

## Kirk's Answers to Problem Set 2 Practice Problems

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### Problem 1A:

residues per turn: 3.5

pitch (distance between turns): 0.51 nm

number of residues of heptad repeats in  $\alpha$  keratin: 300 residues

300 residues / 3.5 residues per turn x 0.51 nm in an  $\alpha$ -helix pitch = 43.7 nm

### Problem 1B:

Does the sequence IQEVERD contain leucine or similar hydrophobic residues at positions 1 and 4? Yes

Does the sequence WQEYERD contain leucine or similar hydrophobic residues at positions 1 and 4? No

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### Problem 2A:

$K_S = \underline{3.67 \times 10^{-4}}$  M

$K_m = \underline{1.16 \times 10^{-3}}$  M

### Problem 2B:

Steady state kinetics because...  $k_2$  is large enough ( $k_2 \approx k_{-1}$ ) to be considered.

### Problem 2C:

What is the  $K_m$  of the *in vivo* concentration for the enzyme? 0.1 mM

How does it compare to the *in vivo* concentration of ATP? 10-100 times less than in vivo [ATP]

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### Problem 3A:

$[S] = \underline{15}$   $\mu$ M

### Problem 3B:

Divide to find the factor:

$\frac{[S]_{0.80} = 60 \mu M}{[S]_{0.20} = 3.75 \mu M} = \underline{16}$  fold increase

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### Problem 4

a) E.C. Classification: oxidoreductase

b) E.C. Classification: ligase

c) E.C. Classification: isomerase

d) E.C. Classification: lyase

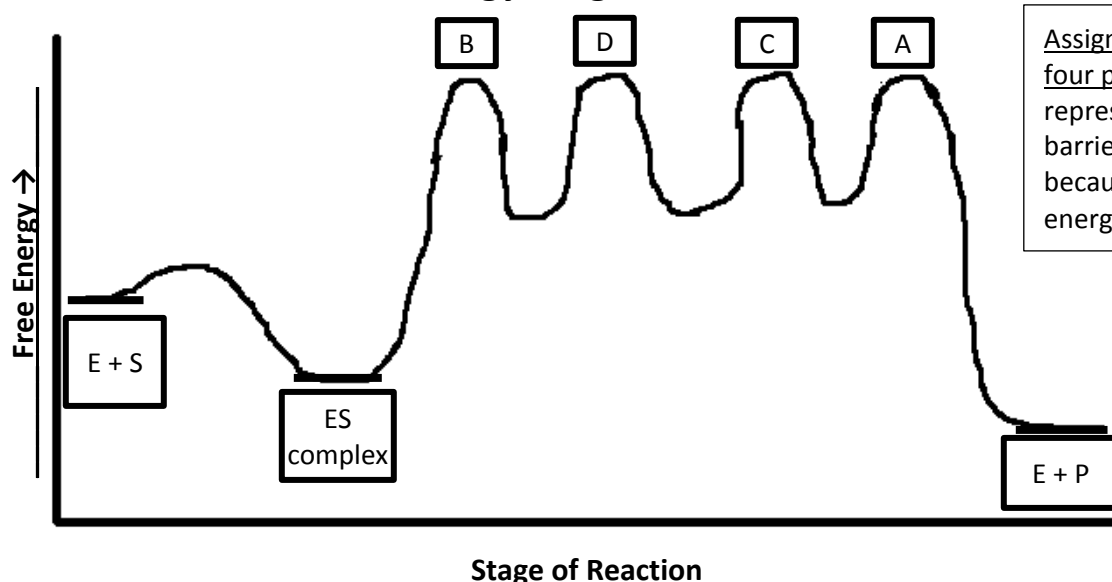
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My favorite color is purrple.  
Remember to relax and enjoy  
a Kit Kat bar before the exam!



## Problem 5A:

## Free Energy Diagram



Assign the four stages to the four peaks. The peaks represent activation energy barriers, at the same height because their relative energy values are not given.

