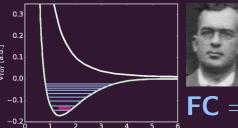
# Electronic structure theory

#### TURBOMOLE





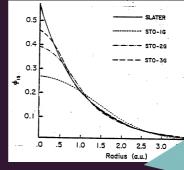


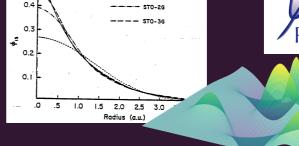
R<sub>AB</sub> [a.u.]

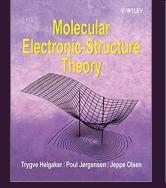


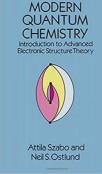


 $\mathsf{FC} = \mathsf{SC}\epsilon$ 

















$$E[\rho] = T_s[\rho] + \int d\mathbf{r} v_{e\times t}(\mathbf{r}) \rho(\mathbf{r}) + E_H[\rho] + E_{\times c}[\rho]$$

## Outline for today

#### Introduction to quantum electronic structure theory

What is electronic structure theory?

From the Schrödinger equation to the Born-Oppenheimer approximation

Wavefunction based methods

Hartree-Fock theory (HF)

Density based methods

Density functional theory (DFT)

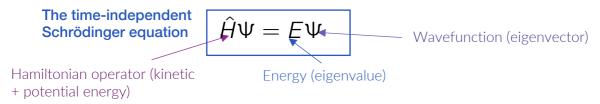
How to run all this in Gaussian?

# Quantum mechanics 101

For systems the size of molecules classical mechanics is not longer accurate

- Energy levels are discrete (quantized energy states)
- Particles behave as waves (wave-particle duality)
- You cannot know both position and momentum exactly at the same time (Heisenberg Uncertainty principle)

Erwin Schrödinger came up with an eigenvector eigenvalue equation to solve all these problems



# Quantum Mechanics is defined in a Hilbert space

- A Hilbert space generalizes the notion of Euclidean space. It extends the methods of vector algebra and calculus from the two/three-dimensional Euclidean space to spaces with any finite or infinite number of dimensions.
- A Hilbert space is an **abstract vector space** possessing the structure of an inner product that allows length and angle to be measured.
- Hilbert spaces are complete: there are enough limits in the space to allow the techniques of calculus to be used.

"Hilbert spaces arise naturally and frequently in mathematics and physics, typically as infinitedimensional function spaces."

# Wavefunctions are complex – why?

arXiv.org > quant-ph > arXiv:0907.0909

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#### **Quantum Physics**

#### Origin of Complex Quantum Amplitudes and Feynman's Rules

Philip Goyal, Kevin H. Knuth, John Skilling

(Submitted on 6 Jul 2009 (v1), last revised 14 Feb 2010 (this version, v3))

Complex numbers are an intrinsic part of the mathematical formalism of quantum theory, and are perhaps its most mysterious feature. In this paper, we show that the complex nature of the quantum formalism can be derived directly from the assumption that a pair of real numbers is associated with each sequence of measurement outcomes, with the probability of this sequence being a real-valued function of this number pair. By making use of elementary symmetry conditions, and without assuming that these real number pairs have any other algebraic structure, we show that these pairs must be manipulated according to the rules of complex arithmetic. We demonstrate that these complex numbers combine according to Feynman's sum and product rules, with the modulus-squared yielding the probability of a sequence of outcomes.

Comments: v2: Clarifications, and minor corrections and modifications. Results unchanged. v3: Minor changes to introduction and conclusion

