \Psi_0 \rangle \quad (2.13)$$

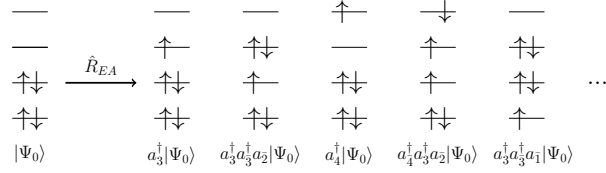


Figure 2.1: EOM-EA.

### 2.2.1 Dyson orbitals

Dyson orbitals are defined as the overlap between the wavefunction of an initial  $N$ -electron state ( $|\Psi_0^N\rangle$ ) and the wavefunction of the final state with  $N \pm 1$  electrons ( $|\Psi_f^{N\pm 1}\rangle$ ).

$$\phi_d = \langle \Psi_0^N | \Psi_f^{N\pm 1} \rangle = \langle \Psi_0^N | \hat{R} \Psi_0^N \rangle \quad (2.14)$$

Because the the terms differ in one electron, the result of the overlap is a vector instead of a scalar, and can be expressed as a linear combination of the molecular orbitals ( $\phi_p(r)$ ) of the reference wavefunction:

$$\phi_{Dyson}(r) = \sum_p \gamma_p \phi_p(r) \quad (2.15)$$

where  $\gamma_p$  are the coefficients that quantify the contribution of each molecular orbital to the Dyson orbital.

Physically, Dyson orbitals can be interpreted as representing the correlated state of the electron that is either removed or added to it. They can be used for the interpretation and prediction of photoelectron spectra as they contain all the information required to calculate differential corss-sections,  $\frac{d\sigma}{d\Omega_k}$ :

$$\frac{d\sigma}{d\Omega_k} = \frac{4\pi^2 k E}{c} |\langle \phi_d | \mu | \Psi_k^{el} \rangle|^2 \quad (2.16)$$

Where where  $k$  is the magnitude of the photoelectron wavevector,  $E$  is the energy of the ionizing radiation, and  $c$  is the speed of light,  $\mu$  is the dipole operator, and  $\Psi_k^{el}$  is the photoelectron wavefunction.





## Chapter 3

# Computational Methods

...

This are my methods

### **Instructions by the Arenberg Doctoral School:**

Appendices: The appendices should include parts of the research which are essential for the work, but which may hamper the readability of the text, e.g. because of their length (mathematical deductions, experimental data, examples, figures, etc.).



## **Chapter 4**

# **Results and Discussion**

...



## Chapter 5

# This is conclusion

...

### **Instructions by the Arenberg Doctoral School:**

An extensive conclusion, including a global discussion of the research results, a discussion of the implications of the PhD research and future perspectives in regards to follow-up research.



## Chapter 6

# Manual

### 6.1 Tips and Tricks

#### 6.1.1 Joint PhD Dissertation

Add the `joint` option to the document class, change the `PARTNER_LOGO.eps` file and fill in the commands such as `\facultypartner` in `thesis.tex`.

#### 6.1.2 Image on the cover page

If you want to place an image on the cover of the dissertation, you can add the code underneath to the template (check with your promotor whether this is allowed).

**Include image:** Search for the `\frontcoverheaderXXIV` command in the `adsphd.cls` file and add the following lines:

```
\begin{textblock*}{56mm}(10mm+#1,15mm)
\includegraphics[width=56mm,height=20mm]{image/filename}
\end{textblock*}
```

Where 56mm is the width, 20mm the height, 10mm the x-location and 15mm the y-location.

**Change cover font color:** Add the command `\color{red}` to the `\frontcoverheaderXXIV` command or enclose specific parts. For example, `{\color{red}\textbf{\@authorf\@author1}}`.

### 6.1.3 Table of contents

To remove list of figures, tables and other preface chapters from the table of contents, search for occurrences of `\addcontentsline` in the file `adsphd.cls` and comment them.

### 6.1.4 Small ebook size

When you add the `epub` option to the `adsphd` class the dissertation is printed to a smaller size to read on a device such as Kindle.

Environments such as tables or `tikZ` pictures are often sized in absolute values and not relative to the size of the output. You can wrap them in a `resizebox` to enforce scaling:

```
\resizebox{\textwidth}{!}{%
  \begin{tabular}{cc}
    ...
  \end{tabular}
}
```

## 6.2 Settings for TeXstudio

If you are working with TeXstudio or other windows latex editors you might want to adjust the editor’s settings to allow a proper compilation of the table of contents and list of figures/tables.

### 6.2.1 Support for TeXstudio structure panel

The chapters do not show up in the TeXstudio structure panel because the `\includechapter` is not recognized. You can replace this command with the following two lines in `thesis.tex` (replace `manual` with the chapter name):

```
% \includechapter{manual}
```



```
\graphicspath{{chapters/manual/image/}}%
\include{chapters/manual/manual}%
```

## 6.2.2 Custom *makeindex* and *makeglossaries* commands

According to the *README.md* the tables are indexed through two custom commands. To edit them in TeXstudio open the *Commands* settings (*Options* → *Configure TeXstudio...*, *Commands* sheet), edit the following fields and press OK.