## Word Bank:

E + P

Free Energy →

ES complex

E + S

Stage of Reaction

A, B, C, D

## Problem 5B:

In the catalytic mechanism shown in stage (b) of the reaction, H-O-H acts as a(n) nucleophile because it attacks an electron-deficient carbon center. The process forms a(n) hydroxide in the transition state and this molecule is what catalyzes the reaction. H-O-H also acts as a(n) acid because it donates an  $H^+$  to molecule A. As a result, molecule A acts as a(n) base because it accepts an  $H^+$  from water. Thus, the catalytic mechanism shown here must be general base catalysis (not in word bank, use table above).

## Problem 5C:

The enzyme catalyzes a hydrolysis reaction so it must be a(n) hydrolase type of enzyme!

## Problem 5D:

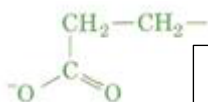
Any amino acid that can donate  $H^+$  will work. Glu, Asp, His are three examples. Cys and Tyr (although poorly) can work as well.

## Problem 6A:

Which EC enzyme class hydrolyzes bonds by adding water across the bond? hydrolase

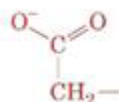
## Problem 6B:

Upper (green) amino acid residue:



Which amino acid has this structure?  
glutamic acid

Lower (red) amino acid residue:



Which amino acid has this structure?  
aspartic acid

## Problem 6C:

Drawing C – general base catalysis

Drawing D – general acid catalysis

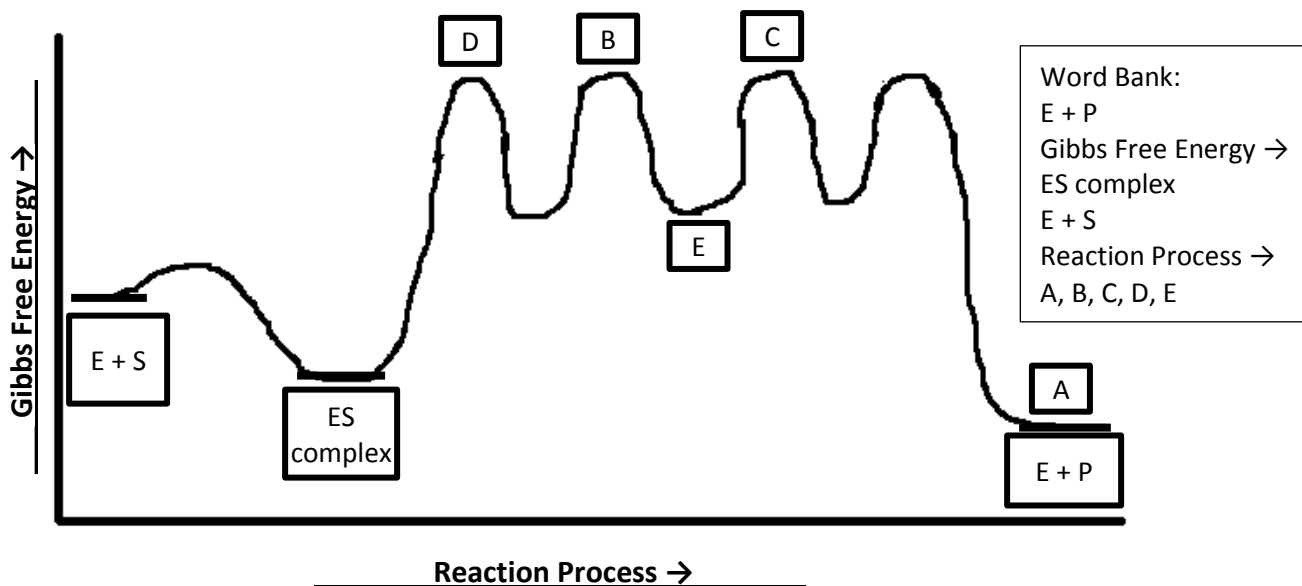
Drawing B or E – covalent catalysis

Show work and solution on your own paper!

v1.1

**Problem 6D:**

Order the reaction steps:

  D   →   B   →   E   →   C   →   A  **Kirk's Example Free Energy Diagram****Calvin and Hobbes by Bill Watterson**

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**Problem Set #2: Due Friday 10/19 at 5:00PM in FO 3.602****Exam #2 Review: TBA****Exam #2: Monday 10/22 at 10:00AM (Lee) in normal classroom****Tuesday 10/23 at 1:00PM (Marsh) in normal classroom**