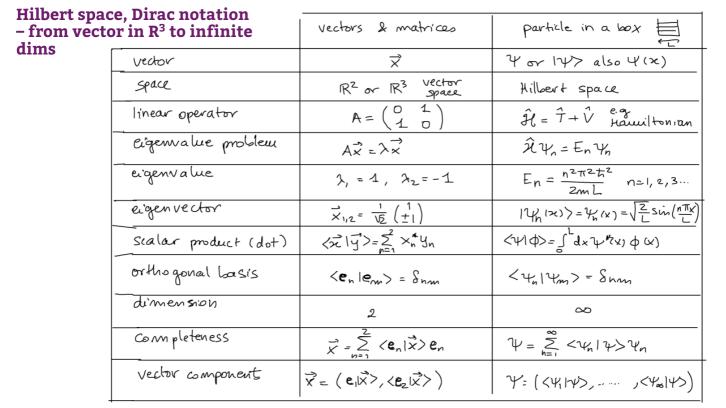
 Subjects:
 Quantum Physics (quant-ph)

 Journal reference:
 Phys. Rev. A 81, 022109 (2010)

 DOI:
 10.1103/PhysRevA.81.022109

 Cite as:
 arXiv:0907.0909 [quant-ph]

(or arXiv:0907.0909v3 [quant-ph] for this version)



## **Dirac notation 101**

The "KET" 
$$\psi = |\psi\rangle = |\psi\rangle$$

Vector

The "BRA" vector  $\psi^* = \langle \psi \rangle$ 
 $(\psi_1^*, \psi_2^*, \dots, \psi_{\infty})$ 

a row vector (complex conjugate transpose of  $|\psi\rangle$ )

So  $(|\psi\rangle)^{\dagger} = \langle \psi \rangle$ 

Tuner product  $\langle \psi | \psi \rangle = \sum_{i=1}^{\infty} \psi_i^* \psi_i + \psi_2^* \psi_2 + \dots$  or integral  $\to$  scalar an operator

Outer product  $|\psi\rangle\langle\psi| = |\psi\rangle\langle\psi| = |$ 

# Expectation value of an operator

# Quantum mechanics 101

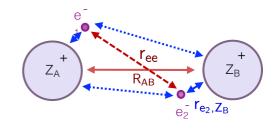
#### What is the Hamiltonian?

"An **operator** which can act on the state of a system, i.e. the wavefunction to provide the value of its energy"

It combines all energy terms, kinetic and potential.

For molecules it takes the following form

$$\begin{split} \hat{H} = & \hat{T}_N + \hat{T}_{el} + \hat{V}_{N,N} + \hat{V}_{N,el} + \hat{V}_{el,el} \\ \hat{H} = & \sum_{i=1}^{N_{\text{nuclei}}} \frac{-\hbar^2}{2m_i} \nabla_i^2 + \sum_{j=1}^{N_{\text{electrons}}} \frac{-\hbar^2}{2m_{el}} \nabla_j^2 + \sum_{k>i}^{N_{\text{nuclei}}} \frac{e^2}{4\pi\epsilon_0} \frac{Z_i Z_k}{R_{ik}} \\ & - \sum_{j}^{N_{\text{nuclei}}} \sum_{i}^{N_{\text{electrons}}} \frac{e^2}{4\pi\epsilon_0} \frac{Z_i}{|\mathbf{R}_i - \mathbf{r}_j|} + \sum_{k>j}^{N_{\text{electrons}}} \frac{e^2}{4\pi\epsilon_0} \frac{1}{r_{kj}} \end{split}$$



Computationally, the Hamiltonian takes the form of a **matrix** ...

# **Quantum mechanics 101**

#### What is the wavefunction?

"A wave function is an infinite dimensional **vector** which describes the **quantum state** of an isolated quantum system. The wave function is **complex-valued**. One can use this wavefunction to find measurable values, e.g. position."

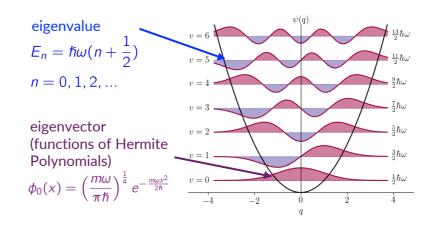
Example - the 1D Harmonic oscillator

$$\hat{H} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + \frac{1}{2} k \hat{x}^2$$