*Makeindex*:

```
"C:/Program Files/MiKTeX 2.9/miktex/bin/x64/makeindex.exe" %.nlo -s nomencl.ist -o %.nls
```

*Makeglossaries*:

```
"C:/Program Files/MiKTeX 2.9/miktex/bin/x64/makeindex.exe" %.glo -s %.ist -t %.glg -o %.gls
```

Now the customized commands can be launched by using *Tools* → *Commands* → *MakeIndex/Makeglossaries*. If you want to automatize it in the standard *Build & View* (F5) and *Compile* (F6) commands look at the following section.

## 6.2.3 Custom Build&View and Compile meta-commands

Open *Options* → *Configure TeXstudio...*, *Build* sheet, edit the following field and press OK.

*Build & View*:

```
txs:///pdflatex | txs:///bibtex | txs:///makeglossaries | txs:///makeindex |
txs:///pdflatex | txs:///pdflatex
```

To view the PDF once created you have to press F7 (or *Tool* → *View*) and the PDF will automatically update in the default viewer when you modify it.

If you prefer to directly view the created PDF **from the beginning** edit the field as follow:

```
txs:///pdflatex | txs:///bibtex | txs:///makeglossaries | txs:///makeindex |
txs:///pdflatex | txs:///pdflatex | txs:///view-pdf
```

## 6.3 Full cover page

**Important:** most printing services will create their own cover page based on the details you send them (title, name, affiliation, ...) and do not supply you with all necessary parameters (e.g., thickness of the paper) because these differ from machine to machine. Therefore, the generated cover page is only indicative and probably not used by your printing server (or even correct).

A full cover page (combining front cover, spine and back cover) can be generated automatically using the command `make cover` or `python3 run.py cover`. This creates a pdf `$(COVERPDF)`; by default this is `cover.pdf`.

The width of the spine is set by redefining `adsphdspinewidth` (9mm by default).

It can be seen in the provided `thesis.tex` that all information necessary to generate a cover page is contained between two markers

```
%%% COVER: Settings %%%
...
%%% COVER: End settings %%%
```

DO NOT REMOVE THESE!! They are used by the Makefile!!

The default front and/or back cover page can be overwritten:

- create a file `mycoverpage.tex`
- redefine the commands `\makefrontcovergeneral` and `\makebackcovergeneral`. For an example and more information, see the provided file `mycoverpage.tex`.

The cover page in the generated pdf has the following structure:

```
<--rbleed--><--backcoverpage--><--lbleed--><--spine width--><--lbleed--><--frontcoverpage--><--rbleed-->
```

The default bleed (both lbleed and rbleed) is 7mm. I suggest not changing this value unless you know what you are doing ;) The latter can be done by redefining `\defaultlbleed` and `\defaultlbleed` respectively.

## Appendix A

# This is myappendix

...

### Instructions by the Arenberg Doctoral School:

Appendices: The appendices should include parts of the research which are essential for the work, but which may hamper the readability of the text, e.g. because of their length (mathematical deductions, experimental data, examples, figures, etc.).



# Bibliography

- [1] FREDERIX, Y., AND ROOSE, D. A drift-filtered approach to diffusion estimation for multiscale processes. In *Coping with complexity: model reduction and data analysis* (2010), vol. 75 of *Lecture Notes in Computational Science and Engineering*, Springer-Verlag.
- [2] MEERT, W. *Inference and Learning for Directed Probabilistic Logic Models*. PhD thesis, Informatics Section, Department of Computer Science, Faculty of Engineering, Mar. 2011. Blockeel, Hendrik (supervisor).
- [3] VAN DEN BROECK, G., TAGHIPOUR, N., MEERT, W., DAVIS, J., AND DE RAEDT, L. Lifted probabilistic inference by first-order knowledge compilation. In *Proceedings of the 22th International Joint Conference on Artificial Intelligence (IJCAI)* (2011).

## Instructions by the Arenberg Doctoral School:

Bibliography: Should be arranged according to the guidelines generally accepted in the relevant research domain.

The thesis needs to be consistent in relation to bibliographic and other references. Either a global list of bibliographic references is provided at the end of the thesis, or each chapter contains bibliographic references at the end, but there cannot be a combination of these two methods of referencing.

# Statement on the use of Generative AI

## Instructions by the Arenberg Doctoral School:

Read the guidelines in relation to GenAI at KU Leuven and add the ‘statement on the use of Generative AI’ in your manuscript.

Uncomment the appropriate sentences in the `\useOfGenAI` command and expand the text where needed to make it more specific and add topics if they are not covered by any of the indicated topics.

The text, code, and images in this thesis are my own (unless otherwise specified). Generative AI has only been used in accordance with the KU Leuven guidelines and appropriate references have been added. I have reviewed and edited the content as needed and I take full responsibility for the content of the thesis.

# List of publications

Input file `chapters/publications/publications.tex` does not exist. Make sure it starts with “`\chapter{List of publications}`”. To not include this chapter in the table of contents, use the starred version of the `\chapter` command. . .

## Instructions by the Arenberg Doctoral School:

List of scientific publications with a clear DOI (digital object identifier) number mentioned:(optional): The list of scientific publications by the doctoral researcher should be arranged according to the guidelines generally accepted in the relevant research domain.







DEPARTMENT OF CHEMISTRY  
DEPARTMENT OF CHEMISTRY  
Celestijnenlaan 200A box 2402  
B-3001 Leuven  
first.name@dept.kuleuven.be  
<https://chem.kuleuven.be/en/research/qcpc/tue/>

