

$$ro := \boxed{roO2} \cdot \frac{kg}{m^3}$$

$$roO2 := 1.225$$

$$vol := 1 \cdot m^3$$

$$volmass := ro \cdot vol \rightarrow kg \cdot roO2$$

$$molarMassO2 := 32 \frac{gm}{mol}$$

$$molMass := molarMassO2$$

$$Nmol := \frac{volmass}{molMass} = 38.281 \text{ mol}$$

$$Npt := Nmol \cdot N_A = 2.305 \cdot 10^{25} \quad \text{count of particles in 1m3 giver RO}$$

$$pszO := 0.299 \cdot 10^{-12} \text{ m}$$

$$psz := pszO$$

$$lw := 600 \text{ nm}$$

$$lwCells := \frac{m}{lw} \quad \text{count of Light cells per 1m}$$

$$lwCellsM3 := lwCells^3 \quad \text{count of Light cells in 1m3}$$

$$ptCellPerLcell := \left( \frac{lw}{psz} \right)^3 = 8.081 \cdot 10^{18} \quad \text{count of particle cells in light cell. per volume}$$

$$cellsM3 := \left( \frac{1 \cdot m}{psz} \right)^3 = 3.741 \cdot 10^{37} \quad \text{count of particle cells in 1m3}$$

$$P_{ptM3} := \frac{N_{pt}}{cellsM3} = 6.162 \cdot 10^{-13}$$

probability to encounter  
particle in one particle cell  
per m<sup>3</sup>

$$P_{ptLc} := P_{ptM3} \cdot ptCellPerLcell = 4.98 \cdot 10^6$$

probability to encounter  
particle in one Light Cell

$$P_m := P_{ptLc} \cdot lwCells = 8.299 \cdot 10^{12}$$

probability for light wave to encounter particles over 1m

$$P_{km} := P_m \cdot 1000 = 8.299 \cdot 10^{15}$$

probability for light wave to encounter particles over 1km

$$trk := 1$$

$$T_{\text{trk}} := 1 - \frac{\textit{psz}}{\textit{lw}} \cdot \textit{trk} \rightarrow -1.0 \cdot \frac{0.49833333333333333333 \cdot 10^{-15} \cdot m}{nm} + 1.0$$

$$ProbPerDist(dist) := P\_km \cdot dist$$

$$Ttrk^{ProbPerDist(1000)} \rightarrow \left( -1.0 \cdot \frac{0.4983333333333333 \cdot 10^{-15} \cdot m}{nm} + 1.0 \right)^{8299262734875000000} = 0$$