$$\hat{V}(x) = \frac{1}{2}m\omega^2\hat{x}^2$$

$$\omega = \sqrt{rac{k}{m}}$$

Need to solve the SE



## What is electronic structure theory?

Area of computational chemistry which focuses on solving the TISE (my acronym for time-independent Schrodinger equation) for the electrons under the **adiabatic** and **Born-Oppenheimer approximations** 

$$\hat{H} = \hat{H}_{nuc}(\mathbf{R}) + \hat{H}_{elec}(\mathbf{r}; \mathbf{R})$$

$$\hat{H}_{\text{nuc}} = \sum_{i=1}^{N_{\text{nuclei}}} \frac{-\hbar^2}{2m_i} \nabla_i^2 + \sum_{k>1}^{N_{\text{nuclei}}} \frac{e^2}{4\pi\epsilon_0} \frac{Z_i Z_k}{R_{ik}}$$

$$\hat{H}_{\text{elec}} = \sum_{j=1}^{N_{\text{electrons}}} \frac{-\hbar^2}{2m_{el}} \nabla_j^2 + \sum_{j}^{N_{\text{nuclei}}} \sum_{i}^{N_{\text{electrons}}} \frac{e^2}{4\pi\epsilon_0} \frac{Z_i}{|\mathbf{R}_i - \mathbf{r}_j|} + \sum_{k>j}^{N_{\text{electrons}}} \frac{e^2}{4\pi\epsilon_0} \frac{1}{r_{kj}}$$

Nucleus

electron

$$\Psi(\textbf{r},\textbf{R}) = \Psi_{el}(\textbf{r};\textbf{R}) \Psi_{nuc}(\textbf{R})$$

$$\hat{H}_{\mathsf{el}}\Psi_{\mathsf{el}}=E_{\mathsf{el}}\Psi_{\mathsf{el}}$$

$$\hat{H}_{nuc}\Psi_{nuc}=E_{nuc}\Psi_{nuc}$$

# What can you obtain by solving the TISE for electrons?

- Ground state geometry of molecules (most stable nuclear arrangement)
- Ground state energy and excited state energy or electrons in molecules (spectroscopy / dynamics etc)
- Orbitals (density of electrons around nuclei)
- Coupling (interaction energy between molecules)
- Minimum energy paths (reaction rate theories)

Etc.

# Wavefunction based methods: Hartree-Fock theory

We are solving the adiabatic electronic TISE

$$\left[\hat{T}_e(\mathbf{r}) + \hat{V}_{eN}(\mathbf{r}; \mathbf{R}) + \hat{V}_{NN}(\mathbf{R}) + \hat{V}_{ee}(\mathbf{r})\right] \Psi(\mathbf{r}; \mathbf{R}) = E_{el} \Psi(\mathbf{r}; \mathbf{R})$$



spin coordinate



For HF, the total electronic wavefunction is assumed to be

$$\Psi_{HP}(\mathbf{r}_1, \mathbf{r}_2, \cdots, \mathbf{r}_N) = \phi_1(\mathbf{r}_1)\phi_2(\mathbf{r}_2)\cdots\phi_N(\mathbf{r}_N)$$

i.e one assumes that the motion of each electron is **separable** from that of the others although we will see each feels the presence of other electrons as a mean field

<u>Problem</u>: electron have spin – wavefunction must be **antisymmetric** to the exchange of any set of space-spin coordinate (**Pauli exclusion principle!**)

We introduce spin-orbitals 
$$\chi_i(\mathbf{x}_i) = \phi(\mathbf{x}_i)\alpha$$
  $\mathbf{x} = \{\mathbf{r}, \omega\}$   $\chi_i(\mathbf{x}_i) = \phi(\mathbf{x}_i)\beta$ 

(\*) Equations taken from C. David. Sherill, Hartree-Fock Theory

# Hartree-Fock theory

So we now have

$$\Psi_{HP}(\mathbf{x}_1, \mathbf{x}_2, \cdots, \mathbf{x}_N) = \chi_1(\mathbf{x}_1)\chi_2(\mathbf{x}_2)\cdots\chi_N(\mathbf{x}_N)$$





