

Problem 1 A logistic difference equation is given by

$$f(x_n) = rx_n(1 - x_n)$$

a) Expand and factor $f(f(x)) = 0$.

Starting at the end goal and working our way backward, we need to show that $f(f(x)) = rx(rx - r + 1)(rx^2 - (1 + r)x + \frac{1+r}{r}) = 0$. If we expand all the terms and use polynomial division we have

$$\begin{aligned} r^3x^4 + r^2x^2 + r^3x^2 - 2r^3x^3 + x - r^2x &= 0 \\ r^2x(rx^3 + x + rx - 2rx^2 - 1) + x & \\ r^2x(1 - x)(-rx^2 + rx - 1) - x & \\ rx(r - rx)(rx^2 - rx + 1) - x = 0 = f(f(x)) &\square \end{aligned}$$

b) Let $x_1 = \frac{1+r+\sqrt{(1+r)(r-3)}}{2r}$ and $x_2 = \frac{1+r-\sqrt{(1+r)(r-3)}}{2r}$. Show that $f(x_1) = x_2$ and $f(x_2) = x_1$.

Let $f(x_n) = rx_n - rx_n^2$.

$$\begin{aligned} f(x_{1,2}) &= r\left(\frac{1+r\pm\sqrt{(1+r)(r-3)}}{2r}\right) - r\left(\frac{1+r\pm\sqrt{(1+r)(r-3)}}{2r}\right)^2 \\ &= \frac{1+r\pm\sqrt{(1+r)(r-3)}}{2} - \frac{(1+r)^2 \pm 2(1+r)\sqrt{(1+r)(r-3)} \pm (1+r)(r-3)}{4r} \\ &= \frac{2r+2r^2 \pm 2r\sqrt{(1+r)(r-3)}}{4r} - \frac{1+2r+r^2 \pm (2+2r)\sqrt{(1+r)(r-3)} \pm (1+r)(r-3)}{4r} \\ &= \frac{-1+r^2 \pm 2\sqrt{(1+r)(r-3)} \pm (3+r^2-2r)}{4r} \\ &= \frac{1+r\pm\sqrt{(1+r)(r-3)}}{2r} = x_{1,2} \end{aligned}$$

c) Show that the derivative $\frac{d}{dx}f(f(x)) = f'(x_1)f'(x_2)$.

To take the derivative of $f(f(x))$, we first use chain rule:

$$\begin{aligned} \frac{d}{dx}f(f(x)) &= f'(x)f'(f(x)) \\ f'(x_{1,2})f'(f(x_{1,2})) &= f'(x_1)f'(x_2) \end{aligned}$$

Assuming $f(x_1) = x_2$ and $f(x_2) = x_1$ from part b). To find the $r > 3$ at which the period 2 orbit becomes unstable, we solve

$$\begin{aligned} |f'(x_1)f'(x_2)| &= 1 \\ f'(x_1)f'(x_2) &= (r - (1 + r + \sqrt{(r+1)(r-3)}))(r - (1 + r - \sqrt{(r+1)(r-3)})) \\ &\quad (-1 - \sqrt{(r+1)(r-3)})(-1 + \sqrt{(r+1)(r-3)}) \\ |1 - (r+1)(r-3)| &= 1 \end{aligned}$$

The quantity above is equal to 1 when $(r + 1)(r - 3) = 0$ and when $(r + 1)(r - 3) = 2$. When $(r + 1)(r - 3) = 0$ we find $r_1 = -1$ and $r_2 = 3$. When $(r + 1)(r - 3) = 2$ we have $r_3 = 1 + \sqrt{6}$ and $r_4 = 1 - \sqrt{6}$. Since $r_3 = 1 + \sqrt{6}$ is the only value greater than 3, the period 2 orbit will become unstable when $r = 1 + \sqrt{6} \approx 3.45$.

Problem 2 A fishery gives a model for the population size versus the number of offspring that survive as

$$N_{n+1} = \frac{RN_n}{1 + [\frac{R-1}{K}]N_n}$$

a) Find the equilibria and determine their stability.

