

STRESS ENERGY OF A GAS OF PARTICLES

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Consider the stress energy tensor of a collection of N particles with 4-momenta p_n^μ and mass m :

$$(1) \quad \begin{aligned} T^{\mu\nu} &= \sum_{n=0}^N \int d\tau \, p_n^\mu u_n^\nu \frac{\delta^4(x^\alpha - x_n^\alpha(\tau))}{\sqrt{|g|}} \\ &= m \sum_{n=0}^N \int \gamma_n dt \, v_n^\mu v_n^\nu \frac{\delta(t - t_n) \delta^3(\vec{x} - \vec{x}_n(t))}{\sqrt{|g|}} \end{aligned}$$

We can get a better idea as to the meaning of this expression by contracting it with the 4-velocity of some stationary observer to measure the total momentum of this gas as seen by the observer. Note that $u_\mu v^\mu = 1$. We have:

$$(2) \quad \begin{aligned} p_{gas}^\mu &= u_\nu T^{\mu\nu} \\ &= m \sum_{n=0}^N \gamma_n v_n^\mu(t_n, \vec{x}) \frac{\delta^3(\vec{x} - \vec{x}_n(t))}{\sqrt{|g|}} \end{aligned}$$

This is simply the sum of the 4-momenta of all the particles, so makes sense. Now let's consider the continuum limit by allowing $N \rightarrow \infty$. This allows us to define an average density of particles:

$$(3) \quad \begin{aligned} T^{\mu\nu} &\rightarrow \lim_{N \rightarrow \infty} T^{\mu\nu} \\ &= \int dt \lim_{N \rightarrow \infty} \sum_{n=0}^N m \frac{\delta^3(\vec{x} - \vec{x}_n(t))}{\sqrt{|g|}} \gamma_n v_n^\mu v_n^\nu \delta(t - t_n) \\ &= \int dt \int d^3\vec{x}' \rho(\vec{x}', t') v^\mu(\vec{x}', t') v^\nu(\vec{x}', t') \delta^3(\vec{x} - \vec{x}') \delta(t - t') \\ &= \rho(\vec{x}, t) v^\mu(\vec{x}, t) v^\nu(\vec{x}, t) \end{aligned}$$

Next we will assume this gas is isotropic, and homogenous. This means the velocities and densities will not depend on position, and that we can take a directional average of all the velocities. We have then:

$$(4) \quad \begin{aligned} v_x &= v \sin \theta \cos \phi & v_y &= v \sin \theta \sin \phi & v_z &= v \cos \theta \\ \langle T^{\mu\nu} \rangle &= \frac{1}{4\pi} \int d\Omega \, T^{\mu\nu} \\ &= \rho \, diag(1, \frac{v^2}{3}, \frac{v^2}{3}, \frac{v^2}{3}) \end{aligned}$$

Under these assumptions we see then that this gas of particles can be interpreted as a perfect fluid with

$$(5) \quad P = \frac{1}{3} \rho v^2$$

Furthermore, taking the classical limit where $v \rightarrow 0$ we see that the gas becomes pressureless.