But to satisfy the antisymmetry principle, i.e. for two electrons we must write

$$\Psi(\mathbf{x}_1, \mathbf{x}_2) = \frac{1}{\sqrt{2}} \left[ \chi_1(\mathbf{x}_1) \chi_2(\mathbf{x}_2) - \chi_1(\mathbf{x}_2) \chi_2(\mathbf{x}_1) \right]$$

Indeed if you swap the coordinates, your find  $\Psi(\mathbf{x}_1,\mathbf{x}_2) = -\Psi(\mathbf{x}_2,\mathbf{x}_1)$ 

Equivalently, 
$$\Psi(\mathbf{x}_1,\mathbf{x}_2) = \frac{1}{\sqrt{2}} \begin{vmatrix} \chi_1(\mathbf{x}_1) & \chi_2(\mathbf{x}_1) \\ \chi_1(\mathbf{x}_2) & \chi_2(\mathbf{x}_2) \end{vmatrix}$$

So for N electrons, we can write the **Slater determinant** 
$$\Psi = \frac{1}{\sqrt{N!}} \begin{bmatrix} \chi_1(\mathbf{x}_1) & \chi_2(\mathbf{x}_1) & \cdots & \chi_N(\mathbf{x}_1) \\ \chi_1(\mathbf{x}_2) & \chi_2(\mathbf{x}_2) & \cdots & \chi_N(\mathbf{x}_2) \\ \vdots & \vdots & \ddots & \vdots \\ \chi_1(\mathbf{x}_N) & \chi_2(\mathbf{x}_N) & \cdots & \chi_N(\mathbf{x}_N) \end{bmatrix}$$

# Hartree-Fock theory

One can write the electronic Hamiltonian  $\hat{T}_e(\mathbf{r}) + \hat{V}_{eN}(\mathbf{r};\mathbf{R}) + \hat{V}_{NN}(\mathbf{R}) + \hat{V}_{ee}(\mathbf{r})$ 

As 
$$\hat{H}_{el} = \sum_i h(i) + \sum_{i < j} v(i,j) + V_{NN}$$
 Ignore  $V_{NN}$  for now

Where (note, we are working in atomic units so  $\hbar=1\;\;;m_{el}=1;\;\;4\pi\epsilon_0=1$ )

$$h(i) = -\frac{1}{2}\nabla_i^2 - \sum_A \frac{Z_A}{r_{iA}}$$
  $v(i,j) = \frac{1}{r_{ij}}$ 

We search for the optimal wavefunctions, i.e. those which lead to a minimum energy

$$E_{el} = \langle \Psi | \hat{H}_{el} | \Psi \rangle \longrightarrow \boxed{E_{el} = \min_{\phi} \left( \langle \Psi | \hat{H}_{e} l | \Psi \rangle \right)}$$





# Hartree Fock Roothaan equations

One can show that for resctricted Hartree-Fock (i.e. no unpaired electrons), solving the TISE can be written as

Roothaan equations

$$\mathbf{FC} = \mathbf{SC}\epsilon \qquad \longrightarrow \qquad \sum_{\nu} F_{\mu\nu} C_{\nu i} = \epsilon_i \sum_{\nu} S_{\mu\nu} C_{\nu i}$$

Fock operator integral

erlap integral 
$$S_{\mu\nu} = \int d\mathbf{x}_1 \tilde{\chi}_{\mu}^*(\mathbf{x}_1) \tilde{\chi}_{\nu}(\mathbf{x}_1), \qquad \chi_i = \sum_{\mu=1}^K C_{\mu i} \tilde{\chi}_{\mu}$$
 
$$F_{\mu\nu} = \int d\mathbf{x}_1 \tilde{\chi}_{\mu}^*(\mathbf{x}_1) f(\mathbf{x}_1) \tilde{\chi}_{\nu}(\mathbf{x}_1). \qquad \text{Coefficients}$$
 ex operator integral

$$\chi_i = \sum_{\mu=1}^K C_{\mu i} \tilde{\chi}_{\mu}$$
 Basis set Coefficients

One electron Fock operator  $f(\mathbf{x}_1) = h(\mathbf{x}_1) + \sum_{i} \mathcal{J}_j(\mathbf{x}_1) - \mathcal{K}_j(\mathbf{x}_1)$ 

