# Topic 8 – Heat

Specific heat capacity (c) - E required to raise t of 1 kg substance by 1 K

Specific latent heat (L) - E required to change state of 1 kg substance at constant t

## Molecule speeds

mean of sum of squares of speeds of individual molecules

## Ideal gas equation

**Conditions: molecules**

1. Negligible size
2. Identical
3. Exert no force on each other except during collisions
4. Random motion

# Topic 9 – Nuclear physics

## Particle interactions

|  |  |  |
| --- | --- | --- |
| **Particle** | **General equation** | **Dangers / effects** |
| Alpha |  | High ionization (outside ok inside dead) |
| Beta |  | Moderate ionization (slight damage) |
| Gamma |  | Minimal ionization |

## Fusion & fission

### Binding energy

Higher more stable as it requires more E to pull the nucleus apart

### Binding energy per nucleon graph

Low N Lowless stableweaker electrostatic force ∴ Fusion

Fe is the most stable element as highest

High N less stable ∴ Fission

Nuclear fusion - Small nuclides that combine together to make larger nuclei, releasing E

**Fusion facts:**

1. High KE and density required to fuse two nuclei, to overcome the electrostatic repulsion between protons
2. Fusion product mass less than sum of reactants as E released

# Topic 10 – Simple harmonic motion (SHM)

**Conditions:**

1. A force continually trying to return object to center position
2. Fdisplacement from center position

Assuming :

## Conservation of energy

Free oscillations have all E conserved.

## Terminologies

### Resonance

When the frequency of the applied force to an oscillating system is equal to its natural frequency, the amplitude of the resulting oscillations increases significantly

### Damping