To find the equilibria, we need to find the points N^* at which $N^* = f(N^*)$. The points that satisfy these conditions are $N^* = 0$ and $N^* = K$. To determine their stability, we first need to compute

$$\begin{aligned} f'(N) &= \frac{(1 + [\frac{R-1}{K}]N_n)R - RN_n(\frac{R-1}{K})}{(1 + [\frac{R-1}{K}]N_n)^2} \\ f'(0) &= \frac{R}{1} = R > 1 \\ f'(K) &= \frac{R^2 - R^2 + R}{R^2} = \frac{1}{R} < 1 \end{aligned}$$

By the stability tests, the point $N^* = 0$ is unstable and the point $N^* = K$ is stable.

b) Find an exact, closed form solution by using the substitution $X_n = \frac{1}{N_n}$.

Using the substitution $X_n = \frac{1}{N_n}$ into the original equation, we have

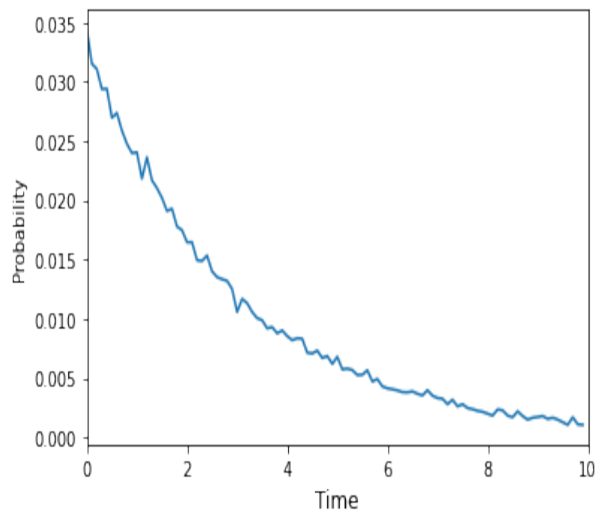
$$\begin{aligned} \frac{1}{X_{n+1}} &= \frac{1 + \frac{R-1}{KX_n}}{R} \cdot \frac{1}{X_n} \\ \frac{1}{X_{n+1}} &= \frac{X_n}{R} + \frac{1}{K} - \frac{1}{RK} \end{aligned}$$

To begin to solve this iteratively, we will use the substitution $B = \frac{1}{K} - \frac{1}{RK}$ for simplicity:

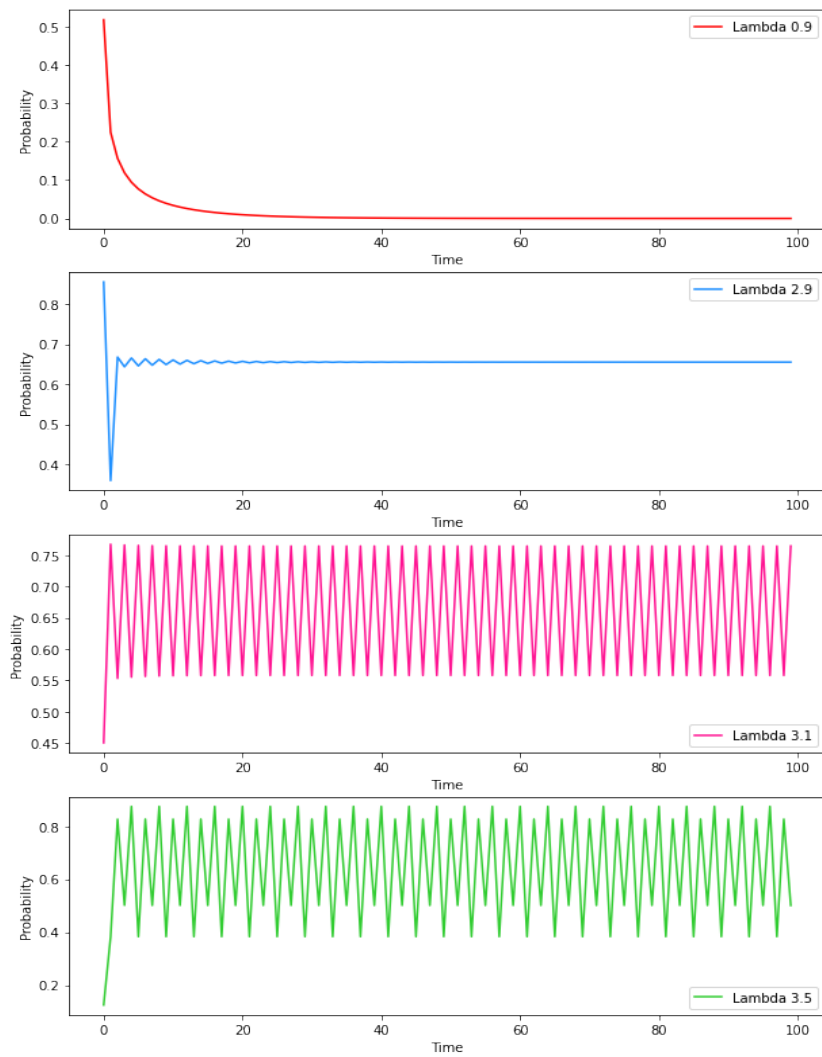
$$\begin{aligned} X_0 &= \frac{X_1}{R} + B \rightarrow X_1 = X_0R - BR \\ X_1 &= \frac{X_2}{R} + B \rightarrow X_2 = X_1R - BR \rightarrow X_2 = X_0R^2 - BR^2 - BR \\ X_2 &= \frac{X_3}{R} + B \rightarrow X_3 = X_2R - BR \rightarrow X_3 = X_0R^3 - BR^3 - BR^2 - BR \\ &\vdots \\ X_n &= X_0R^{n+1} - BR[1 + R + R^2 + \dots + R^n] \\ X_n &= X_0R^{n+1} - (\frac{R}{K} - \frac{1}{K})\frac{1-R^{n+1}}{1-R} \square \end{aligned}$$