# RHF - Roothan equations

$$f(\mathbf{x}_1) = h(\mathbf{x}_1) + \sum_{j} \mathcal{J}_j(\mathbf{x}_1) - \mathcal{K}_j(\mathbf{x}_1)$$

$$\mathcal{J}_j(\mathbf{x}_1) = \int d\mathbf{x}_2 |\chi_j(\mathbf{x}_2)|^2 r_{12}^{-1}$$

**Coulomb integral** gives the average local potential at point x<sub>1</sub> due to the charge distribution from the electron in orbital X<sub>i</sub>

$$\mathcal{K}_j(\mathbf{x}_1) = \int d\mathbf{x}_2 \chi_j^*(\mathbf{x}_2) r_{12}^{-1} \chi_i(\mathbf{x}_2)$$
 Exchange integral no classical analogue - gives the effect of exchanging spin orbitals

effect of exchanging spin orbitals

How does one solve the Roothaan equations?

$$\mathbf{FC} = \mathbf{SC}\epsilon$$

$$\chi_i = \sum_{\mu=1}^K C_{\mu i} \tilde{\chi}_{\mu}$$

Almost an eigenvalue equation, but F depends on it's own solution (the desired orbital coefficients), so the solution to these equations are obtained using a self-consistent-field procedure.

## Self-consistent-field solution of HF

- 1. Initial setup:
- Input nuclear coordinates (R)
- Choose basis set  $\{\phi_i\}_{i=1}^K \longrightarrow \{\chi_i\}_{i=1}^K$
- 2. Compute matrices and matrix elements which only depend on basis set:
- Compute overlap matrix S
- Compute two-electron integrals (these will be multiplied by coefficients afterwards
- Compute kinetic energy matrix elements (in h)
- Compute nuclear potential energy matrix elements (in h)

3. Guess initial coefficient matrix **C** 

4. Compute the Fock matrix **F**Use coefficients of previous iteration to compute Fock matrix of current iteration

5. Solve the Roothaan Equation

$$\mathbf{F}^{(i-1)}\mathbf{C}^{(i)} = \mathbf{SC}^{(i)}\epsilon$$

e.g. use symmetric orthogonalization

#### Converged?

rmsd of current **C matrix** respect to previous iteration / **energy** difference respect to previous iteration

yes Done!

## **Basis sets**

Database with all basis sets available:

https://www.basissetexchange.org/

Split-valence basis set – represent valence orbitals by more than one basis function (double, triple, quadruple)

#### Pople Basis sets: 6-31G\*\*

- 6 number of functions used to construct core atomic orbitals
- 3 number of functions of type 1 used to construct valence atomic orbitals
- 1 number of functions of type 2 used to construct valence atomic orbitals
- G gaussians

First \* - addition of polarized function 'd'

Second \* - addition of polarized function 'p'

# **Choice of basis sets – David Sherrill suggestions**

- The use of minimal basis sets (e.g., STO-3G) is **highly discouraged**.
- Polarized double-zeta basis sets (e.g., cc-pVDZ, DZP, 6-31G\*) are the smallest basis sets which can be counted on to give semiquantitative predictions. **Try to use 6-31G\*** [also called 6-31G(d,p)] when you can instead of 6-31G\* [also called 6-31G(d)].
- Balance your choice of basis with the theoretical method. Usually errors in basis set will largely cancel errors in the treatment of correlation if you pick a proper pair. For Hartree-Fock, use polarized double zeta basis sets. For MP2, CISD, CCSD, use either polarized double zeta or polarized triple zeta basis sets (e.g., TZ2P, cc-pVTZ). For CCSD(T), MRCI, etc, try to use polarized triple or quadruple zeta basis sets (or even cc-pV5Z if possible for tiny molecules).
- **Diffuse functions**: If you have an anion or want to study a Rydberg excited electronic state (or a mixed valence/Rydberg state), you must have diffuse functions. Must. Really. Diffuse functions will be present in Pople basis sets with + signs in them, e.g., 6-31+G\*, 6-31++G\*\*, etc. They are also found in the so-called "augmented" basis sets of Dunning and co-workers, e.g., aug-cc-pVDZ. Diffuse functions may also be useful for electronegative atoms, and should be important in van der Waals complexes.