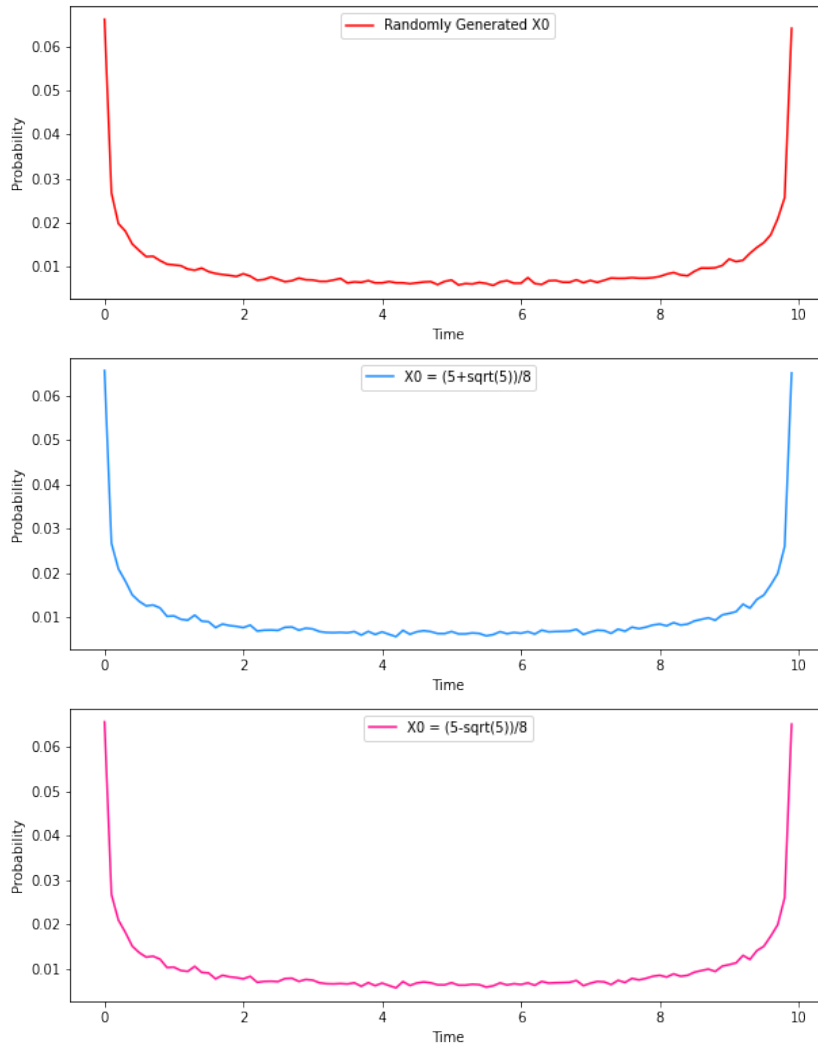
Problem 3:

a)



b)





There are no other special values because a polynomial of degree 2 can only have at most 2 distinct roots .