## **Limitations of Hartree-Fock**

- Does not include electron correlation due to the mean field approximation
- Methods which include electron correlation exist but are computationally more expensive MP2, CCSD(T), CI

#### Infinite Basis set Orbitals) **EXACT** Larger basis set increases Pople basis set flexibility to describe wave 6-31G(d) function/ electron density Valence electrons Increasing an accuracy 6-31+G(d,p) Diffusion function Increasing CPU time Set (Atomic 6-311++G(2df,pd) Polarization function Higher-angular momentum orbitals Dunning basis sets (aug)-cc-pVDZ (aug)-cc-pVTZ (aug)-cc-pVQZ Basis Improving description of electron correlation V-Z Double excitations Perturbative of triple excitations Full electron correlation

CCSD (T)

QCISD (T)

HF/SCF

MP2

MP4

## **Level of Theory**

CCSD (T)-F12 .....

**FULL CI** 

Pople diagram

# Let's test it out in Gaussian

Let's see how to run this on the cluster! Files are on mox in folder

/usr/lusers/valleau/dv/TUTORIALS/tutorial\_Gaussian/HF

# Density based methods: DFT

Given the electronic wavefunction we can define the **electron density** as the probability of finding any electron in position  $\mathbf{r}$  as

$$n(\mathbf{r}) = N \int \Psi_{el}^*(\mathbf{r}, \mathbf{r}_2, ..., \mathbf{r}_N) \Psi_{el}(\mathbf{r}, \mathbf{r}_2, ..., \mathbf{r}_N) d\mathbf{r}_2 d\mathbf{r}_3 ... d\mathbf{r}_N$$

If we integrate  $n(\mathbf{r})$  over all values of  $\mathbf{r}$  we obtain the total number of electrons N.

When looking at the electronic Hamiltonian, we can write

$$\hat{H}_{\text{elec}} = \sum_{j=1}^{N_{\text{electrons}}} \frac{-\hbar^2}{2m_{el}} \nabla_j^2 + \sum_{j}^{N_{\text{nuclei}}} \sum_{i}^{N_{\text{electrons}}} \frac{e^2}{4\pi\epsilon_0} \frac{Z_i}{|\mathbf{R}_i - \mathbf{r}_j|} + \sum_{k>j}^{N_{\text{electrons}}} \frac{e^2}{4\pi\epsilon_0} \frac{1}{r_{kj}}$$

$$H_{\text{elec}} = -\sum_{j} \frac{1}{2} \nabla_i^2 - \sum_{j} \sum_{i} \frac{Z_i}{r_{ik}} + \sum_{k>j} \frac{1}{r_{kj}}$$
Drop hats / go to atomic units
$$H_{\text{elec}} = T + V_{ne} + V_{ee}$$
Could be seen as an external potential

# The Hohenberg-Kohn theorems

**The Hohenberg-Kohn existence theorem**: For any system of interacting particles in an external potential the density is uniquely determined (the external potential is a unique functional of the density)



**The Hohenberg-Kohn variational theorem**: A universal functional for the energy E[n] can be defined as a function of the density. The exact ground state is the global minimum value of this functional.

This means that we can go from solving the Schrodinger equation for 3N electronic coordinates to solving for the electron density **which only depends on 3 coordinates (e.g. x,y,z) !!!** 

The universal functional is **unfortunately unknown**  $\ensuremath{\mathfrak{G}}$ 

$$E[n] = T[n] + V_{ne}[n] + V_{ee}[n]$$

Very difficult to estimate

Hohenberg, P.; Kohn Phys. Rev. 136:B864 (1964)

# Density functional theory: Kohn-Sham equations

To solve the problem, one needs to minimize the energy functional with respect to the density. Successful minimization will lead to the ground-state density  $n_0$ 

**The Kohn-Sham assumption**: a fictitious system of non-interacting electrons subject to an external potential has a ground-state electronic density identical to that of the system of interacting electrons

Kohn-Sham equations 
$$\left( \frac{-\hbar^2}{2m} \nabla^2 + v_{eff}(\mathbf{r}) \right) \phi_i(\mathbf{r}) = \epsilon_i \phi_i(\mathbf{r})$$
Kohn-Sham orbital 
$$n(\mathbf{r}) = \sum_{i}^{N} |\phi_i(\mathbf{r})|^2$$
Kohn-Sham potential 
$$v_{eff}(\mathbf{r}) = v_{ext}(\mathbf{r}) + \int \frac{n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}' + \frac{\delta E_{xc}[n]}{\delta n(\mathbf{r})}$$
Exchange correlation potential 
$$v_{ext}(\mathbf{r}) = \sum_{j}^{N_{\text{nuclei}}} \frac{Z_j}{|\mathbf{R}_j - \mathbf{r}|}$$

# **Kohn-Sham equations**

$$E[n] = T_s[n] + \int d\mathbf{r} v_{ext}(\mathbf{r}) n(\mathbf{r}) + E_H[n] + E_{xc}[n]$$

$$T_s[n] = \sum_{j=1}^{N} \int d\mathbf{r} \phi_i^*(\mathbf{r}) \frac{-\hbar^2}{2m} \nabla^2 \phi(\mathbf{r}) \quad E_H = \frac{1}{2} \int d\mathbf{r} \int d\mathbf{r}' \frac{n(\mathbf{r}) n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|}$$

$$E[n] = \sum_{j=1}^{N} \int d\mathbf{r} \phi_i^*(\mathbf{r}) \frac{-\hbar^2}{2m} \nabla^2 \phi(\mathbf{r}) \quad E_H = \frac{1}{2} \int d\mathbf{r} \int d\mathbf{r}' \frac{n(\mathbf{r}) n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|}$$

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$$E[n] = \sum_{j=1}^{N} \int d\mathbf{r} \phi_i^*(\mathbf{r}) \frac{n(\mathbf{r}) n(\mathbf{r})}{n(\mathbf{r})}$$

$$E[n] = \sum_{j$$

Orbital energies have little physical meaning – however they are related to the electronic energy of the system

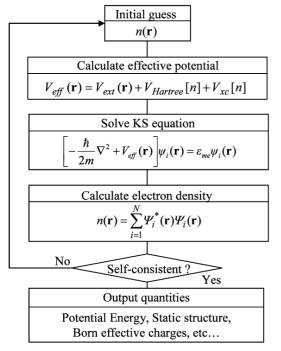
$$E = \sum_{i=1}^{N} \epsilon_i - E_H[n] + E_{xc}[n] - \int \frac{\delta E_{xc}[n]}{\delta n(\mathbf{r})} n(\mathbf{r}) d\mathbf{r}$$

Exchangeknown / term is approximations

## **DFT functionals...**



# How to solve these equations numerically?



Guess of density is often based on choosing an atomic basis set to represent the Kohn-Sham orbitals

The Kohn-Sham wavefunction can be represented by a single slater determinant (non-interaction assumption)

The equation is solved self-consistently as it was in the HF approach

#### **HF versus DFT**

- DFT is exact, but, one needs to know the external potential as a function of *n*.
- Hohenberg and Kohn proved that a functional of n must exist.
- There is no information regarding what the functional should look like.
- DFT fails for charge-transfer states
- DFT and HF fail for non-covalent interactions

#### Contrast between HF and DFT:

- HF approximate theory: solve the relevant equations exactly.
- DFT exact theory: solve the relevant equations approximately because the form of the operator is unknown.

# **DFT** in Gaussian

Files are in folder on mox

/usr/lusers/valleau/dv/TUTORIALS/tutorial\_Gaussian/DFT

## Some recommendations

- This was a very brief introduction to these methods much more complex than what I have described
- There are many other methods out there!
- Choice of basis set (use at least 6-31G\*\*)
- Choice of functional (B3LYP)