Problem 4: A classical chemistry kinetics problem models different concentrations over time:

$$\begin{aligned}\frac{dc_A(t)}{dt} &= -k_1 c_A c_B + k_{-1} c_C, \\ \frac{dc_B(t)}{dt} &= -k_1 c_A c_B + k_{-1} c_C, \\ \frac{dc_C(t)}{dt} &= k_1 c_A c_B - k_{-1} c_C,\end{aligned}$$

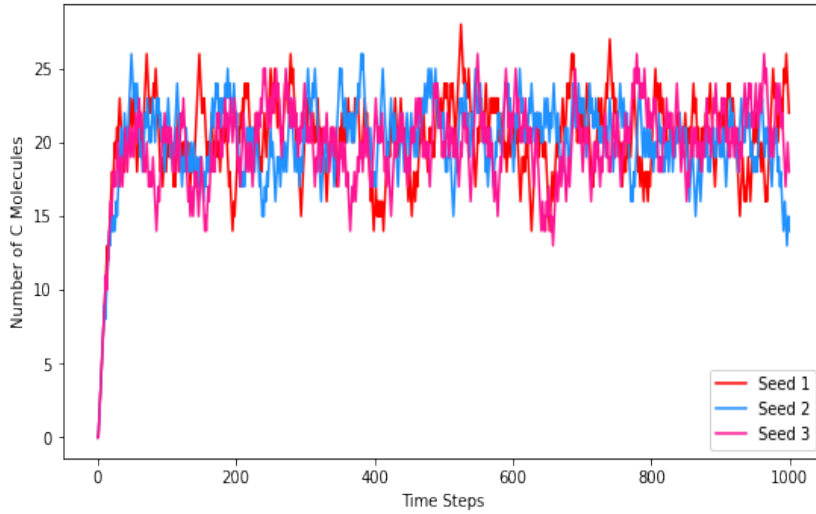
For k_1 and k_{-1} rate constants.

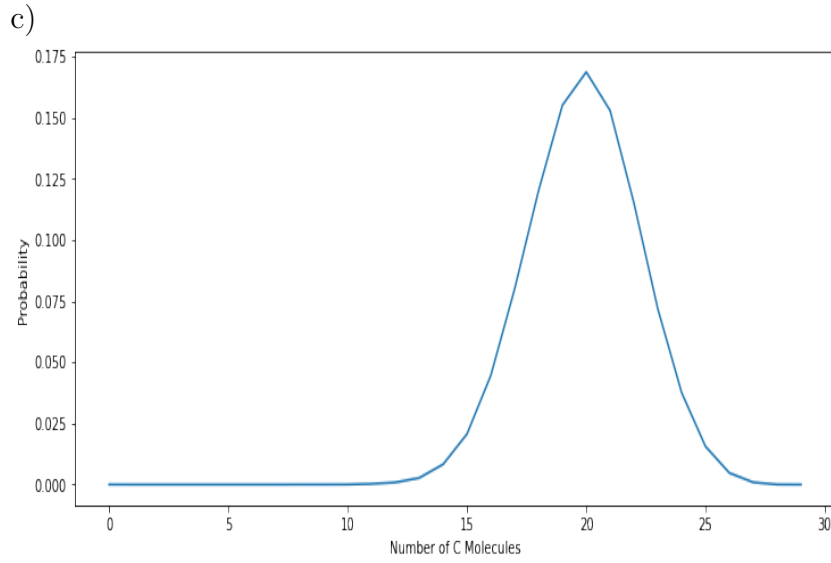
a) Show that $\frac{d}{dt}(c_A(t) + c_C(t)) = 0$ and $\frac{d}{dt}(c_B(t) + c_C(t)) = 0$.

$$\begin{aligned}\frac{d}{dt}(c_A(t) + c_C(t)) &= (-k_1 c_A c_B + k_1 c_A c_B - k_{-1} c_C + k_{-1} c_C) = 0 \\ \frac{d}{dt}(c_B(t) + c_C(t)) &= (-k_1 c_A c_B + k_1 c_A c_B - k_{-1} c_C + k_{-1} c_C) = 0\end{aligned}$$

The change in concentration over time being zero makes sense because an equal number of atoms from substance A and B are being used to create one molecule of substance C , so no particles are lost in the transaction and the change in the number of atoms over time will be zero.

b)





d) The equilibrium of the chemical reaction is when

$$\frac{k_{-1}}{k_1} = \frac{(N_A - n^*)(N_B - n^*)}{n^*}$$

Using the given values, we find that $n^* = 20$. From the graph above, it is easy to see that 20 is the mean number of atoms of C in the reaction, which supports our findings of $n^* = 20$ being an equilibrium point.